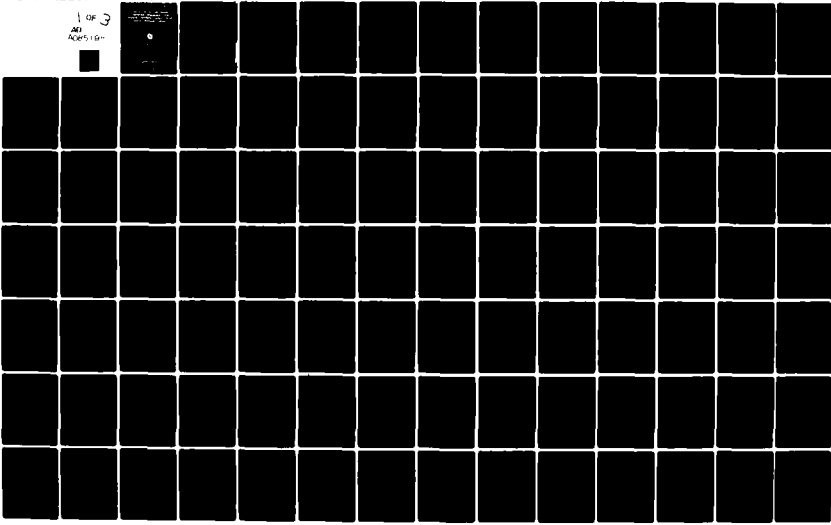


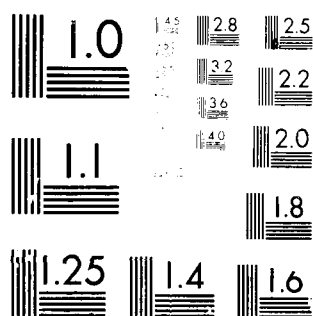
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HIGH ALTITUDE POLLUTION PROGRAM STRATOSPHERIC MEASUREMENT SYSTEM LABORATORY PERFORMANCE CAPABILITY REPORT CHEMICAL CONVERSION TECHNIQUES

Norman H. Macoy and Richard Weingarten
with
Anthony Pires and Sherman Poultney

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February 1980
TECHNICAL REPORT

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U.S. DEPARTMENT OF TRANSPORTATION
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Office of Environment and Energy
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16. Abstract This report documents the laboratory measurements made to resolve specificity questions concerning the chemical conversion techniques for measurement of stratospheric trace species NO_2 , N_2O_5 , and HNO_3 to NO (which is the measurable species). Particular emphasis was placed on the conversion of NO_2 to NO by both catalytic and photolytic converters, thermal conversion of N_2O_5 and HNO_3 to NO_2 with and without possible interfering gases, and the feasibility of measuring total odd-nitrogen concentrations using catalytic thermal conversion to NO. The laboratory measurements were generally carried out at ppm concentrations at STP conditions. The GEARS/EPISODE computer code was used to model both the laboratory reactions and the behavior of the technique at stratospheric conditions. The laboratory measurements and stratospheric simulations lead to the recommen- dation to develop a flight prototype of a Hybrid Gas Conversion System consisting of a number of instrumentation modules. The first module would be a Total Odd-Nitrogen module (and NO) based on high temperature catalytic conversion of the odd-nitrogens and chemiluminescent detection of NO. The second module would be an NO_2 /NO module based on NO_2 photolysis and NO chemiluminescent detection. Other modules could consist of an O_3 UV photometer, an N_2O gas chromatograph, and other appropriate modules that are ready at flight time.	17. Key Words Stratosphere, Stratospheric Measurement Techniques, Photochemical and Pyrolytic Reactions, Trace Gases, Oxides of Nitrogen	18. Distribution Statement This document is available to the public through the National Technical Information Service, Springfield, Virginia 22161.
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SECTION I

INTRODUCTION

This report documents the laboratory measurements made to resolve specificity questions concerning the Hybrid Chemical Conversion Measurement System and to examine the potential of a conversion technique for total odd nitrogen, as recommended in the High Altitude Pollution Program (HAPP) Stratospheric Measurement System Feasibility Study. That study and these laboratory measurements are part of a large study of the High Altitude Pollution Program Stratospheric Measurement System being done by the Perkin-Elmer Corporation for the Department of Transportation, Federal Aviation Administration under Contract DOT-FA77WA-4080. The goal of this larger study is to define and demonstrate a measurement system consisting of a technique or combination of techniques that can be flown on a series of balloon flights to make prescribed, *in situ* measurements of the stratosphere in support of the High Altitude Pollution Program. The other components of the total study include the feasibility of the Tunable Diode Laser Spectrometric System for the prescribed measurements, detailed engineering design documentation for the recommended payload, and a summary report. These other components of the study will be reported separately.

The Feasibility Study (Macoy et al., 1978) had concluded that the feasibility of successfully developing and flying a balloon-borne measurement system was high, based on a careful examination of the candidate system from an engineering point of view. The Study recommended that the system, consisting of a hybrid of analytical techniques, should be developed and deployed pending resolution of certain specificity questions. The selection of the Hybrid Chemical Conversion System was based on a comprehensive and critical review of all measurement techniques. This review included molecular chemical conversion, atomic chemical conversion, ion molecular conversion/mass spectroscopy, *in situ* optical absorption gas chromatography, and resonance fluorescence. Only the Hybrid Chemical Conversion techniques were to be appropriate

for an on-station balloon payload and within current state-of-the-art technology. Even then, certain of these techniques were subject to questions of conversion specificity. The questions did not invalidate those techniques, but might place several prescribed species into one class. The resolution of the specificity questions was therefore necessary and that work is one of the subjects of this report.

As the Feasibility Study was concluding, it became evident that progress in each of the reviewed techniques was not standing still and certain promising techniques should continue to be investigated in parallel to the resolution of specificity questions. The promising techniques were the Tunable Diode Laser Absorption Spectrometer and the Total-Odd Nitrogen Converter. The latter might turn the very lack of specificity of the conversion techniques into a capability to make the important measurement of total odd-nitrogen in the stratosphere. The study of the potential of this converter is another subject of this report. The Tunable Diode Laser Spectrometer technique gave promise that it might be able to measure all of the prescribed species if the IR signatures of heavier molecules cooperated and the requisite S/N performance could be achieved in the field. The study of the feasibility of a Tunable Diode Laser Spectrometer Measurement System is reported in a separate report (Poultney et al., 1979).

As this report is being written, progress in the reviewed techniques is still being made and the list of prescribed species is being re-examined. The ability to take advantage of advances in measurement capability is one great strength of the hybrid system that is recommended. A new technique can replace or supersede a weaker technique in the hybrid system without making the whole system obsolete. One new technique under development in the community for NO_2 promises to increase the feasibility of the Hybrid Gas Conversion system in that manner. This new technique and its potential impact on the realization of the HAPP Stratospheric Measurement System is also addressed in this report. A second strength of the Hybrid System is the ability to add or delete species (or parameters) without making the whole system obsolete. Instruments are being developed to measure these new species and parameters and could be later included in the Hybrid System. We do not here review any of these new instruments (e.g. a J_{NO_2} instrument).

The originally prescribed measurements were NO, NO₂, N₂O₅, HNO₃, N₂O and O₃ at the same time and place in the stratosphere for the full range of concentrations expected during a diurnal cycle. These trace gases play a key role in the present characteristics of the ozone layer. The prescribed measurements would improve our understanding of how aircraft engine emissions might affect the ozone layer and the stratosphere. The Hybrid Gas Conversion System was selected to make these measurements. As discussed in the Feasibility Study, the hybrid system consists of a UV absorption instrument for O₃, a gas chromatography instrument for N₂O, and a chemical conversion/chemiluminescent instrument for HNO₃, N₂O₅, NO₂, and NO. The expected sensitivity for O₃ is 2.7×10^{10} and for N₂O is 3.7×10^9 molecules/cm³. The third instrument consisted of a thermal converter for the HNO₃ and N₂O₅ selective conversions, a photolytic or catalytic converter for the NO₂ selective conversion, and a chemiluminescent detector for the NO detection. The flow of stratospheric gas through the system would be operated on in appropriate sequences by these converters to measure each species in turn using the sensitivity detector. The expected sensitivity for NO is 1.3×10^7 molecules/cm³ for a 1-sec integrating time. The expected sensitivity for the remaining species would be comparable. The accuracy for the remaining species, however depends on the specificity of the conversion techniques for each species in the presence of the others and in the presence of other possible nitrogen oxide species in the stratosphere (e.g. HO₂NO₂ and ClONO₂).

The Feasibility Study recommended development and deployment of the Hybrid Gas Conversion System and, simultaneously, a laboratory study of the specificity issues. This report documents the laboratory measurements made to resolve specificity and other issues. Particular emphasis was placed on (1) the conversion of NO₂ to NO by both catalytic and photolytic converters, (2) thermal conversion of N₂O₅ and HNO₃ to NO₂ with and without possible interfering gases, and (3) the feasibility of measuring total odd-nitrogen concentrations using catalytic thermal conversion (in conjunction with the NO and NO₂ measurements). The conversion of NO₂ to NO by catalytic techniques, if reliable, would mean a significant weight reduction in the balloon payload. In brief, the laboratory measurements consisted of breadboarding the various converters, testing

them with known concentrations of species, and generating as necessary those species which are highly reactive. The laboratory measurements were supported by computer modeling of the conversion process and the flow through the instrument. The modeling was then used to extrapolate the performance of the laboratory instrument to its expected performance in the stratosphere. A summary of the results of the laboratory measurements is given in Section II. Section III describes the measurements and their interpretation in detail. Conclusions and recommendations arising from this study are presented in Section IV.

SECTION II

SUMMARY

Extensive laboratory measurements for the detection and measurement of NO, NO₂, N₂O₅ and HNO₃ have been carried out for various types of chemical conversion instrumentation. Particular emphasis was placed on the conversion of NO₂ to NO by both catalytic and photolytic converters, thermal conversion of N₂O₅ and HNO₃ to NO₂ with and without possible interfering gases, and the feasibility of measuring total odd-nitrogen concentrations using catalytic thermal conversion. Computerized chemical kinetics simulations have also been generated to independently support the laboratory measurements and expected performance of instrumentation at stratospheric float altitudes.

For NO measurement, gas phase titration with O₃ is employed followed by the monitoring of the chemiluminescence of an excited state of NO₂ formed during the titration. This method is well understood, and possesses ample sensitivity to meet the HAPP requirement of 10⁸ molecules/cm³ at stratospheric altitudes.

Analytical methods for the remaining three species employed various techniques for quantitatively converting the molecule directly or indirectly to NO, followed by the chemiluminescent technique cited above.

For NO₂ conversion to NO, four techniques were utilized at a typical laboratory concentration of 1 x 10¹⁴ molecules/cm³ and less. Three techniques employed some form of catalyst. The remaining technique was the photolytic conversion of NO₂ to NO using radiation in the 300-400 nm spectral region. The reactant, NO₂ was also monitored using a high temperature catalytic converter followed by chemiluminescence. The product, NO was monitored by the chemiluminescence technique. Photolytic converter efficiency has been determined to be as high as 57 percent and is a function of photon flux and sample residence time. Efficiency, therefore, is a design engineering tradeoff problem.

This technique was evaluated because specificity with other possible nitrogen oxide species of the stratosphere is not an issue. Of the catalytic conversion techniques evaluated, only the high temperature (800 K) catalytic conversion was found to be very efficient for NO_2 but, unfortunately, the technique cannot be used for NO_2 as N_2O_5 and HNO_3 would be simultaneously converted. Chemical-thermal conversion of NO_2 , an adjunct to thermal-catalytic conversion, although not evaluated in the lab, was reviewed as a technique. Efficiency of conversion for this technique, however, has been found to be low when an oxidant such as ozone is present (Lowenstein, private communication, 1978). Catalytic-chemical conversion of NO_2 was evaluated as a technique using both ferrous sulphate, FeSO_4 , and ferrous ammonium sulphate, $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$. With FeSO_4 as a catalyst, not only was the conversion efficiency found to be low but also substantial quantities of NO_2 were lost, presumably by an adsorption process. This technique is known to be a poor converter of NO_2 when an oxidant (O_3) is present in the sample (Ridley and Schiff, 1978). With $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$ as catalyst, the conversion efficiency was found to be high (greater than 97.5 percent) once procedural problems associated with Viton* were eliminated. Catalytic-sorption converters, either platinum or platinum and palladium, on a support column were found to be ineffective for conversion of NO_2 .

Thermal conversion of N_2O_5 to NO_2 at a typical laboratory concentration of 7×10^{13} molecules/cm³ was found to be straightforward with the decomposition going to completion at a temperature of 475 K. Nitrogen pentoxide which cannot be stored at room temperature for any practical length of time was generated as needed in a kinetic flowing system. Product and reactant species number densities were monitored by IR absorption and, for NO_2 and NO , by catalytic-thermal conversion and chemiluminescence, respectively. The rate limiting reaction rate and stoichiometry were monitored a number of times and found to be in agreement with the other workers in the field.

*Viton is a trade name for the copolymer form of vinylidene fluoride and hexafluoropropylene.

Two procedural issues with N_2O_5 observed during the laboratory studies are related to the generation and decomposition of N_2O_5 . First, if moisture is present on the walls of reaction vessels in use during the generation process, HNO_3 is formed and the N_2O_5 yield is less than unity. Although this is a heterogeneous reaction, the homogeneous formation or desorption of HNO_3 was also observed. This HNO_3 formation was employed for simultaneous species testing of thermal converters. Hence the issue was resolved and the artifacts used as an advantage. Second, for most tests carried out, high purity oxygen was employed for generating ozone so as to oxidize NO_2 to NO_3 and thus generate N_2O_5 . For long term stable storage of NO_2 , the diluent gas selected was high purity air. Upon reacting NO_2 and O_3 , the resulting carrier gas was oxygen rich reaching at times 78 percent which is atypical of air. Thus it is conceivable that significant oxidizing reactions were occurring when the experimental objective was to carry out reducing reactions.

Thermal conversion of HNO_3 at a typical laboratory concentration of 2×10^{14} molecules/cm³ to NO_2 and thermodynamically predicted NO was found to be straightforward with the decomposition going to completion at a temperature as low as 535 K. Product and reactant species number densities were monitored by IR absorption and, for NO_2 and NO , by catalytic-thermal conversion and chemiluminescence, respectively. The decomposition is not promoted by a catalyst but does require a surface presumably to scavenge the product OH radical. Computer analysis, (Appendix B), does confirm a faster decomposition rate when OH is scavenged from a stream operated at sea level conditions. Computer analysis for stratospheric conditions where there is a preponderance of O_3 indicates O_3 does not impact conversion, but also that complete conversion requires a temperature of about 800 K. Nitric acid decomposition tests in the presence of nitrogen, separately in the presence of air, and again separately in the presence of oxygen enriched air are presented. Nitric acid decomposition tests in the presence of relatively high levels of O_3 have also been carried out.

Potential interferents reviewed analytically or theoretically included chlorine nitrate, $ClONO_2$, and pernitric acid, HO_2NO_2 . Both are considered

to be labile with the former undergoing conversion at 350 K and the later at about 300 K. If these species are actually present in the stratosphere at number densities greater than instrumentation detection levels, their presence reduces specificity of a particular thermal conversion.

The lack of specificity of thermal converters can actually be used to advantage for the detection of total odd nitrogen species, NO_x . Rather than be concerned about the possible presence of such species as ClONO_2 and HO_2NO_2 or about the partial decomposition of HNO_3 when N_2O_5 is being totally decomposed, all odd-nitrogen species in the stratosphere would be decomposed to NO using a high-temperature catalytic converter. This possibility was recommended in the Feasibility Study and its feasibility is shown here, based on the above measurements and on analytic modeling. The instrument would consist of two similar channels, an NO channel at 250 K and a total odd-nitrogen channel at 800 K.

One new technique under development for the measurement of NO_2 promises sensitive and specific measurements in a lighter and more compact module. Narrowband UV photolysis of NO_2 leaves NO in an excited state from which its characteristic fluorescence can be measured. Also, the technique could be extended to NO by first converting NO to NO_2 as is done in the chemiluminescent detector. The potential of the technique is critically assessed based on discussions with J. Anderson of Harvard and D. Kley of NOAA.

Finally, the analytic computer models of the conversion techniques developed to model the laboratory measurements have been used to assess the techniques under stratospheric conditions. The modeling results for both laboratory conditions and those for the stratosphere's expected concentrations indicate, with only minor differences, identical conversion trending. This permits valid extrapolation of the laboratory results. Thus, it is concluded that there is no fundamental reason why these techniques would not work at the required sensitivities in the stratosphere. However, sample fidelity and surface reactions which are not amenable to modeling must be given closer attention in laboratory conditions closer to those of the stratosphere.

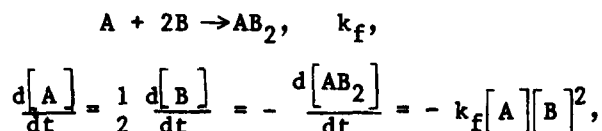
SECTION III

PERFORMANCE TESTING AND EVALUATION

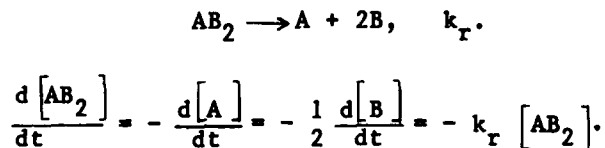
This section is devoted to theoretical considerations and laboratory measurements for various analytical methods for the detection of nitric oxide, nitrogen dioxide, nitrous pentoxide and nitric acid. Generation of the latter two gases at quantitative concentrations is also presented. The final three subsections are devoted to considerations of potential interferents, the feasibility of a total odd-nitrogen measurement, and the promise of a new technique for nitrogen dioxide, NO_2 .

Each subsection presents the appropriate theoretical considerations followed by experimental findings.

As many chemical reactions are listed throughout this report, and their reaction rates evaluated, particular conventions and procedures are employed. Often recourse is made to chemical thermodynamic equilibrium constants. In the main, accepted conventions for notation have been used (Hampson and Garvin, 1978). For example, consider the reaction below with the forward reaction rate coefficient k_f ,



where $[\text{A}]$ is the concentration of constituent A and so forth. Similarly, the backward reaction can be expressed with the reverse reaction rate coefficient k_r ,



At equilibrium, the total time derivatives vanish, leading to the equations

$$k_f [A] [B]^2 = k_r [AB_2];$$

$$K_e = k_f/k_r = \frac{[AB_2]}{[A] [B]^2}, \quad (3-1)$$

where K_e is the equilibrium constant for this reaction expressed in units of concentration (usually moles/liter or molecules/cm³). Usually, only one reaction rate coefficient is known so that an evaluation of K_e must be made in order to derive the opposing reaction rate coefficient. The primary source of information with respect to the equilibrium constant is the set of JANAF Thermochemical Tables. These tables list as a function of temperature the thermochemical properties of most of the compounds of interest in this report. These properties are related to the equilibrium constant by the following expression

$$-R \ln K_p = \Delta(G_T^\circ - H_{ref}^\circ)/T + \Delta(H_{ref}^\circ/T) \quad (3-2)$$

where K_p is the equilibrium constant in units of atmospheres (atm). Δ represents the sum over the products of the reaction minus the sum over the reactants. The quantity $(G_T^\circ - H_{ref}^\circ)/T$ is known as the Gibbs-energy function. The subscript, ref, is the reference point temperature (298.15 K for these tables). ΔH_{ref}° is the heat of formation of the constituent at the reference temperature. The superscript symbol, $^\circ$, signifies atmospheric pressure (760 torr). K_p can be related to K_e .

$$K_p = K_e (RT)^{\Delta n}$$

where Δn is the change in the number of moles of constituents (moles of products minus moles of reactants). Thus,

$$K_e = (RT)^{-\Delta n} \exp \left[-\frac{1}{R} \Delta \left(\frac{G_T^\circ - H_{ref}^\circ}{T} \right) \right] \exp \left[-\frac{1}{RT} \Delta (H_{ref}^\circ) \right].$$

Typically, the Gibbs-energy function changes slowly over the temperature range of interest ($220\text{ K} \leq T \leq 900\text{ K}$). The principal temperature dependence is contained in the second exponential factor. For most reactions $\Delta n = 0$ or ± 1 , so that the effect of temperature in the leading term in the expression is not pronounced. It proved convenient for computer coding to assume that the leading exponential term was independent of temperature.

As noted, usually only one reaction rate coefficient has been measured. A critical review of such coefficients for reactions among atmospheric constituents has been made by Hampson and Garvin (1978). Their recommended values have been used wherever possible in computer modeling codes. Other sources will be cited as required for any reactions not included in the review. If the opposing reaction rate coefficient has also been measured, then it is incorporated directly. Otherwise, the equilibrium constant is evaluated in the manner described above and the opposing reaction rate coefficient is derived, equation 3-2.

Reactions are numbered sequentially throughout the document. The equilibrium constant is denoted as K_{number} , $- \text{number}$. Forward reaction rate coefficients are denoted as k_{number} and reverse reaction rate coefficients as $k_{-\text{number}}$.

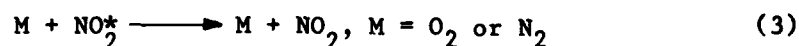
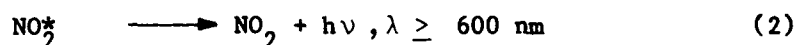
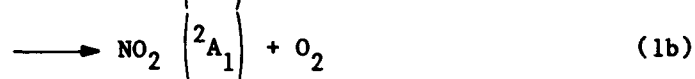
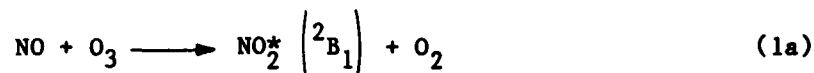
In some cases a laboratory or net stoichiometric reaction is presented and followed by a set of redox reactions or mechanism. As none of the authors are kineticists, Glasstone et al. (1941)-type graphical representations of the mechanism, e.g., potential energy, standard-state molal free energy or enthalpy versus reaction coordinate were considered to be qualitative or too symbolic for use in applying reaction rate theory.

3.1 NITRIC OXIDE CHEMILUMINESCENCE SENSOR

3.1.1 Theory

As a part of CIAP, stratospheric NO was measured by Ridley et al. (1972) using the chemiluminescent method, specifically the ozone oxidation of NO to an optically excited state of NO_2 . The method is also being used by Dr.

Schmeltekopf's group at NOAA for stratospheric measurements and Dr. Stedman's group at NCAR for tropospheric measurements. The method is based upon the reactions:



The bimolecular second order rate constants, k_{1a} and k_{1b} , are given by Clough and Thrush (1967) as:

$$k_{1a} = 7.6 + 1.5 \times 10^{11} \exp(-4180 + 300/RT) \text{ cm}^3\text{-mole}^{-1}\text{-s}^{-1}$$

$$= 1.26 \times 10^{-12} \exp(-4180/RT) \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$$

and

$$k_{1b} = 4.3 + 1.0 \times 10^{11} \exp(-2330 + 150/RT)$$

$$= 7.13 \times 10^{-13} \exp(-2330/RT)$$

Since reaction (1a) has a higher activation energy, luminescence is favored if the temperature is as high as practical. At room temperature, 298 K, the rates k_{1a} and k_{1b} are 1.08×10^{-15} and $1.39 \times 10^{-14} \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$. Therefore, of the NO molecules that react with the O_3 , only $k_{1a}/(k_{1a} + k_{1b})$ or 7.2 percent are converted to NO_2^* . These then radiate via reaction (2) or are quenched via reaction (3).

The intensity of the chemiluminescence (CL) emission is given by Clyne et al. (1964) as:

$$I = \frac{[I_0] \text{ NO } [O_3]}{[M]} \quad (3-3)$$

provided that $k_3 [M] \gg k_2$ with $[M]$ denoting total pressure in molecules/ cm^3 . The value of I_0 is given approximately as $12 \exp(-4180 \pm 300/RT) \text{ photons/s}$

for the spectral region 650 to 875 nm (Fontijn et al., 1970). The broadband output serves to make this a non-specific method for NO, however, experience shows that daytime NO can be reliably measured. Clyne et al. (1964) report a less conservative value of 20.7 for the coefficient. At room temperature, 298 K, $I_0 = 1.0 \times 10^{-2}$ photon/s and at 316 K, 1.54×10^{-2} photon/s. Since $k_2/k_3 = 2.7 \times 10^{14}$ molecules/cm³, the above inequality is valid to pressures as low as 4.5 torr or an altitude of 35 km (Clough and Thrush, 1967).

Achieving linearity with the CL method requires that a negligible net amount of O₃ be consumed by the NO in a flowing system. The combined bimolecular second order reaction rate, k_1 , is given by Johnston and Crosby (1954) as:

$$\begin{aligned} k_1 &= 0.8 \times 10^{12} \exp(-2500/RT) \text{ cm}^3\text{-mole}^{-1}\text{-s}^{-1} \\ &= 0.13 \times 10^{-11} \exp(-2500/RT) \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1} \end{aligned}$$

The rate constant is related to the concentrations through,

$$-\frac{d[O_3]}{dt} = -\frac{d[NO]}{dt} = k_1 [NO] [O_3] \quad (3-4)$$

Assuming that the second reactant gas concentration, $[O_3] \gg [NO]$, the integrated second order rate law is:

$$\frac{1}{[O_3]} \ln \frac{[NO]}{[NO] - \Delta[NO]} = k_1 t = 0.19 \times 10^{-13} t \text{ at } 298 \text{ K} \quad (3-5)$$

or

$$\frac{\Delta[NO]}{[NO]} = 1 - \exp\left(-[O_3] 0.19 \times 10^{-13} t\right) \quad (3-6)$$

If the exponent is made large (4 or 5), then the photomultiplier sensitivity (PMT current per unit NO mass flow rate) is proportional to

$$\left(\frac{k_{1a}}{k_{1a} + k_{1b}} \right) \frac{GQ_{NO}}{P}$$

where G denotes a geometric factor. In terms of reactor volume V, temperature T, reactant flow rate Q_{NO} , and second reactant flow rate Q_{O_3} , the quantity $[O_3]$ after mixing with the sample stream is for a constant volume sampling pump:

$$[O_3] = [O_3'] \frac{Q_{O_3}}{Q_{NO} + Q_{O_3}} \left(\frac{P}{760} \right) \left(\frac{273}{T} \right) \quad (3-7)$$

where $[O_3']$ denotes ozone concentration at the internal generator.

The exponent becomes:

$$\approx 0.19 \times 10^{-13} [O_3'] v \left(\frac{Q_{O_3}}{Q_{NO}} \right)^2 \left(\frac{P}{760} \right)^2 \left(\frac{273}{298} \right)^2$$

Silent discharge O_3 generators can produce as much as 3 percent ozone at a flow rate of 25 scc/s. Using these values and reasonable values for volume and sample flow rate yields:

$$\begin{aligned} [O_3'] &= 3\% \text{ so that } [O_3'] = 8.07 \times 10^{17} \text{ molecules/cm}^3 \\ v &= 1.0 \times 10^3 \text{ cm}^3 \\ Q_{O_3} &= 25 \text{ scc/s} \\ Q_{NO} &= 2.5 \times 10^2 \text{ scc/s} \end{aligned}$$

The exponent and relative extent of the reaction taking place within the reactor can be indicated as a function of altitude, as presented below in Table 3-1. Significant departures from a 100 percent extent of reaction can lead to non-linearities and measurement errors.

TABLE 3-1. $[O_3] k_1 t$ TERM VS ALTITUDE FOR ABOVE VALUES

Altitude, H (km)	Pressure (torr)	$[O_3] k_1 t$	$\Delta[NO]/[NO]$
15	99.0	54.8	1.00
25	21.3	5.9	1.00
35	4.9	86.8	0.934
		4.15	1.00
		0.22	0.98
			0.20

For the parameters selected above, the CL method would be limited to a maximum altitude of about 25 Km. Increased reaction volume, 50 percent, and ozone flow rate, 100 percent, could extend the altitude limit to about 30 Km.

3.1.2 Experimental Procedures and Results

An Aerochem AAS-3S chemiluminescence monitor was extensively employed for general purpose and process control measurements of NO. By reversing the role of the second reactant gas, in this case O₃, ozone could also be monitored. The commercial specifications for this instrument are presented below in Table 3-2.

TABLE 3-2. AEROCHEM RESEARCH AAS-3S SPECIFICATIONS

Useful range	0.1 to 10,000 ppb
Scales	Seven full scales 10, 25, 100, 250, 1000, 2500 and 10,000 ppb
Accuracy	Limited by calibration sample
Linearity	±1% of full scale
Interferences	NH ₃ in NO _x mode
Time Response	11 sec on all scales
Time to Switch from One Gas to Another	Less than 15 seconds
Zero Drift	±2% in 24 hours
Gas Flow Rates	Sample ≈ 2.1 l(STP)min ⁻¹ Air/O ₃ ≈ 0.4 l(STP)min ⁻¹
Normal Temperature Range	15-35°C (59-95°F)
Normal Line Voltage Range	105-125 volts
Electronics	Six inch analog meter on front panel with adjustable recorder output of 0 to 1 volt on rear panel.

Calibration of this type of instrument is based upon round-robin intercomparisons of several instruments and several calibration samples. Field collaborative testing of the chemiluminescence procedure is discussed by Paul C. Constant Jr. et al. (1975).

Design details for the Aerochem monitor, as well as those for the original B.A. Ridley et al. (1972) NO sensor, are given in Table 3-3. Although there are significant design differences between the instruments, the basic detection and measurement of NO is the same. The design differences are related primarily to the internal ambient pressures, reaction volumes, and flow rates and secondarily to the mode of operation used for the photomultiplier tube. The PMT is operated in an analog mode for the laboratory unit whereas for the stratospheric unit it is in a photon counting mode.

Two numerical examples are included in the table for each instrument. For the laboratory unit, an NO mixing ratio of 1 ppb(v) is assumed. For equation 3-3, this yields an intensity of $3 \times 10^5 \text{ hv/cm}^3\text{-s}$. At the anode of the PMT the signal level is given by:

$$S_a = IV_R eG(Q.E.) \quad (3-8)$$

where V_R denotes reaction volume. For the above example, assuming a 10 percent effective quantum efficiency, there is a 3×10^{-7} amp signal level. For the stated anode dark current a noise equivalent detection limit of 7×10^{-3} ppb can easily be derived. For the stratospheric unit, an NO mixing ratio of 4 ppb(v) was adopted. For a reaction temperature of 300 K, equation 3-3 yields an intensity of about $1.9 \times 10^4 \text{ hv/cm}^3\text{-s}$. Since a mixing ratio of 4 ppb yields a signal level of 1.7×10^3 counts/s, the effective emitting volume-quantum efficiency product of 0.088 cm^3 is obtained. Assuming an effective quantum efficiency of 5 percent, an effective emitting volume of about 1.8 cm^3 is obtained, which is considerably less than the actual reaction volume.

With stratospheric instrumentation, a lowest detection limit of 5×10^7 molecules/ cm^3 has been verified (B.A. Ridley et al., 1972).

TABLE 3-3. DESIGN COMPARISON OF NO CHEMILUMINESCENCE SENSORS

Developer	Aerochem Research Lab (1977)	B.A. Ridley et al., (1972)
Type	Laboratory Monitor	Stratospheric Research
Operating Pressure	250 torr	33 torr
Sample Flow Rate	35 scc/s	210 scc/s
Reactant Flow Rate	6.7 scc/s	10 scc/s
Reaction Volume, V_R	10 cm ³	400 cm ³
Reaction Temperature	316 K	≈300 K
Residence Time	0.07 s	1.9 s
Detector	Centronics 4283 RA	EMI 9558 QA ^(b)
Temperature	5°C	-20°C
Gain	6x10 ⁶	3x10 ⁷
Dark Current	2.0 nA	85 cts/s
Conversion Factor ^(a)	9.76x10 ⁻¹³ cm ³ -coul	8.9x10 ⁻² cm ³
NO Mixing Ratio	1 ppb(v)	4 ppb(v) at 33 torr
NO	9.64x10 ⁹ molecules/cm ³	4.25x10 ⁹ molecules/cm ³
O ₃	1.97x10 ¹⁶	4.83x10 ¹⁴
M	7.64x10 ¹⁸	1.06x10 ¹⁸
$I_o[NO][O_3]/[M]$	3.0x10 ⁵ hv/cm ³ -s	1.94x10 ⁴ hv/cm ³ -s
Signal	3.0x10 ⁻⁷ A	1.7x10 ³ cts/s
Responsivity	300 nA/ppb	10 ⁻¹ nA/ppb (360 cts/ppb)
Noise Equivalent Detection Limit	0.007	0.05 ppb
Lowest Detection Limit	0.1 ppb	0.05 ppb
Linearity	0.1-10 ⁴ ppb (1%)	0.05-34 ppb

(a) Conversion factor = $V_R eG(Q.E.)$.

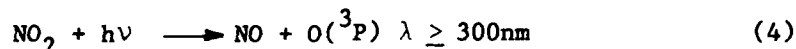
(b) An RCA 31034A photomultiplier was used on the first flight yielding a responsivity of 1855 cts/ppb and a lowest detection limit of 0.02 ppb at an altitude of 22 Km or about 50 torr (Ridley et al., 1974).

3.2 NITROGEN DIOXIDE PHOTOLYTIC CONVERSION

Measurement of stratospheric NO_2 can take place by photolyzing the NO_2 to nitric oxide, followed by gas phase titration with O_3 . This method has been developed and is being used by the NOAA group (MacFarland et al., private communication, 1977). This technique was evaluated because specificity with other possible nitrogen oxide species of the stratosphere is not an issue.

3.2.1 Theory

The mechanism for the photolytic conversion of NO_2 to NO and ground state atomic oxygen is at least a four step process; namely:



The above reaction set is a subset of the set employed by Stedman and Niki (1973) to account for the kinetics and mechanism for the photolysis of NO_2 in air. The expanded set includes O_3 , NO_3 and N_2O_5 reactions. If only reactions 4, 5 and 7 are considered, the net quantum yield is zero. If only reactions 4 and 6 are considered, the net quantum yield is two. The actual yield lies between these values.

For the above set, the rate of change of NO_2 is given by:

$$-\frac{d[\text{NO}_2]}{dt} = k_4 [\text{NO}_2] + k_5 [\text{O}] [\text{NO}_2] [\text{M}] + k_6 [\text{O}] [\text{NO}_2] - 2k_7 [\text{NO}] [\text{NO}_3] \quad (3-9)$$

Since NO_3 will have a rapidly defined (microseconds) stationary value, i.e., $d[\text{NO}_3]/dt = 0$, the equation

$$k_5 [\text{O}] [\text{NO}_2] [\text{M}] = k_7 [\text{NO}_3] [\text{NO}] \quad (3-10)$$

can be used to simplify equation 3-9 to

$$-\frac{d[\text{NO}_2]}{dt} = k_4 [\text{NO}_2] - k_5 [\text{O}] [\text{NO}_2] [\text{M}] + k_6 [\text{O}] [\text{NO}_2] \quad \text{or} \quad (3-11)$$

$$\frac{d \ln [\text{NO}_2]}{dt} = \frac{2k_1 k_6}{k_6 + k_5 [\text{M}]}$$

Equation 3-11 is often simplified to a first order dissociation. Examining the validity of this statement, one notes that $k_5 = 1.0 \times 10^{-31} \text{ cm}^6 \text{ molecule}^{-2} \text{ s}^{-1}$ and $k_6 = 9.2 \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ (Hampson and Garvin, 1978). At 1 atmosphere $M = 2.446 \times 10^{19} \text{ molecules/cm}^3$, so that $k_5 [\text{M}] = 2.446 \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ or about one-fourth that of k_6 . For $k_4 < 0.75 k_6 [\text{O}]$, the simplification is valid and the integrated form of equation 3-11 becomes:

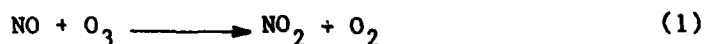
$$\frac{[\text{NO}]_t}{[\text{NO}_2]_0} = 1 - \exp(-I \sigma t) \quad (3-12)$$

$$\Sigma I(\lambda) \sigma(\lambda) d\lambda \quad (3-13)$$

where $I(\lambda)$ corresponds to the photon flux rate and $\sigma(\lambda)$ the absorption cross-section. This expression assumes a quantum yield of unity for the dissociation, a valid assumption for $244 \leq \lambda \leq 398 \text{ nm}$ (Hampson and Garvin, 1973). The cross-section data of Johnston and Graham (1974) and that of NASA Publication 1010 (1977) have been used to obtain a value of k_4 .

Since the conversion efficiency is dependent upon k_4 and sample residence time, t , it is important to maximize their product.

When O_2 is present in the sample stream, the above analysis is inadequate to predict experimental results. Two additional reactions impairing conversion efficiency are:



when significant amounts of O_3 and NO are formed, reaction (1) inhibits the photolytic dissociation process.

The converter design selected used 1 kw a mercury capillary lamp (Illumination Industries, Inc. type AH6-1BC) in conjunction with Corning 7-54 and Pyrex filters to restrict the photon flux to a spectral region defined by approximately 300-400 nm. The photolytic rate for a unit area of radiation is given by

$$k_4 A = \frac{10^7}{hc} \int_{280}^{420} \tau_\lambda (P/\Delta\lambda) \sigma_\lambda d\lambda \text{ cm}^2\text{-s}^{-1} \quad (3-14)$$

where

- λ denotes wavelength in cm
- h equals Planck's constant = 6.62×10^{-27} erg-s
- c equals the velocity of light = 3×10^{10} cm/s
- 10^7 denotes conversion factor for joules to ergs
- τ_λ denotes spectral transmittance of the optical filters
- $P/\Delta\lambda$ denotes total radiated watts per 10 nm increments
- σ_λ denotes NO_2 absorption cross-section in $\text{cm}^2/\text{molecule}$, and
- $d\lambda$ denotes variable of integration

With a new lamp operating at 1 kw and with the filters indicated, evaluation of the above expression yields $k_4 A = 74 \text{ cm}^2\text{-s}^{-1}$ and $k_4 = 0.26 \text{ s}^{-1}$ for this design. The residence time for this cell is approximately 5 seconds so that the conversion efficiency is predicted to be 73 percent. The actual design and measured results for the photolysis of NO_2 is treated in the next two paragraphs.

3.2.2 Stratospheric-Based Instrumentation Modeling

Computer modeling predictions for stratospheric pressure are considered. For the case where $P = 19$ torr (25km), $k_5 [M]$ is reduced by a factor of 40 and the photolytic conversion efficiency increases from 12.8 percent to 44 percent for $k_4 = 0.26 \text{ s}^{-1}$. Thus the performance of the photolytic technique improves with reduced pressure.

3.2.3 NO_2 Photolytic Converter Design

Prior to discussing the procedures and results for the NO_2 photolytic converter, the design of the laboratory converter is presented. The design

of the overall converter was based upon the availability of a photochemical lamp, expedience of attaining the proper spectral filtering and material chemistry aspects.

The high pressure mercury arc capillary lamp is of the straight bore type with a rating of 1 kva. The lamp, part number AH6-1BC, was obtained from Illumination Industries, Inc. To dissipate non-radiative heat, a water cooling jacket was provided. The jacket comprises two concentric cylinders made of quartz and Pyrex. The Pyrex was selected to attenuate radiation below about 300 nm. To suppress radiation above about 400 nm, Corning filter glass 7-54 was employed and configured as a hexagonal cylinder circumscribing the outer cooling jacket. The filter glass elements were held in place by teflon disks supported inside a gold-plated aluminum cylinder which served as the photolytic reaction vessel.

Lamp cooling with high pressure Hg lamps is critical and one is constrained to maintaining the temperature over a finite range so that the proper vapor pressure of mercury is attained while simultaneously minimizing stresses in the quartz envelope. Also steam must be prevented from forming on the quartz envelope. To achieve the above requirements the lamp is used in a horizontal position, chilled de-ionized water is flowed through the water jacket assembly at a rate of 1-1.5 gpm, and an inner jacket or "velocity tube" is employed to minimize the water volume near the lamp envelope, thus increasing the flow velocity.

An a-c power supply, employed for driving the lamp, consists of a line voltage variac, 4 kva transformer, current monitor and voltage monitor. The transformer of a soft iron design has built-in leakage inductance. To assist in starting the lamp at a reasonable voltage close to operating voltage, a small amount of argon is contained within the lamp envelope. The voltage waveform is typically a square wave with a slight overshoot on the leading edge.

Design parameters for the photolytic cell are given in Table 3-4.

TABLE 3-4. NO₂ PHOTOLYTIC CELL DESIGN

Parameter	Value
Lamp Luminous Length	2.54 cm
Lamp Bore Diameter	0.2 cm
Quartz Velocity Tube ID, OD	12, 15 mm
Pyrex Tube ID, OD	36, 41 mm
Corning Filter Facet Width	2.2 cm
Sample Cylinder ID	10.3 cm
Effective Radiative Area	280 cm ²
Radiated Sample Volume	177 cm ³
Sample Flow Rate	2.1 SLPM
Sample Residence Time	5.05 s
Lamp Current	1.2 amps
Lamp Voltage	435-700 V
Lamp Power	552-840 W
Near UV (320-400 nm)	112.3 w @ 1 kva
Radiated Power	
365 nm Radiated	3.6 w/nm
Spectral Power	
Radiated Power/Input Power	0.46
Materials Exposed to NO ₂ Sample	Gold, stainless steel 303, Teflon, Corning glass, Pyrex
NO ₂ Design Conversion Efficiency @ 1 kva	73 percent

3.2.4 Experimental Procedures and Results

Pre-mixed and calibrated samples of NO₂/air were obtained from Airco in aluminum Spectra-Seal[®] cylinders. Airco calibrations of 0.6-10 ppm concentrations were verified using the catalytic thermal converter of the Aerochem CL monitor discussed in paragraph 3.1.

Sample pressure was reduced and regulated using Matheson type regulators containing stainless steel diaphragms. Sample flow rate was set by the Aerochem CL monitor pump and internal sonic orifice. All other plumbing was either stainless steel or teflon[®] (PTFE)*.

*PTFE, polytetrafluorethylene

Relatively new lamps (less than 10 hours of aging) were employed; however no effort was expended in logging the number of on-off cycles which do affect lamp life. In operation the cell was connected in series with the CL monitor. With the Hg lamp in the off state, the CL monitor was set to the NO_x mode to measure the NO_2 entering the photolytic cell. The lamp was then excited and the CL monitor was placed in the NO mode to measure the amount of NO leaving the photolytic cell. If there was no or negligible NO in the supply gas then the ratio of the two measurements was simply the converter efficiency.

The results of the tests are given in Table 3-5 for various lamp power levels. Assuming a linear relation between input power and radiated power, the various anticipated rates and converter efficiencies are included. The measured efficiencies do not correspond directly to the calculated efficiencies but this is expected, since the rates were derived from brochure data and not measured values. The computer derived efficiencies are in large variance with the measured efficiencies. An attempt to account for this variance is given below.

TABLE 3-5. CALCULATED AND MEASURED NO_2 PHOTOLYTIC CONVERTER EFFICIENCIES

Lamp Voltage (Volts)	Lamp Power (Watts)	k_4 (s^{-1})	First Order Calculated Efficiency (Percent)	Computer Code $\Delta[\text{NO}] / \Delta[\text{NO}_2]$	Measured Efficiency (Percent)
435	522	0.14	50	0.58	35
525	630	0.17	57	0.64	45
700	840	0.22	67	0.63	57
-	1000	0.26	73	0.62	-

The oxidation of NO_2 by $\text{O}(^3\text{P})$ has a predominating effect for producing the intermediary, NO_3 with $k_5 M$ being $2.36 \times 10^{-12} \text{ cm}^3/\text{molecule-s}$. The NO_3 is then available to react with NO_2 forming N_2O_5 , or with NO forming NO_2 . Formation of NO_2 , reaction (7), proceeds at a rate about an order of magnitude faster than the formation of N_2O_5 . Thus, as previously mentioned, the quantum

yield is reduced. If the ground state oxygen atom were scavenged by some other means, then the predicted yield would increase.

From this data one notes that efficiency is a design engineering trade relating power with photon flux as well as sample residence time. The residence time is of course related to the selected photolytic cell volume and the flow selected for the high sensitivity NO chemiluminescence monitor.

3.3 NITROGEN DIOXIDE CATALYTIC CONVERSION

Results obtained in employing several types of catalytic converters for NO₂ measurements are reported in this section. The types included are (1) catalytic-thermal converters, (2) chemical-thermal converters, (3) catalytic-chemical converters and (4) catalytic-sorption converters.

Significantly different results have been found for type (1) and type (2) converters when O₃ is present. A type (1) converter used in this study employs a noble metal while the type (2) converters investigated by Dr. Max Lowenstein of NASA employed a metal readily oxidizable. Section 3.3.1.1 attempts to justify the widely disparate results observed with these converters.

3.3.1 Catalytic/Chemical Thermal Converters

3.3.1.1 Theory

Since much of the data reported upon in subsequent sections is based upon catalytic thermal conversion, a review of thermal converters both catalytic and chemical is warranted.

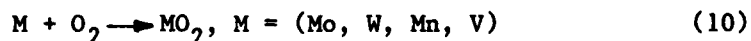
Table 3-6 lists various metals and carbon which have been used in NO_x or NO₂ converters. The listing has been arranged in decreasing value of the Gibbs or free energy formation of the oxide of the metal (or carbon). If the free energy value listed is negative, the reaction can occur. The reverse reaction, decomposition, would then have a positive value and would not occur. The last column corresponds to the minimum temperature required to achieve about 98 percent conversion of NO₂ for non-oxidized surfaces (L.P. Breitenbach and M. Shelef, 1973).

TABLE 3-6. FREE ENERGY OF FORMATION $\Delta F'$ (kcal/OXYGEN ATOM) AND TEMPERATURE FOR 98 PERCENT NO_2 CONVERSION

	300 K	400 K	500 K	T(°C)
Unstable Oxide X ^(a)				600-1000
Ag			+ 0.06	
C	-32.8	-35.0	-37.1	400-600
Ni			-46.1	750 ^(b)
Fe	-59.2	-57.1	-54.9	725-750 ^(c)
Mo	-63.6	-61.4	-59.3	525
W	-63.7	-61.5	-59.3	475
Mn			-83.1	475
Stable Oxide V			-89.0	475
(a) Other noble metals (b) Inconel; 80% Ni, 14% Cr, 6% Fe (c) Monel; Stainless steel 304 and 316				

As noble metal oxides have positive energies of formation, they are unstable and do not readily form in the presence of oxidants. Generally, they act as catalysts for many oxidizing and reducing reactions. In the case of NO_2 conversion, their inertness quality must be traded off against minimum conversion temperature.

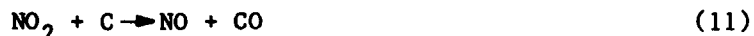
For the remaining elements, NO_2 conversion proceeds according to the reactions:



With increased usage, i.e. surface oxidation, the conversion temperature must be increased to effect the same conversion efficiency.

For the elements with negative free energies of formation, the element acts as an oxygen scavenger. As the free energy of formation becomes more

negative, the conversion temperature can be reduced since the oxidation becomes more preferred. For any given metal, as the temperature is increased, the free energy of oxide formation is shifted toward more positive values, thus reducing its effectiveness for NO_2 reduction. The use of carbon, coke in metallurgical ore reduction, permits lower temperature operation and the conversion proceeds as:



As C is converted to CO or CO_2 in reducing NO_2 or by virtue of the oxygen molecules in an air stream, the surface material, C or M, becomes ineffective.

If a strong oxidant, O_3 , H_2O_2 , HNO_3 , or HNO_4 , is present in the sample stream, the surface material, C or M, is rendered ineffective at a faster rate.

From the above discussion, two facts are evident for NO_x converters. First, thermal converters employing true catalysts, e.g. platinum, stainless steel, and possibly nickel alloys, should be identified as catalytic-thermal converters, whereas the others (eg. Mo) should be identified as chemical-thermal converters. Second, chemical-thermal converters as defined above, can be expected to lose their efficiency if strong oxidants at significant concentration levels are present in the sample. The term "poisoning" is sometimes used to describe this efficiency loss but the definition is loose. Finally, to maintain efficiency, oxidant scavengers or scrubbers are often considered.

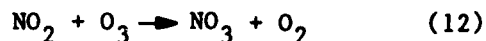
In the absence of any catalytic material, thermal conversion of NO_2 can also occur (Altshuller, 1957 and Breitenbach, 1973). The reaction, (9), is well understood and has been used to set converter temperatures. For oxygen at 152 torr, the ratio of $[\text{NO}_2]$ to $[\text{NO}]$ is given by:

$$\log \frac{[\text{NO}_2]}{[\text{NO}]} = \frac{3002}{T} - 4.2226 \quad (3-15)$$

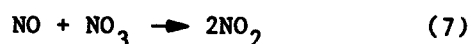
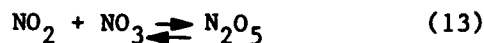
3.3.1.2 Experimental Procedures and Results: Chemical-Thermal Converters

Experimental work by Dr. Max Lowenstein of NASA-ARC (private communication, 1978) has shown that for medium temperature molybdenum-based converter construction, NO_2 conversion may drop to zero if O_3 is present in the sample stream. It is felt that a two-body chemical reaction plus surface chemistry may be occurring.

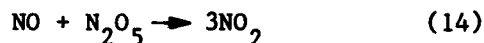
The following two-body mechanism may also be occurring as an abnormal situation. First,



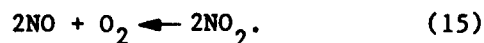
which is very fast, a factor of 50 due to the high temperature, when compared to a 298 K temperature rate. The mechanism continues as follows:



with a net result of



and also

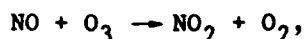


This last reaction is of the third order type and extremely slow with a rate constant

$$k_{15} = 3.3 \times 10^{-39} \exp(-1050/RT) \text{ cm}^3\text{-molecule}^{-2}\text{-s}^{-1}.$$

3.3.1.3 Experimental Procedures and Results: Catalytic-Thermal Converters

The commercial converter in use at Perkin-Elmer is constructed of inconel, alumina and a 90% Pt-10% Rh wire heating filament. As such, it approximates an catalytic-thermal converter. Conversion efficiency of this unit, without O_3 present, was first determined by the accepted Federal Register (1973) method, which is based upon reaction (15) given above and the fast reaction ($k_1 = 1.66 \times 10^{-14} \text{ cm}^3/\text{molecule-s}$);



(1)

both of which produce stoichiometric quantities of NO_2 . For all tests with the converter at its design temperature of 1100°C , conversion was greater than 99.7 percent.

The next set of conversion determinations for the commercial unit were made with about 9 ppm O_3 present and 9.5 ppm NO_2 . The average conversion efficiency was found to be 97.9 percent or about 1.8 percent less, which is in concurrence with findings at NASA-ARC but not of real significance.

Generation of NO from $\text{O}(^3\text{P})$, a product of O_3 thermal decomposition, is a possibility if surface recombination of atomic oxygen does not occur on the platinum surface. A brief series of tests was carried out to assure that the surface recombination mechanism was in fact sufficient to scavenge $\text{O}(^3\text{P})$. The catalytic thermal converter of the Aerochem CL monitor was used at reduced temperature, approximately 750 K, and a stream of ozonized air was passed over the platinum-rhodium heating filament. The ozone concentration level was typically 2.7×10^{13} molecules/cm³ (1.1 ppm). Measurement of NO_x downstream from the converter was typically 2.45×10^{10} molecules/cm³ (1.0 ppb) greater than the measured (NO_x) upstream of the converter. During these tests, the trace contamination NO level of the air source was 1.6×10^{11} molecules/cm³ (6.5 ppb), while the instrument noise level was 2.45×10^9 molecules/cm³ (0.1 ppb). Summarizing this test data one may conclude that 99.9 percent of the $\text{O}(^3\text{P})$ atoms recombined, thereby providing effective scavenging.

It is important to note here that (1) catalytic thermal conversion is more effective for reducing NO_2 to NO than thermodynamic conversion and (2) that large relative levels of O_3 do not seriously impact the conversion process.

3.3.2 Catalytic-Chemical Converters

3.3.2.1 Theory

For this type of converter ferrous sulphate and ferrous ammonium sulphate were employed.

Although the gas-solid phase chemistry is not well known for the action of NO_2 and FeSO_4 , the appropriate acidic solution is well known. Fe^{++} reduces HNO_2 to NO , which in turn combines with Fe^{++} to form $\text{Fe}(\text{NO})^{++}$. This ion has a characteristic dark brown color. If NO_3 is present, in the absence of the NO_2 , reduction occurs, forming as before $\text{Fe}(\text{NO})^{++}$. This latter reaction occurs only when H^+ are prevalent and temperature is relatively higher. These conditions occur at the liquid-liquid boundary layer, and a brown ring forms.

Ferrous ammonium sulphate is the better converter choice since it is known that FeSO_4 selectively removes oxidants, specifically ozone (Miller et al., 1971). The Ridley-Schiff group also confirmed this fact in early 1978 when employing $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ to convert ppb levels of NO_2 when ppm levels of O_3 were present.

3.3.2.2 Experimental Procedures and Results

Tubular flow converters were constructed with glass, quartz wool packing and stainless steel fittings along with the dehydrated form of the sulphate.

The initial converters were sized to yield a space velocity (flow rate/catalyst volume) of about 93 min^{-1} and a superficial linear velocity (flow rate/converter cross-sectional area) of about $2.4 \times 10^3 \text{ cm/min}$. The final converter used extensively with $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$ was designed to yield a space velocity of about 17.5 min^{-1} and superficial linear velocity of about $5 \times 10^2 \text{ cm/min}$.

Initial tests (1 through 3) for each catalyst resulted in large percentage losses of the NO_2 as presented in Tables 3-7 and 3-8. Subsequent testing indicated that viton seals were acting as adsorbents. Particular emphasis was placed upon $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$ since FeSO_4 is a selective remover of O_3 . Procedural changes resulted in no detectable NO_2 loss for the subsequent $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$ tests. Conversions were found to be consistently greater than 97.5 percent.

The use of $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$ is a competitive technique to NO_2 300-400 nm photolysis, however, careful packaging design of the catalyst bed is required

TABLE 3-7. FERROUS SULPHATE CATALYTIC CONVERTER TEST RESULTS

Test	Species	1	2	3
Pre-converter concentration	NO _x	4.2ppm	68ppb	68ppb
	NO ₂	2.4	62	62
	NO	1.8	6	6
Post-converter concentration	NO _x	3.6	33.5	30
	NO ₂	0	19.8	17.5
	NO	3.6	13.7	12.5
NO ₂ converted	Δ NO	1.8	7.7	6.5
NO ₂ unconverted		0.0	19.8	17.5
NO ₂ entrapped	Δ NO _x	0.6	34.5	38.0
NO ₂ % conversion		75	12.4	10.5
NO ₂ % unconverted		0	31.9	28.2
NO ₂ % entrapped		25	55.6	61.3

so as not to reduce sample flow rate significantly nor represent a significant pressure drop.

3.3.3 Catalytic Sorption Converters

For this series of tests, Matthey-Bishop catalysts, 1% Pt and 0.4% Pd with 0.1% Pt on support columns, were contained in glass tubing. The combined weight of the pellets was about 5 gms. Nitric oxide was reacted with O₃ to provide NO, NO₂ and a minimal amount of O₃. The resulting concentrations at the converter input are shown below for two tests using the 1% Pt catalyst.

	Test 1	Test 2
NO	3.15 ppm	5.20 ppm
NO ₂	2.65	2.95
NO _x	5.8	8.15
O ₃	5.5 ppb	0.5 ppb

TABLE 3-8. FERROUS AMMONIUM SULPHATE CATALYTIC CONVERTER TEST RESULTS

Test (a)	Species	1 (ppb)	2 (ppb)	3 (ppb)	4 (ppm)	5 (ppm)	6 (ppm)	7 (ppm)	8 (ppm)	9 (ppm)
Pre-converter concentration	NO _x	437	437	443	2.68	4.40	7.20	4.65	2.74	1.55
	NO ₂	435	435	429	2.68	4.40	7.20	4.65	2.74	1.55
	NO	2	2	14	0.0	0.0	0.0	0.0	0.0	0.0
Post-converter concentration	NO _x	100	80	95	2.68	4.40	7.20	4.65	2.74	1.55
	NO ₂	70	30	47	0.06	0.09	0.09	0.015	0.04	0.03
	NO	30	50	48	2.62	4.31	7.11	4.50	2.70	1.52
NO ₂ converted NO ₂ unconverted NO ₂ Entrapped	NO	28 70 337	48 30 357	34 47 348	2.62 0.06 0.0	4.31 0.09 0.0	7.11 0.09 0.0	4.50 0.15 0.0	2.70 0.04 0.0	1.52 0.03 0.0
	NO ₂ % conversion	6.44	11.0	7.9	97.8	97.9	98.7	96.8	98.5	98.1
	NO ₂ % unconverted	16.1	6.9	10.9	2.2	2.1	1.3	3.2	1.5	1.9
	NO ₂ % entrapped	77.5	82.0	81.1	0.0	0.0	0.0	0.0	0.0	0.0

(a) Tests 1 through 3 employed a converter with a space velocity of about 93 min⁻¹ and a superficial linear velocity of about 2.4 x 10³ cm/min. Tests 4 through 9 employed a converter with a space velocity of about 17.5 min⁻¹ and a superficial linear velocity of about 5 x 10² cm/min. Tests 1 through 3 were conducted with viton seals which caused substantial absorption.

The NO , NO_x concentrations at the output of the converter were found to be equal to each other and equal to 1.42 and 2.80 ppm, respectively. From this data it was concluded that (1) NO_2 either converted, or was adsorbed or both, (2) a portion of the NO was adsorbed, and (3) from this data, it may be concluded that these materials are not considered to be useful. In light of these results, further planned tests were abandoned.

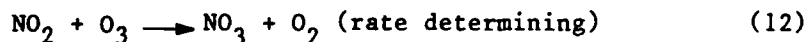
3.4 NITROUS PENTOXIDE (N_2O_5) GENERATION

As nitrous pentoxide (N_2O_5) is reactive and unstable in the gaseous state, it was generated as needed. For N_2O_5 synthesis, excess NO_2 or excess O_3 and the other reactant were mixed in an elementary type flow system. The reactant product NO_3 and NO_2 combined yielding N_2O_5 . This technique approximates the approach used by others.

Process control and determination of initial concentration levels, i.e., before dynamic dilution, were carried out by IR absorption techniques. Measured concentration levels as opposed to predicted levels were desired because of the reactive nature of the particular species. The levels after dilution were then employed in the overall assessment of an appropriate chemical converter ahead of a chemiluminescent monitor. The specific details of the IR measurements are discussed in Appendix A.

3.4.1 Theory

Generation of gas phase N_2O_5 is best carried out by the ozone oxidation of NO_2 as it is reasonably stable in the presence of ozone. The governing reactions are:



where the equilibrium constant, $K = 1.2 \times 10^{-11}$ molecules/ cm^3 at a temperature of 300 K. The reaction can be carried out with either $[\text{NO}_2]$ or $[\text{O}_3]$ in excess. Reaction (12) is about 34 kcal/mole exothermic. Measured stoichiometry $[\Delta\text{NO}_2/\Delta\text{O}_3]$ values are 1.88 and 1.68 for excess O_3 and NO_2 , respectively (Wu et al., 1973), and 1.89 ± 0.08 (standard deviation) (Graham, 1975).

Various experimental rate constant values for k_{12} are shown as an Arrhenius plot in Figure 3-1. The Arrhenius parameters for the results of Johnston and Yost (1949) are

$$k_{12} = 9.82 \times 10^{-12} \exp(-7000/RT) \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$$

and for the results of Graham (1975) are

$$k_{12} = (1.34 \pm 0.11) \times 10^{-13} \exp(-4900 \pm 60/RT) \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$$

where the uncertainties are standard deviation.

For design purposes an average value of $5.37 \pm 1.76 \times 10^{-17} \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$ ($1.32 \pm 0.43 \times 10^{-3} \text{ ppm}^{-1}\text{s}^{-1}$) at 298 K and 760 torr was chosen.

The integrated rate equation for the bimolecular reaction is given by:

$$\frac{1}{[\text{NO}_2]_0 - [\text{O}_3]_0} \ln \frac{[\text{O}_3]_0 ([\text{NO}_2]_0 - [\text{NO}_3])}{[\text{NO}_2]_0 ([\text{O}_3]_0 - [\text{NO}_3])} = k_{12} t (\text{conc})^{-1} \quad (3-16)$$

where the subscript denotes original concentration levels. This expression is plotted in Figure 3-2 for $[\text{NO}_2]_0 = 25 \text{ ppm}$ and $[\text{O}_3]_0 = 10 \text{ ppm}$.

3.4.2 Experimental Procedures

The attendant measurements required to follow the reaction include: (1) initial concentrations and flow rates, (2) the $[\text{O}_3]$ as the sample leaves the reaction vessel so that the extent of the reaction (c.f., Figure 3-2) can be determined, and (3) the $[\text{NO}_2]$ as the sample leaves the reaction vessel so that the stoichiometry can be determined. The latter measurement must be made by a technique that does not destroy either NO_3 or N_2O_5 (e.g., IR absorption). The transit time of the sample between the reaction vessel and the appropriate monitor must be known or be small compared to the residence time of the sample in the reaction vessel.

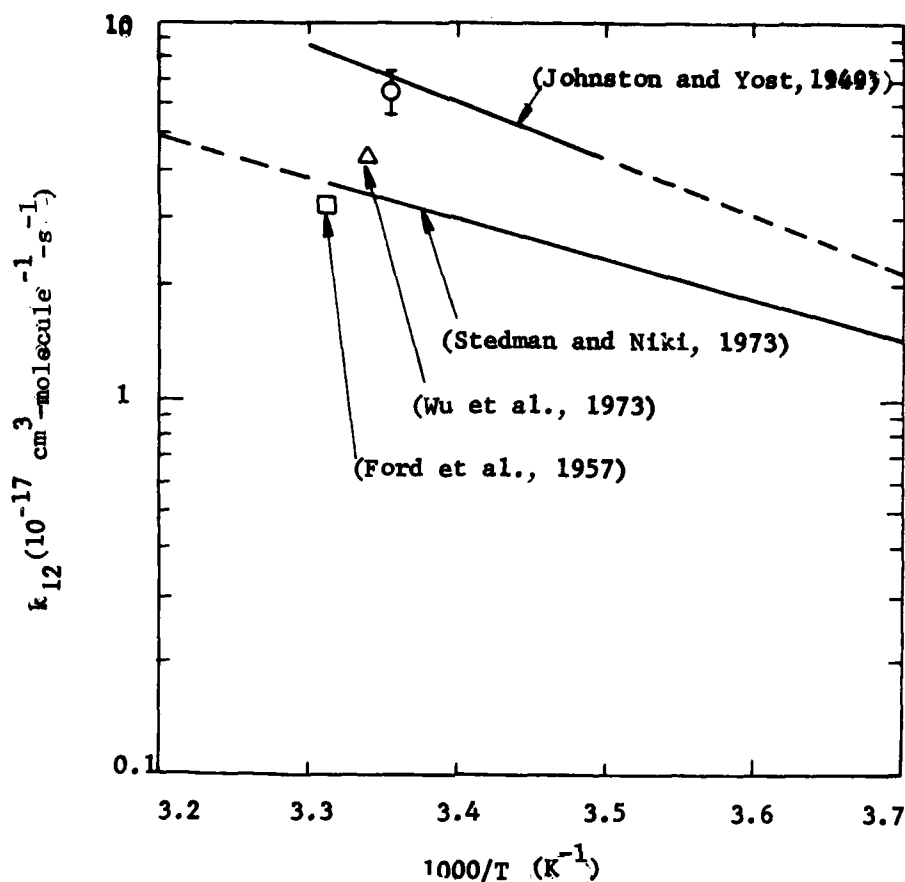


Figure 3-1. Arrhenius Plot For $\text{NO}_2\text{-O}_3$ Reaction

Kinetic runs were carried out initially at 0.25 and 0.50 SLPM with 13.1 ppm NO_2 and 26.0 and 13.0 ppm O_3 , respectively, prior to dilution by the carrier of the other reactant. Subsequent tests were made at a total flow rate of 2.1 SLPM yielding a residence time of about 408 seconds. The reaction vessel volume of 14.3 liters and surface-to-volume ratio of 0.4 cm^{-1} consisted of a 1 liter stainless steel sampling cylinder with concentric counter flow tubing, a 7.8 liter stainless steel vessel and a 5.5 liter PTFE-lined vessel with AgCl windows that also served as a 20 meter White cell for IR absorption measurements. Measurements were usually carried out with a pathlength of 15.75 meters. The stainless steel vessels were vacuum baked prior to ozone conditioning of the walls and flow of the reactants. The above conditioning usually lasted for several hours. Flow rates were controlled by Tylan FC-260 flow controllers. These units were modified slightly by substituting

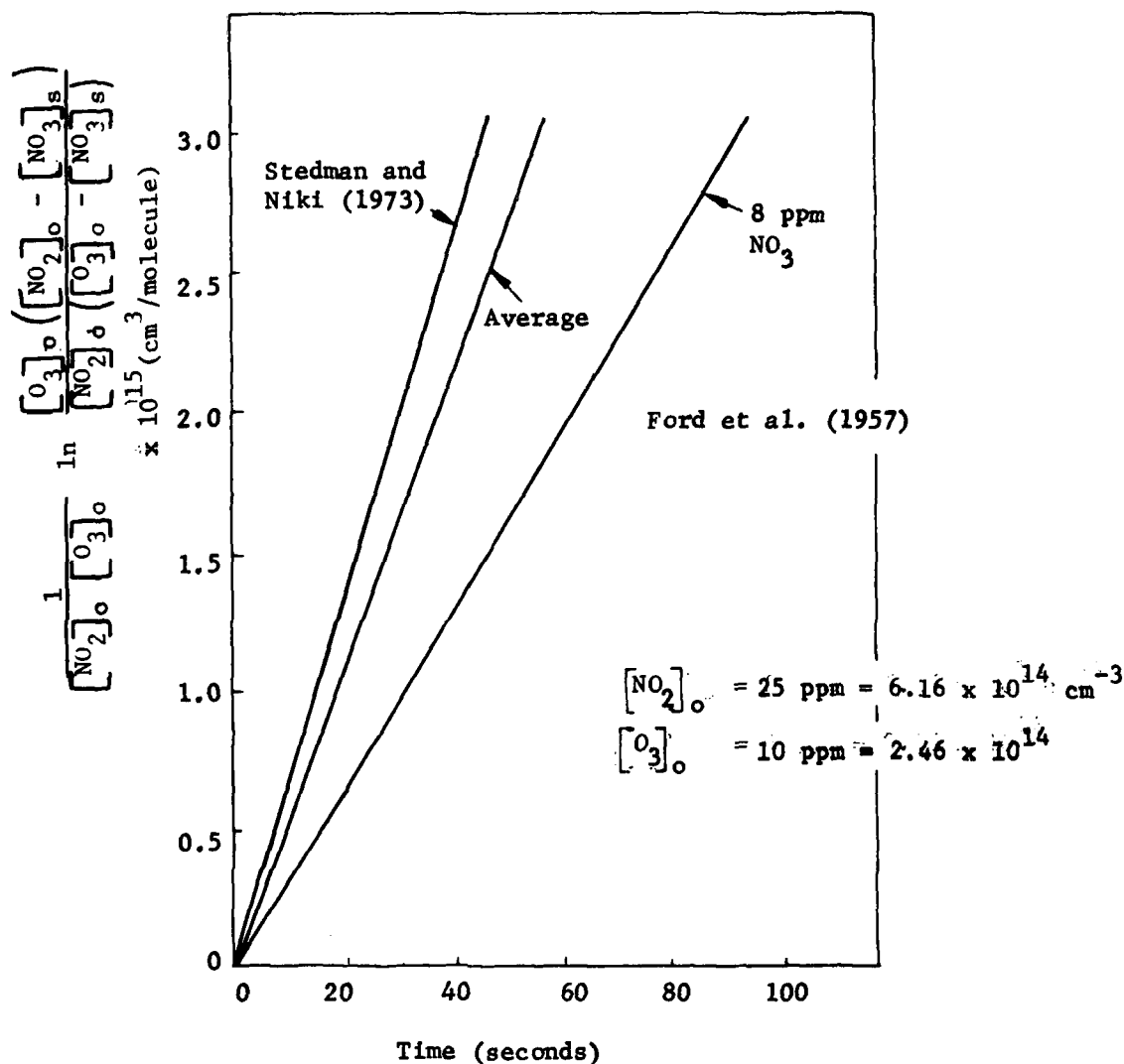


Figure 3-2. NO_3 -Generation versus Time ($T = 298 \text{ K}$)

PTFE O-rings for Viton O-rings. Steady-state flow conditions were usually reached after about 20 minutes. Infrared transmittances over the $1800\text{--}1000 \text{ cm}^{-1}$ spectral range of the White cell contents were employed to follow the course of the reaction. As the flowing system desiccated itself with use, the relative amount of HNO_3 decreased substantially and improved the N_2O_5 yield and standard deviation.

The initial quantitative tests are tabulated in Table 3-9. The measured stoichiometry of 1.86 ± 0.06 is in excellent agreement with the work of Wu

TABLE 3-9. N_2O_5 GENERATION AND STOICHIOMETRY MEASUREMENTS (a)

Expt. No.	$[O_3]_i$	$[O_3]_f$	$[NO_2]_i$	$[NO_2]_f$	$\Delta NO_2 / \Delta O_3$	$[N_2O_5]$	Yield %	$[HNO_3]$	Rate Constant ($ppm^{-1} s^{-1}$)
4	2.66	0.087	15.1	10.0	1.90	0.60	23		
5 (b)	0.0	0.0	15.9	14.2	?	0.37	41		
8	6.45	2.15	12.0	3.70	1.93	2.79	65	0.30	$1.276, 1.323 \times 10^{-3}$
9	7.20	2.40	12.0	2.90	1.90	2.39	50		1.60, 1.69
10	7.10	2.15	12.9 ^(c)	3.94	1.82	2.67	54		1.30, 1.43
6	9.50	6.30	5.63	0.0	1.76	1.42	44		
7	9.40	5.45	7.39	0.0	1.87	1.96	50		
Mean \pm SD					1.86 ± 0.06				$(1.44 \pm 0.17) \times 10^{-3}$

(a) Concentrations in ppm units ($= 2.4 \times 10^{13}$ molecules/cm³).

(b) For this test 10 ppm NO was also present as an initial reactant and consumed 9.4 ppm O_3 introduced as an initial reactant. This left 0.6 ppm NO by calculation vs a measured quantity of 0.62 ppm NO. The source of 0.37 ppm N_2O_5 is not clearly understood.

(c) The average for previous workers is $(1.32 \pm 0.43) \times 10^{-3} ppm^{-1} s^{-1}$ at $T = 298 K$.

et al. (1973) and Graham (1975) shown in Figure 3-3. For a flow tube reactor operated with homogeneous mixing the rate constant for the reaction can be obtained from the steady-state concentrations of the reactions. For the rate limiting oxidation of NO_2 by O_3 the expression for the rate is given by:

$$k_{11} = \frac{-\frac{Q}{V}(-\Delta \text{NO}_2)}{2 [\text{NO}_2]_f [\text{O}_3]_f} = \frac{-\frac{Q}{V}(-\Delta \text{O}_3)}{[\text{NO}_2]_f [\text{O}_3]_f} \quad (3-17)$$

where the factor of 2 denotes the ideal stoichiometry, Q the flow rate, and f the final or steady-state concentration. Where determinant, the two rates are listed in Table 3-9 in the last columns. The value of $(1.44 \pm 0.17) \times 10^{-3} \text{ ppm}^{-1} \text{ s}^{-1}$ is in excellent agreement with the work of others. The measured N_2O_5 yield compared to the theoretical yield based upon the stoichiometry and consumption of reactants has improved from about 23 percent to 65 percent. N_2O_5 is easily hydrolyzed by water forming nitric acid as a product. As it is usually a heterogeneous wall reaction, reconciliation of the odd-nitrogen budget is impossible. Circumvention of this problem to date has used in-situ IR absorption.

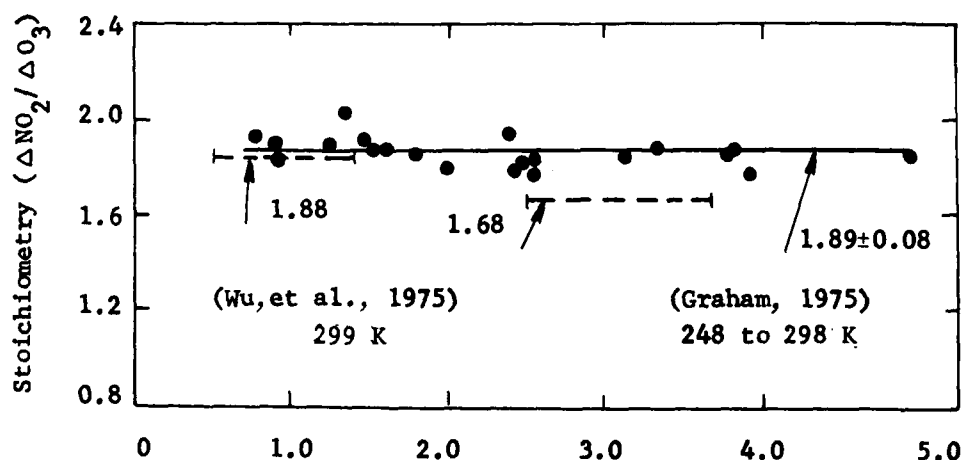
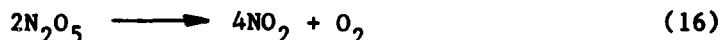


Figure 3-3. Plot of Measured Stoichiometries versus Initial Reactant Ratio For $\text{NO}_2\text{-O}_3$ Experiments

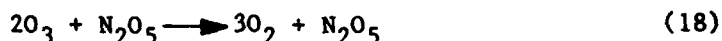
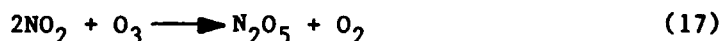
3.5 NITROUS PENTOXIDE (N_2O_5) THERMAL CONVERSION

3.5.1 Theory

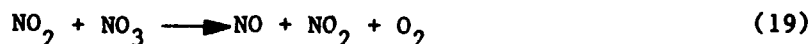
The thermal conversion of N_2O_5 is a complex reaction with a net stoichiometric reaction:



When NO , NO_2 , and O_3 are present N_2O_5 can either be converted, generated, or serve as a catalyst according to the following net stoichiometric reactions:



Reaction 16 will be treated first. The governing mechanistic reactions for reaction 16 are the decomposition and recombination of N_2O_5 followed by two simultaneous bimolecular reaction paths, one of which has a fast sequential reaction converting NO , if present. These reactions are (H.S. Johnston, 1951):



Rate limiting is governed by reactions (19) and (20).

The pyrolytic kinetics of these reactions have been studied by Schott and Davidson (1958) for temperatures greater than 450 K with argon as a diluent. The dissociation reaction of N_2O_5 is unimolecular, close to its second-order low pressure limit. The differential rate equation is given by:

$$\frac{d [N_2O_5]}{dt} = k_{-13} [M] [N_2O_5] \quad (3-18)$$

with $k_{-13} = 5.0 \times 10^{13} \exp(-16,500 + 700/RT) (\text{mol/l})^{-1}\text{s}^{-1}$. At a pressure of 1 atmosphere and a temperature of 600 K, the concentration of the diluent $[M]$ is $2.01 \times 10^{-2} \text{ mol/l}$ and $k_{-13} = 4.85 \times 10^7 (\text{mol/l})^{-1}\text{s}^{-1}$. At a reduced temperature of 400 K, $k_{-13} = 4.77 \times 10^4 (\text{mol/l})^{-1}\text{s}^{-1}$. Integration of the above expression leads to:

$$[\text{NO}_2] = [\text{N}_2\text{O}_5] \left\{ 1 - \exp \left[-k_{-13} [M] t \right] \right\} \quad (3-19)$$

or a half-life of

$$\begin{aligned} \text{Half-Life} &= \frac{\ln 2}{k_{-13} [M]} = 0.70 \text{ s for 600 K and} \\ &= 0.48 \text{ ms for 400 K} \end{aligned} \quad (3-20)$$

For the conditions of the stratosphere, the half-life is given below in Table 3-10.

TABLE 3-10. PREDICTED HALF-LIFE OF N_2O_5 IN A 400 K THERMAL CONVERTER

H (km)	P (atm)	$[M]$ (mol/l)	$\frac{\ln 2}{k_{-13} [M]}$ (s)
10	0.262	8.0×10^{-3}	1.8×10^{-3}
15	0.119	3.6	4.0
20	0.054	1.6	9.0
25	0.025	7.6×10^{-4}	1.9×10^{-2}
30	0.012	3.7	3.9
35	0.0057	1.7	8.5
40	0.0027	8.2×10^{-5}	1.8×10^{-1}

If the O_2 and N_2 molecules of the stratosphere take on the same role as the argon, then nearly complete conversion of N_2O_5 will occur if the residence time in the converter is about one second. A constant volume pump with a flow rate of 2.0-2.5 l/s requires a volume of 2.0 to 2.5 liters.

As cleanliness of the cell is a requirement to render heterogeneous wall reactions to negligible level, the converter volume-to-surface (V/S) ratio becomes important. Assuming a 2.5 cm diameter converter, the length required would be about 2 meters to obtain the necessary volume. If, however, the converter is designed to the 35 km specification limit; the residence time is set to four half-lives (94 percent conversion); and the diameter is increased to 5 cm, then the length becomes 35 cm for a 2 liter/s flow rate. The respective V/S ratios for the above converters are 0.61 and 1.23 cm, respectively.

Returning to the simultaneous bimolecular reactions with $k_7 \gg k_{19}$, the differential rate equation for the intermediary, NO_3 is given by

$$-\frac{d[\text{NO}_3]}{dt} = 2k_{19} [\text{NO}_2] [\text{NO}_3] + 2k_{20} [\text{NO}_3]^2 \quad (3-21)$$

For a given reaction chamber, assuming steady-state conditions, a rate relation is written as:

$$R_{\text{NO}_3} = \frac{Q}{V} ([\text{NO}_3]_i - [\text{NO}_3]_s) \quad (3-22)$$

where R_{NO_3} is the rate of disappearance of NO_3 , Q is the flow rate, V is the reaction volume, and the subscripts i and s denote initial and steady-state conditions. The ratio Q/V has the units of s^{-1} and is inversely related to residence time.

Combining the two rate equations and dividing by $2 [\text{NO}_2] [\text{NO}_3]$ yields,

$$k_{19} + k_{20} \frac{[\text{NO}_3]}{[\text{NO}_2]} = \frac{1}{\tau} \frac{[\text{NO}_3]_i - [\text{NO}_3]_s}{2 \frac{[\text{NO}_3]}{[\text{NO}_2]}} \quad (3-23)$$

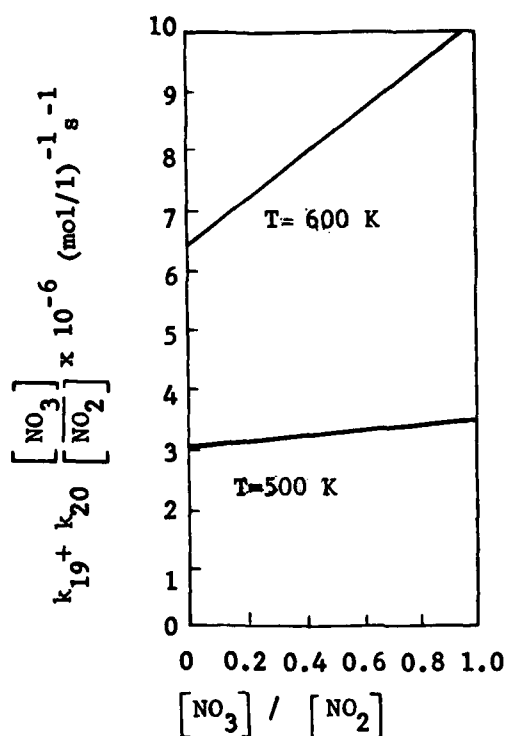
where

$$\log k_{19} = - (965 \pm 150)/T + 8.41 \quad (3-24)$$

and

$$\log k_{19} + k_{20} = - (1403 \pm 60)/T + 9.34 \quad (3-25)$$

The left-hand side of the rate equation is shown graphically in Figure 3-4 as a function of $[\text{NO}_3]/[\text{NO}_2]$ for $T = 500 \text{ K}$ and 600 K . The intercept at



T	k_{19}	$k_{19} + k_{20}$
500	3.02×10^6	3.42×10^6
600	6.33×10^6	10.04×10^6

Figure 3-4. Simultaneous Rate Constants at $T = 500$ and 600 K

a concentration ratio of zero corresponds to k_{19} and the intercept at unity ratio corresponds to $k_{19} + k_{20}$ or the apparent bimolecular rate constant at the beginning of the reaction. The slope of the function is k_{20} .

The rate equation can be rearranged to plot the concentration ratio versus time for given temperature and initial concentration conditions. This ratio is shown in Figure 3-5 for initial conditions of 10 ppm and a total pressure of 1 atmosphere.

For conditions more appropriate to the stratosphere, the terms of the rate equation are presented in Table 3-11 for initial conditions of 20 ppb N_2O_5 , an ambient pressure of 0.1 atm and a 400 K thermal converter.

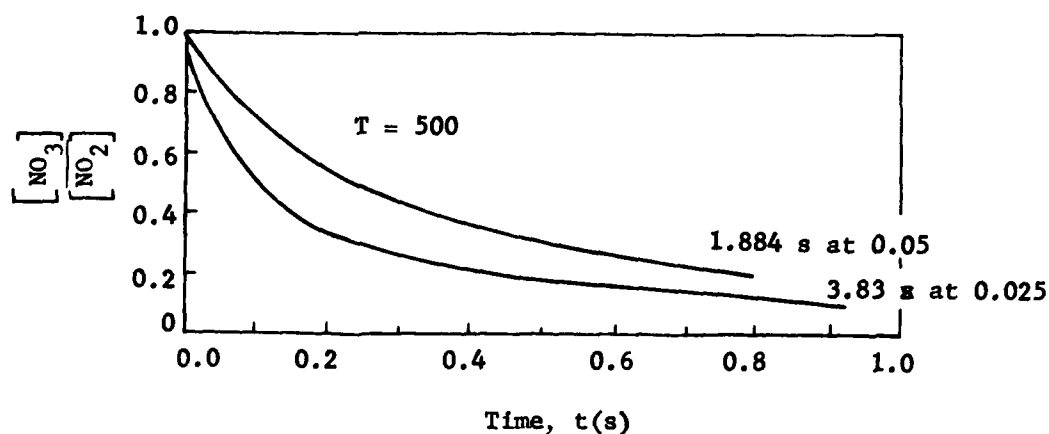


Figure 3-5. $\frac{[\text{NO}_3]}{[\text{NO}_2]}$ Decay versus Time

TABLE 3-11. NO_3 RATE LIMITING DECAY

$\frac{[\text{NO}_3]}{[\text{NO}_2]}$ (ppb)	$\frac{[\text{NO}_2]}{[\text{NO}_3]}$ (ppb)	$\Delta[\text{NO}_3]$ (ppb)	$\frac{[\text{NO}_3]}{[\text{NO}_2]}$	t (s)
10.0	10.0	0	1.0	0
9.473	10.526	0.526	0.9	8.27×10^2
8.889	11.111	1.111	0.8	1.78×10^3
8.235	11.765	1.765	0.7	2.94×10^3
7.500	12.500	2.500	0.6	4.38×10^3
6.667	13.333	3.333	0.5	6.25×10^3

For Figure 3-5 or Table 3-11 it can be clearly seen that the ratio $\frac{[\text{NO}_3]}{[\text{NO}_2]}$ will be approximately unity for residence times of a few seconds or less and concentrations of a few ppb or less. The thermally converted N_2O_5 sample must then be analyzed as NO_2 or NO_3 , or a second conversion of either constituent must occur followed by detection of the end products. The stoichiometry of these conversions, however, must be known.

For reaction 16 the growth of NO_2 is given by:

$$\frac{d[\text{NO}_2]}{dt} = k_{-13} [\text{N}_2\text{O}_5] - k_{13} [\text{NO}_2] [\text{NO}_3] + 2k_{13} [\text{NO}] [\text{NO}_3] \quad (3-26)$$

and the growth of the NO_3 radical is

$$\frac{d[\text{NO}_3]}{dt} = k_{-13} [\text{N}_2\text{O}_5] - (k_{13} + k_{19}) [\text{NO}_2] [\text{NO}_3] - k_{13} [\text{NO}] [\text{NO}_3] \quad (3-27)$$

Since NO will generally be present, the net stoichiometric reaction must also be considered:



For this reaction and for lower temperatures, a steady-state approximation for the $[\text{NO}_3]$ intermediary is made, i.e., $\frac{d[\text{NO}_3]}{dt} = 0$. With this approximation the growth of NO_2 is given by:

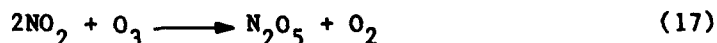
$$\frac{d[\text{NO}_2]}{dt} = k_{-13} [\text{N}_2\text{O}_5] \frac{k_{19} [\text{NO}_2] + 3k_7 [\text{NO}]}{(k_{13} + k_{19}) [\text{NO}_2] + k_7 [\text{NO}]} \quad (3-28)$$

with the knowledge that $k_7 \gg k_{19}$ (pg. 38), this expression can be simplified to;

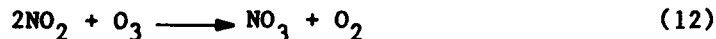
$$\frac{d[\text{NO}_2]}{dt} = 3 k_{-13} [\text{N}_2\text{O}_5] \frac{[\text{NO}]}{[\text{NO}] + k_{13}/k_7 [\text{NO}_2]} \quad (3-29)$$

Since $[\text{NO}] \approx [\text{NO}_2]$ in the stratosphere and k_{13} at 400 K is $\ll k_7$, $k_7 [\text{NO}]$ will be much greater than $k_{13} [\text{NO}_2]$. For these conditions, a first-order rate for reaction 14 will simply be that of the elementary unimolecular rate given for reaction -13.

The third reaction listed in this section is restated;



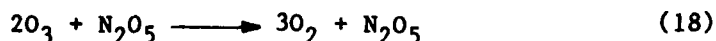
and has been observed (Section 3.4.1) to be second-order in reactants. The mechanism is simply



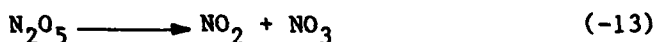
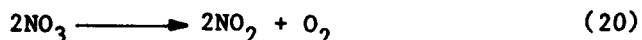
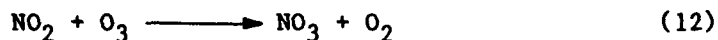
Applying the steady-state assumption for NO_3 , the loss rate for O_3 is simply

$$-\frac{d[\text{O}_3]}{dt} = k_1[\text{NO}_2][\text{NO}_3] \quad (3-30)$$

The fourth reaction listed in this section is restated:



This reaction is the well-known N_2O_5 catalytic decomposition of O_3 . The governing mechanistic reactions for reaction (18) are:



If the pseudo-reference species, NO_2 and NO_3 , are equilibrated with N_2O_5 , they are treated kinetically as though they were intermediaries in a consecutive process. This leads to the following differential rate equation:

$$-\frac{d[\text{O}_3]}{dt} = 2 \left(\frac{k_{-13}^2 k_{11}^2 k_{19}}{4k_{13}} \right)^{1/3} [\text{N}_2\text{O}_5]^{2/3} [\text{O}_3]^{2/3} \quad (3-31)$$

3.5.2 Stratospheric-Based Instrument Modeling

The above reactions plus the ozone oxidation of NO to NO_2 and NO_2 to NO_3 have been modeled using the EPISODE adaptation of the GEARS code (Gear, 1971 and Hindmarsh, 1975). These codes are chemical kinetics programs that solve coupled stiff differential equations. The solutions are obtained by implicit linear multisteps. With the GEAR code, fixed-step formulae are employed with changes in step, when required, by interpolation. In contrast, the EPISODE code is based upon formulae that are of variable step size. This feature

lends stability to the solution. Given the correct set of chemical reactions, the solutions tend to be an exact representation of the products and reactants for homogeneous reactions.

Table 3-12 provides printouts for a 400 K converter with input concentrations set for $H = 15$ km and $H = 25$ km. The initial concentrations can be read from the respective first lines at $t = 0$. Within the first half-second most of the N_2O_5 is converted to NO_2 and NO_3 . Subsequently $[NO_2]$ increases to a steady-state level of 9.6×10^9 molecules/cm³ and 7.2×10^9 molecules/cm³; $[NO_3]$ increases at a slower rate to a steady-state level of 3.5×10^8 molecules/cm³ and 1.1×10^9 molecules/cm³; and $[NO]$ decays. As an example of the $[NO_x]$ budgeting, refer to the case for $H = 25$ km and $t = 5.0$ s. These values are given in Table 3-13.

Although the data of Table 3-13 shows acceptable accounting or budgeting of $\Delta[NO_x] = \Delta[NO_2] + \Delta[NO]$ from redox reactions the data does not permit an accurate measurement of $[N_2O_5]$. The principal reason is that some of the unmeasured $[NO_3]$ is converted to NO_2 via reaction -13. This in effect alters specificity. If the measurement could be carried out in 2 seconds, the $\Delta[NO_3]$ error would be reduced by about 33 percent for the case cited.

From the example cited in Table 3-13, setting aside for the moment the above mentioned problem, a quantitative measure of $[N_2O_5]$ may be obtained, if the following conditions are met: (1) N_2O_5 is not lost to the instrumentation walls, (2) $NO_x = NO + NO_2$ was determined accurately, (3) the $[NO_3]$ from both the ambient stream and as an N_2O_5 decomposition product is quantitatively converted to NO_x prior to being measured as NO , and (4) the $NO-NO_2$ instrumentation did not convert any of the ambient N_2O_5 during sample residence. This latter statement is a specificity condition. The third condition is treated next.

Of the intermediary, NO_3 entering an NO_2 photolytic cell, some may be converted according to the spectral energy available as given by Johnston and Graham (1974):

TABLE 3-12. HAPP RESIDENCE TIME STUDY

T = 400 K

HAPP RESIDENCE TIME STUDY H = 15 km

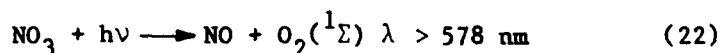
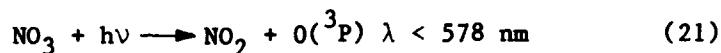
TIME(S)	H2O5	NO2	NO3	NO	O3	O2	HN03	HO	H2O
0.0	1.0000 07	5.0000 09	2.0000 06	5.0000 09	3.0000 12	8.1000 17	1.3000 09	1.0000 06	6.0000 12
0.50	1.1860 02	5.4120 09	1.3490 07	4.5970 09	3.0000 12	8.1000 17	1.3000 09	8.3900 05	6.0000 12
1.00	1.4180 02	5.7810 09	1.5090 07	4.2260 09	2.9990 12	8.1000 17	1.3000 09	7.0460 05	6.0000 12
1.50	1.6740 02	6.1200 09	1.6020 07	3.8860 09	2.9990 12	8.1000 17	1.3000 09	5.9410 05	6.0000 12
2.00	1.9520 02	6.4310 09	1.8670 07	3.5720 09	2.9990 12	8.1000 17	1.3000 09	5.0330 05	6.0000 12
2.50	2.2530 02	6.7170 09	2.0610 07	3.2840 09	2.9980 12	8.1000 17	1.3000 09	4.2860 05	6.0000 12
3.00	2.5740 02	6.9800 09	2.2660 07	3.0140 09	2.9980 12	8.1000 17	1.3000 09	3.6700 05	6.0000 12
3.50	2.9150 02	7.2210 09	2.4800 07	2.7760 09	2.9980 12	8.1000 17	1.3000 09	3.1630 05	6.0000 12
4.00	3.2750 02	7.4430 09	2.7030 07	2.5520 09	2.9980 12	8.1000 17	1.3000 09	2.7450 05	6.0000 12
4.50	3.6540 02	7.6470 09	2.9350 07	2.3460 09	2.9970 12	8.1000 17	1.3000 09	2.4000 05	6.0000 12
5.00	4.0490 02	7.8340 09	3.1740 07	2.1570 09	2.9970 12	8.1000 17	1.3000 09	2.1150 05	6.0000 12
5.50	4.4600 02	8.0050 09	3.4200 07	1.9820 09	2.9970 12	8.1000 17	1.3000 09	1.8790 05	6.0000 12
6.00	4.8860 02	8.1630 09	3.6740 07	1.8220 09	2.9970 12	8.1000 17	1.3000 09	1.6840 05	6.0000 12
6.50	5.3260 02	8.3070 09	3.9350 07	1.6750 09	2.9970 12	8.1000 17	1.3000 09	1.5220 05	6.0000 12
7.00	5.7800 02	8.4400 09	4.2020 07	1.5400 09	2.9970 12	8.1000 17	1.3000 09	1.3870 05	6.0000 12
7.50	6.2450 02	8.5620 09	4.4750 07	1.4150 09	2.9960 12	8.1000 17	1.3000 09	1.2760 05	6.0000 12
8.00	6.7220 02	8.6730 09	4.7540 07	1.3010 09	2.9960 12	8.1000 17	1.3000 09	1.1830 05	6.0000 12
8.50	7.2090 02	8.7760 09	5.0390 07	1.1960 09	2.9960 12	8.1000 17	1.3000 09	1.1050 05	6.0000 12
9.00	7.7060 02	8.8690 09	5.3290 07	1.0990 09	2.9960 12	8.1000 17	1.3000 09	1.0400 05	6.0000 12
9.50	8.2130 02	8.9550 09	5.6240 07	1.0100 09	2.9960 12	8.1000 17	1.3000 09	9.8540 04	6.0000 12
10.00	8.7280 02	9.0340 09	5.9240 07	9.2860 08	2.9960 12	8.1000 17	1.3000 09	9.3960 04	6.0000 12

HAPP RESIDENCE TIME STUDY H = 25 km

TIME(S)	H2O5	NO2	NO3	NO	O3	O2	HN03	HO	H2O
0.0	7.0000 08	6.2000 09	2.0000 06	7.0000 08	4.3000 12	1.7000 17	3.0000 09	1.0000 06	6.0000 12
0.50	9.1280 05	6.9820 09	7.0120 08	6.1730 08	4.3000 12	1.7000 17	3.0000 09	8.3840 05	6.0000 12
1.00	9.2520 03	7.0560 09	7.0220 08	5.4380 08	4.3000 12	1.7000 17	3.0000 09	6.5560 05	6.0000 12
1.50	8.1640 03	7.1200 09	7.0270 08	4.7910 08	4.3000 12	1.7000 17	3.0000 09	5.1650 05	6.0000 12
2.00	8.2390 03	7.1760 09	7.0370 08	4.2210 08	4.3000 12	1.7000 17	3.0000 09	4.1060 05	6.0000 12
2.50	8.3120 03	7.2250 09	7.0500 08	3.7190 08	4.3000 12	1.7000 17	3.0000 09	3.3000 05	6.0000 12
3.00	8.3810 03	7.2670 09	7.0670 08	3.2770 08	4.3000 12	1.7000 17	3.0000 09	2.6860 05	6.0000 12
3.50	8.4480 03	7.3040 09	7.0870 08	2.8880 08	4.3000 12	1.7000 17	3.0000 09	2.2190 05	6.0000 12
4.00	8.5130 03	7.3360 09	7.1100 08	2.5450 08	4.3000 12	1.7000 17	3.0000 09	1.8620 05	6.0000 12
4.50	8.5750 03	7.3640 09	7.1350 08	2.2430 08	4.2990 12	1.7000 17	3.0000 09	1.5910 05	6.0000 12
5.00	8.6360 03	7.3880 09	7.1620 08	1.9770 08	4.2990 12	1.7000 17	3.0000 09	1.3840 05	6.0000 12
5.50	8.6940 03	7.4090 09	7.1900 08	1.7430 08	4.2990 12	1.7000 17	3.0000 09	1.2260 05	6.0000 12
6.00	8.7520 03	7.4260 09	7.2200 08	1.5360 08	4.2990 12	1.7000 17	3.0000 09	1.1060 05	6.0000 12
6.50	8.8080 03	7.4410 09	7.2520 08	1.3540 08	4.2990 12	1.7000 17	3.0000 09	1.0140 05	6.0000 12
7.00	8.8630 03	7.4540 09	7.2840 08	1.1940 08	4.2990 12	1.7000 17	3.0000 09	9.4410 04	6.0000 12
7.50	8.9170 03	7.4650 09	7.3180 08	1.0530 08	4.2990 12	1.7000 17	3.0000 09	8.9050 04	6.0000 12
8.00	8.9710 03	7.4740 09	7.3530 08	9.2900 07	4.2990 12	1.7000 17	3.0000 09	8.4960 04	6.0000 12
8.50	9.0230 03	7.4810 09	7.3890 08	8.1970 07	4.2990 12	1.7000 17	3.0000 09	8.1830 04	6.0000 12
9.00	9.0750 03	7.4870 09	7.4250 08	7.2330 07	4.2990 12	1.7000 17	3.0000 09	7.9440 04	6.0000 12
9.50	9.1260 03	7.4920 09	7.4620 08	6.3850 07	4.2990 12	1.7000 17	3.0000 09	7.7610 04	6.0000 12
10.00	9.1760 03	7.4960 09	7.4990 08	5.6380 07	4.2990 12	1.7000 17	3.0000 09	7.6200 04	6.0000 12

TABLE 3-13. NO_x BUDGET
(for Table 3-12, H = 25 km, T = 400 K, t = 5.0 s)

Species	Decomposition	Species	Combination
ΔN_2O_5	-7.000×10^8	ΔNO_x	$+ 6.86 \times 10^8$
ΔNO_2	7.000×10^8		
ΔNO_3	7.142×10^8		
$\Delta N_2O_5 - \Delta NO_3$	-0.142×10^8		
Total ΔNO_x	6.858×10^8	Total NO _x	$+ 6.86 \times 10^8$



There is conjecture, however, that NO + O₂(¹Σ) products also may be formed for wavelengths less than 578 nm.

Photoabsorption cross-sections to $\lambda \geq 470$ nm and quantum yields in the visible have been determined for reactions 21 and 22 (Graham, 1975). The quantum yields, wavelength averaged cross-sections, and light distributions are given below in Table 3-14. From the fourth column of data it is apparent that the conjecture of the previous paragraph is valid. For a solar spectral distribution and zero zenith angle, Graham's computed rates are $J_{21} = 0.099 \pm 0.02 \text{ s}^{-1}$ and $J_{22} = 0.040 \pm 0.02 \text{ s}^{-1}$.

The products of reaction 22 will rapidly react as follows:



where

$$k_6 = 3.5 \times 10^{-12} \text{ cm}^3\text{-molecule}^{-1}\text{s}^{-1} \text{ (Kaufman, 1961)}$$

Since the NO₂ photolytic converter of paragraph 3.2 does not have any substantial flux at $\lambda > 470$ nm reaction 21 is not anticipated. NO₃ absorption cross-sections for $\lambda < 400$ nm, however, have not been measured.

TABLE 3-14. NO₃ FREE RADICAL PHOTOLYTIC TERMS

Parameter	Reaction					
	J ₂₁	J ₂₁	J ₂₁	J ₂₁	J ₂₂	J ₂₂
Quantum Yield	0.14	0.23	0.049	0.85	0.63	
Avg σ NO ₃ : $\lambda \leq 580$ nm (cm ² /molec)	1.88x10 ⁻¹⁸	2.51x10 ⁻¹⁸		1.88x10 ⁻¹⁸	2.51x10 ⁻¹⁸	
Avg σ NO ₃ $\lambda > 580$	2.99	3.11	2.17x10 ⁻¹⁸	2.99	3.11	
Percent light, $\lambda \leq 580$ nm	93	37	0	93	37	
Percent light, $\lambda > 580$	7	63	100	7	63	
Integrated light intensity (10 ¹⁶ h ν /cm ² -s)	1.55	1.07	0.33	1.55	1.07	

Although an N_2O_5 measurement does not appear feasible due to specificity problems, some of which are deferred to paragraph 3.8, Potential Interferents, the thermal conversion of N_2O_5 was experimentally treated in the lab and is discussed in the next paragraph.

3.5.3 Experimental Procedures and Results

The thermal decomposition of nitrous pentoxide was studied using a stainless steel converter between the N_2O_5 generating apparatus and the long path absorption cell of the Model 580 spectrophotometer. The spectrophotometer enabled observation of the major constituents involved in the thermal conversion process. A schematic of the apparatus is shown in Figure 3-6.

N_2O_5 was created by the gas phase reaction of NO_2 and O_3 in the presence of excess NO_2 in the same manner as described in paragraph 3.5.1. Several minor changes, however, were made to further reduce water contamination. Two test runs are discussed in detail.

The core of the converter consisted of a $2\frac{1}{2}$ -inch diameter x 2-inch long stainless steel block. The gas passed through a $\frac{1}{2}$ -inch diameter hole in the center of the block. A sintered stainless disk pressed into the gas channel permitted complete thermalization of the gas. The temperature of the block was monitored using platinum resistance sensors. A cartridge heater inserted into the block was the heat source. The block temperature was not regulated but stayed within ± 1 K during the course of each data run.

For the initial test run, pertinent conversion data of the reactants are presented in Table 3-15. The initial concentrations of $[\text{NO}_2]$ and $[\text{O}_3]$ prior to mixing were 23.5×10^{13} and 6.5×10^{13} molecules/ cm^3 , respectively. The final concentrations of $[\text{NO}_2]$ and $[\text{O}_3]$ were 11.3×10^{13} and zero molecules/ cm^3 . The ratio $\Delta[\text{NO}_2]/\Delta[\text{O}_3]$ is therefore 1.88, as expected. Although the IR scan of Figure 3-7 for $T = 23^\circ\text{C}$ shows little evidence of H_2O being present, the measured yield of N_2O_5 for this test, assuming a stoichiometry of 1.88, is only 43 percent. The remaining 57 percent appears as nitric acid which is evident at 1340 and 1315 cm^{-1} . The expected amount of HNO_3 based upon the 57 percent loss of N_2O_5 is 7.44×10^{13} molecules/ cm^3 . From Table 3-15, it can be seen about 71 percent of the HNO_3 appeared in the gas phase.

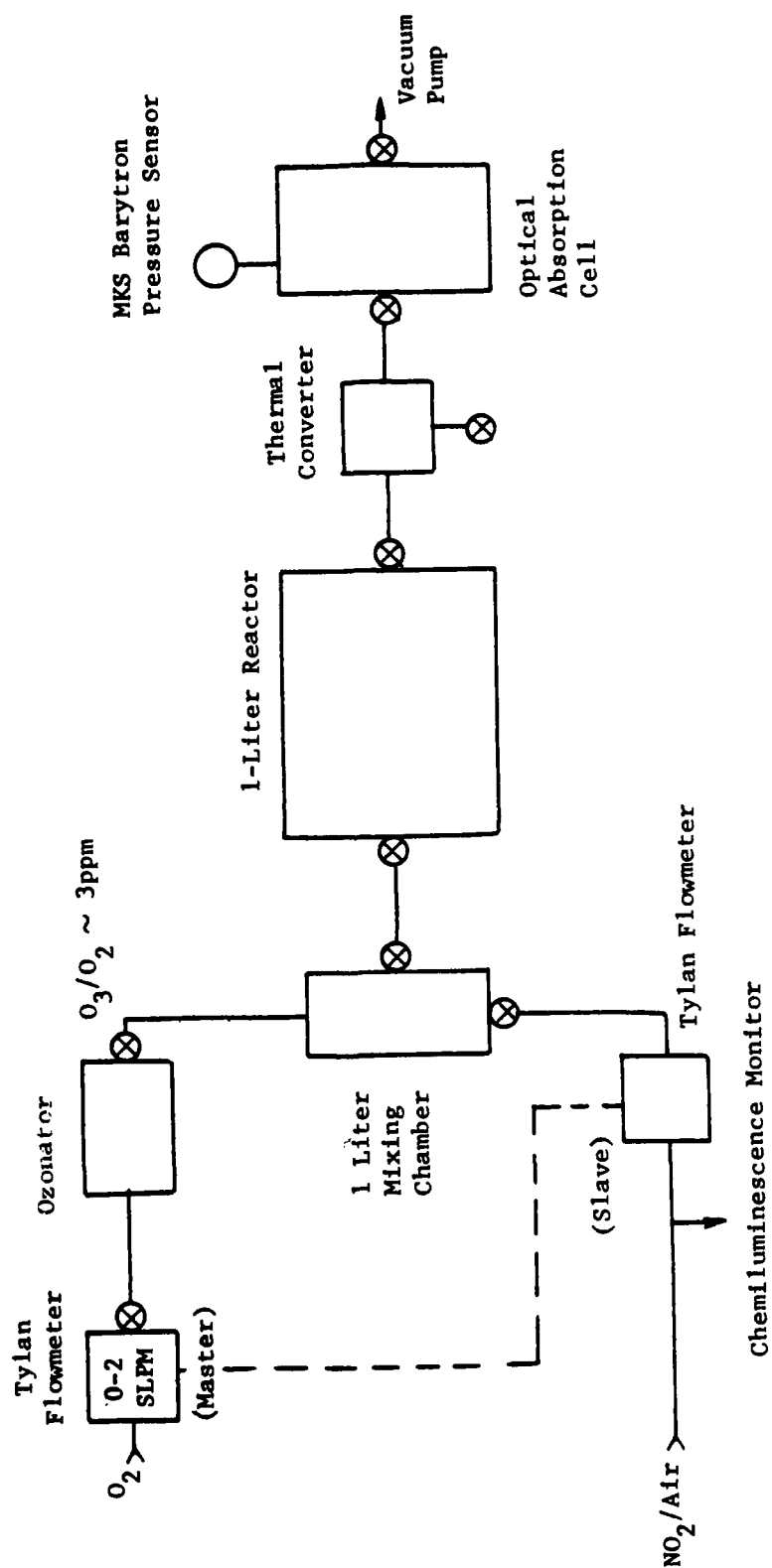


Figure 3-6. Setup for Experiment To Observe Thermal Decomposition of N_2O_5

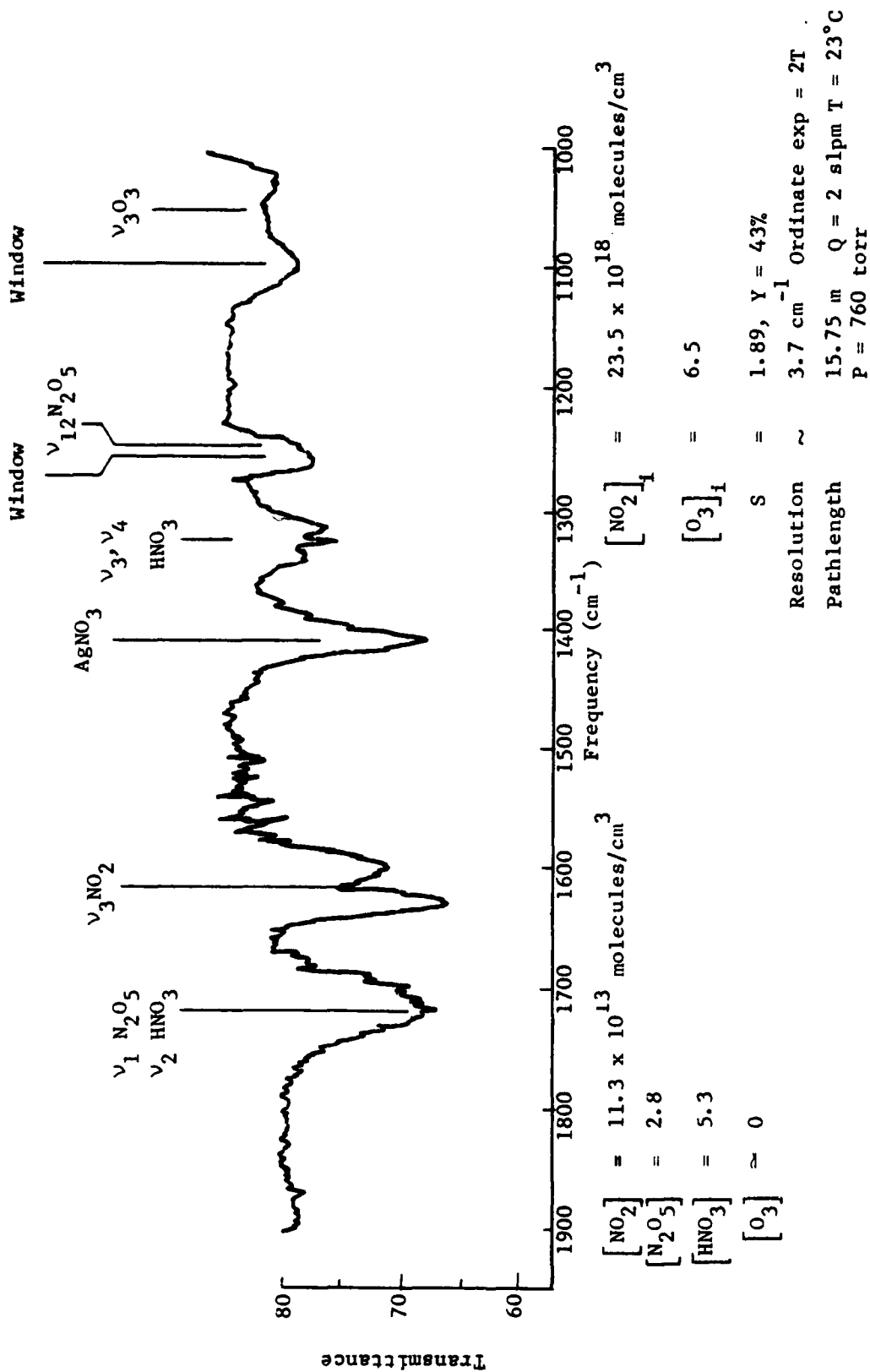


Figure 3-7. N_2O_5 Generation Spectra

TABLE 3-15. NO_2 , N_2O_5 AND HNO_3 MEASURED CONCENTRATIONS
VS THERMAL CONVERTER TEMPERATURE

Temp. (K)	NO_2	N_2O_5 $\times 10^{13}$ (molecules/cm ³)	HNO_3
298	11.3	2.78	5.32
373	13.1	1.63	7.67
473	18.8	0	5.62
523	17.8		5.96
573	21.7		3.87
623	21.9		1.84
673	25.3		0.71

For the initial testing of the N_2O_5 converter it was decided that a series of thermal tests at differing temperatures should be carried out. From the above table and a sample residence time of about 0.5 second, it can be seen that all of the N_2O_5 accounted for had decomposed at a temperature of 473 K. At this temperature, however, the data indicates that $\Delta[\text{NO}_2]$ is considerably greater than expected from a $\Delta[\text{N}_2\text{O}_5]$ of -2.78×10^{13} molecules/cm³.

In addition, a review of the HNO_3 concentration versus temperature indicates that possibly HNO_3 is being desorbed from the converter walls at a temperature of 373 K. As the converter had been used on the two previous days, it is feasible that residual HNO_3 was present.

Finally, the high temperature NO_2 and HNO_3 data are of interest for the conversion of HNO_3 which is described in detail in paragraph 3.6. At a temperature of 673 K the NO_2 concentration was determined to be 25.3×10^{13} molecules/cm³ as compared to the initial concentration of 23.5×10^{13} molecules/cm³ prior to reaction with O_3 or about 1.8×10^{13} molecules/cm³ too high. Undoubtedly this discrepancy contains experimental measurement errors but could also have arisen from the apparent desorbed HNO_3 which from the first two HNO_3 entries is seen to be about 2.3×10^{13} molecules/cm³. Considering

the 523-673 K temperature range and making the $[\text{NO}_2] = 1.8 \times 10^{13}$ molecules/cm³ correction, one has a $\Delta[\text{NO}_2]$ of 5.7×10^{13} molecules/cm³ as compared to a $\Delta[\text{HNO}_3]$ of 5.25×10^{13} molecules/cm³. Paragraph 3.6.1, to follow, discusses theoretically a stoichiometry of unity for the decomposition of HNO_3 to NO_2 .

For the second test run emphasis was placed on repeatability and closer monitoring of thermal trends. The concentration of NO_2 prior to dilution was 58 ppm as determined by the chemiluminescence monitor. When diluted with the O_3/O_2 flow, the concentration of NO_2 was 15.7 ppm. The concentration determined by IR absorption was found to be 10 ppm. Ozone was generated by a standard UV lamp ozonator and grade 4.5 oxygen. The spectrum was analyzed using the data of Pitts (1976), c.f. Appendix A. The result was an ozone concentration of 3.3 ppm (NTP) or 3.2 ppm under laboratory conditions of 765 Torr and 26°C. The change in NO_2 concentration following reaction with the ozone was 8.0 ppm with all of the ozone being consumed. The ratio $[\text{NO}_2]/[\text{O}_3] = 2.5$ for these experiments.

The absorption spectrum of the reacted gases is shown in Figure 3-8. In addition to the NO_2 band at 1620 cm^{-1} , bands attributable to HNO_3 and N_2O_5 are clearly visible. The spectral run of Figure 3-8 extends the spectrum to lower frequencies to present the ν_{13} and ν_{14} bands of N_2O_5 . Using the absorption crosssections given in Appendix A for HNO_3 at 1315 cm^{-1} and 1340 cm^{-1} , the concentration of HNO_3 was found to be 3.2×10^{13} molecules/cm³ or 1.4 ppm. The maximum amount of N_2O_5 that could be generated would be equal to the amount of O_3 consumed which was 3.2 ppm. The nitric acid is believed to be formed by heterogeneous wall reactions of N_2O_5 with adsorbed water. The net reaction is



Thus, 1.4 ppm of HNO_3 represents the reaction of 0.7 ppm of N_2O_5 . Using the absorption cross sections given in Appendix A for N_2O_5 at 1246 cm^{-1} , the concentration of N_2O_5 was found to be 4.6×10^{13} molecules/cm³ or 2.0 ppm. This represents a yield of 59 percent with the loss of 0.7 ppm to heterogeneously and homogeneously formed HNO_3 . Thus the nominal N_2O_5 yield is $2.7/3.4 = 0.8$ which is quite good.

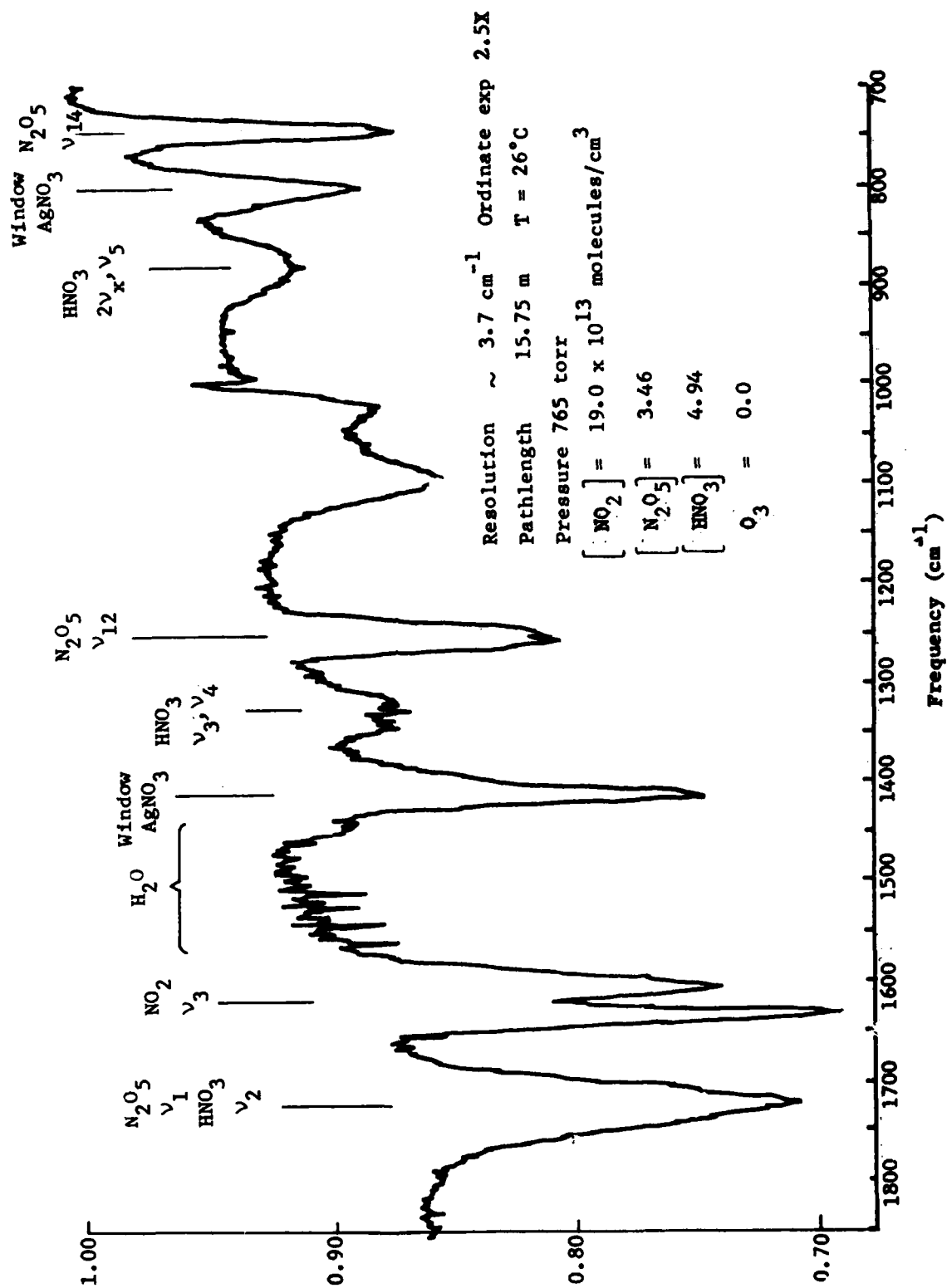


Figure 3-8. Absorption Spectra of N_2O_5 and HNO_3

The recorded data consisted of a least three complete spectral scans, similar to Figure 3-8, taken after the block had reached thermal equilibrium. The transmission at various distinct spectral features was recorded and compared to a reference spectrum (pure O_2). The absorbance was plotted as a function of temperature for these features of N_2O_5 , HNO_3 , and NO_2 (Figure 3-9). In addition, the N_2O_5 - HNO_3 mixed feature at 1720 cm^{-1} was plotted (Figure 3-10). Examining these graphs, it is seen that the amount of N_2O_5 declines steadily with increasing temperatures. The nitric acid concentration peaks near 475 K, while the yield of NO_2 increases steadily to within 1 percent of its unreacted level. The "missing" NO_2 has probably been converted to NO although this was not explicitly checked. The standard deviation of the data was typically 0.1 to 0.3 percent indicating stable conditions during the successive spectral measurements.

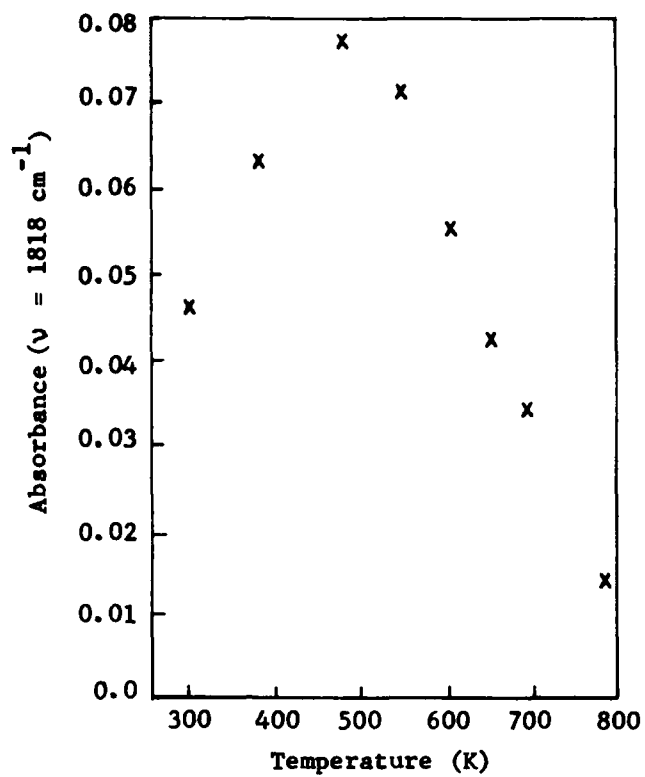
The complicated thermal behavior exhibited in these graphs is related to several factors. Primary among these is the time that the gas spends in the long path absorption cell (≈ 300 sec). This time is long enough to allow some reverse reactions to occur, especially in the oxygen-rich environment of these experiments. Also, the gas temperature in the absorption cell was not well determined. Nevertheless, almost total recovery of the NO_2 has been achieved, thus indicating > 99 percent conversion efficiency.

3.6 NITRIC ACID (HNO_3) GENERATION

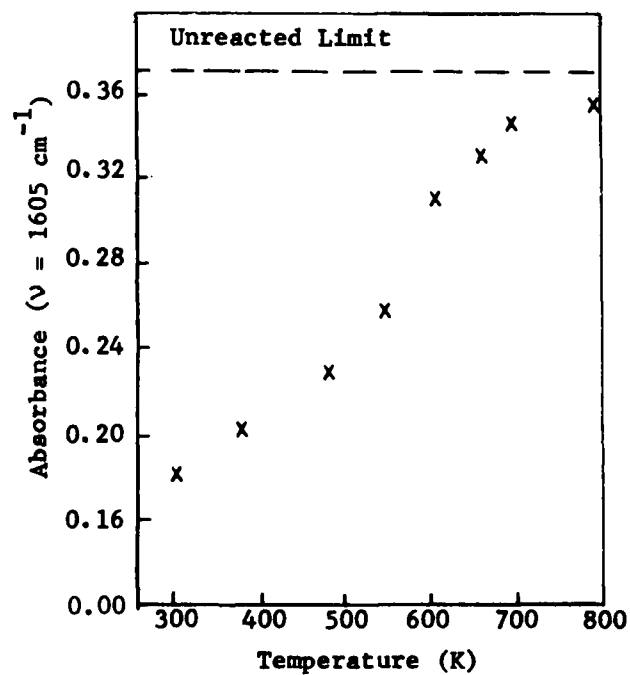
For nitric acid (HNO_3) synthesis, vapor diffusion of HNO_3 from an $HNO_3/H_2SO_4/H_2O$ mixture was employed. This technique approximates the approach used by Stedman (1977).

3.6.1 Theory

Generation of HNO_3 is best carried out by the low pressure distillation of HNO_3/H_2SO_4 (Wilson and Miles, 1940) combined with a diffusion flow technique (Stedman, 1977). A ratio of 100 to 33.3 parts of concentrated 70% HNO_3 /30% H_2O and H_2SO_4 was used to obtain weight percentages of $HNO_3 = 49$, $H_2SO_4 = 30$, and $H_2O = 21$. The vapor composition was greater than 98% HNO_3 . A stream of dry nitrogen or dry air is then blown by the mouth of the flask to create ~ 3 ppm concentrations of HNO_3 in the flowing gas stream.

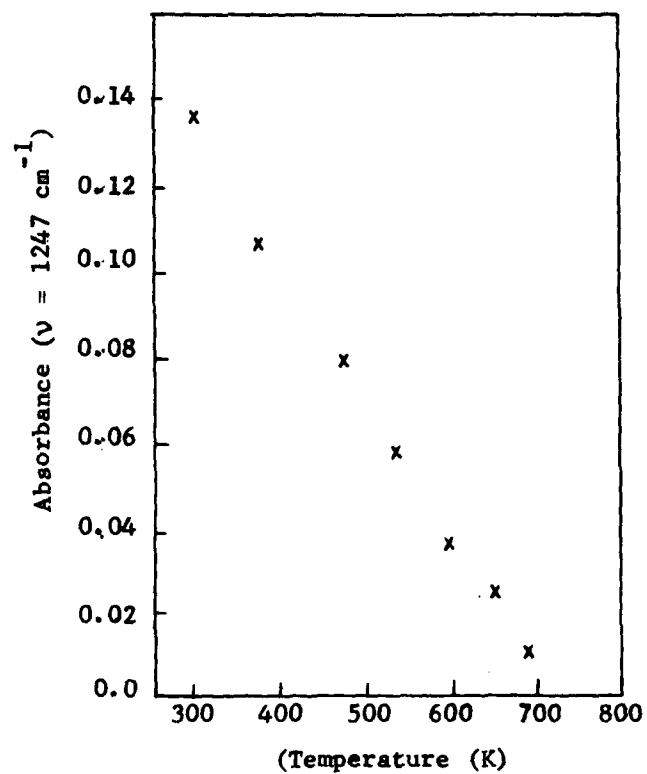


a. HNO₃ Absorbance

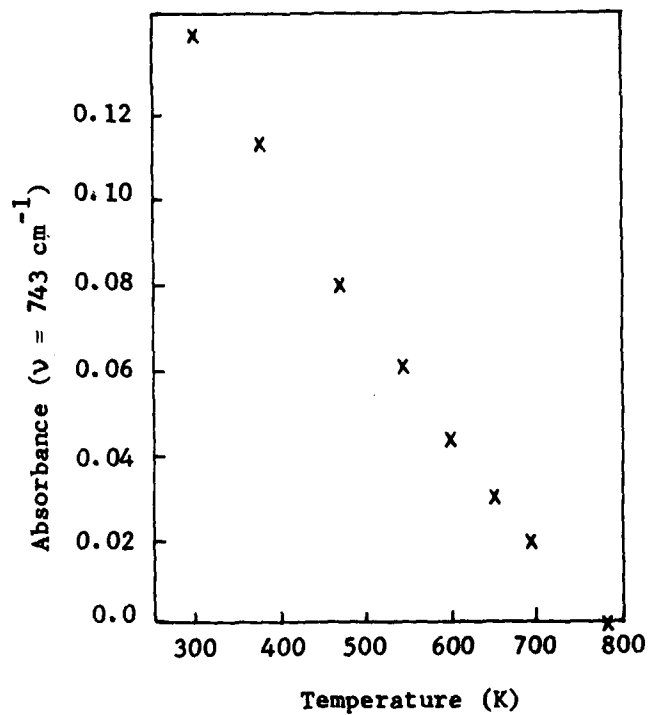


b. NO₂ Absorbance

Figure 3-9. Absorbances at Converter Output (1 of 2)



c. N_2O_5 Absorbance



d. N_2O_5 Absorbance

Figure 3-9. Absorbances at Converter Output (2 of 2)

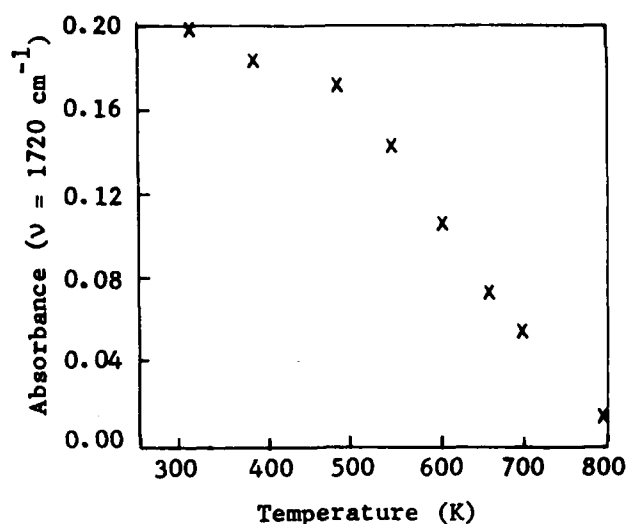


Figure 3-10. HNO_3 and N_2O_5 Absorbance at Converter Output

The vapor pressure of the HNO_3 was derived from boiling point data on Gibbs type ternary composition diagrams and plotted versus $1000/T$ shown in Figure 3-11. For diffusion flow samples of HNO_3 , the solution was held at 35°C yielding a partial pressure of 34 torr. For high spectral resolution measurements at temperatures simulating those of the stratosphere, a differential pumping technique was employed and the Pascal-Saposchnikow data was extrapolated according to the integrated Clausius-Clapeyron expression:

$$\log P = \frac{0.05223A}{T} + B(\text{torr}),$$

where $A = 45,568$ and $B = 9.259$. This expression permits one to derive from total pressure measurements the saturated partial pressure of HNO_3 as a function of temperature.

Under diffusion flow, the predicted concentration at the upper end of the diffusion tube is given by the expression

$$C = \frac{10^6 \times q_d}{Q} \text{ (ppm).}$$

where

$$q_d = \frac{DA}{L} \ln \frac{P}{P - P_{\text{HNO}_3}},$$

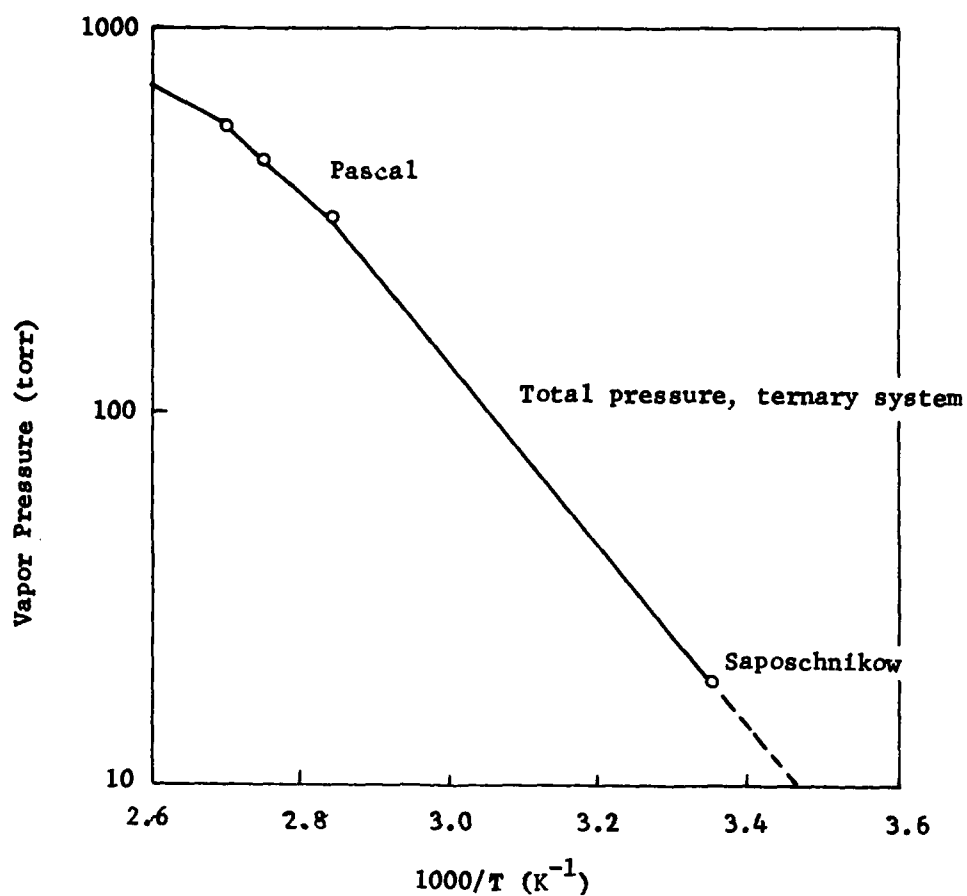


Figure 3-11. HNO_3 Vapor Pressure vs Reciprocal Temperature

and:

Q = flow rate (ml/s)

q_d = diffusion rate (ml/s)

D = diffusion coefficient $\approx 0.09 \left(\frac{306}{298} \right) \left(\frac{760}{747} \right) \text{ cm}^2/\text{sec}$

P = pressure in the diffusion cell

p = partial pressure of the diffusing vapor

A = diffusion tube cross-sectional area (cm^2)

L = Diffusion tube length (cm)

With 3/16-inch ID tubing, A/L was equal to 0.0124 cm. At a temperature of $T = 35^{\circ}\text{C}$ (308 K) and $P = 747$ torr, $q_d = 5.6 \times 10^{-5}$ ml/sec. At an initial flow rate of 0.25 SLPM, one has a computed concentration of 13.4 ppm. Maximum flow rate was set to 2.1 SLPM yielding a computed concentration of 1.6 ppm.

3.6.2 Experimental Procedures

Once diffusion was initiated, equilibrium was attained in about 24 hours and remained constant for weeks. Internal pressure disturbances above the solution cause only short-termed fluctuations in concentration.

Materials employed in the diffusion generator included a triple necked pyrex distillation flask fitted with $\frac{1}{2}$ 24/40 PTFE stoppers lightly coated with a Hooker Chemical Company fluorolube grease (GR-90), stainless steel fittings and PTFE tubing. Temperature was maintained by an agitated temperature controlled water bath.

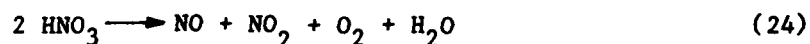
3.7 NITRIC ACID (HNO_3) THERMAL CONVERSION

Thermal conversion of HNO_3 as an analytical method for stratospheric monitoring is, as yet, untried. The method, however, has been laboratory tested by D. H. Stedman (private communication, 1977) for tropospheric monitoring and by the authors as discussed in this section. The converter is constructed of glass and contains glass beads for heat transfer. Temperature is maintained at about 550 K. The kinetics of the various reactions of this section relating to HNO_3 have been carefully reviewed. Decomposition of HNO_3 is often heterogeneous, requiring a surface and its formation rate does not lend itself to description by Arrhenius' equation. The function of the surface appears to be to act as an OH scavenger.

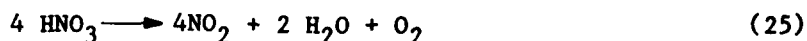
3.7.1 Theory

Conversion of HNO_3 has been experimentally treated by Frejacques (1951) and extensively by Johnston et al. (1951, 1953, 1955).

The postulated reaction of Frejacques is the following:



with N_2O_5 as an intermediary product. Subsequent measurements by colleagues of Johnston and the authors indicate that the thermal conversion follows the Johnston path. The thermal conversion is a complex reaction with the net stoichiometric reaction:



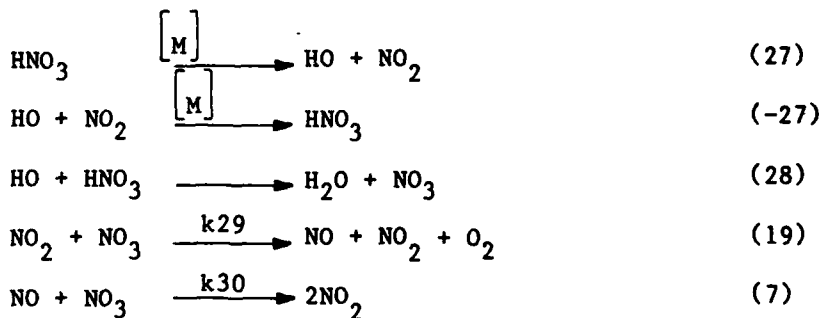
A consecutive side reaction is the thermal conversion of NO_2 to NO which was treated in Section 3.3.1.1. The ensuing analysis does not consider this side reaction.

In the presence of NO , the thermal conversion is also a complex reaction with the net stoichiometric reaction:



This reaction is treated after a discussion of reaction (25).

The governing mechanistic reactions for reaction (25) are the decomposition and recombination of HNO_3 followed by a product-reactant reaction followed by the two N_2O_5 simultaneous bimolecular reaction paths, one of which has a fast-sequential reaction converting NO , if present. These reactions are (H. S. Johnston, 1955):



The earlier work of Johnston et al. (1951) indicated that decomposition of HNO_3 was first order for temperatures above 670 K and given approximately as $k_{25} = 1.34 \times 10^9 \exp(-32,600/RT) \text{ s}^{-1}$. For temperatures below 570 K, the decomposition proceeded initially as a first-order reaction followed by a rate slower than first-order. Appearance of this transition was also a function of the reaction vessel size.

The later work of Johnston et al. (1955) concerning HNO_3 indicated that for temperatures above 650 K and a 2-liter Vycor bulb, the decomposition of pure HNO_3 followed reaction (25) with a first-order rate constant proportional to initial concentration. The proportionality is an empirical second-order rate constant given by $k_{25} = 9.3 \times 10^{-17} \exp(-38,300/RT) \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$. Using argon and oxygen as foreign gasses in separate experiments, the first-order rate constant was found to be about $5.38 \times 10^4 [M] + 0.006 \text{ s}^{-1}$ for $[M] < 2.4 \times 10^{14} \text{ molecules/cm}^3$, a temperature of 673 K and initial HNO_3 concentration of $1.29 \times 10^{14} \text{ molecules/cm}^3$. At temperatures of interest, the second-order rates are

$$\begin{aligned} k_{25} &= 7.3 \times 10^{-35} \text{ cm}^3\text{-molecules}^{-1}\text{-s}^{-1} \quad (298 \text{ K}) \\ &= 5.5 \times 10^{-22} \quad (550 \text{ K}) \end{aligned}$$

The primary step in all the reactions of HNO_3 bimolecular decomposition is that of reaction (27), the formation of nitrogen dioxide and the hydroxyl radical. At low total pressures (<5 torr) and, therefore, slower rates, nitric oxide and nitrogen dioxide accelerate and inhibit, respectively, the decomposition.

Atkinson et al. (1976) along with others have studied the formation rate of HNO_3 , i.e., reaction (-27). The data is summarized in the NASA Reference Publication 1010 (1977). The recommended form for k_{-27} is very complex (Hampson and Garvin 1978) but a reasonable approximation can be found in Anderson et al. (1974). The expression they obtain is:

$$\begin{aligned} k_{-26} &= 2.3 \times 10^{-30} \times (295/T)^{2.5} \text{ cm}^6\text{-molecule}^{-1}\text{-s}^{-1} \\ &295 \leq T \leq 450 \text{ K} \end{aligned}$$

The forward decomposition reaction rate, k_{27} , can be derived using the equilibrium constant for the reaction, $K_{27,-27}$

$$K_{27,-27} = (1.3 \times 10^{30}/T \exp[-24970/T]) \text{ molecules-cm}^{-3},$$

where $K_{27,-27}$ has been evaluated at room temperature using the JANAF tables. The functional form of $K_{27,-27}$ should not change appreciably with temperature over the range of interest. Thus, the form for the forward reaction can be found

$$k_{27} = K_{27,-27} \times k_{-27} = (3.0/T) (295/T)^{2.5} \exp \left(-24970/T \right) \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1}$$

$$k_{27M} = 4.6 \times 10^{-20} \text{ s}^{-1} \quad (298 \text{ K}, 1 \text{ atm})$$

$$= 0.36 \text{ s}^{-1} \quad (650 \text{ K}, 1 \text{ atm})$$

$$\text{and } k_{-27M} = 6.2 \times 10^{-11} \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1}$$

$$- 8.6 \times 10^{-12} \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1} \quad (650 \text{ K}, 1 \text{ atm})$$

Similarly, for reaction (28), the forward rate has been evaluated (Hampson and Garvin, 1978),

$$k_{28} = 8 \times 10^{-14} \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1} \quad (240 \leq T \leq 406 \text{ K}).$$

The equilibrium constant, $K_{28,-28}$, derived from the JANAF tables satisfies the equation

$$K_{28,-28} = 2.6 \exp \left[9160/T \right].$$

Thus, the backward reaction rate, k_{-28} , becomes:

$$k_{-28} = (k_{28}/K_{28,-28}) = 2.4 \times 10^{-14} \exp \left(-9160/T \right) \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1}$$

$$= 2.6 \times 10^{-41} \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1} \quad (298 \text{ K})$$

$$= 1.8 \times 10^{-20} \text{ cm}^3 \text{ -molecule}^{-1} \text{ -s}^{-1} \quad (650 \text{ K})$$

For reaction (25) the growth of NO_2 is given by:

$$\frac{d[\text{NO}_2]}{dt} = k_{25} [\text{HNO}_3] - k_{25} [\text{HO}] [\text{NO}_2] + 2 k_{28} [\text{NO}] [\text{NO}_3]. \quad (3-32)$$

Noting that HO, NO_3 and NO are the temporary or intermediary molecules and employing successive steady-state approximations until a solution is reached, we consider the rate of change of HO with respect to time first. This is given by;

$$\frac{d[\text{HO}]}{dt} = k_{27} [\text{HNO}_3] - k_{-27} [\text{HO}] [\text{NO}_2] - k_{28} [\text{HO}] [\text{HNO}_3] \quad (3-33)$$

Accepting the statement that HO will be short-lived at high temperature, the generation and removal rates will be comparable, so that $d[\text{HO}]/dt$ will be zero. This is a statement equivalent to the steady-state approximation for kinetic reactions. With this in mind,

$$k_{27} [\text{HNO}_3] = k_{-27} [\text{HO}] [\text{NO}_2] + k_{28} [\text{HO}] [\text{HNO}_3] \quad (3-34)$$

or that

$$[\text{HO}] = \frac{k_{27} [\text{HNO}_3]}{k_{-27} [\text{NO}_2] + k_{28} [\text{HNO}_3]} \quad (3-34a)$$

The rate of change or growth of the radical NO_3 is given by;

$$\frac{d[\text{NO}_3]}{dt} = k_{28} [\text{HO}] [\text{HNO}_3] - k_{29} [\text{NO}_2] [\text{NO}_3] - k_{30} [\text{NO}] [\text{NO}_3] \quad (3-35)$$

Making the same steady-state approximation for NO_3 as with HO, one has;

$$k_{28} [\text{HO}] [\text{HNO}_3] = k_{29} [\text{NO}_2] [\text{NO}_3] + k_{30} [\text{NO}] [\text{NO}_3] \quad (3-36)$$

or that;

$$[\text{NO}_3] = \frac{k_{28} [\text{HO}] [\text{HNO}_3]}{k_{29} [\text{NO}_2] + k_{30} [\text{NO}]} \quad (3-36a)$$

Combining equations (3-36a) and (3-34) one has that;

$$k_{27} [\text{HNO}_3] = k_{-27} [\text{HO}] [\text{NO}_2] + k_{29} [\text{NO}_2] [\text{NO}_3] + k_{30} [\text{NO}] [\text{NO}_3] \quad (3-37)$$

when equations (3-37) and (3-32) are combined, one has after factoring;

$$\frac{d[\text{NO}_2]}{dt} = \text{NO}_3 (k_{29} [\text{NO}_2] + k_{30} [\text{NO}]) \quad (3-38)$$

From equation (3-36) one has;

$$[\text{NO}_3] = \frac{k_{28} [\text{HO}] [\text{HNO}_3]}{k_{29} [\text{NO}_2] + k_{30} [\text{NO}]} \quad (3-39)$$

or since,

$$[\text{HO}] = \frac{k_{28} [\text{HNO}_3]}{k_{-27} [\text{NO}_2] + k_{28} [\text{HNO}_3]} \quad (3-40)$$

from equation (3-34), NO_3 can be written as,

$$[\text{NO}_3] = \frac{k_{28} k_{27} [\text{HNO}_3]^2}{k_{-27} [\text{NO}_2] + k_{28} [\text{HNO}_3]} \quad (3-41)$$

The final result desired is obtained by combining equations (3-41) and (3-38) to yield;

$$\frac{d[\text{NO}_2]}{dt} = \frac{k_{28} k_{27} [\text{HNO}_3]^2}{k_{-27} [\text{NO}_2] + k_{28} [\text{HNO}_3]} \times \frac{k_{29} [\text{NO}_2] + 3 k_{30} [\text{NO}]}{k_{29} [\text{NO}_2] + k_{30} [\text{NO}]} \quad (3-42)$$

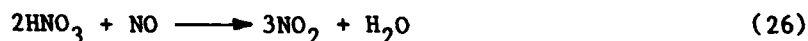
which can be simplified, since $k_7 \gg k_{29}$ (pg 39), to;

$$\frac{d[\text{NO}_2]}{dt} = \frac{3k_{28} k_{27} [\text{HNO}_3]^2}{k_{-27} [\text{NO}_2] + k_{28} [\text{HNO}_3]} \quad (3-43)$$

For reaction (28) the rate of loss for HNO_3 is given by:

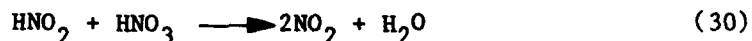
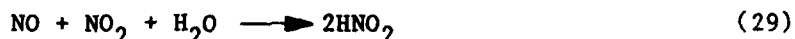
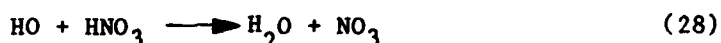
$$-\frac{d[\text{HNO}_3]}{dt} = \frac{2k_{27} k_{28} [\text{HNO}_3]^2}{k_{-27} [\text{NO}_2] + k_{28} [\text{HNO}_3]} \quad (3-44)$$

With NO present, one has the following net stoichiometric reaction:



so that the final concentration of NO_2 is 1.5 times that of the HNO_3 . For low pressure Johnston et al. (1955) has postulated that five steps are required. The governing mechanistic reactions for reaction (26) are the decomposition of HNO_3 followed by a product-reactant reaction followed by a fast

bimolecular reaction converting NO to NO₂ followed by the synthesis of nitrous acid as an intermediary and finally a reaction to consume the nitrous acid. These reactions are:



3.7.2 Stratospheric-Based Instrument Modeling

The above set of reactions along with the N₂O₅ set of reactions given in paragraph 3.5.1 have been modeled using the EPISODE code to analyze and review concentration-time profiles of the various molecules and radicals. For laboratory conditions the profiles indicate that OH scavenging and elevated temperature promote the decomposition of HNO₃. For stratospheric conditions the profiles (c.f., Appendix B-2) indicate that only increased temperature promotes the decomposition of HNO₃, i.e., the intermediary radical, OH does not impede decomposition.

3.7.3 Experimental Procedures and Results

The thermal decomposition of nitric acid was investigated in the laboratory with emphasis placed on a quantitative understanding and investigation of the thermal converter. In addition, ozone was introduced into the HNO₃ vapor stream to determine whether it had an effect on the thermal conversion process.

The thermal converter based on a design by Stedman (private communication, 1977), consisted of 40 cm long by 1/2-inch O.D. stainless steel tubing, packed with 1/8-inch diameter Pyrex balls. (The Pyrex balls served to raise the temperature of the gas to that of the tubing.) The tubing was wrapped with electrical heating tape and placed in an insulated cylinder. The temperature of the tubing, which could be raised as high as 500°C, was monitored

by a thermocouple inserted into the gas stream exiting the converter. The temperature could also be monitored by observing the $[\text{NO}_2]/[\text{NO}]$ ratio of the exiting gas stream when NO_2/air was passed through the converter.

The gas leaving the thermal converter was directed into an absorption cell, placed within a Perkin-Elmer Model 580A spectrophotometer. The internal optics of the absorption cell were set to an optical absorption path length of 15.75 m. The infrared absorption properties of HNO_3 , NO_2 , and O_3 are discussed in Appendix A of this report. By combining the known optical properties of these molecules with the observed IR absorptions, the number density of each specie was determined.

From the absorption cell, the flowing gas was directed to the Aerochem chemiluminescence monitor. The monitor was primarily used to detect NO (whose IR absorption signature is especially weak) and to verify the spectroscopically determined, NO_2 concentrations. A comparison of the analytical methods using the CL monitor as opposed to those of the spectrophotometer was made by using an NO_2/air span gas diluted by additional nitrogen to maintain a constant flow rate (Figure 3-12). Agreement between the two instruments was quite good. The CL monitor response also was determined as a function of gas flow rate (Figure 3-13). Since the CL monitor has a constant volume bellows pump within it to draw in the gas sample, the pressure in the sample line will vary as the flow rate is externally controlled, but the response remains proportional to the flow rate.

It was shown in paragraph 3.6.1 that the concentration of HNO_3 depended inversely on the gas flow rate. Therefore, to vary the concentration of HNO_3 , the flow rate was varied. However, pressure in the absorption cell was maintained at constant value by adjusting a throttle valve located at the exit port of the cell. CL monitor readings were "normalized" to take into account the varying flow rate.

The results of a set of measurements of the thermal conversion of HNO_3 are given in Table 3-16. The absorption data is based on the average of three or more successive spectra recorded after equilibrium was established. The standard deviation of the data is also shown. The CL monitor response for

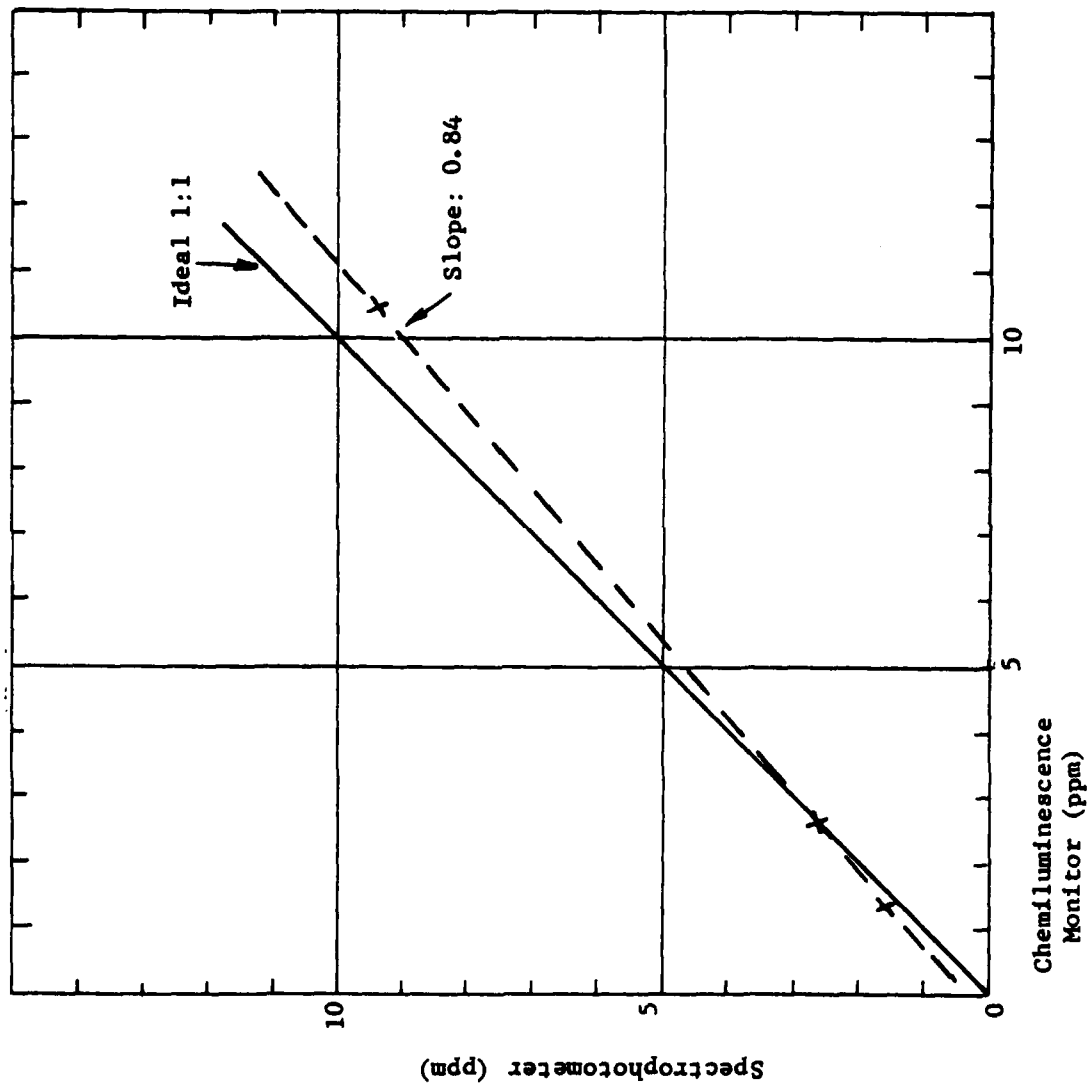


Figure 3-12. Response of Perkin-Elmer Model 580 Spectrophotometer vs. Chemiluminescence Monitor

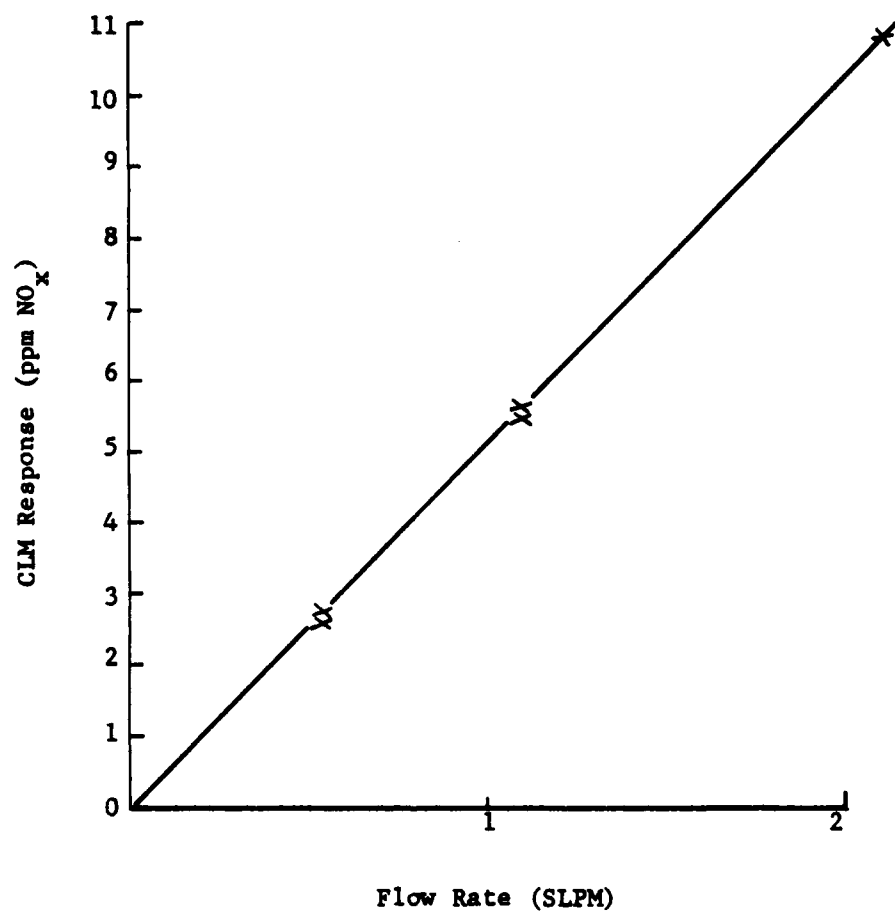


Figure 3-13. Chemiluminescence Monitor Performance vs. Flow Rate

TABLE 3-16. HNO_3 THERMAL CONVERTER (a) PERFORMANCE CHARACTERISTICS WITH CROSS CALIBRATION HIGH TEMPERATURE CATALYTIC CONVERSION VS CHEMILUMINESCENCE MEASUREMENTS

<u>Apparatus</u>	<u>Sampling Point</u>	<u>Concentration</u>	<u>Analysis Techniques</u>
Low Temperature Converter, $T = 535 \pm K$	Inlet	$[\text{HNO}_3] = 7.2 \pm 0.4 \text{ ppm}$	IR Absorption
	Outlet	$[\text{NO}_2] = 7.1 \pm 0.4 \text{ ppm}$	IR Absorption
		$[\text{NO}_2] = 7.8 \text{ ppm}$	Thermal-CLM
		$[\text{NO}] = 0.6 \text{ ppm}$	CLM
		$[\text{NO}_2] / [\text{NO}] = 14.0^{(b)}$	Ratio
High Temperature Catalytic Converter and CLM	Inlet	$[\text{HNO}_3] = 7.2 \pm 0.4 \text{ ppm}$	IR Absorption
	Outlet Sample	" $[\text{NO}_2]$ " = 7.7	Thermal-CLM

(a) Tubular Pyrex Converter with 3 mm Pyrex Spheres

(b) Equivalent to Thermodynamic Conversion Ratio Expected for $T = 535 \text{ K}$

the NO_x mode was scaled to compensate for the changing flow rate. The NO and NO_x -NO density data were derived from the NO_x data and from NO concentration data which was also recorded by the monitor.

Comparison of the second and third lines of the table shows that the amount of NO (from the CL monitor) plus NO_2 (from the spectrophotometer) is approximately 7.6×10^{13} molecules/cm³, which is very good agreement. It is also apparent by comparing lines 3 and 4 that the Aerochem instrument responds accurately to HNO_3 vapor. Comparison of the CL monitor with the Model 580A (Figure 3-12) indicated a 5 to 15 percent disagreement. This uncertainty limits our ability to compare data to an accuracy of no more than 10 percent. Finally, the $[\text{NO}_2]/[\text{NO}]$ ratio indicates a gas temperature in the 520 to 550 K range.* The equilibrium constant, $K_{25,-25}$, at this temperature indicates that the HNO_3 should be entirely decomposed.

In an earlier experiment, dry nitrogen rather than air was used as the carrier gas for the HNO_3 vapor. The products of the decomposition were NO_2 and NO, with their ratio being $[\text{NO}_2]/[\text{NO}] = 3.48 \pm 0.91$; the HNO_3 concentration was 5.68×10^{13} molecules/cm³. The optically detected NO_2 density was 4.52×10^{13} . Taking into account the measured ratio of $[\text{NO}_2]$ to $[\text{NO}]$, the sum of $[\text{NO}]$ and $[\text{NO}_2]$ was 5.82×10^{13} molecules/cm³ which is, again, very close to the original HNO_3 number density. The uncertainty associated with these measurements was on the order of ± 10 percent. The reason the $[\text{NO}_2]$ to $[\text{NO}]$ ratio differed from the data discussed previously is related to the use of nitrogen. The NO_2 decomposition reaction to form NO has been discussed (paragraph 3.3.1.1); the equilibrium constant is independent of the carrier gas (air or nitrogen) while the actual ratio of $[\text{NO}_2]$ to $[\text{NO}]$ is not. This explains the difference in the ratio for the two cases. For either carrier gas, the thermal decomposition of HNO_3 was total to within experimental error. The maximum flow rate used was 2.1 SLPM.

*For determination of the $[\text{NO}_2]/[\text{NO}]$ ratio, the following expression has been used;

$$\log \frac{[\text{NO}_2]}{[\text{NO}]} = \frac{3002}{T} - 4.223 \quad (3-15)$$

and is applicable at standard pressure and $p\text{O}_2 = 152$ torr.

3.8 POTENTIAL INTERFERENTS (Specificity)

Potential interferents that could affect the performance of a hybrid chemical conversion system include members of both the ClO_xNO_x family of compounds and the HO_xNO_x family of compounds. These potential interferents would effect the performance through either photolytic effects or thermal lability effects. This program has (1) analyzed photolytic effects of the ClO_xNO_x compounds, (2) experimentally treated ClONO_2 until accidental loss of all material occurred, and (3) analyzed thermal lability effects of ClONO_2 and HO_2NO_2 .

3.8.1 ClO_xNO_x Family Compounds

This section treats the potential interference of nitrosyl chloride (ClNO), nitryl chloride, chlorine nitrite (ClONO), and chlorine nitrate (ClONO_2) with particular emphasis on ClONO_2 since it is believed to be the most predominant member of the ClO_xNO_x family with the stratosphere.

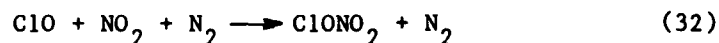
3.8.1.1 ClONO_2 Generation and Handling

Chlorine nitrate has been synthesized by three methods. The first method (M. Schmeisser, 1963) is based upon the reaction:



The boiling point of nitrate is about 18 to 22°C and the melting point about -107°C. Decomposition occurs slowly at room temperature, but since Cl_2O is very sensitive to heat, ClONO_2 is commonly stored at -78°C or at 77 K. For research quantities, much of the ClONO_2 used in the United States has been generated and distributed by Dr. Louis C. Glasgow of E.I. DuPont de Nemours and Company. Containerization employs glass ampules or 1 liter stainless steel sampling cylinders.

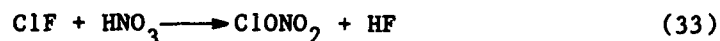
A second method (Birks et al., 1977) is based upon the reaction:



The initial reactants employed are 0.25 percent Cl_2 in He, O_2 for the generation of O_3 and NO_2 in air. Two mixing stations are required; the first

downstream from a microwave cavity, which is used for generating atomic chlorine, and an ozonator. The second is downstream for the ClO and NO₂ sources.

The third method (Schack, 1967) is based upon the reaction:



The technique avoids the use of hazardous chlorine oxides since the initial reactants are separately condensed at 77 K into a stainless steel container. Bulb to bulb cryogenic vacuum distillation separates the ClONO₂ from the hydrofluoric acid.

The ClONO₂ used for this program was generated by Dr. Glasgow, on February 3, 1977, using the first method. The batch (about 5 gms material at 77 K) underwent four bulb to bulb distillations before delivery and a fifth distillation to remove further accumulated impurities was carried out at Perkin-Elmer on September 12, 1978. The material was held at 77 K between the last two distillations. Impurities generally include Cl₂, NO₂, N₂O₅, HNO₃, and possibly ClO₂. Absorption spectra in the UV region indicated that the transferred batch contained substantial Cl₂, but no discernible NO₂.

The vapor pressure of ClONO₂ is given by the expression (Schack, 1967):

$$\log P = \frac{-0.05223 A}{T} + B \text{ (torr)}$$

where A = 28,900 and B = 7.9892. This expression versus 1000/T is shown in Figure 3-14 along with similar data from International Critical Tables for the possible impurities Cl₂, NO₂, N₂O₅, and HNO₃. For the fifth distillation, Cl₂ was pumped off at 163 K using a slush of methyl butane (isopentane) and liquid nitrogen. The batch was then elevated in temperature to about 223 K and the ClONO₂ was allowed to enter an IR absorption cell at a pressure of about 14 torr or condensed in a one-liter stainless steel container.

Materials in contact with the sample included glass, teflon,^{(R)*} stainless steel, KBr and halofluorocarbon grease, GR-90.

* PTFE, Polytrifluoromonoethylen

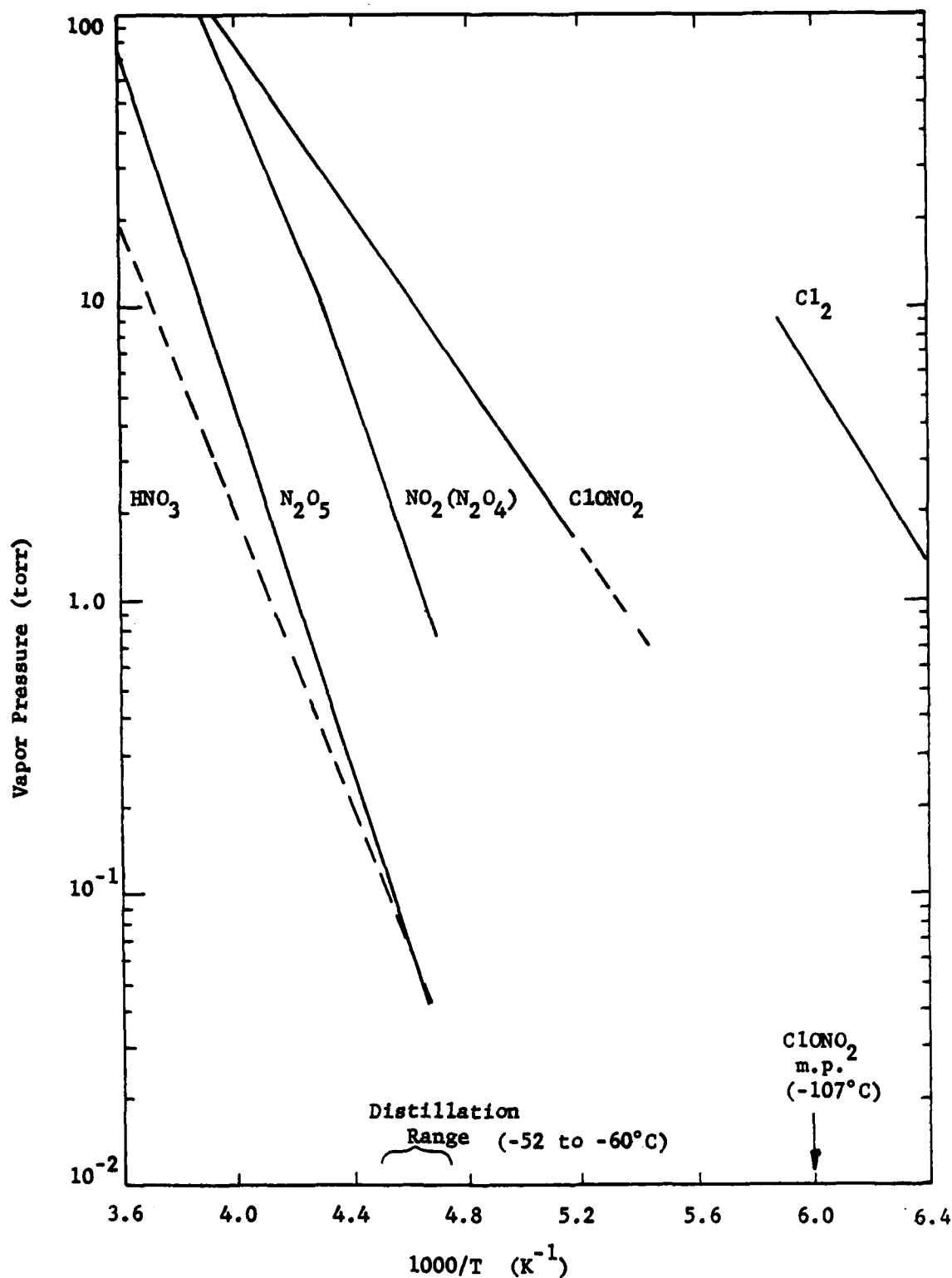


Figure 3-14. Vapor Pressure Diagram For ClONO₂ and Suspected Impurities

The ClONO_2 absorbance was then measured on a Perkin-Elmer Model 521 spectrometer using a preset resolution of 3 cm^{-1} for survey over the region 1800 to 650 cm^{-1} . During transfer of the cell to the spectrometer, the sample was held at about 273 K and protected from room lighting by Wratten-type filter material.

Two lengthy survey scans were performed and analyzed as the ClONO_2 thermally decomposed at a temperature close to 306 K. The decay of ClONO_2 is shown in Figure 3-15 and a summary of the data is given in Table 3-17. Due to the large absorbance of ClONO_2 , the initial fill pressure of about 14 torr was reduced to a pressure of about 4.5 torr. The unused portion of ClONO_2 was diluted with nitrogen and stored at LN_2 temperature for later use. Failure of the automatic LN_2 transfer system during an unattended period caused the ClONO_2 to attain room temperature with a subsequent thermal decay.

3.8.1.2 ClO_xNO_x Photolytic Interferents

It is possible that, should an NO_2 photolytic converter be used prior to chemiluminescence of NO with O_3 , the ClNO , ClNO_2 , ClONO , and ClONO_2 will be photolyzed to yield a higher reading of NO_2 . This possibility is not considered to be significant as is shown below, where what is believed to be typical concentration levels and a white-light photolytic source appropriately filtered with Corning C.S. 7-54 and 320-nm glass filters are assumed.

Although three members of the ClO_xNO_x family are not considered among the "most important" trace species of the stratosphere and, therefore, have received very little attention, they do have significant near-UV absorption cross-sections as is seen shown in Figure 3-16. The fourth species, ClONO_2 is considered to be an important sink as it does not appear to react with O_3 and reacts only slowly with O.

By employing Corning C.S. 7-54 and 320-nm filters, an NO_2 photolytic technique was developed that rendered NO_3 a negligible interferent. The same analysis was applied to the ClO_xNO_x family to test the specificity of the technique. The results of the analysis are shown below in Table 3-18 with the appropriate absorption cross-sections taken from Figure 3-16.

• ClONO_2
 Δ NO_3
 \circ HNO_3
 \square NO_2

Curve A $P \sim 14$ Torr
 Curve B $P \sim 4.5$ Torr

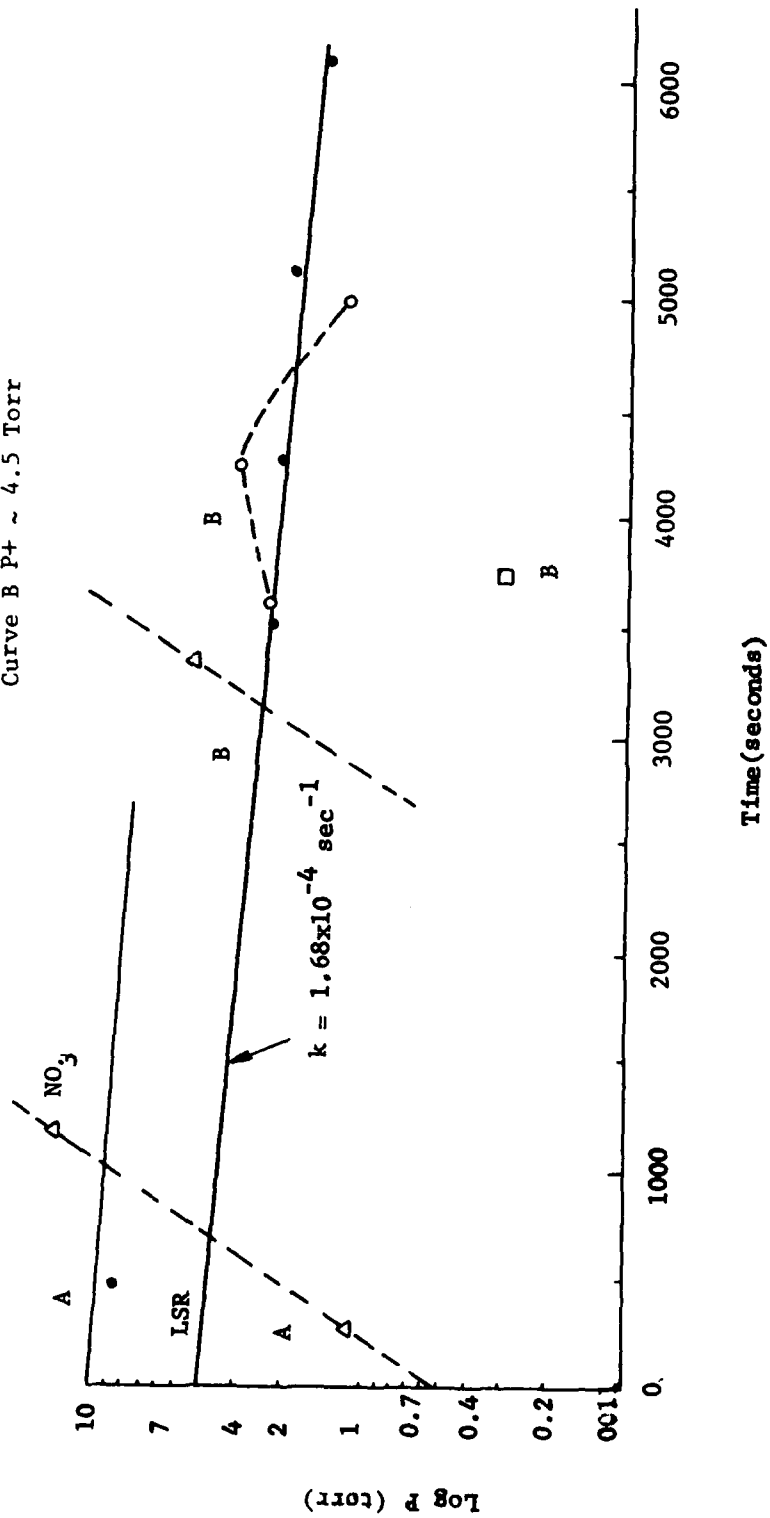
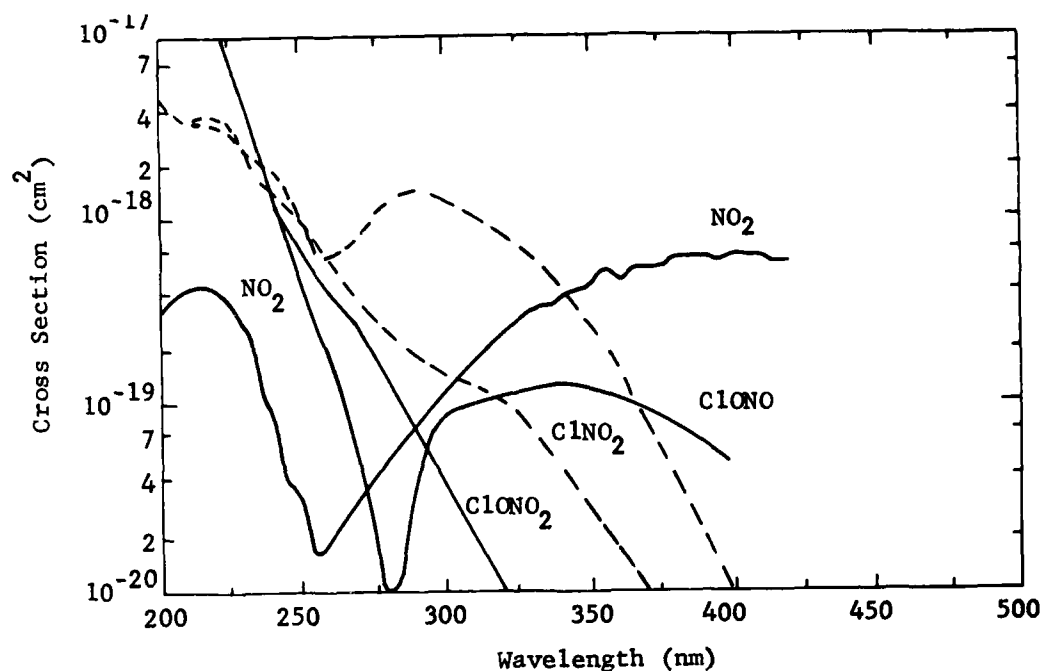


Figure 3-15. Thermal Decomposition of ClONO_2 , $T \sim 306 \text{ K}$

TABLE 3-17. ClONO_2 THERMAL DECOMPOSITION

Time (Sec)	Species	Analytical Band (cm^{-1})	Number Density (Molec/ cm^3)	Partial Pressure (Torr)	Total Pressure
285	NO_3	1860	0.04×10^{18}	1.13	~14 Torr
480	ClONO_2	(ν_1)1735	0.27	8.5	
1160	NO_3	1345	0.48	15	
1250	ClONO_2	(ν_2)1292	0.34	10.6	
1335	N_2O_5	(ν_{12})1246	0.0049	0.154	
3330	NO_3	1860	0.14	4.29	~4.5 Torr
3520	ClONO_2	(ν_1)1735	0.068	2.14	
3580	HNO_3	(ν_2)1706	0.071	2.22	
3760	NO_2	(ν_3)1600	0.0091	0.29	
4280	HNO_3	(ν_4)1311	0.095	2.98	
4290	ClONO_2	1292 (ν_2)(1288)	0.064	2.03	
4970	HNO_3	890	0.037	1.17	
5140	ClONO_2	(ν_4) 780	0.056	1.77	
6090	ClONO_2	(ν_1)1735	0.044	1.39	



NO₂ Curve Adapted from H.S. Johnston and R. Graham, Can. J. Chem 52, 1415 (1974). Other Curves from NASA Reference Publ. 1010, Chlorofluoromethanes and the Stratosphere, edited by R. Hudson, Aug 1977

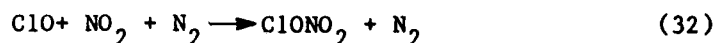
Figure 3-16. NO₂ and ClO_xNO_x Absorption Cross-Sections

TABLE 3-18. ClO_xNO_x-NO₂ PHOTOLYSIS SPECIFICITY WITH SPECTRAL DISCRIMINATION

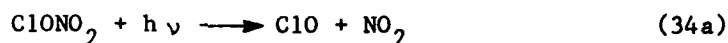
Species	$\sum_{300}^{420} \sigma_{\lambda} \tau_{\lambda} d\lambda \times 10^{19} \text{ cm}^2\text{-nm}$	$\sum_{300}^{420} \sigma_{\lambda} \tau_{\lambda} d\lambda \sum_{300}^{420} \sigma_{\text{NO}_2} \tau_{\lambda} d\lambda$
Nitrogen dioxide, NO ₂	278	1.00
Nitrosyl chloride, ClNO	74	0.27
Nitryl chloride, ClNO ₂	25.2	0.091
Chlorine nitrite, ClONO	289	1.04
Chlorine nitrate, ClONO ₂	2.7	0.01

Almost nothing is known about the stratospheric concentrations or mixing ratios of the ClO_xNO_x species cited above. Modeling efforts to date extend only to ClONO_2 (F.M. Luther, 1976; and NAS, 1976), and an upper limit of 2 ppb using the limb sensing techniques has been set (D.G. Murcray et al., 1977).

At stratospheric altitudes chlorine nitrate is produced at night by the combination of the radical ClO with NO_2 according to



During periods of high insolation, ClONO_2 is photolyzed via



At high altitude, i.e. $H > 40$ km, the photolysis takes place over the 200 to 220 nm spectral region, where the absorption cross-section is $\sim 375 \times 10^{-20} \text{ cm}^2/\text{molecule}$. At lower altitudes, 20 to 30 km, the diurnal change is much less, with photolysis occurring with radiation greater than 300 nm.

The number density profile of NO_2 and ClONO_2 are shown below in Table 3-19. The ClONO_2 concentrations are taken from F.M. Luther (1976). As the chlorine nitrate density is well below that of NO_2 , the 0.01 of Table 3-18 is further reduced. ClONO_2 photolyzation is not considered to be a significant problem.

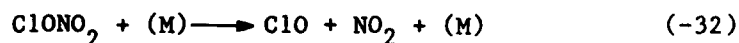
TABLE 3-19. NO_2 AND ClONO_2 CONCENTRATIONS

H (km)	NO_2 (30 N) (molecules/cm ³)	ClONO_2 (midnight) (molecules/cm ³)	ClONO_2 (noon) (molecules/cm ³)
20	1.3×10^9	1.0×10^8	8.0×10^7
25	2.2	4.0	3.0×10^8
30	2.2	1.8	1.0
35	1.4	5.0×10^7	1.0×10^7

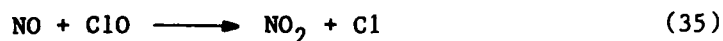
3.8.1.3 ClONO₂ Thermal Lability

The thermal lability of ClONO₂ can impact laboratory handling of the material as well as the specificity of a chemical conversion system that relies on thermal converters. A study by Cox et al. (1977), which employed enthalpy data for ClONO₂, led to a high pressure limiting rate given by $10^{14} \exp(-12480/T) \text{ s}^{-1}$ for the decomposition of ClONO₂. This rate is about 3500 times slower than the data of Knauth cited below. The most recent work concerning ClONO₂ kinetics is that of H.D. Knauth (1978). Thermal decompositions were studied in the presence of NO, ClNO, and N₂ over the temperature range 303 to 363 K at pressures from 3 to 380 torr.

The unimolecular dissociation of ClONO₂ given by:



is followed by the fast reaction:



and chlorine consuming steps.

For pN₂ less than 220 torr, an empirical first-order linear relationship between k₋₃₂ and [N₂], $k_{-32} = k_{-32a} + k_{-32}'' [\text{N}_2]$, was determined and suggests that the reaction takes place in the low pressure region or that the departures from the low pressure region are quite small. The rates are given as:

$$k_{-32a} = 2.743 \times 10^6 \exp(-3350/RT) \text{ sec}^{-1} \text{ and}$$

$$k_{-32}'' = 1.318 \times 10^{-5} \exp(6030/RT) \text{ cm}^3\text{-mole}^{-1}\text{-sec}^{-1}$$

The Knauth data for $T \geq 333 \text{ K}$ is shown in Figure 3-17 in log-log form. The data for $T \leq 300 \text{ K}$ has been extrapolated from the higher temperature data. The solid square points ($T = 263 \text{ K}$) correspond to reasonable N₂ diluent storage conditions, but half-life values of only 257 and 82 minutes (at the knee) result. At a temperature of -78°C , the half-life becomes $1.7 \times 10^8 \text{ sec}$, or $2 \times 10^3 \text{ days}$.

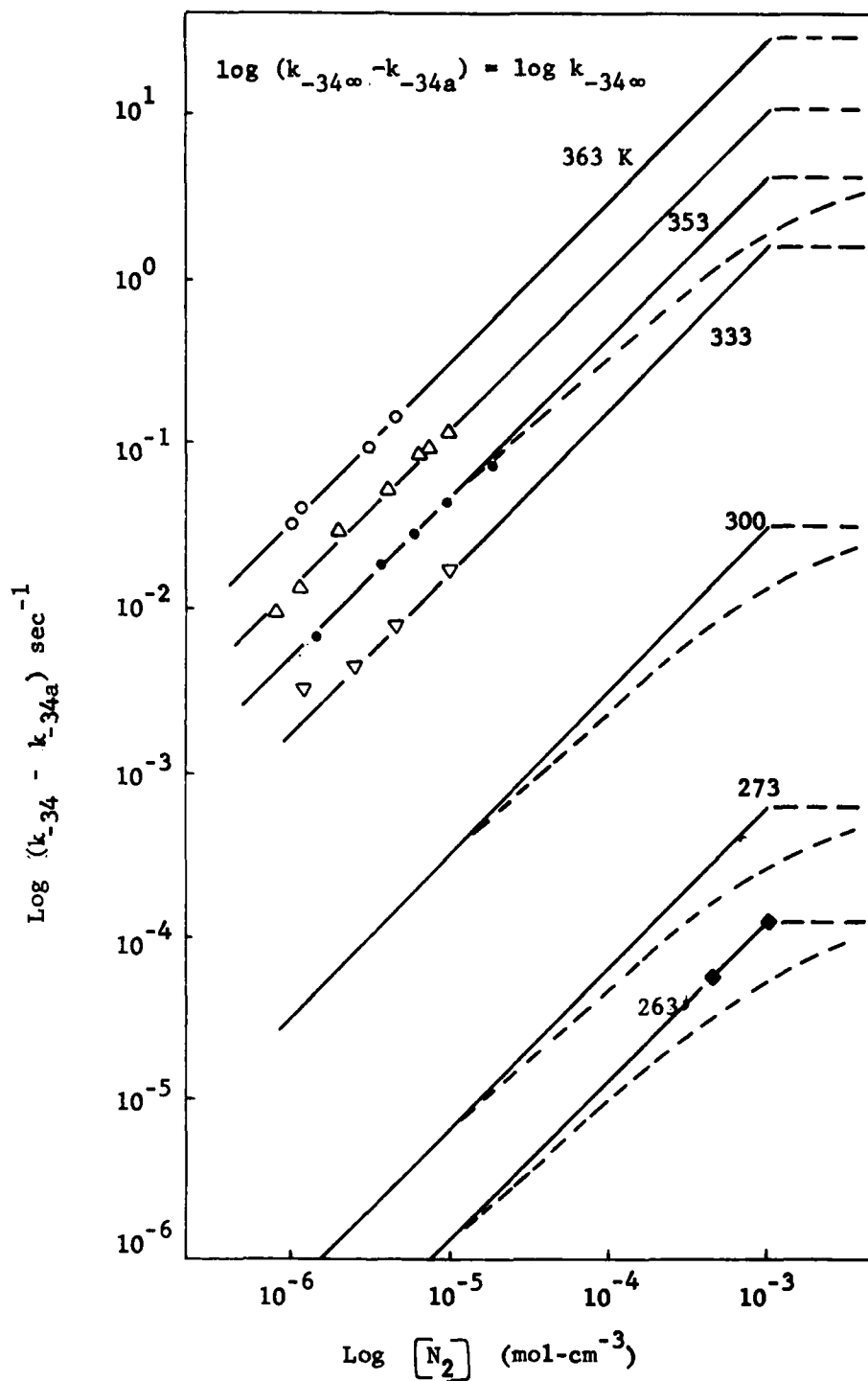
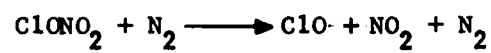
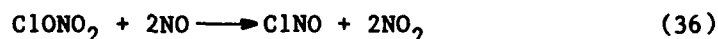


Figure 3-17. Chlorine Nitrate-Nitrogen Dissociation Rates

For the region of the stratosphere of interest (N_2) $\sim 3 \times 10^{18}$ molecules- cm^{-3} and less, so that a 300 K monitoring system with a sample residence time of several seconds would not cause appreciable thermal decay of $ClONO_2$. On the other hand, a 400 K converter for N_2O_5 would result in a $ClONO_2$ half-life of about 0.2 second, causing a lack of specificity for N_2O_5 conversion. Impact on the HNO_3 conversion accuracy due to the relative concentrations would be minimal.

The reaction of $NO + ClONO_2$ leading to an intermediate product but ultimately to $ClNO$ and NO_2 or



was also studied by Knauth. The rate was found to be $2.09 \times 10^{-12} \exp(-11,850/RT)$ cm^3 -molecule $^{-1}$ -s $^{-1}$ over the 303 to 343 K temperature range. At 300 K, Knauth's value of 4.8×10^{-21} cm^3 -molecule $^{-1}$ -s $^{-1}$ is in general agreement with that of Rowland et al. (1976). At 800K, the highest anticipated temperature for any of the converters, the rate is still small, 1.2×10^{-15} cm^3 -molecule $^{-1}$ -s $^{-1}$, so NO will not be consumed by the $ClONO_2$. However, the ClO product from thermal decomposition will consume NO .

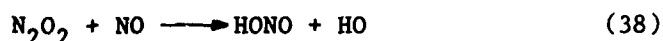
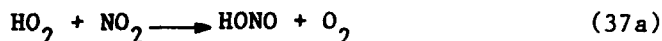
3.8.2 HO NO Family Compounds

This section treats the potential interference of nitrous acid ($HONO$) and pernitric acid (HO_2NO_2); nitric acid ($HONO_2$) having been treated in Section 3.4. Concern with $HONO$ is dismissed by review of pertinent reactions for its generation and high degree of photodissociation during daylight hours. Concern with HO_2NO_2 is not dismissed.

3.8.2.1 Nitrous Acid ($HONO$)

Nitrous acid is not expected to be an interferent for reasons cited below. It can be formed by four methods. These reactions are:



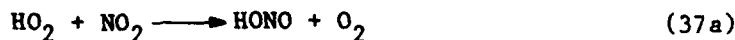


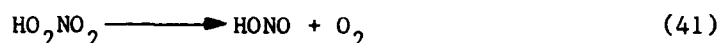
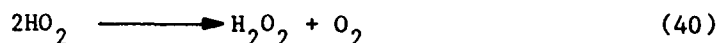
Reaction 31 is favored within power plant stacks and near automobile exhausts. Room temperature kinetic studies have been carried out by Chan et al. (1976). Rate constant estimates that use 0.5 cm^{-1} spectroscopy to follow the kinetics are $k_{31} = 6 \times 10^{-38} \text{ cm}^6\text{-molecule}^{-2}\text{-s}^{-1}$ and $k_{-29} = 9.4 \times 10^{-19} \text{ cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$. Since this is a third-order rate equation, it will not proceed to the right unless the conditions are very favorable. Furthermore, if HONO is formed it will be photolyzed with $J > 0$, since $\sigma > 1 \times 10^{-19} \text{ cm}^2/\text{molecule}$ for $340 \leq \lambda \leq 384 \text{ nm}$ (NASA Publ. 1010). Hydridizing nitrogen sesquioxide, reaction 38, has not received much attention. The free radical HO_2 combination with NO_2 , reaction 39, favors the formation of HO_2NO_2 since k_{37a}/k_{37b} is about 10^{-3} (Graham et al., 1977). The kinetics of this reaction have also been studied by Simonaitis and Heicklen, (1974).

3.8.2.2 Pernitric Acid (HO_2NO_2)

The formation of HO_2NO_2 via a two-body or three-body mechanism proceeds well enough that it has been modeled by Jesson et al. (1977) with peak concentrations occurring in the 1- to 3-ppb range. It is therefore considered to be a potential interferent.

The reaction kinetics of HO_2 and NO_2 have been studied recently by a number of investigators: Simonaitis and Heicklen (1974, 1976, and 1977), Howard and Evenson (1977), Niki et al. (1977), Jesson et al. (1977), Graham et al. (1977) and Graham et al. (1978). The reactions most often considered are:





Reactions (39a) and (39b) are occasionally written as a three-body reaction. The rate constants are given in Table 3-20 in the usual Arrhenius' form. The disparity in term values for k_{-37} will be shown below to have little significance for temperatures greater than 298 K.

As the thermal conversion technique is being considered for N_2O_5 and HNO_3 species, the above cited first-order rates for reaction (-39b) have been developed for laboratory ambient temperature, 400 K and 550 K. These rates, half-lives, and NO_2 concentrations are presented in Table 3-21 for resident times of 1, 2, and 5 seconds. For temperatures above 298 K, conversion is complete in less than a second and yields NO_2 . This conversion reduces the specificity of N_2O_5 and HNO_3 measurements. For an NO_2 measurement at an instrument temperature of 298 K, lack of specificity is also present, but to a lesser degree. Straightforward solution to this problem, if the published rates have been correctly extrapolated, is not apparent at this time.

Finally, since the thermal decomposition of HO_2NO_2 changes from first order to second order with reduced pressure, k_{-37b} should be investigated at stratospheric temperatures and pressures $760 \gg p > 7$ torr.

3.9 NO_z FEASIBILITY*

For a selective hybrid chemical conversion (SMS), thermal partitioning of thermally labile molecules, N_2O_5 and HNO_3 , has led to ambiguous results and therefore such a system is not recommended. The method is also particularly susceptible to interfering species as mentioned in Paragraph 3.8. A non-selective or total odd nitrogen SMS, however, is feasible and described below.

* NO_z is defined to be the sum of NO , NO_2 , N_2O_5 , HNO_3 , ClONO_2 , and HO_2NO_2 . Adsorbed NO_z on particulate matter is not specifically treated in the text. Three modes of sampling are believed to be possible. First by using a fluoropore-type filter only, a gas phase sample will be accepted. Second, by employing no filter, both gas phase and adsorbed NO_z will be accepted. A third conceivable mode would be a time shared combination of the first two modes.

TABLE 3-20. HO_xNO_x REACTION RATES

Reaction	Rate (cm ³ -molecule ⁻¹ -s ⁻¹)	Rate (295-300 K) (cm ³ -molecule ⁻¹ -s ⁻¹)	Reference
(41a)	$1.2 \times 10^{-11} \exp [-(1400+500)/RT]$	1.14×10^{-12}	Simonaitis and Heicklen (1977)
(41b)	$3.0 \times 10^{-11} \exp (-775/RT)$	8.03×10^{-12}	Howard and Evenson (1977)
(39a)	$1.7 \times 10^{-12} \exp (-1400/RT)$	2.0×10^{-15}	Simonaitis and Heicklen (1976)
(39b)	Temperature Independent	4.5×10^{-30}	Howard and Evenson (1977)
(-39b)	$6 \times 10^{17} \exp (-26000/RT) \text{ s}^{-1}$	1.63×10^{-13}	Simonaitis and Heicklen (1977)
	$1.4 \times 10^{14} \exp (-20700/RT) \text{ s}^{-1}$	4.2×10^{-13}	Simonaitis and Heicklen (1977)
	$5.2 \times 10^{-6} \exp (-19,900/RT)$	$1.97 \times 10^{-31} \text{ cm}^6$	Howard and Evenson (1977)
(42)	Temperature Independent	$5.16 \times 10^{-2} \text{ s}^{-1}$	Simonaitis and Heicklen (1977)
(43)	$1 \times 10^8 \exp (-14000/RT) \text{ s}^{-1}$	$9.04 \times 10^{-2} \text{ s}^{-1}$	Graham et al, (1978)
		1.30×10^{-20}	Graham et al, (1978)
		3.3×10^{-12}	Simonaitis and Heicklen (1977)
		$5.4 \times 10^{-3} \text{ s}^{-1}$	Simonaitis and Heicklen (1977)

TABLE 3-21. FIRST-ORDER HO₂NO₂ RATE CONSTANTS AND PREDICTED THERMALLY YIELDED NO₂ AS A FUNCTION OF SAMPLE RESIDENCE TIME (a)

Simonaitis and Heicklen Decay Rate					Graham et al., Decay Rate				
T (K)	k _{-39b} (s ⁻¹)	-39b (s)	t _R (s)	NO ₂ ^(a) (ppb)	k _{-39b} (s ⁻¹)	-39b (s)	t _R (s)	NO ₂ ^(a) (ppb)	
220	8.86x10 ⁹	7.8x10 ⁷	-	-	3.76x10 ⁻⁷	1.84x10 ⁶	-	-	
298	5.16x10 ⁻²	13.43	1	0.10	9.04x10 ⁻²	7.66	1	0.18	
			2	0.20			2	0.13	
			5	0.45			5	0.72	
400	3.80x10 ³	1.8x10 ⁻⁴	1	2	6.75x10 ²	1.03x10 ⁻³	1	2	
			2	2			2	2	
			5	2			5	2	
550	2.79x10 ⁷	2.5x10 ⁻⁸	1	2	8.20x10 ⁵	8.45x10 ⁻⁷	1	2	
			2	2			2	2	
			5	2			5	2	

(a) Assumes an initial HO₂NO₂ of 2 ppb at 25 km (from Jesson et al., 1977).

(a) Assumes an initial HO₂NO₂ of 2 ppb at 25 km (from Jesson et al., 1977).

The proposed NO₂ instrumentation includes a NO chemiluminescence monitor in series with a high temperature catalytic converter (NO₂ channel) along with a second NO chemiluminescence monitor operated in parallel. This instrumentation would permit NO, NO₂, and NO₂-NO measurements. To maintain equivalent flow parameters, e.g. identical residence times, the NO channel would be made equivalent to the NO₂ channel except that a uniform low temperature would be employed. Experimental data for an NO₂ type channel was presented in paragraphs 3.3.1 and 3.5.2. To better understand the sample reaction kinetics, stratospheric simulations of the sample passing through the instrumentation have been modeled and analyzed by the GEARS/EPISODE code. An understanding of these analyses is required to determine the feasibility of an NO₂ or total odd-nitrogen monitor.

Finally, as part of the overall feasibility of NO₂ type of instrumentation, wall effects and the propagation of very low levels of certain species are considered by drawing upon vacuum technology in an attempt to develop a physical description of sorption processes. Simulations are treated first, while a baseline design for the sample handling is treated last.

3.9.1 Stratospheric-Based Instrument Modeling

The instrument will naturally induce changes in the incoming ambient air sample. These changes must be minimized if an accurate determination of the unperturbed atmosphere is to be made. These internal changes depend in part on temperature, pressure, specie concentrations, reaction rates and sample flow rate. The calculations have been carried out assuming inviscid flow of a perfect gas at low Reynolds number for several pressures. The instrumentation has been assumed to be athermalized to pre-selected model values and heterogenous reactions have been assumed to be negligible.

The EPISODE code was employed to analyze NO₂, N₂O₅, HNO₃, C₂ONO₂, and HO₂NO₂ conversions. Criteria for reaction selection was based upon known mechanisms, a reasonable knowledge of the rates, the magnitude of the rate and the reactant number density. Appropriate O₃ reactions were included since $[O_3] \sim 10^{12}$ molecules/cm³.

The reaction set for N_2O_5 thermal decomposition was listed at the beginning of Paragraph 3.5.1 with the exception of the ozone oxidation of NO to NO_2 .

The reaction set for HNO_3 thermal decomposition consists of the Johnston mechanisms listed in Paragraph 3.7.1 for reactions (25) and (26), i.e., HNO_3 in the presence of NO, and the ozone oxidations of NO and NO_2 to NO_2 and NO_3 .

Both reaction sets, 16 reactions in total, have been combined and computer printouts of concentration versus time for a 25 km altitude situation are provided in Appendix B. Instrumentation temperatures of 250, 300, 400, 550, 600, 700, and 800 K have been selected. The first two values correspond to instrumentation temperatures where no active thermal heating would be occurring. The last temperature value corresponds to an upper limit value for the conversion of HNO_3 and nearly complete conversion of NO_2 .

A portion of the profile data of Appendix B has been selected and the gradient of appropriate species is presented in Table 3-22 for an assumed residence time of 5 seconds and initial concentrations representative of 25 km. These initial concentrations are; $\text{NO} = 7.0 \times 10^8$, $\text{NO}_2 = 6.2 \times 10^9$, $\text{N}_2\text{O}_5 = 7.0 \times 10^8$, $\text{HNO}_3 = 3.0 \times 10^9$, $\text{O}_3 = 4.3 \times 10^{12}$ and $\text{H}_2\text{O} = 6.0 \times 10^{12}$ molecules/ cm^3 . At reduced temperatures, e.g., 250 and 300 K, where one is interested in an $[\text{NO}]$ determination, NO is oxidized by O_3 forming NO_2 . The loss or measurement bias can be determined with a knowledge of the ambient $[\text{O}_3]$, the measurement of $[\text{NO}]$, instrumentation temperature and sample flow rate. At the intermediate temperatures tabulated, where conversion of HNO_3 is only partial, the gradient ratio given in the last column shows significant departure from unity. At the higher temperatures tabulated, the ratio converges to unity. The small departure from unity is attributable to a small increase in the $[\text{NO}_3]$ and represents a measurement bias error since NO_3 is not measured.

TABLE 3-22. NET CHANGE OF SPECIES (molecules/cm³) VS TEMPERATURE FOR RESIDENCE TIME OF 5 SECONDS

T(K)	ΔNO	ΔNO_2	$\Delta 2\text{N}_2\text{O}_5 + \text{HNO}_3$	$-\Delta \text{NO}_2 + 2\text{N}_2\text{O}_5 + \text{HNO}_3 / \Delta \text{NO}$
250	-0.090×10^9	$+0.090 \times 10^9$	-0.0	1.000
300	-0.212	+0.219	-0.023×10^9	0.924
400	-0.502	0.119	-1.400	2.551
550	+0.538	+0.042	-1.400	2.524
600	+3.115	-2.487	-1.403	1.249
700	+7.110	-5.554	-1.947	1.054
800	10.18	-6.058	-4.400	1.027

From the data tabulated in this table and various instrumentation parameters cited above, it may be concluded that $[\text{NO}]$ can certainly be determined and that the summation term $[\text{NO}_2 + 2\text{N}_2\text{O}_5 + \text{HNO}_3]$ can be determined at 800 K by differencing the two $[\text{NO}]$ measurements after correction for the NO oxidized during instrument transit time. For short duration flights, the use of dry ice is practical for further reducing the oxidation.

A design analysis or tradeoff study employing the EPISODE code to determine altitude, temperature, and residence time effects was conducted to verify the engineering feasibility of a high temperature NO_z converter. Figure 3-18 presents the parametric results of this analysis where the gradient ratio defined in Table 3-22 is plotted against altitude for temperatures of 800 and 900 K and residence times of 1.0, 2.5, and 5.0 seconds.

The sigmoidal shapes of the parameter curves of Figure 3-18 is an artifact of the representative species concentration profiles of the stratosphere selected for the calculation. No analysis was carried out for grossly perturbed profiles.

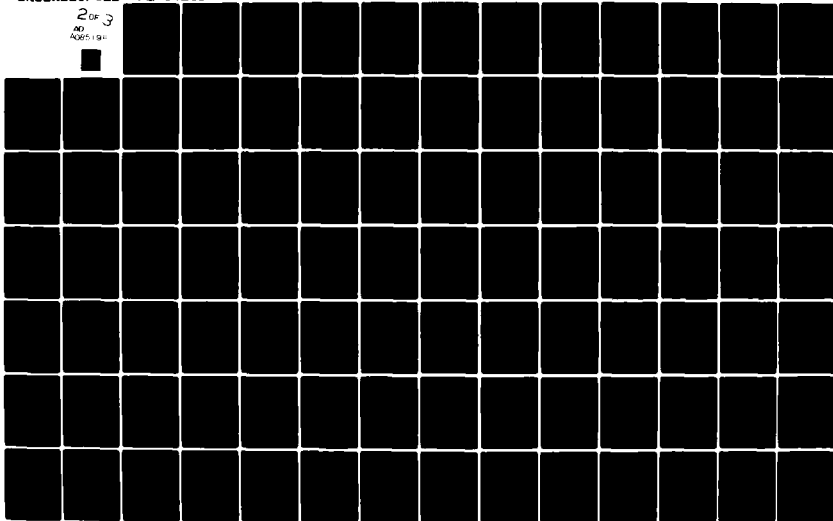
To minimize the NO_3 measurement bias error, the NO_z high temperature converter should be operated at as high a temperature as practical and retain

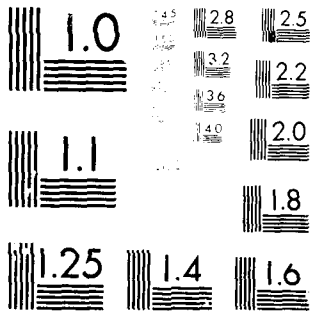
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PERKIN-ELMER CORP NORWALK CONN ELECTRO-OPTICAL DIV F/G 13/2
HIGH ALTITUDE POLLUTION PROGRAM STRATOSPHERIC MEASUREMENT SYSTE--ETC(U)
FEB 80 N H MACOY, R WEINGARTEN, A PIRES DOT-FA77WA-4080
PE-14262 FAA/EE-80-11 NL

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MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

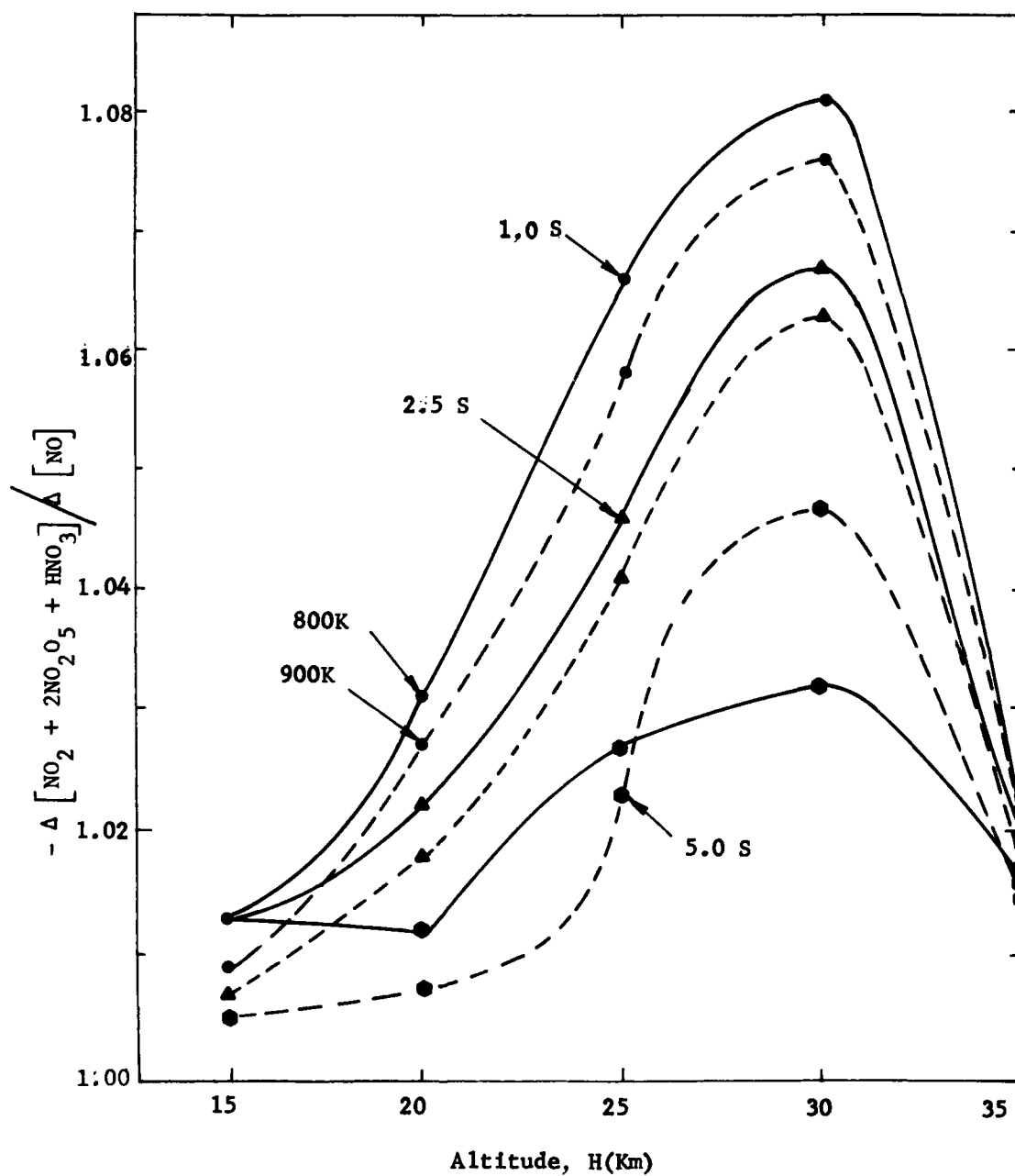
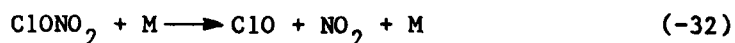


Figure 3-18. NO_3 Measurement Bias Error

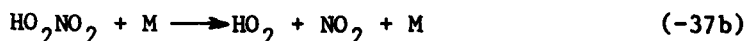
the sample for as long as practical. Based upon a temperature of 800 K and a residence time of 5 seconds, Figure 3-18 suggests that the measurement uncertainty will be less than 4 or 5 percent.

The thermally labile species, ClONO_2 and HO_2NO_2 , were then considered and their decomposition reactions added to the EPISODE code file. The reactions and rates are listed below for reference. The dimensional units are $\text{cm}^3\text{-molecule}^{-1}\text{-s}^{-1}$.



$$k_{-32} = k_{-34a} + 1.32 \times 10^{-5} \exp(-11,980/T) [\text{N}_2]$$

$$k_{-32a} = 2.74 \times 10^6 \exp(-3350/T)$$



$$k_{-37b} = 6 \times 10^{17} \exp(-2600/RT)$$

Computer printouts for the expanded set are included in Appendix B for temperatures of 250, 300, 700 and 800 K.

3.9.2 Heterogenous Sample Changes

Internal sample changes will also result if adsorption-desorption phenomena are occurring. Molecular adsorption takes place until a condition of equilibrium with desorption is realized. The following treatment, based on both kinetic theory and activation energy, is adopted from the work of Santeler et al. (1966). For the major species of the stratosphere, H_2O will be of most concern. For water with a molecular weight of 18, the surface concentration (ideal monolayer) S_m is, from kinetic theory,

$$S_m(300 \text{ K}, 18) = 3.7 \times 10^{-5} \text{ torr-liter/cm}^2 = 1.2 \times 10^{13} \text{ molecules/cm}^2$$

$$S_m(800 \text{ K}, 18) = 6.0 \times 10^{-5} \text{ torr-liter/cm}^2 = 2.0 \times 10^{13} \text{ molecules/cm}^2$$

for temperatures of 300 and 800 K, respectively. At a pressure of 0.1 atm, the time needed to form a monolayer is about 33 nS.

Applying the activation energy concept for residence time gives

$$t_r = t_0 \exp (\Delta E/RT)$$

where t_0 = the vibrational period of the lattice, 10^{-13} s at 300 K

E = characteristic energy, 24,000 cal/gm-mole

R = gas constant, 1.986 cal/deg-mole

T = temperature, 300 or 800 K

Therefore, $t_r = 3 \times 10^4$ and 3.6×10^{-7} seconds for the low and high temperatures. Thus one can expect that the room temperature NO instrumentation ahead of the gas phase titration region will scavenge N_2O_5 and possibly HNO_3 from the sample stream. Conversely, the 800 K NO_z instrumentation ahead of the gas phase titration region will not scavenge either N_2O_5 or HNO_3 from the sample stream. A recommended preflight task, however, is a thermal vacuum cycling of the high temperature converter.

Finally, it is recommended that all materials exposed to the sample during handling be restricted to glass; quartz; corrosion resistant steels, such as hastelloy, monel, inconel, nickel, gold, or Durimet 20TM; or the austenitic steels, such as stainless steel 316 (Grade CF8M) for general use or stainless steel 347 (Grade CF8C) should welding be required; and finally polytetrafluoroethylene (PTFE). Glasses and metals should be scrupulously cleaned in benzol and distilled acetone while the polymer materials should be used in the as-processed condition. The use of polymer materials is a controversial subject. In studying part per trillion levels of NO, NOAA scientists have found that, at that level, NO may or may not interact with PTFE, depending upon the ambient humidity (MacFarland, private communication, 1979). Reactivity is reduced to acceptable levels when humidity is low.

3.9.3 NO_z Baseline Design

This section considers the sample handling design for an NO_z converter and NO monitor. A discussion of payload consumables, signal processing, and many other engineering details are therefore missing.

The basic design includes three reaction volumes, sample ingress and egress ports, and reaction volume interconnections. The three reactions volumes are defined to be: (1) the high temperature NO_2 converter, (2) an instrument zeroing or background volume where O_3 is injected periodically to consume ambient NO, so as to establish a zero level signal, and (3) a chemiluminescence volume where gas phase titration of O_3 and NO occur. From the data of Figure 3-18, a baseline flow rate of 1 SLPS is selected. This rate would be derived from a servo controlled motor operating a high efficiency-zero-head lobe pump such as employed by NOAA (Drummond, private communication, 1979). This method of sample transport has two immediate advantages: (1) the flow rate is independent of ambient pressure and (2) once the pump and motor are characterized, an on-board flow meter is not required.

A general description follows. Upon sample entry to the instrumentation, the sample is exposed to a high temperature catalytic furnace or reaction vessel where NO_2 is converted to NO. For monitoring of only NO, the sample passes through a similar and parallel vessel whose temperature is maintained below 300 K. Each channel contains a pre-reaction vessel where O_3 may be periodically injected to generate a zero-reference level. Provision for injecting known levels of NO and NO_2 are also included for calibration purposes.

The design layout for the NO_2 converter/NO monitor is shown in Figure 3-19. The NO monitor reaction vessel design closely approximates the design used by NOAA for tropospheric sensing of NO. The design has been altered for stratospheric sensing.

The NO_2 moderate temperature reaction vessel is spherically shaped to maximize the volume to surface ratio. The volume, 4070 cm^3 , was selected to provide a residence time of about 5 seconds, actually 4 seconds at a flow rate of 1 SLPS. A platinum-rhodium heating filament on an alumina support and sample ingress-egress deflection baffles of stainless steel material will be located within the sphere. The platinum filament serves two functions. First, as catalyst, it promotes the conversion of NO_2 to NO as discussed in paragraph 3.3.1. Secondly, it tends to scavenge atomic oxygen by surface

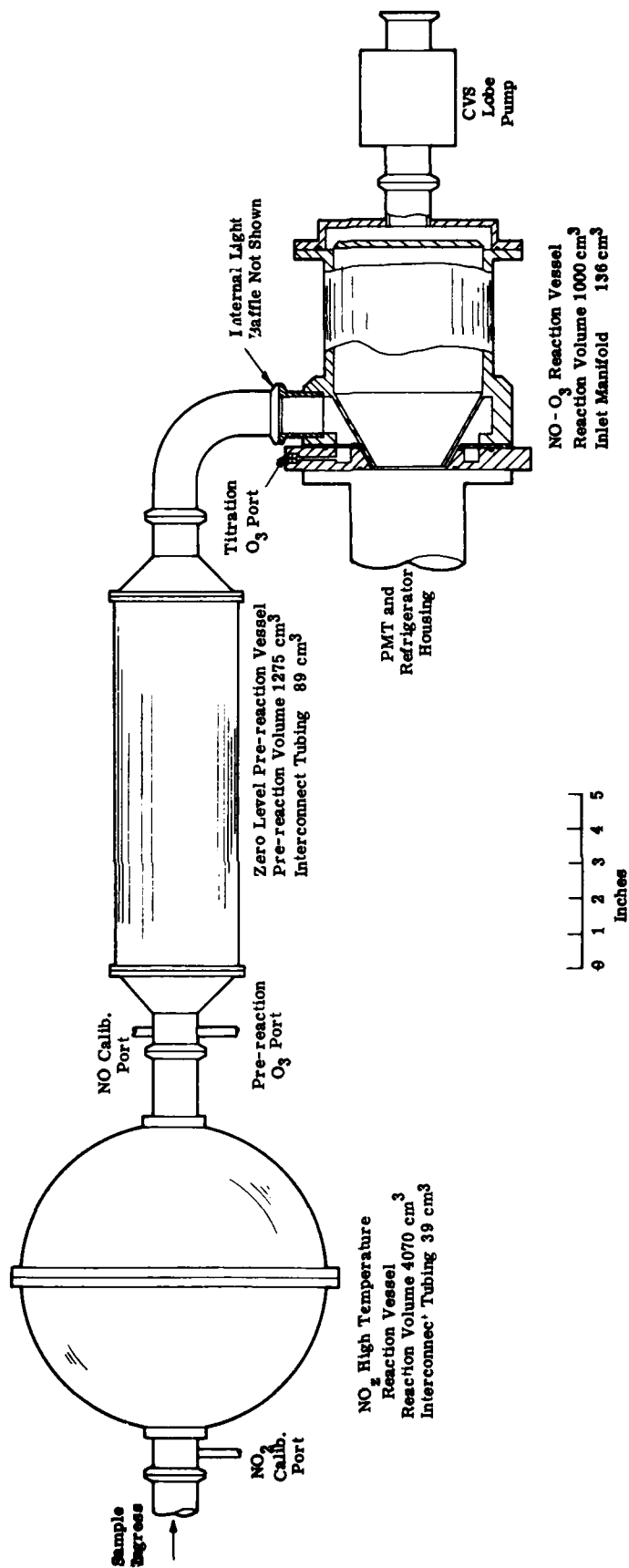


Figure 3-19. NO₂ Converter - NO Sensor Flow Layout

recombination produced by the thermal decomposition of O_3 . To minimize conductive heat losses, the sample ingress and egress flanges and tubing would be constructed of gold plated, thin walled, commercially pure titanium since titanium has a low heat conductivity.

The design details for the high temperature NO_2 catalytic converter used in the laboratory as well as the moderate temperature NO_2 catalytic converter are given in Table 3-23.

The function of the zero level pre-reaction vessel permits chemiluminescence to occur before the sample reaches the photomultiplier tube, thereby allowing the PMT counter to accumulate a base level dark count. This count is then used as the zero reference level. The ozone is injected 1500 cm^3 ahead of the titration ozone manifold. At a sample flow rate of 0.25 SLPS, the residence time is 6 seconds. Using expressions 3-6 and 3-7, and the O_3 parameters of paragraph 3.1.1, yields

$$\frac{\Delta[NO]}{[NO]} = 1 - \exp(-4.9) = 0.983$$

at an altitude of 25 km so that all but 0.7 percent of the NO is consumed ahead of the $NO-O_3$ reaction vessel. At reduced altitudes the reaction is more complete.

The $NO-O_3$ reaction vessel contains an ozone inlet manifold and a sample inlet manifold. Mixing is initiated concentrically about the axis of the vessel near the PMT end of the vessel. At the opposite end of the vessel, a gold coated mirror is placed in conjunction with the gold plated cylindrical vessel to form a photon integrating cavity. A sample pumping port and lobe motor pump are located behind the mirror. For a sample residence time of 4 seconds, the data of Table 3-1 predicts total conversion to NO_2 of which about 7.2 percent is converted to the optically active state 2B_1 .

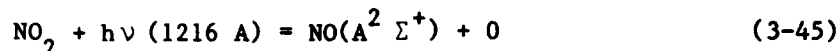
3.10 POTENTIAL OF A NEW TECHNIQUE FOR NO_2 MEASUREMENTS

J.G. Anderson of Harvard University (private communication, 1979) is making laboratory measurements on a new technique for the measurement of NO_2 .

TABLE 3-23. HIGH TEMPERATURE REACTION
VESSEL DESIGN COMPARISON

Parameter	Aerochem NO ₂ Converter	Perkin-Elmer Converter	Relative Factor
Flow Rate	0.035 SLPS	1.0 SLPS	28.6
Temperature	~ 1373 K	~ 800-900 K	-
Ambient pressure	1000 mb	6 to 120 mb	-
Filament material	Pt 10% Rh	Pt 10% Rh	-
Filament casing	McDanel Refractory 99% alumina	Omega Engineering 99% alumina	-
Bonding material	Aremco Ultrabond 552	Sodium Silicate cement or Ultra- bond 552	-
Filament size	Avg No.30 (0.010")	Avg No. 24 (0.020")	-
Filament length	120 inches	600 inches	5.0
Contiguous sample volume	1.14 cm ³	26 cm ³	22.7
Contiguous sample residence time	33 mS	26 mS	0.8
Contiguous filament surface area	27.9 cm ²	235 cm ²	8.4
Heat equivalent power	14 watts	11 watts @ 50 mb	-

This new technique is very similar to the technique recently used for water vapor measurements in the stratosphere (Kley and Stone, 1978) and was suggested in the same report. The technique consists of the photodissociation of NO₂ using Lyman alpha (1216 A) light and the detection of the characteristic gamma-band fluorescence of the excited NO produced.



The detected signal in cts/sec-molecule/cc is proportional to the fluorescence intensity multiplied by optical efficiencies and by geometrical factors.

The fluorescence intensity can be expressed by

$$I = \frac{[\text{NO}_2] \cdot J \cdot \phi \cdot A}{A + \text{air } K_q^{\text{air}}} \quad (3-47)$$

The photodissociation process is described by ϕ , the quantum yield, and by J , the photodissociation coefficient. This coefficient is related to the light source photon flux ψ and the absorption cross section, σ_λ ,

$$J = \psi_\lambda \sigma_\lambda \quad (3-48)$$

The fluorescence process is described by A , the transition probability per second, and k_q^{air} , the quenching rate coefficient for air.

Preliminary results of Anderson indicate that the sensitivity of the technique is 5×10^{-8} cts/sec-molecule/cc or about 10^8 molecules/cc of NO_2 at a S/N of 1. His results indicate a quantum yield smaller than 1% for all gamma-bands. This photoyield is smaller than that of H_2O for $\text{OH}(\text{A}^2 \Sigma^+)$ and smaller than anticipated (Kley and Stone, 1978). Anderson is considering means to improve the sensitivity, and Kley (private communication, 1979) will also make laboratory measurements. These means are, to select a better photolysis wavelength where either the cross section or quantum yield is larger or to utilize a more intense source according to 3-47 and 3-48. The present sensitivity is adequate to meet the HAPP requirements.

Several specificity issues remain to be studied, but are not expected to affect the NO_2 measurements. The photolyzing radiation will act on all stratospheric species and could produce interfering fluorescence or secondary fluorescence. The low quantum yields mean that secondary fluorescence (i.e., NO_2 fluorescence from an NO_2 product of another dissociation) will be very

small. One does not expect interfering fluorescence from other product species. On the contrary, these product species may themselves be measured at the same time using other spectral channels. Complete identifications of their parents may, however, require use of several excitation wavelengths (Kley and Stone, 1978). Finally, the need to view as much fluorescence from excited NO as possible could cause acceptance of scattered sunlight.

The demonstration of this dissociation/fluorescence technique for NO₂ would also lead to a more specific technique for NO. One would convert the NO to NO₂ using titration with O₃ as is done in the chemiluminescence technique for NO. Its drawback is that NO₂ concentrations are usually larger than NO and so one would lack sensitivity for NO concentrations much lower than NO₂ concentrations (e.g., during sunrise, sunset, and night).

The new technique for NO₂ promises both sensitive and specific measurements of NO₂ in a lighter and more compact module than the broadband dissociation module. If all these potentials are demonstrated, the dissociation/fluorescence technique should replace the broadband technique in the Hybrid Gas Conversion System. Flight demonstration may be carried out in 1979 by Anderson in his configuration of a fast-flow, parachute drop payload. This configuration is not essential to the technique, which is also applicable to the slow-flow sampling envisioned for the HAPP Stratospheric Measurement System. Slow-flow sampling and calibration techniques have been demonstrated for NO in the stratosphere (B.A. Ridley, et al., 1972).

SECTION IV

CONCLUSIONS AND RECOMMENDATIONS

This section is devoted to conclusions drawn from the experimental work as well as the analytical work undertaken during this program. A summary of this work is given in Section II.

Recommendations are also presented in the context that only hybrid chemical conversion techniques were reported upon. The validity of the hybrid technique for carrying out the scientific and engineering objectives of the HAPP Stratospheric Measurement System is paramount. The merits of only this system are considered in developing a recommended position. Where appropriate, further areas of investigation, again only for this type of system, are recommended.

4.1 CONCLUSIONS

Conclusions for the laboratory findings associated with molecular chemical conversion techniques are arranged in the same order as the evaluation of the technique appeared in Section III.

Measurement of stratospheric NO is basically carried out by gas phase titration with O_3 immediately followed by the sensing of the chemiluminescence of the product molecule, NO_2^* . With laboratory instrumentation used in this study, a sensitivity of 1.6×10^{10} molecules/cm³ could be routinely measured. With stratospheric instrumentation, a lowest detection limit of 5×10^7 molecules/cm³ has been verified (B.A. Ridley, et al., 1972).

Conversion of NO_2 to NO and $O(^3P)$ by the photolytic process is complex and may result in substantial changes of the relative concentrations of other species in the sample stream. These changes in turn can bias the readings. In addition, dissipated heat from the photolytic lamp could assist in the

premature thermal decomposition of N_2O_5 , and certainly ClONO_2 and HO_2NO_2 , the latter two species being particularly labile and believed to be present in the stratosphere. Further, as shown in the Feasibility Study Report, this method requires substantial amounts of electrical power derived from light weight primary cells or heavier secondary cells. Reduction of the premature decomposition of N_2O_5 , ClONO_2 and HO_2NO_2 to negligible levels would require that the photolytic cell be packed in dry ice. A lowest detection limit of about 1×10^8 molecules/cm³ is achievable but depends upon engineering trades relating electrical power and radiated flux.

Concerning the measurement of N_2O_5 the conclusions may be grouped into categories of (1) generation and handling of N_2O_5 in known quantities and (2) conversion of N_2O_5 into product molecules which can be quantitatively measured. Firstly, ppm quantities of N_2O_5 can be easily generated, the process followed by IR absorption techniques followed in turn by stoichiometric and rate limiting evaluations. Handling of N_2O_5 is more difficult because water tends to combine, thus forming HNO_3 . This artifact of the generation and handling of N_2O_5 can be routinely evaluated by monitoring the IR signature of HNO_3 . Secondly, thermal conversion of N_2O_5 is complete at temperatures in the 450-473 K range. Obtaining a measurable decomposition product quantitatively, however, is not straightforward. For example, computer analysis shows that at an altitude of 15 km, the net change of NO_3 exceeds the net change of N_2O_5 ; while at a altitude of 25 km, the net change of NO_3 is less than the net change of N_2O_5 . Computer analysis and laboratory tests at higher temperatures, up to and including 800 K, indicate that the thermal conversion process becomes substantially more quantitative. At this point on the temperature scale, however, specificity to HNO_3 is lost, leading to the concept of NO_2 instrumentation.

Thermal conversion of HNO_3 is a non-catalytic process, but one that requires a surface. Surface materials employed during this program included Pyrex and stainless steel. Temperature ranges employed were 520-550 K for Pyrex and up to 673 K for stainless steel. Thermal decomposition of HNO_3 was found to be complete at the lower temperature range with the principal

product being NO_2 . A quantitative portion of the NO_2 is thermally converted to NO. Stoichiometry of the conversion to NO_2 prior to allowance for NO_2 thermal decomposition was found to be unity within experimental error. In summary, the conversion of HNO_3 to NO_2 and NO was found to be straightforward for concentrations as high as 1.7×10^{14} molecules/ cm^3 .

Interferents impacting the specificity of thermal conversion instrumentation include ClONO_2 and HO_2NO_2 to the extent that they are present in the stratosphere. Conversion of HO_2NO_2 will be present for temperatures ≥ 300 K. Conversion of ClONO_2 will be present for temperatures ≥ 350 K.

Conversion of total odd-nitrogen, to measurable NO, without regard to specificity, can be carried out using the high temperature catalytic thermal technique. The stoichiometry of the various molecules, HNO_3 , N_2O_5 , ClONO_2 , HO_2NO_2 and NO_2 is well understood. Passage of the sample through a converter operated at 500-800 K would decompose the heavier molecules, including NO_2 , at stratospheric pressures. The resulting product, NO, would then be measured by the GPT-chemiluminescence method. Measurement of only NO would be carried out in an identical, parallel path but without thermal conversion.

The narrow-band photolysis/fluorescence technique for NO_2 is being investigated by J. Anderson in the laboratory. His best result thus far indicates a sensitivity of about 10^8 molecules/cc, which meets the HAPP NO_2 specification. Specificity is not expected to be a problem. Conversion of NO to NO_2 by titration with ozone may allow the same technique to measure NO to the same precision. Anderson tentatively plans a demonstration flight during 1979 in his unique parachute package. The same technique is applicable to the slow-flow sampling envisioned for the HAPP Stratospheric Measurement System. In both cases, on-board calibration is advisable.

The GEARS/EPISODE computer code was found to be a valuable analytic modeling tool for developing laboratory apparatus as well as for developing design and performance parameters for instrumentation intended for stratospheric conditions. The modeling results for both laboratory conditions and those

for expected concentrations of the stratosphere indicate, with only minor differences, identical conversion trending. This permits valid extrapolation of the laboratory results. The code provides a rigorous analysis that will also benefit pre-launch and in-flight calibration tasks.

Sample fidelity, transport, radical chemistry and surface reactions which are not amenable to modeling must be given particular attention in the instrumentation design, use and interpretation of data.

4.2 RECOMMENDATIONS

From the results of the Feasibility Study and the results of the laboratory performance studies reported here, Perkin-Elmer recommends the development of a flight prototype of the Hybrid Gas Conversion Measurement System. It is recommended that the flight prototype consist of a number of instrumentation modules that can be improved, changed, or added to achieve the scientific objectives of the High Altitude Pollution Program. Candidate modules are the total odd-nitrogen (including NO) module, the NO_2/NO module, the O_3 module, and the N_2O module. Measurements with these modules would meet all requirements on the Stratospheric Measurement System except the partitioning of the heavier odd-nitrogen species. Each of these modules is considered to be feasible from an engineering point of view. It may be possible to add a partitioning module at a later time, depending on the technology of stratospheric measurements. In no case, however, would a module be flown before testing in the laboratory under stratospheric conditions.

Perkin-Elmer recommends that the development of a flight prototype of the Total Odd-Nitrogen Instrumentation Module be initiated. Functional design would be based upon chemiluminescence detection for NO and high-temperature catalytic conversion for the HNO_3 , N_2O_5 , NO_2 , and other odd-nitrogen species. As documented in this report, the measurement technique is demonstrated under laboratory conditions, no new technology is being advanced, and none of the module components are state-of-the-art. This development should be carried out jointly by the system contractor and a stratospheric chemistry laboratory group.

Perkin-Elmer further recommends that the development of a flight prototype NO₂/NO Instrumentation Module be initiated. This module is essential for obtaining the important measurements of NO₂. Functional design would be based upon chemiluminescence detection of NO and the narrowband photolysis fluorescence technique for NO₂ if that technique is demonstrated in the stratosphere as expected. In this case, a balloon version of the latter technique would have to be developed. If the narrow-band photolysis techniques does not work as expected, the broad-band photolysis technique should be further optimized for NO₂ detection as indicated in this report.

Perkin-Elmer also recommends that two other Instrument Modules be considered as part of the flight prototype at this time. These modules are the flyable gas chromatograph for N₂O and the UV photometer for O₃. Both modules are in states of advanced development and testing. The flyable gas chromatograph is being developed by Valco Instruments and Baseline Instruments and is being tested by NOAA in Boulder. The flyable UV photometer is made by DASIBI Environmental Corporation and has been tested by NASA Johnson Space Center.

Perkin-Elmer finally recommends that provision be included in a flight prototype for one or two other instrumentation modules. The volume set aside could be utilized for batteries, permitting longer flights until additional candidate modules are developed.

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APPENDIX A

INFRARED ABSORPTION MEASUREMENTS

A.1 N₂O₅ AND HNO₃ CONCENTRATION DETERMINATION

Prior to evaluating a particular analytical method for a species, as well as following the course of a reaction, IR band spectra were recorded to obtain the quantitative concentration of the species.

The analytical method (Lambert-Beer's Law) was employed using band model absorption cross sections. The method utilized either a Perkin-Elmer Model 521 or Model 580 spectrophotometer equipped with an ambient temperature Foxboro/Wilks 20-meter cell lined with PTFE. Because of the corrosive nature of HNO₃, the KBr windows usually used were replaced with AgCl windows. In all cases, the sample and diluent carrier were allowed to flow freely through the cell. Measurements were carried out at ambient pressure, 747-755 torr and temperature, 296-298 K.

The number density, N, or partial pressure p of the species was calculated on the basis of the Lambert-Beer Law:

$$N = \frac{\ln\left(\frac{I_0}{\sigma I}\right)}{\sigma L} \quad (\text{molecules/cm}^3) \text{ and}$$

$$p = \frac{RT \ln\left(\frac{I_0}{I}\right)}{N_{Av} L} \quad (\text{torr})$$

where: $\ln\left(\frac{I_0}{I}\right)$ corresponds to the base e absorbance

σ = effective cross section in cm²/molecule

L = pathlength in cm

$R = \text{gas constant} = 62366 \text{ torr-cm}^3/\text{mole-K}$

$T = \text{temperature of the sample during measurement}$

$N_{Av} = \text{Avogadro's number} = 6.023 \times 10^{23} \text{ molecules/mole}$

N_2O_5 and HNO_3 have similar molecular structures, as shown in Figure A-1 (Hisatune et al., 1962 and McGraw et al., 1965). Therefore, low resolution spectra may occur at the same frequencies. The ν_1 asymmetric N-O stretch, A_1 symmetry, and ν_{11} asymmetric N-O stretch, B_2 symmetry, modes of N_2O_5 are shown schematically. These modes are observed at a frequency of about 1728 cm^{-1} . The ν_2 asymmetric N-O stretch mode (R-branch) of HNO_3 is observed at about 1720 cm^{-1} .

If both species are present, then the ν_2 band for HNO_3 cannot be used for quantitative measurements without other spectral information*. In cases where two materials are present in a sample mixture, each absorbing at the same frequency or frequencies, the mixture can be analyzed for the concentration of each component by determining the absorbance for the mixture at two or more frequencies.

$$A_{e,\nu} = \sigma_{A,\nu} N_A L + \sigma_{B,\nu} N_B L$$

where $A_{e,\nu}$ denotes absorbance to the base e at frequency ν and the subscripts A,B denote N_2O_5 and HNO_3 . The cross section, σ , as obtained from the data of Nightingale et al., (1954); Goldman et al., (1971); Goldman et al., (1975); and Graham (1975) are tabulated in Table A-1.

* With the synthesis and handling of N_2O_5 the possibility exists for heterogeneous wall reactions if adsorbed water is present. The product of the reaction is HNO_3 since H_2O_5 is the anhydride of HNO_3 .

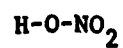
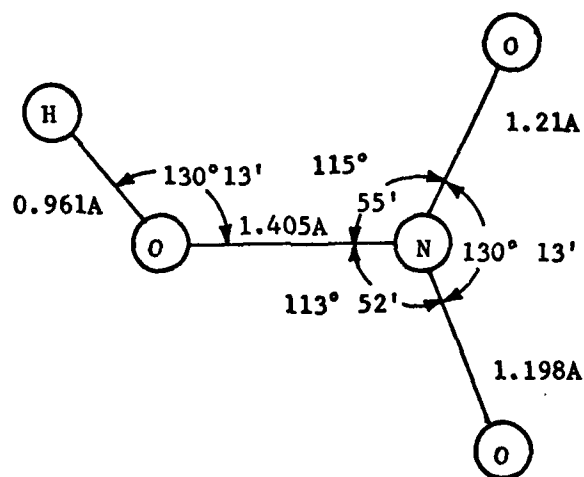
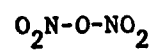
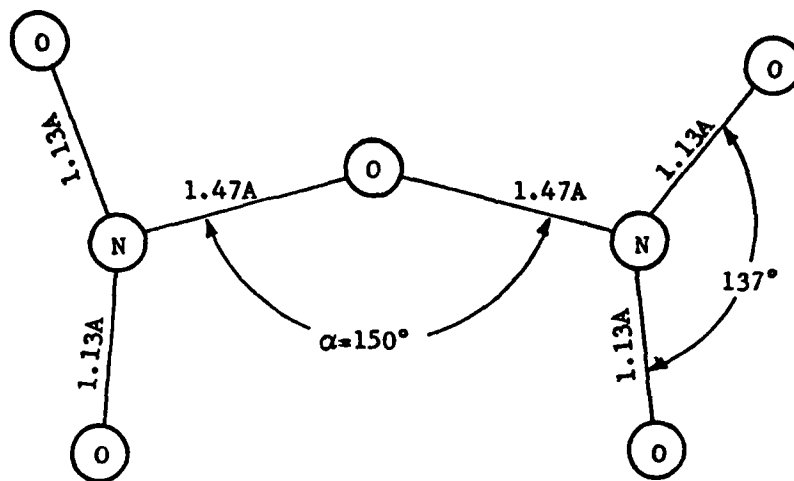


Figure A-1. Molecular Structures for Nitrous Pentoxide and Nitric Acid

TABLE A-1. INFRARED ABSORPTION BAND CROSS SECTIONS FOR
N₂O₅ AND HNO₃ (cm²/molecule)

	Frequency (cm ⁻¹)			
	1728	1340	1315	1246
A = N ₂ O ₅	2.4 x 10 ⁻¹⁸	0.15 x 10 ¹⁸	0.02 x 10 ¹⁸	1.75 x 10 ⁻¹⁸
B = HNO ₃	0.82	0.76	0.98	0.00

From the general expression for absorbance and the data of Table A-1, one may derive the working expressions for number densities as:

$$N_A = \frac{A_{1728} - 1.08A_{1340}}{2.28 \times 10^{-18}L} = \frac{1.19A_{1728} - A_{1315}}{2.85 \times 10^{-18}L} = \frac{A_{1246}}{1.75 \times 10^{-18}L}$$

$$N_B = \frac{A_{1728} - 1.37A_{1246}}{0.82 \times 10^{-18}L} = \frac{A_{1340} - 0.086A_{1246}}{0.76 \times 10^{-18}L} = \frac{A_{1315} - 0.011A_{1246}}{0.98 \times 10^{-18}L}$$

A.2 BAND MODEL CROSS SECTIONS

For reference purposes the band model cross sections for the various gases considered during the program are listed. This data was utilized when low resolution IR spectroscopy was employed as a laboratory analytical method for monitoring reactions and concentrations. The spectral resolution of the instrument is also provided. In all cases the data presented corresponds to room temperature data. Where possible, the most accepted data is employed. A summary chart of the medium strength and stronger bands is presented in Figure A-2.

A.2.1 Nitric Oxide

The data presented for the ν_1 band was taken from laboratory measurements utilizing a Perkin-Elmer model 580 spectrometer set at resolutions of 2.8 and 1.0 cm⁻¹. The 1.0 cm⁻¹ data corresponds to the envelope of the resolved lines of the P and R branches. The NO gas at 6 torr was pressure broadened by N₂ with a total pressure of 760 torr. A path length of 8.25 meters was employed.

ν (cm ⁻¹)	σ (cm ² /molecule) $\Delta\nu = 2.8$ cm ⁻¹	ν (cm ⁻¹)	σ (cm ² /molecule) $\Delta\nu = 1.0$ cm ⁻¹
1950	1.30 x 10 ⁻²¹	1950	0.44 x 10 ⁻²¹
1945	2.22	1945	0.94
1940	3.23	1940	1.55
1935	4.24	1935	2.23
1930	5.41	1930	3.06
1925	6.78	1930	4.17
1920	7.49	1920	4.91
1915	8.03	1915	6.17
1910	8.17	1910	6.95
1905	8.32	1905	7.29
1900	7.63	1900	7.13
1895	7.46	1895	6.28
1890	6.58	1890	5.25
1885	4.21	1885	3.81
1880	1.99	1880	0.04
1876(Q)	7.53	1876(Q)	10.18
1870	3.81	1870	0.49
1865	7.04	1865	5.44
1860	8.68	1860	6.61
1855	6.47	1855	6.72
1850	7.55	1850	6.01
1845	10.3	1845	6.01
1840	7.07	1840	4.49
1835	6.83	1835	4.04
1830	6.41	1830	3.97
1825	6.02	1825	4.22
1820	4.46	1820	2.60
1815	3.48	1815	2.12
1810	2.51	1810	1.60
1805	2.51	1805	1.96
1800	2.27	1800	1.44

A.2.2 Nitrogen Dioxide

The data presented for the ν_3 band was taken from Goldman et al., (1975), and corresponds to a resolution of about 10 cm^{-1} .

$\nu \text{ (cm}^{-1}\text{)}$	$S^\circ/d \text{ (cm}^{-1}\text{-atm}^{-1}\text{)}$	$\sigma \text{ (cm}^2\text{/molecule)}$
1650	3.0	0.128×10^{-18}
1645	7.7	0.330
1640	18.0	0.768
1635	25.0	1.066
1630	31.5	1.343
1625	27.0	1.151
1620	17.5	0.746
1615	15.4	0.659
1610	19.0	0.811
1605	23.0	0.981
1600	25.0	1.066
1595	23.0	0.984
1590	18.0	0.768
1585	12.5	0.533
1580	8.7	0.373
1575	5.7	0.245
1570	3.5	0.149

A.2.3 Nitrogen Trioxide Radical

Nitrogen Trioxide is a free radical intermediate in the $\text{N}_2\text{O}_5 - \text{O}_3$ system and was also observed in room temperature decomposition of ClONO_2 . The data presented for the 1360 cm^{-1} band was extracted from Cramarossa and Johnston (1965), and corresponds to a resolution of about 3 cm^{-1} .

$\nu \text{ (cm}^{-1}\text{)}$	$\sigma \text{ (cm}^2\text{/molecule)}$
1410	0.029×10^{-18}
1400	0.038
1390	0.054

ν (cm ⁻¹)	σ (cm ² /molecule)
1380	0.072
1370	0.098
1360	0.150
1350	0.088
1340	0.060
1330	0.040
1320	0.022
1310	0.016

A.2.4 Nitrous Pentoxide

The data presented for the ν_1 band was extracted from Nightingale et al., (1954) and the ν_2 cross-section data given below. The data presented for the ν_2 and ν_{12} bands was extracted from Graham (1975) and corresponds to a resolution of 5.10 cm⁻¹.

ν (cm ⁻¹)	σ (cm ² /molecule)	ν (cm ⁻¹)	σ (cm ² /molecule)	ν (cm ⁻¹)	σ (cm ² /molecule)
1780	0.338 x 10 ⁻¹⁸	1370	0.065 x 10 ⁻¹⁸	1260	0.50 x 10 ⁻¹⁸
1770	0.595	1365	0.080	1255	1.30
1760	0.910	1360	0.097	1250	1.52
1750	1.295	1355	0.113	1245	1.73
1740	1.824	1350	0.127	1240	1.58
1730	2.380	1345	0.142	1235	1.01
1720	2.169	1340	0.149	1230	0.24
1710	1.546	1335	0.135		
1700	1.082	1330	0.111		
1690	0.732	1325	0.089	1252	1.55
1680	0.393	1320	0.068	1246	1.75

A.2.6 Chlorine Nitrate

The data presented for the ν_1 and ν_2 bands was taken from Graham et al., (1977), and corresponds to a resolution of about 0.0625 cm^{-1} . The cross-section values are independent of resolution and pressure.

$\nu \text{ (cm}^{-1}\text{)}$	$\sigma \text{ (cm}^2\text{/molecule)}$	$\nu \text{ (cm}^{-1}\text{)}$	$\sigma \text{ (cm}^2\text{/molecule)}$
1760	0.186×10^{-18}	1310	0.186×10^{-18}
1755	0.335	1305	0.595
1750	0.744	1300	1.116
1745	1.339	1295	0.476
1740	0.893	1290	0.707
1735	1.041	1285	0.986
1730	1.320	1280	0.558
1725	1.004	1275	0.186
1720	0.707	1270	0.074
1715	0.446		
1710	0.242		
1731(P)	1.361	1286(P)	1.004
1738(Q)	1.153	1292(Q)	1.897
1744(R)	1.413	1300(R)	1.116

A.2.7 Ozone

The data presented for the ν_3 band was extracted from Pitts et al., (1976) and corresponds to a resolution of about 1.3 cm^{-1} .

$\nu \text{ (cm}^{-1}\text{)}$	$\sigma \text{ (cm}^2\text{/molecule)}$
1065	0.0614×10^{-18}
1060	0.1235
1055	0.1717

A.2.5 Nitric Acid

The data presented for the ν_2 and ν_3 bands was taken from Goldman et al., (1971), and corresponds to a resolution of $\sim 0.5 \text{ cm}^{-1}$.

$\nu \text{ (cm}^{-1}\text{)}$	$S^\circ/d \text{ (cm}^{-1}\text{-atm}^{-1}\text{)}$	$\sigma \text{ (cm}^2\text{/molecule)}$
1735	9.420	0.4017×10^{-18}
1730	17.68	0.7539
1725	24.79	1.057
1720	25.84	1.102
1715	26.99	1.151
1710	21.09	0.899
1705	23.22	0.990
1700	23.58	1.005
1695	23.31	0.994
1690	19.83	0.845
1685	14.12	0.602
1680	8.803	0.375
1350	8.716	0.3717
1345	14.87	0.6340
1340	17.87	0.7620
1335	16.42	0.7001
1330	16.82	0.7172
1325	21.32	0.9091
1320	21.68	0.9244
1315	23.09	0.9845
1310	21.51	0.9172
1305	18.00	0.7675
1300	13.74	0.5859
1395	10.51	0.4481
1290	8.759	0.3735

<u>ν (cm⁻¹)</u>	<u>σ (cm²/molecule)</u>
1050	0.1376
1045	0.0530
1040	0.0793
1035	0.1200
1030	0.1306
1025	0.1323
1020	0.1253
1015	0.0956
1010	0.0819
1005	0.0716
1000	0.0530

APPENDIX B

COMPUTER SIMULATION OF NO_z THERMAL CONVERSION IN THE STRATOSPHERE

The computer code was developed to evaluate the performance of ideal thermal converters operating on stratospheric gas samples. The intent was to follow the temporal evolution of the sample to determine the decomposition paths and the significance of the final product abundances. The model included most of the major odd-nitrogen compounds and other important stratospheric constituents. The constituent set was extended until it included chlorine nitrate and chlorine oxide. At that point, the set was fixed because further expansion would have involved the entire family of chlorine compounds and their associated reactions. Such an expansion would have been beyond the scope of this study. Table B-1 contains a list of the molecules considered and the abundances used as the initial conditions for the computer code. In all, nineteen constituents were used together with forty reactions among them. Reactions with unknown products were ignored. Also, an NO₂ photolysis reaction was included but not used for this computation.

The evolution of the abundances was determined using the EPISODE version of the GEARS code. EPISODE is a computer algorithm for the solution of ordinary differential equations. Reaction rates were taken from the literature (principally the review by Hampson and Garvin, 1978). When both forward and backward rates were known, they were used directly. Otherwise, the equilibrium constant was evaluated and used to determine the "missing" rate (cf; Section 3.1).

The output format begins with a listing of the reactions and their evaluated rates at the initial conditions of temperature and pressure. The third body or "M" dependence is included in the reaction rates both as a computational convenience and as an aid to highlighting the significant rates. The temporal evolution of each constituent then follows.

TABLE B-1. MOLECULAR NUMBER DENSITIES USED FOR SAMPLE MODELING WITHIN INSTRUMENTATION

MOLECULE	T(K) Pmb(Torr) H(km)	217 120 mb(91) 15	217 55(42) 20	222 25(19) 25	227 12(9.1) 30	237 6(4.6) 35	251 2.6(2.0) 40
O ₃		2.5 x 10 ¹²	4.5 x 10 ¹²	4.2 x 10 ¹²	2.5 x 10 ¹²	1.3 x 10 ¹²	6.0 x 10 ¹¹
NO		5.0 x 10 ⁹	1.5 x 10 ⁹	7.0 x 10 ⁸	4.5 x 10 ⁸	5.5 x 10 ⁸	7.0 x 10 ⁸
NO ₂		4.5 x 10 ⁹	8.0 x 10 ⁹	6.5 x 10 ⁹	4.2 x 10 ⁹	2.2 x 10 ⁹	7.0 x 10 ⁸
N ₂ O ₅		1.0 x 10 ⁷	4.0 x 10 ⁸	7.0 x 10 ⁸	4.0 x 10 ⁸	5.5 x 10 ⁷	3.5 x 10 ⁶
HNO ₃		1.5 x 10 ⁹	4.0 x 10 ⁹	2.0 x 10 ⁹	4.0 x 10 ⁸	4.9 x 10 ⁷	
ClONO ₂		5.0 x 10 ⁷ (est)	1.0 x 10 ⁸	4.0 x 10 ⁸	2.0 x 10 ⁸	4.0 x 10 ⁷	
HO ₂ NO ₂		2.0 x 10 ⁹	4.0 x 10 ⁹	3.5 x 10 ⁹	1.5 x 10 ⁹	2.3 x 10 ⁸	
HO ₂		1.5 x 10 ⁷	2.0 x 10 ⁷	2.3 x 10 ⁷	2.5 x 10 ⁷	1.7 x 10 ⁷	8.0 x 10 ⁶
NO ₃		2.0 x 10 ⁴	5.0 x 10 ⁵	2.0 x 10 ⁶	3.0 x 10 ⁶	2.0 x 10 ⁶	1.0 x 10 ⁶
O ₂		8.2 x 10 ¹⁷	3.8 x 10 ¹⁷	1.7 x 10 ¹⁷	7.8 x 10 ¹⁶	3.8 x 10 ¹⁶	
HO		2.0 x 10 ⁶	1.0 x 10 ⁶	1.5 x 10 ⁶	3.0 x 10 ⁶	5.0 x 10 ⁶	1.0 x 10 ⁷
H ₂ O		7.3 x 10 ¹²	3.8 x 10 ¹²	1.9 x 10 ¹²	1.2 x 10 ¹²	5.3 x 10 ¹¹	
O		1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	
H		1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	
H ₂		1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	
H ₂ O ₂		8.5 x 10 ⁸	1.3 x 10 ⁹	1.8 x 10 ⁹	1.2 x 10 ⁹	2.8 x 10 ⁸	4.0 x 10 ⁷
HNO		1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	1.0 x 10 ⁶	
HNO ₂		6.0 x 10 ⁶	1.0 x 10 ⁶	8.0 x 10 ⁵	8.0 x 10 ⁵	7.0 x 10 ⁵	6.0 x 10 ⁵
ClO		2.0 x 10 ⁷ (est)	3.5 x 10 ⁷	5.7 x 10 ⁷	4.5 x 10 ⁷	5.0 x 10 ⁷	

T=250 K, H=15 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> NO2 + NO3 + M	1.940-05	6.070-13
2	2*N03 >>> 2*N02 + O2	4.710-17	7.467-62
3	NO2 + NO3 >>> NO2 + NO + O2	4.210-15	1.190-34
4	NO3 + NO >>> 2*N02	1.900-11	3.290-32
5	NO + O3 >>> NO2 + O2	6.360-15	1.080-56
6	NO2 + O3 >>> NO3 + O2	6.650-18	8.330-39
7	HN03 + M >>> HO + NO2 + M	2.580-27	1.180-11
8	HN03 + HO >>> H2O + NO3	8.000-14	2.800-29
9	O + O + M >>> O2 + M	1.290-14	3.310-56
10	O + O2 + M >>> O3 + M	2.770-15	7.760-12
11	O + O3 >>> 2*O2	1.920-15	0.0
12	O + NO + M >>> NO2 + M	5.440-13	4.030-47
13	O + NO2 >>> NO + O2	5.120-12	6.270-53
14	O + NO2 + M >>> NO3 + M	3.390-13	2.720-23
15	HO + HO >>> H2O + O	1.090-12	9.770-27
16	O2 + 2*NO >>> 2*N02	2.750-38	3.880-35
17	NO2 + H-NU >>> NO + O	0.0	0.0
18	O + HO >>> H + O2	4.200-11	7.730-25
19	O + HO2 >>> HO + O2	1.080-11	5.450-60
20	O2 + H + M >>> HO2 + M	1.340-13	1.710-30
21	O3 + H >>> HO + O2	1.270-11	0.0
22	O3 + HO >>> HO2 + O2	2.750-14	1.040-48
23	O3 + HO2 >>> HO + 2*O2	4.450-16	2.410-68
24	H + HO + M >>> H2O + M	2.720-12	0.0
25	H + HO2 >>> 2*HO	9.400-12	1.620-46
26	H + HO2 >>> H2 + O2	1.040-11	2.550-61
27	H + H2O >>> H2 + HO	2.340-28	1.140-15
28	H + H2O2 >>> H2 + HO2	8.360-15	5.620-29
29	H + H2O2 >>> HO + H2O	1.090-14	1.350-75

30	2*H0 + M >>> H2O2 + M	1.550-12	4.170-31
31	H0 + H02 >>> H2O + O2	1.120-11	3.890-74
32	2*H02 >>> H2O2 + O2	2.300-12	8.580-49
33	H02 + H2O >>> H2O2 + H0	1.020-39	5.490-13
34	N0 + H + M >>> HNO + M	8.450-14	1.320-32
35	N0 + H0 >>> N02 + H	3.050-38	3.010-11
36	N0 + H0 + M >>> HNO2 + M	1.110-11	4.210-28
37	N0 + H02 >>> N02 + H0	1.650-13	7.470-21
38	H + H + M >>> H2 + M	3.390-15	0.0
39	HNO4 + M >>> H02 + N02 + M	3.640-05	6.730-13
40	CLN03 + M >>> CLO + N02 + M	1.570-06	6.110-13

TIME (S)	N205	N02	N03	N0	U3	02	HNO3	H0	H20
0.0	1.0000 07	4.5000 09	2.0000 04	5.0000 09	2.5000 12	8.2000 17	1.5000 09	2.0000 06	7.3000 12
0.20	1.0000 07	4.5160 09	3.4990 04	4.9440 09	2.5000 12	8.2000 17	1.5000 09	1.9360 06	7.3000 12
0.40	1.0000 07	4.5320 09	4.9740 04	4.8680 09	2.5000 12	8.2000 17	1.5000 09	1.8750 06	7.3000 12
0.60	1.0000 07	4.5470 09	6.4250 04	4.9520 09	2.5000 12	8.2000 17	1.5000 09	1.8150 06	7.3000 12
0.80	1.0000 07	4.5630 09	7.8550 04	4.9370 09	2.5000 12	8.2000 17	1.5000 09	1.7580 06	7.3000 12
1.00	1.0000 07	4.5790 09	9.2620 04	4.9210 09	2.5000 12	8.2000 17	1.5000 09	1.7020 06	7.3000 12
1.20	1.0000 07	4.5940 09	1.0650 05	4.9050 09	2.5000 12	8.2000 17	1.5000 09	1.6490 06	7.3000 12
1.40	1.0000 07	4.6100 09	1.2010 05	4.8900 09	2.5000 12	8.2000 17	1.5000 09	1.5970 06	7.3000 12
1.60	1.0000 07	4.6250 09	1.3360 05	4.8740 09	2.5000 12	8.2000 17	1.5000 09	1.5470 06	7.3000 12
1.80	1.0000 07	4.6410 09	1.4680 05	4.8590 09	2.5000 12	8.2000 17	1.5000 09	1.4990 06	7.3000 12
2.00	1.0000 07	4.6560 09	1.5990 05	4.8430 09	2.5000 12	8.2000 17	1.5000 09	1.4530 06	7.3000 12
2.20	1.0000 07	4.6710 09	1.7280 05	4.8280 09	2.5000 12	8.2000 17	1.5000 09	1.4080 06	7.3000 12
2.40	1.0000 07	4.6870 09	1.8550 05	4.8130 09	2.5000 12	8.2000 17	1.5000 09	1.3650 06	7.3000 12
2.60	1.0000 07	4.7020 09	1.9800 05	4.7970 09	2.5000 12	8.2000 17	1.5000 09	1.3230 06	7.3000 12
2.80	1.0000 07	4.7170 09	2.1030 05	4.7820 09	2.5000 12	8.2000 17	1.5000 09	1.2830 06	7.3000 12
3.00	1.0000 07	4.7320 09	2.2250 05	4.7670 09	2.5000 12	8.2000 17	1.5000 09	1.2440 06	7.3000 12
3.20	1.0000 07	4.7470 09	2.3450 05	4.7520 09	2.5000 12	8.2000 17	1.5000 09	1.2060 06	7.3000 12
3.40	1.0000 07	4.7620 09	2.4640 05	4.7370 09	2.5000 12	8.2000 17	1.5000 09	1.1700 06	7.3000 12
3.60	1.0000 07	4.7770 09	2.5810 05	4.7220 09	2.5000 12	8.2000 17	1.5000 09	1.1350 06	7.3000 12
3.80	1.0000 07	4.7920 09	2.6960 05	4.7070 09	2.5000 12	8.2000 17	1.5000 09	1.1020 06	7.3000 12
4.00	1.0000 07	4.8070 09	2.8100 05	4.6920 09	2.5000 12	8.2000 17	1.5000 09	1.0690 06	7.3000 12
4.20	1.0000 07	4.8220 09	2.9230 05	4.6770 09	2.5000 12	8.2000 17	1.5000 09	1.0380 06	7.3000 12
4.40	1.0000 07	4.8370 09	3.0340 05	4.6620 09	2.5000 12	8.2000 17	1.5000 09	1.0070 06	7.3000 12
4.60	1.0000 07	4.8520 09	3.1430 05	4.6470 09	2.5000 12	8.2000 17	1.5000 09	9.7820 05	7.3000 12
4.80	1.0000 07	4.8670 09	3.2520 05	4.6320 09	2.5000 12	8.2000 17	1.5000 09	9.4990 05	7.3000 12
5.00	1.0000 07	4.8810 09	3.3590 05	4.6180 09	2.5000 12	8.2000 17	1.5000 09	9.2270 05	7.3000 12
5.20	1.0000 07	4.8960 09	3.4650 05	4.6030 09	2.5000 12	8.2000 17	1.5000 09	8.9640 05	7.3000 12
5.40	1.0000 07	4.9100 09	3.5690 05	4.5880 09	2.5000 12	8.2000 17	1.5000 09	8.7100 05	7.3000 12
5.60	1.0000 07	4.9250 09	3.6730 05	4.5740 09	2.5000 12	8.2000 17	1.5000 09	8.4650 05	7.3000 12
5.80	1.0000 07	4.9390 09	3.7750 05	4.5590 09	2.5000 12	8.2000 17	1.5000 09	8.2290 05	7.3000 12
6.00	1.0000 07	4.9540 09	3.8760 05	4.5450 09	2.5000 12	8.2000 17	1.5000 09	8.0010 05	7.3000 12
6.20	1.0000 07	4.9680 09	3.9760 05	4.5300 09	2.5000 12	8.2000 17	1.5000 09	7.7800 05	7.3000 12
6.40	1.0000 07	4.9830 09	4.0750 05	4.5160 09	2.5000 12	8.2000 17	1.5000 09	7.5680 05	7.3000 12
6.60	1.0000 07	4.9970 09	4.1720 05	4.5010 09	2.5000 12	8.2000 17	1.5000 09	7.3620 05	7.3000 12
6.80	1.0000 07	5.0110 09	4.2690 05	4.4870 09	2.4990 12	8.2000 17	1.5000 09	7.1640 05	7.3000 12
7.00	1.0000 07	5.0250 09	4.3650 05	4.4730 09	2.4990 12	8.2000 17	1.5000 09	6.9730 05	7.3000 12
7.20	1.0000 07	5.0400 09	4.4600 05	4.4590 09	2.4990 12	8.2000 17	1.5000 09	6.7890 05	7.3000 12
7.40	1.0000 07	5.0540 09	4.5530 05	4.4450 09	2.4990 12	8.2000 17	1.5000 09	6.6100 05	7.3000 12
7.60	1.0000 07	5.0680 09	4.6460 05	4.4300 09	2.4990 12	8.2000 17	1.5000 09	6.4380 05	7.3000 12
7.80	1.0000 07	5.0820 09	4.7380 05	4.4160 09	2.4990 12	8.2000 17	1.5010 09	6.2720 05	7.3000 12
8.00	1.0000 07	5.0960 09	4.8290 05	4.4020 09	2.4990 12	8.2000 17	1.5010 09	6.1120 05	7.3000 12
8.20	1.0010 07	5.1100 09	4.9190 05	4.3880 09	2.4990 12	8.2000 17	1.5010 09	5.9570 05	7.3000 12
8.40	1.0010 07	5.1240 09	5.0090 05	4.3740 09	2.4990 12	8.2000 17	1.5010 09	5.8080 05	7.3000 12
8.60	1.0010 07	5.1380 09	5.0970 05	4.3600 09	2.4990 12	8.2000 17	1.5010 09	5.6640 05	7.3000 12
8.80	1.0010 07	5.1510 09	5.1850 05	4.3470 09	2.4990 12	8.2000 17	1.5010 09	5.5250 05	7.3000 12
9.00	1.0010 07	5.1650 09	5.2720 05	4.3330 09	2.4990 12	8.2000 17	1.5010 09	5.3900 05	7.3000 12
9.20	1.0010 07	5.1790 09	5.3580 05	4.3190 09	2.4990 12	8.2000 17	1.5010 09	5.2610 05	7.3000 12
9.40	1.0010 07	5.1930 09	5.4430 05	4.3050 09	2.4990 12	8.2000 17	1.5010 09	5.1350 05	7.3000 12
9.60	1.0010 07	5.2060 09	5.5280 05	4.2920 09	2.4990 12	8.2000 17	1.5010 09	5.0140 05	7.3000 12
9.80	1.0010 07	5.2200 09	5.6120 05	4.2780 09	2.4990 12	8.2000 17	1.5010 09	4.8980 05	7.3000 12
10.00	1.0010 07	5.2330 09	5.6960 05	4.2640 09	2.4990 12	8.2000 17	1.5010 09	4.7850 05	7.3000 12

TIME (S)	C	H	H2	H02	H202	HNO	HN02	HN04	CLN03
0.0	1.0000 06	1.0000 06	1.0000 06	1.5000 07	8.5000 08	1.0000 06	6.0000 06	2.0000 09	5.0000 07
0.20	1.0340-02	2.0150-08	1.0000 06	1.6030 07	8.5000 08	1.0000 06	6.0220 06	2.0000 09	5.0010 07
0.40	1.0230-02	1.9500-08	1.0000 06	1.6050 07	8.5000 08	1.0000 06	6.0430 06	2.0000 09	5.0020 07
0.60	1.0120-02	1.8890-08	1.0000 06	1.6070 07	8.5000 08	1.0000 06	6.0630 06	2.0000 09	5.0030 07
0.80	1.0020-02	1.8290-08	1.0000 06	1.6100 07	8.5000 08	1.0000 06	6.0830 06	2.0000 09	5.0040 07
1.00	9.9330-03	1.7710-08	1.0000 06	1.6120 07	8.5000 08	1.0000 06	6.1020 06	2.0000 09	5.0060 07
1.20	9.8470-03	1.7150-08	1.0000 06	1.6140 07	8.5000 08	1.0000 06	6.1200 06	2.0000 09	5.0070 07
1.40	9.7670-03	1.6620-08	1.0000 06	1.6160 07	8.5000 08	1.0000 06	6.1380 06	2.0000 09	5.0080 07
1.60	9.6920-03	1.6100-08	1.0000 06	1.6180 07	8.5000 08	1.0000 06	6.1550 06	2.0000 09	5.0090 07
1.80	9.6210-03	1.5600-08	1.0000 06	1.6200 07	8.5000 08	1.0000 06	6.1710 06	2.0000 09	5.0100 07
2.00	9.5560-03	1.5120-08	1.0000 06	1.6220 07	8.5000 08	1.0000 06	6.1870 06	2.0000 09	5.0110 07
2.20	9.4940-03	1.4650-08	1.0000 06	1.6240 07	8.5000 08	1.0000 06	6.2020 06	2.0000 09	5.0120 07
2.40	9.4370-03	1.4200-08	1.0000 06	1.6250 07	8.5000 08	1.0000 06	6.2170 06	2.0000 09	5.0130 07
2.60	9.3830-03	1.3770-08	1.0000 06	1.6270 07	8.5000 08	1.0000 06	6.2320 06	2.0000 09	5.0150 07
2.80	9.3330-03	1.3350-08	1.0000 06	1.6280 07	8.5000 08	1.0000 06	6.2460 06	2.0000 09	5.0160 07
3.00	9.2860-03	1.2940-08	1.0000 06	1.6300 07	8.5000 08	1.0000 06	6.2590 06	2.0000 09	5.0170 07
3.20	9.2420-03	1.2550-08	1.0000 06	1.6310 07	8.5000 08	1.0000 06	6.2720 06	2.0000 09	5.0180 07
3.40	9.2010-03	1.2180-08	1.0000 06	1.6330 07	8.5000 08	1.0000 06	6.2840 06	2.0000 09	5.0190 07
3.60	9.1620-03	1.1810-08	1.0000 06	1.6340 07	8.5000 08	1.0000 06	6.2970 06	2.0000 09	5.0200 07
3.80	9.1260-03	1.1460-08	1.0000 06	1.6360 07	8.5000 08	1.0000 06	6.3080 06	2.0000 09	5.0210 07
4.00	9.0920-03	1.1120-08	1.0000 06	1.6370 07	8.5000 08	1.0000 06	6.3200 06	2.0000 09	5.0230 07
4.20	9.0600-03	1.0800-08	1.0000 06	1.6380 07	8.5000 08	1.0000 06	6.3310 06	2.0000 09	5.0240 07
4.40	9.0310-03	1.0480-08	1.0000 06	1.6390 07	8.5000 08	1.0000 06	6.3410 06	2.0000 09	5.0250 07
4.60	9.0030-03	1.0180-08	1.0000 06	1.6400 07	8.5000 08	1.0000 06	6.3510 06	2.0000 09	5.0260 07
4.80	8.9770-03	9.8830-09	1.0000 06	1.6410 07	8.5000 08	1.0000 06	6.3610 06	2.0000 09	5.0270 07
5.00	8.9520-03	9.6000-09	1.0000 06	1.6420 07	8.5000 08	1.0000 06	6.3710 06	2.0000 09	5.0280 07
5.20	8.9290-03	9.3260-09	1.0000 06	1.6430 07	8.5000 08	1.0000 06	6.3800 06	2.0000 09	5.0300 07
5.40	8.9080-03	9.0620-09	1.0000 06	1.6440 07	8.5000 08	1.0000 06	6.3890 06	2.0000 09	5.0310 07
5.60	8.8870-03	8.8070-09	1.0000 06	1.6450 07	8.5000 08	1.0000 06	6.3980 06	2.0000 09	5.0320 07
5.80	8.8680-03	8.5610-09	1.0000 06	1.6460 07	8.5000 08	1.0000 06	6.4060 06	2.0000 09	5.0330 07
6.00	8.8510-03	8.3240-09	1.0000 06	1.6470 07	8.5000 08	1.0000 06	6.4150 06	2.0000 09	5.0340 07
6.20	8.8340-03	8.0950-09	1.0000 06	1.6480 07	8.5000 08	1.0000 06	6.4230 06	2.0000 09	5.0360 07
6.40	8.8180-03	7.8730-09	1.0000 06	1.6490 07	8.5000 08	1.0000 06	6.4300 06	2.0000 09	5.0370 07
6.60	8.8040-03	7.6600-09	1.0000 06	1.6490 07	8.5000 08	1.0000 06	6.4380 06	2.0000 09	5.0380 07
6.80	8.7900-03	7.4540-09	1.0000 06	1.6500 07	8.5000 08	1.0000 06	6.4450 06	2.0000 09	5.0390 07
7.00	8.7770-03	7.2550-09	1.0000 06	1.6510 07	8.5000 08	1.0000 06	6.4520 06	2.0000 09	5.0400 07
7.20	8.7650-03	7.0630-09	1.0000 06	1.6510 07	8.5000 08	1.0000 06	6.4590 06	2.0000 09	5.0420 07
7.40	8.7530-03	6.8770-09	1.0000 06	1.6520 07	8.5000 08	1.0000 06	6.4660 06	2.0000 09	5.0430 07
7.60	8.7420-03	6.6980-09	1.0000 06	1.6530 07	8.5000 08	1.0000 06	6.4720 06	2.0000 09	5.0440 07
7.80	8.7320-03	6.5260-09	1.0000 06	1.6530 07	8.5000 08	1.0000 06	6.4780 06	2.0000 09	5.0450 07
8.00	8.7230-03	6.3590-09	1.0000 06	1.6540 07	8.5000 08	1.0000 06	6.4840 06	2.0000 09	5.0460 07
8.20	8.7140-03	6.1980-09	1.0000 06	1.6540 07	8.5000 08	1.0000 06	6.4900 06	2.0000 09	5.0480 07
8.40	8.7050-03	6.0430-09	1.0000 06	1.6550 07	8.5000 08	1.0000 06	6.4960 06	2.0000 09	5.0490 07
8.60	8.6970-03	5.8930-09	1.0000 06	1.6550 07	8.5000 08	1.0000 06	6.5010 06	2.0000 09	5.0500 07
8.80	8.6900-03	5.7480-09	1.0000 06	1.6560 07	8.5000 08	1.0000 06	6.5070 06	2.0000 09	5.0510 07
9.00	8.6830-03	5.6080-09	1.0000 06	1.6560 07	8.5000 08	1.0000 06	6.5120 06	2.0000 09	5.0520 07
9.20	8.6760-03	5.4730-09	1.0000 06	1.6560 07	8.5000 08	1.0000 06	6.5170 06	2.0000 09	5.0540 07
9.40	8.6700-03	5.3430-09	1.0000 06	1.6570 07	8.5000 08	1.0000 06	6.5220 06	2.0000 09	5.0550 07
9.60	8.6640-03	5.2170-09	1.0000 06	1.6570 07	8.5000 08	1.0000 06	6.5270 06	2.0000 09	5.0560 07
9.80	8.6580-03	5.0950-09	1.0000 06	1.6580 07	8.5000 08	1.0000 06	6.5320 06	2.0000 09	5.0570 07
10.00	8.6530-03	4.9780-09	1.0000 06	1.6580 07	8.5000 08	1.0000 06	6.5360 06	2.0000 09	5.0590 07

T=300 K, H=15 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> NO2 + NO3 + M	1.570-02	2.850-13
2	2*N03 >>> 2*N02 + O2	2.410-16	1.750-5A
3	NO2 + NO3 >>> NO2 + NO + O2	8.210-15	4.920-35
4	NO3 + NO >>> 2*N02	1.900-11	6.980-29
5	NO + O3 >>> NO2 + O2	1.670-14	2.850-49
6	NO2 + O3 >>> NO3 + O2	3.410-17	2.000-34
7	HN03 + M >>> HO + NO2 + M	1.920-20	6.240-12
8	HN03 + HO >>> H2O + NO3	8.000-14	1.250-26
9	O + O + M >>> O2 + M	5.910-15	1.260-56
10	O + O2 + M >>> O3 + M	1.640-15	1.320-08
11	O + O3 >>> 2*O2	8.900-15	0.0
12	O + NO + M >>> NO2 + M	3.070-13	1.200-37
13	O + NO2 >>> NO + O2	6.250-12	3.730-46
14	O + NO2 + M >>> NO3 + M	2.830-13	2.260-23
15	HO + HO >>> H2O + O	1.570-12	4.630-24
16	O2 + 2*N0 >>> 2*N02	1.930-38	2.260-31
17	NO2 + H-NU >>> NO + O	0.0	0.0
18	O + HO >>> H + O2	4.200-11	2.160-22
19	O + HO2 >>> HO + O2	1.510-11	8.510-52
20	O2 + H + M >>> HO2 + M	7.980-14	6.510-24
21	O3 + H >>> HO + O2	1.790-11	1.620-69
22	O3 + HO >>> HO2 + O2	5.350-14	8.980-43
23	O3 + HO2 >>> HO + 2*O2	1.040-15	1.530-63
24	H + HO + M >>> H2O + M	1.410-12	1.140-74
25	H + HO2 >>> 2*HO	1.770-11	1.140-40
26	H + HO2 >>> H2 + O2	1.310-11	6.800-53
27	H + H2O >>> H2 + HO	2.180-25	6.410-15
28	H + H2O2 >>> H2 + HO2	2.130-14	2.960-26
29	H + H2O2 >>> HO + H2O	2.760-14	2.950-65

30	2*H0 + M >>> H2O2 + M	7.100-13	5.180-24
31	H0 + H02 >>> H2O + O2	1.570-11	1.980-63
32	2*H02 >>> H2O2 + O2	3.210-12	1.980-42
33	H02 + H2O >>> H2O2 + H0	6.110-35	8.570-17
34	N0 + H + M >>> HNO + M	5.770-14	1.480-25
35	N0 + H0 >>> N02 + H	7.190-34	4.920-11
36	N0 + H0 + M >>> HNO2 + M	4.420-12	2.480-21
37	N0 + H02 >>> N02 + H0	3.660-13	3.740-19
38	H + H + M >>> H2 + M	2.930-15	2.430-6A
39	HNO4 + M >>> H02 + N02 + M	2.520-02	3.020-13
40	CLN03 + M >>> CLO + N02 + M	2.280-03	3.070-13

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	N0	U3	O2	HNO3	HO	H2O
0.0	1.0000 07	4.5000 09	2.0000 04	5.0000 09	2.5000 12	8.2000 17	1.5000 09	2.0000 06	7.3000 12
0.20	9.9690 06	4.5520 09	1.2730 05	4.9580 09	2.5000 12	8.2000 17	1.5000 09	1.9460 06	7.3000 12
0.40	9.9370 06	4.6030 09	2.3340 05	4.9170 09	2.5000 12	8.2000 17	1.5000 09	1.9020 06	7.3000 12
0.60	9.9060 06	4.6540 09	3.3820 05	4.8760 09	2.5000 12	8.2000 17	1.5000 09	1.8680 06	7.3000 12
0.80	9.8750 06	4.7040 09	4.4190 05	4.8360 09	2.5000 12	8.2000 17	1.5000 09	1.8440 06	7.3000 12
1.00	9.8440 06	4.7540 09	5.4460 05	4.7950 09	2.5000 12	8.2000 17	1.5000 09	1.8290 06	7.3000 12
1.20	9.8140 06	4.8040 09	6.4610 05	4.7550 09	2.5000 12	8.2000 17	1.5000 09	1.8220 06	7.3000 12
1.40	9.7830 06	4.8530 09	7.4670 05	4.7160 09	2.5000 12	8.2000 17	1.5000 09	1.8240 06	7.3000 12
1.60	9.7530 06	4.9020 09	8.4630 05	4.6760 09	2.5000 12	8.2000 17	1.5000 09	1.8330 06	7.3000 12
1.80	9.7220 06	4.9510 09	9.4490 05	4.6370 09	2.5000 12	8.2000 17	1.5000 09	1.8500 06	7.3000 12
2.00	9.6920 06	4.9990 09	1.0430 06	4.5990 09	2.5000 12	8.2000 17	1.5000 09	1.8740 06	7.3000 12
2.20	9.6620 06	5.0470 09	1.1400 06	4.5600 09	2.5000 12	8.2000 17	1.5000 09	1.9040 06	7.3000 12
2.40	9.6320 06	5.0940 09	1.2360 06	4.5220 09	2.5000 12	8.2000 17	1.5000 09	1.9410 06	7.3000 12
2.60	9.6020 06	5.1410 09	1.3310 06	4.4850 09	2.4990 12	8.2000 17	1.5000 09	1.9840 06	7.3000 12
2.80	9.5720 06	5.1880 09	1.4250 06	4.4470 09	2.4990 12	8.2000 17	1.5000 09	2.0320 06	7.3000 12
3.00	9.5430 06	5.2340 09	1.5190 06	4.4100 09	2.4990 12	8.2000 17	1.5000 09	2.0860 06	7.3000 12
3.20	9.5130 06	5.2800 09	1.6120 06	4.3730 09	2.4990 12	8.2000 17	1.5000 09	2.1450 06	7.3000 12
3.40	9.4840 06	5.3260 09	1.7050 06	4.3370 09	2.4990 12	8.2000 17	1.5000 09	2.2090 06	7.3000 12
3.60	9.4550 06	5.3710 09	1.7970 06	4.3010 09	2.4990 12	8.2000 17	1.5000 09	2.2770 06	7.3000 12
3.80	9.4260 06	5.4160 09	1.8880 06	4.2650 09	2.4990 12	8.2000 17	1.5000 09	2.3500 06	7.3000 12
4.00	9.3970 06	5.4610 09	1.9790 06	4.2290 09	2.4990 12	8.2000 17	1.5000 09	2.4260 06	7.3000 12
4.20	9.3680 06	5.5050 09	2.0690 06	4.1940 09	2.4990 12	8.2000 17	1.5000 09	2.5060 06	7.3000 12
4.40	9.3390 06	5.5490 09	2.1580 06	4.1590 09	2.4990 12	8.2000 17	1.5000 09	2.5900 06	7.3000 12
4.60	9.3100 06	5.5930 09	2.2470 06	4.1240 09	2.4990 12	8.2000 17	1.5000 09	2.6770 06	7.3000 12
4.80	9.2820 06	5.6360 09	2.3360 06	4.0900 09	2.4990 12	8.2000 17	1.5000 09	2.7670 06	7.3000 12
5.00	9.2540 06	5.6790 09	2.4240 06	4.0560 09	2.4990 12	8.2000 17	1.5000 09	2.8590 06	7.3000 12
5.20	9.2250 06	5.7210 09	2.5120 06	4.0220 09	2.4990 12	8.2000 17	1.5000 09	2.9550 06	7.3000 12
5.40	9.1970 06	5.7640 09	2.5990 06	3.9880 09	2.4990 12	8.2000 17	1.5000 09	3.0530 06	7.3000 12
5.60	9.1690 06	5.8060 09	2.6860 06	3.9550 09	2.4990 12	8.2000 17	1.5000 09	3.1530 06	7.3000 12
5.80	9.1420 06	5.8470 09	2.7720 06	3.9220 09	2.4990 12	8.2000 17	1.5000 09	3.2550 06	7.3000 12
6.00	9.1140 06	5.8890 09	2.8580 06	3.8890 09	2.4990 12	8.2000 17	1.5000 09	3.3590 06	7.3000 12
6.20	9.0860 06	5.9300 09	2.9440 06	3.8570 09	2.4990 12	8.2000 17	1.5000 09	3.4650 06	7.3000 12
6.40	9.0590 06	5.9700 09	3.0300 06	3.8240 09	2.4990 12	8.2000 17	1.5010 09	3.5730 06	7.3000 12
6.60	9.0310 06	6.0110 09	3.1150 06	3.7920 09	2.4990 12	8.2000 17	1.5010 09	3.6820 06	7.3000 12
6.80	9.0040 06	6.0510 09	3.2000 06	3.7610 09	2.4990 12	8.2000 17	1.5010 09	3.7920 06	7.3000 12
7.00	8.9770 06	6.0900 09	3.2840 06	3.7290 09	2.4990 12	8.2000 17	1.5010 09	3.9040 06	7.3000 12
7.20	8.9500 06	6.1300 09	3.3690 06	3.6980 09	2.4990 12	8.2000 17	1.5010 09	4.0160 06	7.3000 12
7.40	8.9230 06	6.1690 09	3.4530 06	3.6670 09	2.4990 12	8.2000 17	1.5010 09	4.1300 06	7.3000 12
7.60	8.8960 06	6.2080 09	3.5370 06	3.6370 09	2.4990 12	8.2000 17	1.5010 09	4.2440 06	7.3000 12
7.80	8.8700 06	6.2460 09	3.6200 06	3.6060 09	2.4990 12	8.2000 17	1.5010 09	4.3600 06	7.3000 12
8.00	8.8430 06	6.2840 09	3.7040 06	3.5760 09	2.4990 12	8.2000 17	1.5010 09	4.4750 06	7.3000 12
8.20	8.8170 06	6.3220 09	3.7870 06	3.5460 09	2.4990 12	8.2000 17	1.5010 09	4.5920 06	7.3000 12
8.40	8.7900 06	6.3600 09	3.8700 06	3.5160 09	2.4990 12	8.2000 17	1.5010 09	4.7080 06	7.3000 12
8.60	8.7640 06	6.3970 09	3.9530 06	3.4870 09	2.4980 12	8.2000 17	1.5010 09	4.8250 06	7.3000 12
8.80	8.7380 06	6.4340 09	4.0360 06	3.4580 09	2.4980 12	8.2000 17	1.5010 09	4.9420 06	7.3000 12
9.00	8.7120 06	6.4710 09	4.1190 06	3.4290 09	2.4980 12	8.2000 17	1.5010 09	5.0600 06	7.3000 12
9.20	8.6860 06	6.5080 09	4.2010 06	3.4000 09	2.4980 12	8.2000 17	1.5010 09	5.1770 06	7.3000 12
9.40	8.6610 06	6.5440 09	4.2840 06	3.3720 09	2.4980 12	8.2000 17	1.5010 09	5.2950 06	7.3000 12
9.60	8.6350 06	6.5800 09	4.3660 06	3.3440 09	2.4980 12	8.2000 17	1.5010 09	5.4120 06	7.3000 12
9.80	8.6100 06	6.6150 09	4.4480 06	3.3160 09	2.4980 12	8.2000 17	1.5010 09	5.5290 06	7.3000 12
10.00	8.5840 06	6.6510 09	4.5310 06	3.2880 09	2.4980 12	8.2000 17	1.5010 09	5.6460 06	7.3000 12

TIME (S)	N	H	H2	H02	H2O2	HNO	HNO2	HNO4	CLN03
0.0	1.0000 06	1.0000 06	1.0000 06	1.5000 07	8.5000 08	1.0000 06	6.0000 06	2.0000 09	5.0000 07
0.20	2.4480 01	2.2110-07	1.0000 06	2.6070 07	8.5000 08	1.0000 06	6.0090 06	1.9900 09	4.9980 07
0.40	2.4480 01	2.1610-07	1.0000 06	3.6080 07	8.5000 08	1.0000 06	6.0170 06	1.9800 09	4.9970 07
0.60	2.4480 01	2.1220-07	1.0000 06	4.6090 07	8.5000 08	1.0000 06	6.0250 06	1.9700 09	4.9950 07
0.80	2.4480 01	2.0940-07	1.0000 06	5.5910 07	8.5000 08	1.0000 06	6.0330 06	1.9600 09	4.9930 07
1.00	2.4480 01	2.0770-07	1.0000 06	6.5730 07	8.5000 08	1.0000 06	6.0410 06	1.9500 09	4.9910 07
1.20	2.4480 01	2.0700-07	1.0000 06	7.5480 07	8.5000 08	1.0000 06	6.0490 06	1.9410 09	4.9900 07
1.40	2.4480 01	2.0710-07	1.0000 06	8.5180 07	8.5000 08	1.0000 06	6.0560 06	1.9310 09	4.9880 07
1.60	2.4480 01	2.0820-07	1.0000 06	9.4810 07	8.5000 08	1.0000 06	6.0640 06	1.9210 09	4.9860 07
1.80	2.4480 01	2.1010-07	1.0000 06	1.0440 08	8.5000 08	1.0000 06	6.0710 06	1.9120 09	4.9850 07
2.00	2.4480 01	2.1280-07	1.0000 06	1.1390 08	8.5000 08	1.0000 06	6.0790 06	1.9020 09	4.9830 07
2.20	2.4480 01	2.1630-07	1.0000 06	1.2330 08	8.5000 08	1.0000 06	6.0870 06	1.8920 09	4.9810 07
2.40	2.4470 01	2.2050-07	1.0000 06	1.3270 08	8.5000 08	1.0000 06	6.0940 06	1.8830 09	4.9800 07
2.60	2.4470 01	2.2530-07	1.0000 06	1.4200 08	8.5010 08	1.0000 06	6.1020 06	1.8740 09	4.9780 07
2.80	2.4470 01	2.3080-07	1.0000 06	1.5130 08	8.5010 08	1.0000 06	6.1100 06	1.8640 09	4.9770 07
3.00	2.4470 01	2.3700-07	1.0000 06	1.6050 08	8.5010 08	1.0000 06	6.1180 06	1.8550 09	4.9750 07
3.20	2.4470 01	2.4370-07	1.0000 06	1.6970 08	8.5010 08	1.0000 06	6.1260 06	1.8460 09	4.9730 07
3.40	2.4470 01	2.5090-07	1.0000 06	1.7870 08	8.5010 08	1.0000 06	6.1350 06	1.8360 09	4.9720 07
3.60	2.4470 01	2.5860-07	1.0000 06	1.8780 08	8.5010 08	1.0000 06	6.1430 06	1.8270 09	4.9700 07
3.80	2.4470 01	2.6690-07	1.0000 06	1.9670 08	8.5020 08	1.0000 06	6.1520 06	1.8180 09	4.9680 07
4.00	2.4470 01	2.7540-07	1.0000 06	2.0570 08	8.5020 08	1.0000 06	6.1610 06	1.8090 09	4.9670 07
4.20	2.4470 01	2.8460-07	1.0000 06	2.1450 08	8.5020 08	1.0000 06	6.1700 06	1.8000 09	4.9650 07
4.40	2.4470 01	2.9410-07	1.0000 06	2.2330 08	8.5030 08	1.0000 06	6.1890 06	1.7910 09	4.9620 07
4.60	2.4470 01	3.0400-07	1.0000 06	2.3200 08	8.5030 08	1.0000 06	6.1990 06	1.7820 09	4.9620 07
4.80	2.4480 01	3.1420-07	1.0000 06	2.4070 08	8.5030 08	1.0000 06	6.2090 06	1.7730 09	4.9610 07
5.00	2.4480 01	3.2480-07	1.0000 06	2.4930 08	8.5040 08	1.0000 06	6.2200 06	1.7640 09	4.9590 07
5.20	2.4480 01	3.3560-07	1.0000 06	2.5790 08	8.5040 08	1.0000 06	6.2300 06	1.7560 09	4.9570 07
5.40	2.4480 01	3.4670-07	1.0000 06	2.6640 08	8.5040 08	1.0000 06	6.2410 06	1.7470 09	4.9560 07
5.60	2.4480 01	3.5810-07	1.0000 06	2.7480 08	8.5050 08	1.0000 06	6.2520 06	1.7380 09	4.9540 07
5.80	2.4480 01	3.6970-07	1.0000 06	2.8320 08	8.5050 08	1.0000 06	6.2640 06	1.7300 09	4.9530 07
6.00	2.4480 01	3.8150-07	1.0000 06	2.9160 08	8.5060 08	1.0000 06	6.2750 06	1.7210 09	4.9510 07
6.20	2.4480 01	3.9360-07	1.0000 06	3.0010 08	8.5070 08	1.0000 06	6.2870 06	1.7130 09	4.9500 07
6.40	2.4480 01	4.0580-07	1.0000 06	3.0810 08	8.5080 08	1.0000 06	6.2990 06	1.7040 09	4.9480 07
6.60	2.4480 01	4.1820-07	1.0000 06	3.1620 08	8.5080 08	1.0000 06	6.3120 06	1.6960 09	4.9470 07
6.80	2.4480 01	4.3070-07	1.0000 06	3.2430 08	8.5090 08	1.0000 06	6.3250 06	1.6870 09	4.9450 07
7.00	2.4480 01	4.4340-07	1.0000 06	3.3240 08	8.5090 08	1.0000 06	6.3380 06	1.6790 09	4.9440 07
7.20	2.4480 01	4.5620-07	1.0000 06	3.4040 08	8.5100 08	1.0000 06	6.3510 06	1.6710 09	4.9420 07
7.40	2.4480 01	4.6910-07	1.0000 06	3.4830 08	8.5110 08	1.0000 06	6.3640 06	1.6620 09	4.9410 07
7.60	2.4480 01	4.8210-07	1.0000 06	3.5620 08	8.5120 08	1.0000 06	6.3780 06	1.6540 09	4.9390 07
7.80	2.4480 01	4.9520-07	1.0000 06	3.6410 08	8.5130 08	1.0000 06	6.3920 06	1.6460 09	4.9380 07
8.00	2.4480 01	5.0830-07	1.0000 06	3.7180 08	8.5140 08	1.0000 06	6.4060 06	1.6380 09	4.9360 07
8.20	2.4490 01	5.2150-07	1.0000 06	3.7960 08	8.5150 08	1.0000 06	6.4210 06	1.6300 09	4.9350 07
8.40	2.4490 01	5.3480-07	1.0000 06	3.8720 08	8.5160 08	1.0000 06	6.4360 06	1.6220 09	4.9340 07
8.60	2.4490 01	5.4810-07	1.0000 06	3.9480 08	8.5170 08	1.0000 06	6.4510 06	1.6140 09	4.9320 07
8.80	2.4490 01	5.6140-07	1.0000 06	4.0240 08	8.5180 08	1.0000 06	6.4660 06	1.6060 09	4.9310 07
9.00	2.4490 01	5.7470-07	1.0000 06	4.0990 08	8.5190 08	1.0000 06	6.4810 06	1.5980 09	4.9290 07
9.20	2.4490 01	5.8810-07	1.0000 06	4.1740 08	8.5200 08	1.0000 06	6.4970 06	1.5900 09	4.9280 07
9.40	2.4490 01	6.0140-07	1.0000 06	4.2480 08	8.5210 08	1.0000 06	6.5130 06	1.5820 09	4.9260 07
9.60	2.4490 01	6.1480-07	1.0000 06	4.3210 08	8.5220 08	1.0000 06	6.5290 06	1.5740 09	4.9250 07
9.80	2.4490 01	6.2810-07	1.0000 06	4.3940 08	8.5230 08	1.0000 06	6.5450 06	1.5670 09	4.9240 07
10.00	2.4500 01	6.4140-07	1.0000 06	4.4670 08	8.5240 08	1.0000 06	6.5610 06	1.5590 09	4.9220 07

T=700 K, H=15 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	2.31D-06	2.37D-14
2	2*N03 >>> 2*N02 + O2	2.57D-14	7.44D-49
3	N02 + N03 >>> N02 + N0 + O2	5.51D-14	3.97D-36
4	N03 + N0 >>> 2*N02	1.90D-11	2.23D-19
5	N0 + O3 >>> N02 + O2	2.65D-13	4.54D-28
6	N02 + O3 >>> N03 + O2	3.62D-15	6.56D-22
7	HN03 + M >>> H0 + N02 + M	1.92D-01	3.21D-13
8	HN03 + H0 >>> H2O + N03	8.00D-14	4.74D-19
9	O + O + M >>> O2 + M	4.56D-16	1.43D-19
10	O + O2 + M >>> O3 + M	2.66D-16	1.61D-01
11	O + O3 >>> 2*O2	7.11D-13	8.63D-43
12	O + N0 + M >>> N02 + M	4.33D-14	1.03D-10
13	O + N02 >>> N0 + O2	1.11D-11	8.50D-27
14	O + N02 + M >>> N03 + M	1.21D-13	9.70D-24
15	H0 + H0 >>> H2O + O	4.53D-12	2.04D-16
16	O2 + 2*N0 >>> 2*N02	7.04D-39	9.37D-21
17	N02 + H-NU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.20D-11	2.11D-15
19	O + H02 >>> H0 + O2	3.91D-11	2.19D-28
20	O2 + H + M >>> H02 + M	1.32D-14	2.96D-05
21	O3 + H >>> H0 + O2	4.78D-11	5.62D-37
22	O3 + H0 >>> H02 + O2	3.59D-13	8.31D-26
23	O3 + H02 >>> H0 + 2*O2	1.18D-14	1.12D-49
24	H + H0 + M >>> H2O + M	6.68D-14	1.61D-26
25	H + H02 >>> 2*H0	1.08D-10	5.87D-24
26	H + H02 >>> H2 + O2	2.55D-11	8.03D-29
27	H + H2O >>> H2 + H0	6.56D-17	8.90D-13
28	H + H2O2 >>> H2 + H02	3.06D-12	1.77D-18
29	H + H2O2 >>> H0 + H2O	3.98D-13	1.02D-35

30	2*H0 + M >>> H2O2 + M	5.480-14	5.020-04
31	H0 + H02 >>> H2O + O2	4.060-11	7.730-33
32	2*H02 >>> H2O2 + O2	4.320-12	3.020-24
33	H02 + H2O >>> H2O2 + H0	2.720-21	3.070-12
34	N0 + H + M >>> HNO + M	1.400-14	1.050-05
35	N0 + H0 >>> N02 + H	2.230-21	2.020-10
36	N0 + H0 + M >>> HNO2 + M	2.280-13	2.840-02
37	N0 + H02 >>> N02 + H0	3.600-12	2.680-14
38	H + H + M >>> H2 + M	1.210-15	1.010-24
39	HNO4 + M >>> H02 + N02 + M	2.490 06	2.330-14
40	CLN03 + M >>> CLO + N02 + M	1.800 06	1.140-14

HAPP RESIDENCE TIME STUDY

TIME (s)	N205	N02	N03	N0	U3	O2	HNO3	HO	H2O
0.0	1.0000-07	4.5000 09	2.0000 04	5.0000 09	2.5000 12	8.2000 17	1.5000 09	2.0000 06	7.3000 12
0.20	2.2460-03	5.1850 09	4.2140 07	6.3990 09	2.2260 12	8.2000 17	1.4440 09	1.0120 09	7.3000 12
0.40	2.9720-03	4.3500 09	6.6480 07	7.2640 09	2.1280 12	8.2000 17	1.3890 09	1.1400 09	7.3000 12
0.60	3.3880-03	3.8530 09	8.5580 07	7.7940 09	2.0390 12	8.2000 17	1.3370 09	1.2020 09	7.3000 12
0.80	3.6950-03	3.5520 09	1.0120 08	8.1280 09	1.9570 12	8.2000 17	1.2870 09	1.2540 09	7.3000 12
1.00	3.9610-03	3.3690 09	1.1440 08	8.3460 09	1.8810 12	8.2000 17	1.2390 09	1.3020 09	7.3000 12
1.20	4.2120-03	3.2560 09	1.2580 08	8.4930 09	1.8100 12	8.2000 17	1.1920 09	1.3460 09	7.3000 12
1.40	4.4530-03	3.1870 09	1.3590 08	8.5970 09	1.7450 12	8.2000 17	1.1480 09	1.3860 09	7.3000 12
1.60	4.6850-03	3.1460 09	1.4490 08	8.6720 09	1.6830 12	8.2000 17	1.1050 09	1.4240 09	7.3000 12
1.80	4.9070-03	3.1220 09	1.5290 08	8.7290 09	1.6260 12	8.2000 17	1.0630 09	1.4580 09	7.3000 12
2.00	5.1180-03	3.1080 09	1.6020 08	8.7740 09	1.5720 12	8.2000 17	1.0230 09	1.4890 09	7.3000 12
2.20	5.3180-03	3.1030 09	1.6680 08	8.8110 09	1.5220 12	8.2000 17	9.8490 08	1.5170 09	7.3000 12
2.40	5.5050-03	3.1010 09	1.7270 08	8.8430 09	1.4740 12	8.2000 17	9.4800 08	1.5430 09	7.3000 12
2.60	5.6800-03	3.1030 09	1.7810 08	8.8700 09	1.4290 12	8.2000 17	9.1250 08	1.5660 09	7.3000 12
2.80	5.8420-03	3.1070 09	1.8290 08	8.8950 09	1.3870 12	8.2000 17	8.7840 08	1.5870 09	7.3000 12
3.00	5.9920-03	3.1130 09	1.8730 08	8.9170 09	1.3470 12	8.2000 17	8.4550 08	1.6050 09	7.3000 12
3.20	6.1300-03	3.1190 09	1.9120 08	8.9380 09	1.3090 12	8.2000 17	8.1390 08	1.6220 09	7.3000 12
3.40	6.2570-03	3.1260 09	1.9480 08	8.9580 09	1.2730 12	8.2000 17	7.8350 08	1.6370 09	7.3000 12
3.60	6.3720-03	3.1330 09	1.9790 08	8.9760 09	1.2390 12	8.2000 17	7.5420 08	1.6490 09	7.3000 12
3.80	6.4770-03	3.1400 09	2.0070 08	8.9940 09	1.2060 12	8.2000 17	7.2610 08	1.6610 09	7.3000 12
4.00	6.5710-03	3.1470 09	2.0320 08	9.0110 09	1.1750 12	8.2000 17	6.9900 08	1.6700 09	7.3000 12
4.20	6.6560-03	3.1540 09	2.0530 08	9.0270 09	1.1460 12	8.2000 17	6.7290 08	1.6780 09	7.3000 12
4.40	6.7310-03	3.1610 09	2.0720 08	9.0430 09	1.1180 12	8.2000 17	6.4790 08	1.6850 09	7.3000 12
4.60	6.7980-03	3.1680 09	2.0880 08	9.0580 09	1.0910 12	8.2000 17	6.2370 08	1.6910 09	7.3000 12
4.80	6.8560-03	3.1750 09	2.1010 08	9.0720 09	1.0650 12	8.2000 17	6.0050 08	1.6950 09	7.3000 12
5.00	6.9070-03	3.1820 09	2.1120 08	9.0860 09	1.0410 12	8.2000 17	5.7820 08	1.6980 09	7.3000 12
5.20	6.9500-03	3.1880 09	2.1210 08	9.1000 09	1.0170 12	8.2000 17	5.5670 08	1.7000 09	7.3000 12
5.40	6.9860-03	3.1940 09	2.1280 08	9.1130 09	9.9460 11	8.2000 17	5.3600 08	1.7020 09	7.3000 12
5.60	7.0150-03	3.2010 09	2.1330 08	9.1250 09	9.7290 11	8.2000 17	5.1610 08	1.7020 09	7.3000 12
5.80	7.0380-03	3.2060 09	2.1360 08	9.1380 09	9.5220 11	8.2000 17	4.9700 08	1.7010 09	7.3000 12
6.00	7.0560-03	3.2120 09	2.1370 08	9.1500 09	9.3220 11	8.2000 17	4.7860 08	1.7000 09	7.3000 12
6.20	7.0670-03	3.2180 09	2.1370 08	9.1610 09	9.1300 11	8.2000 17	4.6080 08	1.6980 09	7.3000 12
6.40	7.0740-03	3.2230 09	2.1350 08	9.1720 09	8.9450 11	8.2000 17	4.4380 08	1.6960 09	7.3000 12
6.60	7.0750-03	3.2280 09	2.1320 08	9.1830 09	8.7670 11	8.2000 17	4.2740 08	1.6920 09	7.3000 12
6.80	7.0720-03	3.2340 09	2.1280 08	9.1940 09	8.5960 11	8.2000 17	4.1160 08	1.6890 09	7.3000 12
7.00	7.0650-03	3.2380 09	2.1230 08	9.2040 09	8.4300 11	8.2000 17	3.9640 08	1.6850 09	7.3000 12
7.20	7.0540-03	3.2430 09	2.1160 08	9.2140 09	8.2710 11	8.2000 17	3.8180 08	1.6800 09	7.3000 12
7.40	7.0390-03	3.2480 09	2.1090 08	9.2240 09	8.1170 11	8.2000 17	3.6770 08	1.6750 09	7.3000 12
7.60	7.0200-03	3.2520 09	2.1000 08	9.2330 09	7.9680 11	8.2000 17	3.5410 08	1.6690 09	7.3000 12
7.80	6.9980-03	3.2570 09	2.0910 08	9.2420 09	7.8240 11	8.2000 17	3.4110 08	1.6630 09	7.3000 12
8.00	6.9730-03	3.2610 09	2.0810 08	9.2510 09	7.6850 11	8.2000 17	3.2860 08	1.6570 09	7.3000 12
8.20	6.9450-03	3.2650 09	2.0700 08	9.2600 09	7.5510 11	8.2000 17	3.1650 08	1.6510 09	7.3000 12
8.40	6.9140-03	3.2690 09	2.0580 08	9.2680 09	7.4210 11	8.2000 17	3.0490 08	1.6440 09	7.3000 12
8.60	6.8810-03	3.2720 09	2.0460 08	9.2760 09	7.2950 11	8.2000 17	2.9370 08	1.6370 09	7.3000 12
8.80	6.8460-03	3.2760 09	2.0330 08	9.2840 09	7.1730 11	8.2000 17	2.8300 08	1.6300 09	7.3000 12
9.00	6.8090-03	3.2790 09	2.0200 08	9.2920 09	7.0540 11	8.2000 17	2.7260 08	1.6220 09	7.3000 12
9.20	6.7690-03	3.2830 09	2.0060 08	9.2990 09	6.9390 11	8.2000 17	2.6270 08	1.6150 09	7.3000 12
9.40	6.7280-03	3.2860 09	1.9920 08	9.3060 09	6.8280 11	8.2000 17	2.5310 08	1.6070 09	7.3000 12
9.60	6.6850-03	3.2890 09	1.9780 08	9.3140 09	6.7200 11	8.2000 17	2.4390 08	1.5990 09	7.3000 12
9.80	6.6410-03	3.2920 09	1.9630 08	9.3200 09	6.6150 11	8.2000 17	2.3500 08	1.5910 09	7.3000 12
10.00	6.5950-03	3.2950 09	1.9470 08	9.3270 09	6.5130 11	8.2000 17	2.2640 08	1.5830 09	7.3000 12

TIME (\$)	N	H	H2	M02	M202	MNO	MNO2	MNO4	CLN03
0.0	1.0000 06	1.0000 06	1.0000 06	1.5000 07	8.5000 08	1.0000 06	6.0000 06	2.0000 09	5.0000 07
0.20	1.6310 11	5.4730 05	1.0020 06	1.1370 09	8.5290 08	1.0000 06	6.1500 06	5.5110-02	2.2890-03
0.40	1.5600 11	5.9040 05	1.0060 06	1.1310 09	8.5440 08	1.0000 06	6.4570 06	4.5990-02	1.9200-03
0.60	1.4950 11	5.9640 05	1.0090 06	1.1840 09	8.5590 08	1.0000 06	6.8230 06	4.2630-02	1.7010-03
0.80	1.4350 11	5.9760 05	1.0130 06	1.2370 09	8.5760 08	1.0000 06	7.2310 06	4.1060-02	1.5680-03
1.00	1.3800 11	5.9660 05	1.0160 06	1.2860 09	8.5950 08	1.0000 06	7.6700 06	4.0490-02	1.4870-03
1.20	1.3280 11	5.9390 05	1.0200 06	1.3310 09	8.6160 08	1.0000 06	8.1350 06	4.0530-02	1.4380-03
1.40	1.2800 11	5.8990 05	1.0240 06	1.3730 09	8.6390 08	1.0000 06	8.6210 06	4.0910-02	1.4070-03
1.60	1.2350 11	5.8460 05	1.0280 06	1.4120 09	8.6630 08	1.0000 06	9.1250 06	4.1510-02	1.3890-03
1.80	1.1930 11	5.7840 05	1.0320 06	1.4470 09	8.6890 08	1.0000 06	9.6450 06	4.2210-02	1.3780-03
2.00	1.1540 11	5.7140 05	1.0360 06	1.4790 09	8.7160 08	1.0000 06	1.0180 07	4.2970-02	1.3720-03
2.20	1.1170 11	5.6370 05	1.0400 06	1.5080 09	8.7440 08	1.0000 06	1.0720 07	4.3730-02	1.3700-03
2.40	1.0820 11	5.5550 05	1.0440 06	1.5340 09	8.7740 08	1.0000 06	1.1280 07	4.4480-02	1.3690-03
2.60	1.0490 11	5.4680 05	1.0490 06	1.5580 09	8.8050 08	1.0000 06	1.1840 07	4.5200-02	1.3700-03
2.80	1.0180 11	5.3780 05	1.0530 06	1.5800 09	8.8370 08	1.0000 06	1.2410 07	4.5880-02	1.3720-03
3.00	9.8920 10	5.2860 05	1.0570 06	1.5990 09	8.8690 08	1.0000 06	1.2990 07	4.6520-02	1.3740-03
3.20	9.6140 10	5.1910 05	1.0610 06	1.6160 09	8.9030 08	1.0000 06	1.3570 07	4.7110-02	1.3770-03
3.40	9.3510 10	5.0950 05	1.0650 06	1.6310 09	8.9370 08	1.0000 06	1.4160 07	4.7650-02	1.3800-03
3.60	9.1010 10	4.9990 05	1.0690 06	1.6440 09	8.9720 08	1.0000 06	1.4750 07	4.8150-02	1.3830-03
3.80	8.8630 10	4.9020 05	1.0730 06	1.6560 09	9.0080 08	1.0000 06	1.5340 07	4.8600-02	1.3860-03
4.00	8.6370 10	4.8050 05	1.0770 06	1.6660 09	9.0440 08	1.0000 06	1.5940 07	4.9000-02	1.3890-03
4.20	8.4210 10	4.7080 05	1.0800 06	1.6740 09	9.0800 08	1.0000 06	1.6540 07	4.9360-02	1.3930-03
4.40	8.2150 10	4.6120 05	1.0840 06	1.6810 09	9.1170 08	1.0000 06	1.7140 07	4.9670-02	1.3960-03
4.60	8.0180 10	4.5160 05	1.0880 06	1.6860 09	9.1540 08	1.0000 06	1.7730 07	4.9950-02	1.3990-03
4.80	7.8300 10	4.4220 05	1.0910 06	1.6910 09	9.1920 08	1.0000 06	1.8330 07	5.0180-02	1.4020-03
5.00	7.6500 10	4.3290 05	1.0950 06	1.6940 09	9.2290 08	1.0000 06	1.8930 07	5.0390-02	1.4050-03
5.20	7.4770 10	4.2370 05	1.0980 06	1.6960 09	9.2670 08	1.0000 06	1.9530 07	5.0550-02	1.4080-03
5.40	7.3120 10	4.1470 05	1.1020 06	1.6980 09	9.3040 08	1.0000 06	2.0120 07	5.0690-02	1.4100-03
5.60	7.1530 10	4.0580 05	1.1050 06	1.6980 09	9.3420 08	1.0000 06	2.0710 07	5.0800-02	1.4130-03
5.80	7.0010 10	3.9710 05	1.1080 06	1.6970 09	9.3790 08	1.0000 06	2.1310 07	5.0870-02	1.4160-03
6.00	6.8540 10	3.8850 05	1.1110 06	1.6960 09	9.4170 08	1.0000 06	2.1890 07	5.0930-02	1.4180-03
6.20	6.7130 10	3.8010 05	1.1140 06	1.6940 09	9.4540 08	1.0000 06	2.2480 07	5.0950-02	1.4210-03
6.40	6.5780 10	3.7190 05	1.1170 06	1.6910 09	9.4910 08	1.0000 06	2.3060 07	5.0960-02	1.4230-03
6.60	6.4470 10	3.6390 05	1.1200 06	1.6880 09	9.5280 08	1.0000 06	2.3640 07	5.0940-02	1.4250-03
6.80	6.3220 10	3.5600 05	1.1230 06	1.6840 09	9.5650 08	1.0000 06	2.4210 07	5.0900-02	1.4280-03
7.00	6.2000 10	3.4830 05	1.1260 06	1.6800 09	9.6020 08	1.0000 06	2.4780 07	5.0840-02	1.4300-03
7.20	6.0830 10	3.4080 05	1.1280 06	1.6750 09	9.6380 08	1.0000 06	2.5350 07	5.0770-02	1.4320-03
7.40	5.9700 10	3.3350 05	1.1310 06	1.6690 09	9.6740 08	1.0000 06	2.5910 07	5.0680-02	1.4340-03
7.60	5.8610 10	3.2630 05	1.1330 06	1.6630 09	9.7090 08	1.0000 06	2.6460 07	5.0570-02	1.4360-03
7.80	5.7560 10	3.1940 05	1.1360 06	1.6570 09	9.7450 08	1.0000 06	2.7010 07	5.0450-02	1.4380-03
8.00	5.6540 10	3.1250 05	1.1380 06	1.6510 09	9.7790 08	1.0000 06	2.7560 07	5.0320-02	1.4400-03
8.20	5.5550 10	3.0590 05	1.1410 06	1.6440 09	9.8140 08	1.0000 06	2.8100 07	5.0170-02	1.4410-03
8.40	5.4590 10	2.9940 05	1.1430 06	1.6370 09	9.8480 08	1.0000 06	2.8640 07	5.0020-02	1.4430-03
8.60	5.3670 10	2.9310 05	1.1450 06	1.6320 09	9.8820 08	1.0000 06	2.9170 07	4.9850-02	1.4450-03
8.80	5.2770 10	2.8690 05	1.1470 06	1.6220 09	9.9150 08	1.0000 06	2.9690 07	4.9680-02	1.4460-03
9.00	5.1900 10	2.8090 05	1.1490 06	1.6140 09	9.9480 08	1.0000 06	3.0210 07	4.9490-02	1.4480-03
9.20	5.1060 10	2.7510 05	1.1510 06	1.6070 09	9.9810 08	1.0000 06	3.0730 07	4.9300-02	1.4490-03
9.40	5.0240 10	2.6940 05	1.1530 06	1.5980 09	1.0010 09	1.0000 06	3.1230 07	4.9100-02	1.4510-03
9.60	4.9450 10	2.6390 05	1.1550 06	1.5900 09	1.0040 09	1.0000 06	3.1740 07	4.8900-02	1.4520-03
9.80	4.8680 10	2.5840 05	1.1570 06	1.5820 09	1.0080 09	1.0000 06	3.2230 07	4.8680-02	1.4540-03
10.00	4.7930 10	2.5320 05	1.1590 06	1.5730 09	1.0110 09	1.0000 06	3.2730 07	4.8470-02	1.4550-03

T= 800 K, H=15 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> NO2 + NO3 + M	1.270 07	1.780-14
2	2*N03 >>> 2*N02 + O2	3.980-14	5.950-48
3	NO2 + NO3 >>> NO2 + NO + O2	6.590-14	3.130-36
4	NO3 + NO >>> 2*N02	1.900-11	1.740-18
5	NO + O3 >>> NO2 + O2	3.430-13	4.420-26
6	NO2 + O3 >>> NO3 + O2	5.610-15	9.780-21
7	HN03 + M >>> HO + NO2 + M	9.110 00	2.010-13
8	HN03 + HO >>> H2O + NO3	8.000-14	2.430-18
9	O + O + M >>> O2 + M	3.400-16	4.320-15
10	O + O2 + M >>> O3 + M	2.130-16	1.080 02
11	O + O3 >>> 2*O2	1.070-12	6.110-39
12	O + NO + M >>> NO2 + M	3.410-14	3.250-08
13	O + NO2 >>> NO + O2	1.170-11	5.550-25
14	O + NO2 + M >>> NO3 + M	1.060-13	8.480-24
15	HO + HO >>> H2O + O	5.000-12	1.060-15
16	O2 + 2*N0 >>> 2*N02	6.400-39	8.780-20
17	NO2 + H-NU >>> NO + O	0.0	0.0
18	O + HO >>> H + O2	4.200-11	9.550-15
19	O + HO2 >>> HO + O2	4.280-11	3.430-24
20	O2 + H + M >>> HO2 + M	1.060-14	1.580-03
21	O3 + H >>> HO + O2	5.250-11	6.310-34
22	O3 + HO >>> HO2 + O2	4.300-13	3.240-24
23	O3 + HO2 >>> HO + 2*O2	1.480-14	2.350-48
24	H + HO + M >>> H2O + M	4.130-14	4.090-22
25	H + HO2 >>> 2*H0	1.280-10	2.160-22
26	H + HO2 >>> H2 + O2	2.710-11	1.450-26
27	H + H2O >>> H2 + HO	4.090-16	1.410-12
28	H + H2O2 >>> H2 + HO2	3.930-13	9.470-18
29	H + H2O2 >>> HO + H2O	5.110-13	6.020-33

30	2*H0 + M >>> H2O2 + M	4.080-14	3.370-02
31	H0 + H02 >>> H2O + O2	4.440-11	5.700-30
32	2*H02 >>> H2O2 + O2	9.100-12	1.530-22
33	H02 + H2O >>> H2O2 + H0	5.190-20	3.460-12
34	NO + H + M >>> HNO + M	1.160-14	6.880-04
35	NO + H0 >>> NO2 + H	3.300-20	2.300-10
36	NO + H0 + M >>> HNO2 + M	1.640-13	1.560-00
37	NO + H02 >>> NO2 + H0	4.460-12	7.660-14
38	H + H + M >>> H2 + M	1.060-15	1.120-20
39	HNO4 + M >>> H02 + NO2 + M	1.330 07	1.750-14
40	CLN03 + M >>> CLO + NO2 + M	1.160 07	6.670-15

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	N0	O3	O2	HNO3	HO	H2O
0.0	1.000 07	4.500 09	2.000 04	5.000 09	2.500 12	8.200 17	1.500 09	2.000 06	7.300 12
0.20	1.8710-04	1.7590 09	7.6210 07	1.0990 10	1.2190 12	8.200 17	2.4270 08	3.8750 09	7.2990 12
0.40	1.1750-04	9.1570 08	9.1900 07	1.2020 10	9.9910 11	8.200 17	3.9330 07	5.2230 09	7.2980 12
0.60	9.8950-05	7.2420 08	9.7840 07	1.2290 10	8.4140 11	8.200 17	6.4410 06	6.1480 09	7.2970 12
0.80	9.5250-05	6.7680 08	1.0080 08	1.2290 10	7.2270 11	8.200 17	1.1250 06	6.8090 09	7.2970 12
1.00	9.5380-05	6.6740 08	1.0230 08	1.2300 10	6.3040 11	8.200 17	2.7050 05	7.2770 09	7.2960 12
1.20	9.6300-05	6.6940 08	1.0300 08	1.2300 10	5.5660 11	8.200 17	1.3670 05	7.5970 09	7.2960 12
1.40	9.7150-05	6.7490 08	1.0310 08	1.2290 10	4.9640 11	8.200 17	1.1870 05	7.8020 09	7.2960 12
1.60	9.7730-05	6.8160 08	1.0270 08	1.2280 10	4.4650 11	8.200 17	1.1860 05	7.9200 09	7.2960 12
1.80	9.7990-05	6.8860 08	1.0190 08	1.2280 10	4.0460 11	8.200 17	1.2050 05	7.9700 09	7.2960 12
2.00	9.7950-05	6.9570 08	1.0080 08	1.2270 10	3.6890 11	8.200 17	1.2210 05	7.9690 09	7.2960 12
2.20	9.7660-05	7.0260 08	9.9530 07	1.2260 10	3.3820 11	8.200 17	1.2300 05	7.9300 09	7.2960 12
2.40	9.7130-05	7.0940 08	9.8040 07	1.2260 10	3.1160 11	8.200 17	1.2340 05	7.8630 09	7.2960 12
2.60	9.6400-05	7.1600 08	9.6410 07	1.2250 10	2.8840 11	8.200 17	1.2340 05	7.7740 09	7.2960 12
2.80	9.5490-05	7.2240 08	9.4660 07	1.2250 10	2.6790 11	8.200 17	1.2300 05	7.6700 09	7.2960 12
3.00	9.4450-05	7.2860 08	9.2830 07	1.2240 10	2.4980 11	8.200 17	1.2230 05	7.5560 09	7.2960 12
3.20	9.3280-05	7.3460 08	9.0930 07	1.2240 10	2.3360 11	8.200 17	1.2140 05	7.4350 09	7.2960 12
3.40	9.2000-05	7.4030 08	8.8990 07	1.2230 10	2.1920 11	8.200 17	1.2040 05	7.3100 09	7.2960 12
3.60	9.0640-05	7.4590 08	8.7010 07	1.2230 10	2.0620 11	8.200 17	1.1930 05	7.1820 09	7.2960 12
3.80	8.9200-05	7.5130 08	8.5020 07	1.2230 10	1.9440 11	8.200 17	1.1800 05	7.0540 09	7.2960 12
4.00	8.7710-05	7.5660 08	8.3020 07	1.2230 10	1.8370 11	8.200 17	1.1670 05	6.9260 09	7.2960 12
4.20	8.6170-05	7.6160 08	8.1020 07	1.2220 10	1.7400 11	8.200 17	1.1540 05	6.8000 09	7.2960 12
4.40	8.4590-05	7.6660 08	7.9020 07	1.2220 10	1.6520 11	8.200 17	1.1400 05	6.6760 09	7.2960 12
4.60	8.2980-05	7.7130 08	7.7040 07	1.2220 10	1.5700 11	8.200 17	1.1270 05	6.5540 09	7.2970 12
4.80	8.1360-05	7.7590 08	7.5090 07	1.2220 10	1.4960 11	8.200 17	1.1130 05	6.4350 09	7.2970 12
5.00	7.9720-05	7.8040 08	7.3150 07	1.2210 10	1.4270 11	8.200 17	1.0990 05	6.3190 09	7.2970 12
5.20	7.8080-05	7.8480 08	7.1240 07	1.2210 10	1.3630 11	8.200 17	1.0860 05	6.2070 09	7.2970 12
5.40	7.6430-05	7.8900 08	6.9370 07	1.2210 10	1.3040 11	8.200 17	1.0720 05	6.0970 09	7.2970 12
5.60	7.4790-05	7.9310 08	6.7530 07	1.2210 10	1.2490 11	8.200 17	1.0590 05	5.9910 09	7.2970 12
5.80	7.3160-05	7.9710 08	6.5720 07	1.2210 10	1.1980 11	8.200 17	1.0460 05	5.8880 09	7.2970 12
6.00	7.1540-05	8.0100 08	6.3950 07	1.2200 10	1.1510 11	8.200 17	1.0330 05	5.7890 09	7.2970 12
6.20	6.9930-05	8.0480 08	6.2220 07	1.2200 10	1.1060 11	8.200 17	1.0210 05	5.6930 09	7.2970 12
6.40	6.8340-05	8.0850 08	6.0530 07	1.2200 10	1.0640 11	8.200 17	1.0090 05	5.6000 09	7.2970 12
6.60	6.6770-05	8.1220 08	5.8870 07	1.2200 10	1.0250 11	8.200 17	9.9680 04	5.5100 09	7.2970 12
6.80	6.5220-05	8.1570 08	5.7260 07	1.2200 10	9.8840 10	8.200 17	9.8530 04	5.4230 09	7.2970 12
7.00	6.3700-05	8.1910 08	5.5690 07	1.2190 10	9.5390 10	8.200 17	9.7400 04	5.3380 09	7.2970 12
7.20	6.2200-05	8.2250 08	5.4160 07	1.2190 10	9.2130 10	8.200 17	9.6290 04	5.2570 09	7.2980 12
7.40	6.0730-05	8.2580 08	5.2660 07	1.2190 10	8.9050 10	8.200 17	9.5220 04	5.1780 09	7.2980 12
7.60	5.9280-05	8.2900 08	5.1210 07	1.2190 10	8.6140 10	8.200 17	9.4180 04	5.1020 09	7.2980 12
7.80	5.7870-05	8.3220 08	4.9800 07	1.2190 10	8.3390 10	8.200 17	9.3160 04	5.0280 09	7.2980 12
8.00	5.6480-05	8.3530 08	4.8420 07	1.2190 10	8.0780 10	8.200 17	9.2170 04	4.9570 09	7.2980 12
8.20	5.5120-05	8.3830 08	4.7090 07	1.2180 10	7.8300 10	8.200 17	9.1200 04	4.8880 09	7.2980 12
8.40	5.3790-05	8.4130 08	4.5790 07	1.2180 10	7.5950 10	8.200 17	9.0260 04	4.8210 09	7.2980 12
8.60	5.2490-05	8.4420 08	4.4530 07	1.2180 10	7.3720 10	8.200 17	8.9350 04	4.7560 09	7.2980 12
8.80	5.1220-05	8.4700 08	4.3300 07	1.2180 10	7.1590 10	8.200 17	8.8460 04	4.6940 09	7.2980 12
9.00	4.9980-05	8.4980 08	4.2110 07	1.2180 10	6.9560 10	8.200 17	8.7590 04	4.6330 09	7.2980 12
9.20	4.8770-05	8.5260 08	4.0960 07	1.2180 10	6.7620 10	8.200 17	8.6750 04	4.5740 09	7.2980 12
9.40	4.7590-05	8.5530 08	3.9840 07	1.2170 10	6.5780 10	8.200 17	8.5930 04	4.5170 09	7.2980 12
9.60	4.6430-05	8.5800 08	3.8750 07	1.2170 10	6.4010 10	8.200 17	8.5130 04	4.4610 09	7.2980 12
9.80	4.5310-05	8.6060 08	3.7700 07	1.2170 10	6.2330 10	8.200 17	8.4350 04	4.4070 09	7.2980 12
10.00	4.4210-05	8.6320 08	3.6680 07	1.2170 10	6.0710 10	8.200 17	8.3590 04	4.3550 09	7.2980 12

TIME (S)	N	H	H2	H02	H202	HNO	HNO2	HN04	CLNO3
0.0	1.0000 06	1.0000 06	1.0000 06	1.5000 07	8.5000 08	1.0000 06	6.0000 06	2.0000 09	5.0000 07
0.20	7.5660 11	7.4370 06	1.0480 06	1.8930 09	8.4670 08	1.0000 06	5.1290 06	4.3570 03	7.0520 05
0.40	6.2020 11	8.2250 06	1.1500 06	2.6250 09	8.4800 08	1.0000 06	5.2780 06	3.1460 03	3.6720 05
0.60	5.2230 11	8.1570 06	1.2830 06	3.1150 09	8.5440 08	1.0000 06	5.8270 06	2.9520 03	2.9040 05
0.80	4.4870 11	7.7640 06	1.4280 06	3.4610 09	8.6480 08	1.0000 06	6.5150 06	3.0660 03	2.7140 05
1.00	3.9140 11	7.2400 06	1.5760 06	3.7030 09	8.7860 08	1.0000 04	7.2150 06	3.2350 03	2.6770 05
1.20	3.4560 11	6.6760 06	1.7200 06	3.8670 09	8.9460 08	1.0000 04	7.8610 06	3.3880 03	2.6840 05
1.40	3.0820 11	6.1170 06	1.8570 06	3.9700 09	9.1220 08	1.0000 06	8.4220 06	3.5070 03	2.7060 05
1.60	2.7730 11	5.5860 06	1.9830 06	4.0270 09	9.3060 08	1.0000 06	8.8850 06	3.5920 03	2.7330 05
1.80	2.5120 11	5.0940 06	2.0990 06	4.0480 09	9.4930 08	1.0000 04	9.2500 06	3.6480 03	2.7620 05
2.00	2.2910 11	4.6450 06	2.2040 06	4.0430 09	9.6790 08	1.0000 06	9.5230 06	3.6810 03	2.7900 05
2.20	2.1000 11	4.2390 06	2.2990 06	4.0170 09	9.8600 08	1.0000 06	9.7150 06	3.6940 03	2.8180 05
2.40	1.9350 11	3.8720 06	2.3850 06	3.9770 09	1.0030 09	1.0000 06	9.8340 06	3.6930 03	2.8450 05
2.60	1.7910 11	3.5430 06	2.4610 06	3.9260 09	1.0200 09	1.0000 06	9.8930 06	3.6790 03	2.8710 05
2.80	1.6640 11	3.2490 06	2.5300 06	3.8670 09	1.0360 09	1.0000 06	9.9020 06	3.6560 03	2.8970 05
3.00	1.5510 11	2.9840 06	2.5910 06	3.8030 09	1.0500 09	1.0000 06	9.8700 06	3.6270 03	2.9220 05
3.20	1.4510 11	2.7470 06	2.6460 06	3.7360 09	1.0640 09	1.0000 06	9.8050 06	3.5920 03	2.9460 05
3.40	1.3610 11	2.5340 06	2.6950 06	3.6660 09	1.0770 09	1.0000 06	9.7130 06	3.5520 03	2.9690 05
3.60	1.2810 11	2.3420 06	2.7390 06	3.5960 09	1.0880 09	1.0000 06	9.6020 06	3.5100 03	2.9910 05
3.80	1.2080 11	2.1690 06	2.7790 06	3.5250 09	1.0990 09	1.0000 06	9.4760 06	3.4640 03	3.0130 05
4.00	1.1420 11	2.0130 06	2.8140 06	3.4550 09	1.1090 09	1.0000 06	9.3390 06	3.4210 03	3.0340 05
4.20	1.0810 11	1.8720 06	2.8460 06	3.3860 09	1.1180 09	1.0000 05	9.1950 06	3.3750 03	3.0540 05
4.40	1.0260 11	1.7440 06	2.8750 06	3.3180 09	1.1260 09	1.0000 05	9.0450 06	3.3280 03	3.0740 05
4.60	9.7580 10	1.6280 06	2.9010 06	3.2510 09	1.1330 09	1.0000 05	8.8930 06	3.2820 03	3.0930 05
4.80	9.2940 10	1.5230 06	2.9240 06	3.1860 09	1.1390 09	1.0000 05	8.7400 06	3.2360 03	3.1120 05
5.00	8.8660 10	1.4270 06	2.9450 06	3.1230 09	1.1450 09	1.0000 05	8.5870 06	3.1900 03	3.1300 05
5.20	8.4700 10	1.3390 06	2.9640 06	3.0610 09	1.1500 09	1.0000 05	8.4360 06	3.1440 03	3.1470 05
5.40	8.1030 10	1.2540 06	2.9810 06	3.0020 09	1.1540 09	1.0000 05	8.2860 06	3.1000 03	3.1640 05
5.60	7.7630 10	1.1840 06	2.9960 06	2.9440 09	1.1580 09	1.0000 05	8.1390 06	3.0560 03	3.1810 05
5.80	7.4460 10	1.1170 06	3.0100 06	2.8880 09	1.1610 09	1.0000 05	7.9960 06	3.0130 03	3.1970 05
6.00	7.1500 10	1.0540 06	3.0230 06	2.8340 09	1.1640 09	1.0000 05	7.8560 06	2.9710 03	3.2120 05
6.20	6.8730 10	9.9660 05	3.0340 06	2.7810 09	1.1660 09	1.0000 05	7.7190 06	2.9300 03	3.2280 05
6.40	6.6140 10	9.4340 05	3.0450 06	2.7300 09	1.1680 09	1.0000 05	7.5870 06	2.8890 03	3.2430 05
6.60	6.3720 10	8.9410 05	3.0540 06	2.6810 09	1.1690 09	1.0000 05	7.4580 06	2.8500 03	3.2570 05
6.80	6.1430 10	8.4850 05	3.0620 06	2.6340 09	1.1700 09	1.0000 05	7.3330 06	2.8120 03	3.2710 05
7.00	5.9280 10	8.0610 05	3.0700 06	2.5880 09	1.1700 09	1.0000 05	7.2120 06	2.7740 03	3.2850 05
7.20	5.7260 10	7.6670 05	3.0770 06	2.5440 09	1.1700 09	1.0000 05	7.0950 06	2.7380 03	3.2980 05
7.40	5.5350 10	7.3000 05	3.0830 06	2.5010 09	1.1700 09	1.0000 05	6.9820 06	2.7030 03	3.3120 05
7.60	5.3540 10	6.9580 05	3.0890 06	2.4590 09	1.1690 09	1.0000 05	6.8720 06	2.6680 03	3.3250 05
7.80	5.1830 10	6.6390 05	3.0940 06	2.4190 09	1.1680 09	1.0000 05	6.7660 06	2.6340 03	3.3370 05
8.00	5.0210 10	6.3400 05	3.0980 06	2.3800 09	1.1670 09	1.0000 05	6.6630 06	2.6020 03	3.3500 05
8.20	4.8670 10	6.0600 05	3.1020 06	2.3420 09	1.1650 09	1.0000 05	6.5630 06	2.5700 03	3.3620 05
8.40	4.7210 10	5.7980 05	3.1060 06	2.3060 09	1.1640 09	1.0000 05	6.4670 06	2.5380 03	3.3740 05
8.60	4.5820 10	5.5520 05	3.1090 06	2.2700 09	1.1620 09	1.0000 05	6.3740 06	2.5080 03	3.3850 05
8.80	4.4500 10	5.3210 05	3.1120 06	2.2360 09	1.1590 09	1.0000 05	6.2830 06	2.4790 03	3.3970 05
9.00	4.3240 10	5.1030 05	3.1140 06	2.2030 09	1.1570 09	1.0000 05	6.1960 06	2.4500 03	3.4080 05
9.20	4.2040 10	4.8980 05	3.1170 06	2.1700 09	1.1540 09	1.0000 05	6.1110 06	2.4220 03	3.4190 05
9.40	4.0890 10	4.7050 05	3.1190 06	2.1390 09	1.1520 09	1.0000 05	6.0290 06	2.3940 03	3.4300 05
9.60	3.9790 10	4.5230 05	3.1200 06	2.1080 09	1.1490 09	1.0000 05	5.9500 06	2.3670 03	3.4410 05
9.80	3.8740 10	4.3500 05	3.1220 06	2.0790 09	1.1450 09	1.0000 05	5.8720 06	2.3410 03	3.4510 05
10.00	3.7740 10	4.1870 05	3.1230 06	2.0500 09	1.1420 09	1.0000 05	5.7980 06	2.3160 03	3.4620 05

T= 250 K, H= 20 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	8.900-06	2.780-13
2	2*N03 >>> 2*N02 + O2	4.710-17	7.460-62
3	N02 + N03 >>> N02 + N0 + O2	4.210-15	1.190-34
4	N03 + N0 >>> 2*N02	1.900-11	3.290-32
5	N0 + O3 >>> N02 + O2	6.360-15	1.080-56
6	N02 + O3 >>> N03 + O2	6.650-18	8.330-39
7	HN03 + M >>> H0 + N02 + M	1.180-27	5.410-12
8	HN03 + H0 >>> H2O + N03	8.000-14	2.800-29
9	O + O + M >>> O2 + M	5.920-15	6.950-57
10	O + O2 + M >>> O3 + M	1.270-15	3.550-12
11	O + O3 >>> 2*O2	1.920-15	0.0
12	O + N0 + M >>> N02 + M	2.490-13	1.850-47
13	O + N02 >>> N0 + O2	5.120-12	6.270-53
14	O + N02 + M >>> N03 + M	1.560-13	1.240-23
15	H0 + H0 >>> H2O + O	1.090-12	9.770-27
16	O2 + 2*N0 >>> 2*N02	2.750-38	3.880-35
17	N02 + H-NU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.200-11	7.730-25
19	O + H02 >>> H0 + O2	1.080-11	5.450-60
20	O2 + H + M >>> H02 + M	6.130-14	7.850-31
21	O3 + H >>> H0 + O2	1.270-11	0.0
22	O3 + H0 >>> H02 + O2	2.750-14	1.040-48
23	O3 + H02 >>> H0 + 2*O2	4.450-16	2.410-68
24	H + H0 + M >>> H2O + M	1.250-12	0.0
25	H + H02 >>> 2*H0	9.400-12	1.620-46
26	H + H02 >>> H2 + O2	1.040-11	2.550-61
27	H + H2O >>> H2 + H0	2.340-28	1.140-15
28	H + H2O2 >>> H2 + H02	8.360-15	5.620-29
29	H + H2O2 >>> H0 + H2O	1.090-14	1.350-75

30	2*H0 + M >>> H2O2 + M	7.110-13	1.910-31
31	H0 + H02 >>> H2O + O2	1.120-11	3.890-74
32	2*H02 >>> H2O2 + O2	2.300-12	8.580-49
33	H02 + H2O >>> H2O2 + H0	1.020-39	5.490-13
34	N0 + H + M >>> HNO + M	3.870-14	6.060-33
35	N0 + H0 >>> N02 + H	3.050-34	3.010-11
36	N0 + H0 + M >>> HNO2 + M	5.090-12	1.930-28
37	N0 + H02 >>> N02 + H0	1.650-13	7.470-21
38	H + H + M >>> H2 + M	1.560-15	0.0
39	HNO4 + M >>> H02 + N02 + M	2.060-05	3.790-13
40	CLN03 + M >>> CLO + N02 + M	7.180-07	3.180-13

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	N0	O3	O2	H003	H0	H20
0.0	4.0000 08	8.0000 09	5.0000 05	1.5000 09	4.5000 12	3.8000 17	4.0000 09	1.0000 06	3.8000 12
0.20	4.0000 08	8.0080 09	5.4560 05	1.4910 09	4.5000 12	3.8000 17	4.0000 09	9.7710 05	3.8000 12
0.40	4.0000 08	8.0170 09	5.9100 05	1.4830 09	4.5000 12	3.8000 17	4.0000 09	9.5260 05	3.8000 12
0.60	4.0000 08	8.0250 09	6.3620 05	1.4740 09	4.5000 12	3.8000 17	4.0000 09	9.2900 05	3.8000 12
0.80	4.0000 08	8.0340 09	6.8120 05	1.4660 09	4.5000 12	3.8000 17	4.0000 09	9.0620 05	3.8000 12
1.00	4.0000 08	8.0420 09	7.2600 05	1.4580 09	4.5000 12	3.8000 17	4.0000 09	8.8420 05	3.8000 12
1.20	4.0000 08	8.0500 09	7.7060 05	1.4490 09	4.5000 12	3.8000 17	4.0000 09	8.6290 05	3.8000 12
1.40	4.0000 08	8.0580 09	8.1500 05	1.4410 09	4.5000 12	3.8000 17	4.0000 09	8.4240 05	3.8000 12
1.60	4.0000 08	8.0670 09	8.5920 05	1.4330 09	4.5000 12	3.8000 17	4.0000 09	8.2260 05	3.8000 12
1.80	4.0000 08	8.0750 09	9.0320 05	1.4250 09	4.5000 12	3.8000 17	4.0000 09	8.0340 05	3.8000 12
2.00	4.0000 08	8.0830 09	9.4700 05	1.4170 09	4.5000 12	3.8000 17	4.0000 09	7.8500 05	3.8000 12
2.20	4.0000 08	8.0910 09	9.9070 05	1.4080 09	4.5000 12	3.8000 17	4.0000 09	7.6710 05	3.8000 12
2.40	4.0000 08	8.0990 09	1.0340 06	1.4000 09	4.5000 12	3.8000 17	4.0000 09	7.4990 05	3.8000 12
2.60	4.0000 08	8.1070 09	1.0770 06	1.3920 09	4.5000 12	3.8000 17	4.0000 09	7.3330 05	3.8000 12
2.80	4.0000 08	8.1150 09	1.1210 06	1.3840 09	4.5000 12	3.8000 17	4.0000 09	7.1720 05	3.8000 12
3.00	4.0000 08	8.1220 09	1.1640 06	1.3770 09	4.5000 12	3.8000 17	4.0000 09	7.0170 05	3.8000 12
3.20	4.0000 08	8.1300 09	1.2060 06	1.3690 09	4.5000 12	3.8000 17	4.0000 09	6.8680 05	3.8000 12
3.40	4.0000 08	8.1380 09	1.2490 06	1.3610 09	4.5000 12	3.8000 17	4.0000 09	6.7230 05	3.8000 12
3.60	4.0000 08	8.1460 09	1.2910 06	1.3530 09	4.5000 12	3.8000 17	4.0000 09	6.5840 05	3.8000 12
3.80	4.0000 08	8.1530 09	1.3340 06	1.3450 09	4.5000 12	3.8000 17	4.0000 09	6.4490 05	3.8000 12
4.00	4.0000 08	8.1610 09	1.3760 06	1.3380 09	4.5000 12	3.8000 17	4.0000 09	6.3190 05	3.8000 12
4.20	4.0000 08	8.1690 09	1.4180 06	1.3300 09	4.5000 12	3.8000 17	4.0000 09	6.1940 05	3.8000 12
4.40	4.0000 08	8.1770 09	1.4600 06	1.3220 09	4.5000 12	3.8000 17	4.0000 09	6.0730 05	3.8000 12
4.60	4.0000 08	8.1840 09	1.5020 06	1.3150 09	4.5000 12	3.8000 17	4.0000 09	5.9560 05	3.8000 12
4.80	4.0000 08	8.1910 09	1.5430 06	1.3070 09	4.5000 12	3.8000 17	4.0000 09	5.8430 05	3.8000 12
5.00	4.0000 08	8.1980 09	1.5850 06	1.2920 09	4.5000 12	3.8000 17	4.0000 09	5.7330 05	3.8000 12
5.20	4.0000 08	8.2060 09	1.6260 06	1.2850 09	4.5000 12	3.8000 17	4.0000 09	5.6280 05	3.8000 12
5.40	4.0000 08	8.2130 09	1.6670 06	1.2780 09	4.5000 12	3.8000 17	4.0000 09	5.5260 05	3.8000 12
5.60	4.0000 08	8.2200 09	1.7080 06	1.2700 09	4.5000 12	3.8000 17	4.0000 09	5.4280 05	3.8000 12
5.80	4.0000 08	8.2280 09	1.7490 06	1.2630 09	4.5000 12	3.8000 17	4.0000 09	5.3330 05	3.8000 12
6.00	4.0000 08	8.2350 09	1.7900 06	1.2560 09	4.5000 12	3.8000 17	4.0000 09	5.2420 05	3.8000 12
6.20	4.0000 08	8.2420 09	1.8310 06	1.2500 09	4.5000 12	3.8000 17	4.0000 09	5.1530 05	3.8000 12
6.40	4.0000 08	8.2490 09	1.8710 06	1.2490 09	4.5000 12	3.8000 17	4.0000 09	5.0680 05	3.8000 12
6.60	4.0000 08	8.2560 09	1.9110 06	1.2420 09	4.5000 12	3.8000 17	4.0000 09	4.9850 05	3.8000 12
6.80	4.0000 08	8.2630 09	1.9520 06	1.2350 09	4.5000 12	3.8000 17	4.0000 09	4.9050 05	3.8000 12
7.00	4.0000 08	8.2700 09	1.9920 06	1.2270 09	4.5000 12	3.8000 17	4.0000 09	4.8280 05	3.8000 12
7.20	4.0000 08	8.2770 09	2.0320 06	1.2200 09	4.5000 12	3.8000 17	4.0000 09	4.7540 05	3.8000 12
7.40	4.0000 08	8.2840 09	2.0720 06	1.2140 09	4.5000 12	3.8000 17	4.0000 09	4.6820 05	3.8000 12
7.60	4.0000 08	8.2910 09	2.1120 06	1.2070 09	4.5000 12	3.8000 17	4.0000 09	4.6130 05	3.8000 12
7.80	4.0000 08	8.2980 09	2.1520 06	1.2000 09	4.5000 12	3.8000 17	4.0000 09	4.5460 05	3.8000 12
8.00	4.0000 08	8.3050 09	2.1910 06	1.1930 09	4.5000 12	3.8000 17	4.0000 09	4.4810 05	3.8000 12
8.20	4.0000 08	8.3110 09	2.2310 06	1.1860 09	4.5000 12	3.8000 17	4.0000 09	4.4190 05	3.8000 12
8.40	4.0000 08	8.3180 09	2.2700 06	1.1790 09	4.5000 12	3.8000 17	4.0000 09	4.3580 05	3.8000 12
8.60	4.0000 08	8.3250 09	2.3100 06	1.1720 09	4.5000 12	3.8000 17	4.0000 09	4.3000 05	3.8000 12
8.80	4.0000 08	8.3310 09	2.3490 06	1.1660 09	4.5000 12	3.8000 17	4.0000 09	4.2440 05	3.8000 12
9.00	4.0000 08	8.3380 09	2.3880 06	1.1590 09	4.5000 12	3.8000 17	4.0000 09	4.1890 05	3.8000 12
9.20	4.0000 08	8.3450 09	2.4270 06	1.1530 09	4.5000 12	3.8000 17	4.0000 09	4.1370 05	3.8000 12
9.40	4.0000 08	8.3510 09	2.4660 06	1.1460 09	4.5000 12	3.8000 17	4.0000 09	4.0860 05	3.8000 12
9.60	4.0000 08	8.3580 09	2.5050 06	1.1390 09	4.5000 12	3.8000 17	4.0000 09	4.0370 05	3.8000 12
9.80	4.0000 08	8.3640 09	2.5440 06	1.1330 09	4.5000 12	3.8000 17	4.0000 09	3.9900 05	3.8000 12
10.00	4.0000 08	8.3700 09	2.5830 06	1.1260 09	4.5000 12	3.8000 17	4.0000 09	3.9440 05	3.8000 12

TIME (s)	O	H	H2	H02	H2O2	HNO	HNO2	HNO4	CLN03
0.0	1.0000 06	1.0000 06	1.0000 06	2.0000 07	1.3000 09	1.0000 04	1.0000 06	4.0000 09	1.0000 08
0.20	3.5340-02	4.7800-08	1.0000 06	2.1020 07	1.3000 09	1.0000 04	1.0020 06	4.0000 09	1.0000 08
0.40	3.5240-02	4.6610-08	1.0000 06	2.1030 07	1.3000 09	1.0000 06	1.0030 06	4.0000 09	1.0000 08
0.60	3.5140-02	4.5450-08	1.0000 06	2.1050 07	1.3000 09	1.0000 06	1.0040 06	4.0000 09	1.0010 08
0.80	3.5040-02	4.4330-08	1.0000 06	2.1070 07	1.3000 09	1.0000 06	1.0060 06	4.0000 09	1.0010 08
1.00	3.4950-02	4.3260-08	1.0000 06	2.1080 07	1.3000 09	1.0000 06	1.0070 06	4.0000 09	1.0010 08
1.20	3.4870-02	4.2220-08	1.0000 06	2.1100 07	1.3000 09	1.0000 04	1.0080 06	4.0000 09	1.0010 08
1.40	3.4790-02	4.1210-08	1.0000 06	2.1110 07	1.3000 09	1.0000 04	1.0100 06	4.0000 09	1.0010 08
1.60	3.4720-02	4.0240-08	1.0000 06	2.1130 07	1.3000 09	1.0000 04	1.0110 06	4.0000 09	1.0010 08
1.80	3.4650-02	3.9310-08	1.0000 06	2.1140 07	1.3000 09	1.0000 06	1.0120 06	4.0000 09	1.0020 08
2.00	3.4580-02	3.8400-08	1.0000 06	2.1150 07	1.3000 09	1.0000 06	1.0130 06	4.0000 09	1.0020 08
2.20	3.4520-02	3.7530-08	1.0000 06	2.1170 07	1.3000 09	1.0000 06	1.0140 06	4.0000 09	1.0020 08
2.40	3.4460-02	3.6690-08	1.0000 06	2.1180 07	1.3000 09	1.0000 06	1.0150 06	4.0000 09	1.0020 08
2.60	3.4400-02	3.5870-08	1.0000 06	2.1190 07	1.3000 09	1.0000 05	1.0160 06	4.0000 09	1.0020 08
2.80	3.4350-02	3.5090-08	1.0000 06	2.1200 07	1.3000 09	1.0000 06	1.0170 06	4.0000 09	1.0020 08
3.00	3.4300-02	3.4330-08	1.0000 06	2.1210 07	1.3000 09	1.0000 06	1.0180 06	4.0000 09	1.0030 08
3.20	3.4250-02	3.3600-08	1.0000 06	2.1230 07	1.3000 09	1.0000 04	1.0190 06	4.0000 09	1.0030 08
3.40	3.4210-02	3.2890-08	1.0000 06	2.1240 07	1.3000 09	1.0000 06	1.0200 06	4.0000 09	1.0030 08
3.60	3.4170-02	3.2210-08	1.0000 06	2.1250 07	1.3000 09	1.0000 06	1.0210 06	4.0000 09	1.0030 08
3.80	3.4130-02	3.1550-08	1.0000 06	2.1260 07	1.3000 09	1.0000 06	1.0220 06	4.0000 09	1.0030 08
4.00	3.4090-02	3.0920-08	1.0000 06	2.1260 07	1.3000 09	1.0000 06	1.0230 06	4.0000 09	1.0040 08
4.20	3.4050-02	3.0300-08	1.0000 06	2.1270 07	1.3000 09	1.0000 06	1.0240 06	4.0000 09	1.0040 08
4.40	3.4020-02	2.9710-08	1.0000 06	2.1280 07	1.3000 09	1.0000 06	1.0250 06	4.0000 09	1.0040 08
4.60	3.3990-02	2.9140-08	1.0000 06	2.1290 07	1.3000 09	1.0000 06	1.0260 06	4.0000 09	1.0040 08
4.80	3.3960-02	2.8580-08	1.0000 06	2.1300 07	1.3000 09	1.0000 06	1.0270 06	4.0000 09	1.0040 08
5.00	3.3930-02	2.8050-08	1.0000 06	2.1310 07	1.3000 09	1.0000 06	1.0280 06	4.0000 09	1.0050 08
5.20	3.3900-02	2.7530-08	1.0000 06	2.1310 07	1.3000 09	1.0000 04	1.0290 06	4.0000 09	1.0050 08
5.40	3.3880-02	2.7040-08	1.0000 06	2.1320 07	1.3000 09	1.0000 06	1.0290 06	4.0000 09	1.0050 08
5.60	3.3850-02	2.6560-08	1.0000 06	2.1330 07	1.3000 09	1.0000 05	1.0300 06	4.0000 09	1.0050 08
5.80	3.3830-02	2.6090-08	1.0000 06	2.1330 07	1.3000 09	1.0000 06	1.0310 06	4.0000 09	1.0050 08
6.00	3.3810-02	2.5640-08	1.0000 06	2.1340 07	1.3000 09	1.0000 06	1.0310 06	4.0000 09	1.0060 08
6.20	3.3790-02	2.5210-08	1.0000 06	2.1350 07	1.3000 09	1.0000 06	1.0320 06	4.0000 09	1.0060 08
6.40	3.3770-02	2.4790-08	1.0000 06	2.1350 07	1.3000 09	1.0000 06	1.0330 06	4.0000 09	1.0060 08
6.60	3.3750-02	2.4390-08	1.0000 06	2.1360 07	1.3000 09	1.0000 06	1.0330 06	4.0000 09	1.0060 08
6.80	3.3730-02	2.4000-08	1.0000 06	2.1360 07	1.3000 09	1.0000 06	1.0340 06	4.0000 09	1.0060 08
7.00	3.3710-02	2.3620-08	1.0000 06	2.1370 07	1.3000 09	1.0000 06	1.0340 06	4.0000 09	1.0060 08
7.20	3.3700-02	2.3240-08	1.0000 06	2.1370 07	1.3000 09	1.0000 06	1.0350 06	4.0000 09	1.0070 08
7.40	3.3680-02	2.2910-08	1.0000 06	2.1380 07	1.3000 09	1.0000 06	1.0360 06	4.0000 09	1.0070 08
7.60	3.3670-02	2.2570-08	1.0000 06	2.1380 07	1.3000 09	1.0000 04	1.0370 06	4.0000 09	1.0070 08
7.80	3.3650-02	2.2240-08	1.0000 06	2.1390 07	1.3000 09	1.0000 06	1.0370 06	4.0000 09	1.0070 08
8.00	3.3640-02	2.1920-08	1.0000 06	2.1390 07	1.3000 09	1.0000 06	1.0380 06	4.0000 09	1.0080 08
8.20	3.3630-02	2.1620-08	1.0000 06	2.1400 07	1.3000 09	1.0000 06	1.0390 06	4.0000 09	1.0080 08
8.40	3.3620-02	2.1320-08	1.0000 06	2.1400 07	1.3000 09	1.0000 04	1.0390 06	4.0000 09	1.0080 08
8.60	3.3600-02	2.1040-08	1.0000 06	2.1410 07	1.3000 09	1.0000 06	1.0400 06	4.0000 09	1.0080 08
8.80	3.3590-02	2.0760-08	1.0000 06	2.1410 07	1.3000 09	1.0000 06	1.0400 06	4.0000 09	1.0080 08
9.00	3.3580-02	2.0500-08	1.0000 06	2.1420 07	1.3000 09	1.0000 06	1.0400 06	4.0000 09	1.0080 08
9.20	3.3570-02	2.0240-08	1.0000 06	2.1420 07	1.3000 09	1.0000 06	1.0410 06	4.0000 09	1.0080 08
9.40	3.3560-02	1.9990-08	1.0000 06	2.1420 07	1.3000 09	1.0000 06	1.0410 06	4.0000 09	1.0090 08
9.60	3.3560-02	1.9750-08	1.0000 06	2.1420 07	1.3000 09	1.0000 06	1.0410 06	4.0000 09	1.0090 08
9.80	3.3550-02	1.9520-08	1.0000 06	2.1420 07	1.3000 09	1.0000 04	1.0420 06	4.0000 09	1.0090 08
10.00	3.3540-02	1.9290-08	1.0000 06	2.1430 07	1.3000 09	1.0000 04	1.0420 06	4.0000 09	1.0090 08

T=300 K, H= 20 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	7.200-03	1.310-13
2	2*N03 >>> 2*N02 + O2	2.410-16	1.750-58
3	N02 + N03 >>> N02 + N0 + O2	8.210-15	4.920-35
4	N03 + N0 >>> 2*N02	1.900-11	6.980-29
5	N0 + O3 >>> N02 + O2	1.670-14	2.850-49
6	N02 + O3 >>> N03 + O2	3.410-17	2.000-34
7	HN03 + M >>> H0 + N02 + M	8.820-21	2.860-12
8	HN03 + H0 >>> H2O + N03	8.000-14	1.250-26
9	O + O + M >>> O2 + M	2.710-15	2.650-57
10	O + O2 + M >>> O3 + M	7.520-16	6.040-09
11	O + O3 >>> 2*O2	8.900-15	0.0
12	O + N0 + M >>> N02 + M	1.410-13	5.520-38
13	O + N02 >>> N0 + O2	6.250-12	3.730-46
14	O + N02 + M >>> N03 + M	1.300-13	1.040-23
15	H0 + H0 >>> H2O + O	1.570-12	4.630-24
16	O2 + 2*N0 >>> 2*N02	1.930-38	2.260-31
17	N02 + H-NU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.200-11	2.160-22
19	O + H02 >>> H0 + O2	1.510-11	8.510-57
20	O2 + H + M >>> H02 + M	3.660-14	2.980-24
21	O3 + H >>> H0 + O2	1.790-11	1.620-69
22	O3 + H0 >>> H02 + O2	5.350-14	8.980-43
23	O3 + H02 >>> H0 + 2*O2	1.040-15	1.530-63
24	H + H0 + M >>> H2O + M	6.470-13	5.240-75
25	H + H02 >>> 2*H0	1.770-11	1.140-40
26	H + H02 >>> H2 + O2	1.310-11	6.800-53
27	H + H2O >>> H2 + H0	2.180-25	6.410-15
28	H + H2O2 >>> H2 + H02	2.130-14	2.960-26
29	H + H2O2 >>> H0 + H2O	2.760-14	2.950-65

30	2*H0 + M >>> H2O2 + M	3.250-13	2.370-24
31	H0 + H02 >>> H2O + O2	1.570-11	1.980-63
32	2*H02 >>> H2O2 + O2	3.210-12	1.980-42
33	H02 + H2O >>> H2O2 + H0	6.110-35	8.570-13
34	NO + H + M >>> HNO + M	2.640-14	6.770-26
35	NO + H0 >>> NO2 + H	7.190-34	4.920-11
36	NO + H0 + M >>> HNO2 + M	2.020-12	1.140-21
37	NO + H02 >>> NO2 + H0	3.660-13	3.740-19
38	H + H + M >>> H2 + M	1.300-15	1.120-68
39	HNO4 + M >>> H02 + NO2 + M	1.400-02	1.680-13
40	CLN03 + M >>> CL0 + NO2 + M	1.050-03	1.570-13

HAPP RESIDENCE TIME STUDY

TIME (s)	N205	N02	N03	N0	03	02	HN03	H0	H20
0.0	4.0000 08	8.0000 09	5.0000 05	1.5000 09	4.5000 12	3.9000 17	4.0000 09	1.0000 06	3.8000 12
0.20	3.9940 08	8.0340 09	1.3160 06	1.4780 09	4.5000 12	3.8000 17	4.0000 09	9.8030 05	3.8000 12
0.40	3.9880 08	8.0680 09	2.1280 06	1.4560 09	4.5000 12	3.8000 17	4.0000 09	9.6750 05	3.8000 12
0.60	3.9830 08	8.1010 09	2.9360 06	1.4340 09	4.5000 12	3.8000 17	4.0000 09	9.6660 05	3.8000 12
0.80	3.9770 08	8.1340 09	3.7390 06	1.4120 09	4.5000 12	3.8000 17	4.0000 09	9.7680 05	3.8000 12
1.00	3.9710 08	8.1660 09	4.5390 06	1.3910 09	4.5000 12	3.8000 17	4.0000 09	9.9750 05	3.8000 12
1.20	3.9660 08	8.1980 09	5.3350 06	1.3700 09	4.5000 12	3.8000 17	4.0000 09	1.0280 06	3.8000 12
1.40	3.9600 08	8.2300 09	6.1260 06	1.3500 09	4.5000 12	3.8000 17	4.0000 09	1.0640 06	3.8000 12
1.60	3.9540 08	8.2620 09	6.9150 06	1.3300 09	4.5000 12	3.8000 17	4.0000 09	1.1170 06	3.8000 12
1.80	3.9490 08	8.2930 09	7.6990 06	1.3100 09	4.5000 12	3.8000 17	4.0000 09	1.1730 06	3.8000 12
2.00	3.9430 08	8.3240 09	8.4810 06	1.2900 09	4.5000 12	3.8000 17	4.0000 09	1.2380 06	3.8000 12
2.20	3.9370 08	8.3540 09	9.2590 06	1.2710 09	4.5000 12	3.8000 17	4.0000 09	1.3090 06	3.8000 12
2.40	3.9320 08	8.3840 09	1.0030 07	1.2520 09	4.5000 12	3.8000 17	4.0000 09	1.3880 06	3.8000 12
2.60	3.9260 08	8.4140 09	1.0910 07	1.2330 09	4.5000 12	3.8000 17	4.0000 09	1.4720 06	3.8000 12
2.80	3.9200 08	8.4440 09	1.1570 07	1.2150 09	4.5000 12	3.8000 17	4.0000 09	1.5630 06	3.8000 12
3.00	3.9150 08	8.4730 09	1.2340 07	1.1960 09	4.5000 12	3.8000 17	4.0000 09	1.6590 06	3.8000 12
3.20	3.9090 08	8.5020 09	1.3100 07	1.1780 09	4.5000 12	3.8000 17	4.0000 09	1.7600 06	3.8000 12
3.40	3.9040 08	8.5310 09	1.3870 07	1.1610 09	4.5000 12	3.8000 17	4.0000 09	1.8660 06	3.8000 12
3.60	3.8980 08	8.5590 09	1.4620 07	1.1430 09	4.5000 12	3.8000 17	4.0000 09	1.9760 06	3.8000 12
3.80	3.8920 08	8.5870 09	1.5380 07	1.1260 09	4.5000 12	3.8000 17	4.0000 09	2.0910 06	3.8000 12
4.00	3.8870 08	8.6150 09	1.6130 07	1.1090 09	4.5000 12	3.8000 17	4.0000 09	2.2090 06	3.8000 12
4.20	3.8810 08	8.6430 09	1.6880 07	1.0930 09	4.5000 12	3.8000 17	4.0000 09	2.3310 06	3.8000 12
4.40	3.8760 08	8.6700 09	1.7630 07	1.0760 09	4.5000 12	3.8000 17	4.0000 09	2.4570 06	3.8000 12
4.60	3.8700 08	8.6970 09	1.8380 07	1.0600 09	4.5000 12	3.8000 17	4.0000 09	2.5850 06	3.8000 12
4.80	3.8650 08	8.7240 09	1.9130 07	1.0440 09	4.5000 12	3.8000 17	4.0000 09	2.7160 06	3.8000 12
5.00	3.8590 08	8.7500 09	1.9870 07	1.0280 09	4.5000 12	3.8000 17	4.0000 09	2.8500 06	3.8000 12
5.20	3.8540 08	8.7760 09	2.0610 07	1.0130 09	4.5000 12	3.8000 17	4.0000 09	2.9860 06	3.8000 12
5.40	3.8480 08	8.8020 09	2.1350 07	9.9780 08	4.4990 12	3.8000 17	4.0000 09	3.1250 06	3.8000 12
5.60	3.8430 08	8.8280 09	2.2090 07	9.8280 08	4.4990 12	3.8000 17	4.0000 09	3.2650 06	3.8000 12
5.80	3.8370 08	8.8530 09	2.2820 07	9.6800 08	4.4990 12	3.8000 17	4.0000 09	3.4070 06	3.8000 12
6.00	3.8320 08	8.8780 09	2.3560 07	9.5350 08	4.4990 12	3.8000 17	4.0000 09	3.5510 06	3.8000 12
6.20	3.8260 08	8.9030 09	2.4290 07	9.3910 08	4.4990 12	3.8000 17	4.0000 09	3.6970 06	3.8000 12
6.40	3.8210 08	8.9280 09	2.5020 07	9.2500 08	4.4990 12	3.8000 17	4.0000 09	3.8430 06	3.8000 12
6.60	3.8150 08	8.9520 09	2.5750 07	9.1110 08	4.4990 12	3.8000 17	4.0000 09	3.9910 06	3.8000 12
6.80	3.8100 08	8.9760 09	2.6480 07	8.9740 08	4.4990 12	3.8000 17	4.0000 09	4.1410 06	3.8000 12
7.00	3.8040 08	9.0000 09	2.7200 07	8.8380 08	4.4990 12	3.8000 17	4.0000 09	4.2910 06	3.8000 12
7.20	3.7990 08	9.0240 09	2.7930 07	8.7050 08	4.4990 12	3.8000 17	4.0000 09	4.4410 06	3.8000 12
7.40	3.7940 08	9.0470 09	2.8650 07	8.5740 08	4.4990 12	3.8000 17	4.0000 09	4.5930 06	3.8000 12
7.60	3.7880 08	9.0710 09	2.9380 07	8.4450 08	4.4990 12	3.8000 17	4.0000 09	4.7450 06	3.8000 12
7.80	3.7830 08	9.0940 09	3.0100 07	8.3180 08	4.4990 12	3.8000 17	4.0000 09	4.8980 06	3.8000 12
8.00	3.7770 08	9.1170 09	3.0820 07	8.1920 08	4.4990 12	3.8000 17	4.0000 09	5.0510 06	3.8000 12
8.20	3.7720 08	9.1390 09	3.1540 07	8.0690 08	4.4990 12	3.8000 17	4.0010 09	5.2050 06	3.8000 12
8.40	3.7670 08	9.1610 09	3.2260 07	7.9470 08	4.4990 12	3.8000 17	4.0010 09	5.3590 06	3.8000 12
8.60	3.7610 08	9.1840 09	3.2970 07	7.8270 08	4.4990 12	3.8000 17	4.0010 09	5.5130 06	3.8000 12
8.80	3.7560 08	9.2060 09	3.3690 07	7.7090 08	4.4990 12	3.8000 17	4.0010 09	5.6670 06	3.8000 12
9.00	3.7510 08	9.2270 09	3.4410 07	7.5930 08	4.4990 12	3.8000 17	4.0010 09	5.8210 06	3.8000 12
9.20	3.7450 08	9.2490 09	3.5120 07	7.4780 08	4.4990 12	3.8000 17	4.0010 09	5.9760 06	3.8000 12
9.40	3.7400 08	9.2700 09	3.5840 07	7.3650 08	4.4990 12	3.8000 17	4.0010 09	6.1300 06	3.8000 12
9.60	3.7350 08	9.2910 09	3.6550 07	7.2540 08	4.4990 12	3.8000 17	4.0010 09	6.2840 06	3.8000 12
9.80	3.7300 08	9.3120 09	3.7260 07	7.1440 08	4.4990 12	3.8000 17	4.0010 09	6.4380 06	3.8000 12
10.00	3.7240 08	9.3330 09	3.7970 07	7.0360 08	4.4990 12	3.8000 17	4.0010 09	6.5910 06	3.8000 12

TIME (S)	O	M	H ²	H ² O ₂	H ² O	HNO ₂	HNO ₄	CLNO ₃
0.0	1.0000 06	1.0000 06	1.0000 06	2.0000 07	1.3000 09	1.0000 06	1.0000 09	1.0000 08
0.20	9.5040 01	7.2940-07	1.0000 06	3.2200 07	1.3000 09	1.0000 06	4.0000 09	9.9990 07
0.40	9.5040 01	7.1980-07	1.0000 06	4.3360 07	1.3000 09	1.0000 06	3.9780 09	9.9980 07
0.60	9.5040 01	7.1910-07	1.0000 06	5.4480 07	1.3000 09	1.0000 06	3.9670 09	9.9960 07
0.80	9.5040 01	7.2670-07	1.0000 06	6.5540 07	1.3000 09	1.0000 06	3.9550 09	9.9950 07
1.00	9.5040 01	7.4220-07	1.0000 06	7.6560 07	1.3000 09	1.0000 06	3.9440 09	9.9940 07
1.20	9.5040 01	7.6490-07	1.0000 06	8.7540 07	1.3000 09	1.0000 06	3.9330 09	9.9930 07
1.40	9.5040 01	7.9460-07	1.0000 06	9.8470 07	1.3000 09	1.0000 06	3.9220 09	9.9920 07
1.60	9.5040 01	8.3040-07	1.0000 06	1.0930 08	1.3000 09	1.0000 06	3.9110 09	9.9900 07
1.80	9.5040 01	8.7300-07	1.0000 06	1.2020 08	1.3000 09	1.0000 06	3.9010 09	9.9890 07
2.00	9.5040 01	9.2090-07	1.0000 06	1.3100 08	1.3000 09	1.0000 06	3.8900 09	9.9880 07
2.20	9.5040 01	9.7420-07	1.0000 06	1.4170 08	1.3000 09	1.0000 06	3.8790 09	9.9870 07
2.40	9.5040 01	1.0330-06	1.0000 06	1.5240 08	1.3000 09	1.0000 06	3.8680 09	9.9860 07
2.60	9.5040 01	1.0950-06	1.0000 06	1.6310 08	1.3000 09	1.0000 06	3.8570 09	9.9850 07
2.80	9.5050 01	1.1630-06	1.0000 06	1.7370 08	1.3000 09	1.0000 06	3.8460 09	9.9830 07
3.00	9.5050 01	1.2340-06	1.0000 06	1.8430 08	1.3000 09	1.0000 06	3.8360 09	9.9820 07
3.20	9.5050 01	1.3100-06	1.0000 06	1.9480 08	1.3000 09	1.0000 06	3.8250 09	9.9810 07
3.40	9.5050 01	1.3880-06	1.0000 06	2.0530 08	1.3000 09	1.0000 06	3.8140 09	9.9800 07
3.60	9.5050 01	1.4710-06	1.0000 06	2.1570 08	1.3000 09	1.0000 06	3.8040 09	9.9790 07
3.80	9.5050 01	1.5560-06	1.0000 06	2.2610 08	1.3000 09	1.0000 06	3.7930 09	9.9780 07
4.00	9.5060 01	1.6440-06	1.0000 06	2.3640 08	1.3000 09	1.0000 06	3.7830 09	9.9770 07
4.20	9.5060 01	1.7350-06	1.0000 06	2.4680 08	1.3000 09	1.0000 06	3.7720 09	9.9750 07
4.40	9.5060 01	1.8280-06	1.0000 06	2.5720 08	1.3000 09	1.0000 06	3.7620 09	9.9740 07
4.60	9.5070 01	1.9230-06	1.0000 06	2.6760 08	1.3000 09	1.0000 06	3.7510 09	9.9730 07
4.80	9.5070 01	2.0210-06	1.0000 06	2.7740 08	1.3000 09	1.0000 06	3.7410 09	9.9720 07
5.00	9.5070 01	2.1210-06	1.0000 06	2.8750 08	1.3000 09	1.0000 06	3.7300 09	9.9710 07
5.20	9.5080 01	2.2220-06	1.0000 06	2.9760 08	1.3010 09	1.0000 06	3.7200 09	9.9700 07
5.40	9.5080 01	2.3250-06	1.0000 06	3.0770 08	1.3010 09	1.0000 06	3.7100 09	9.9690 07
5.60	9.5080 01	2.4300-06	1.0000 06	3.1770 08	1.3010 09	1.0000 06	3.6990 09	9.9680 07
5.80	9.5090 01	2.5360-06	1.0000 06	3.2760 08	1.3010 09	1.0000 06	3.6890 09	9.9670 07
6.00	9.5090 01	2.6430-06	1.0000 06	3.3760 08	1.3010 09	1.0000 06	3.6790 09	9.9650 07
6.20	9.5100 01	2.7510-06	1.0000 06	3.4740 08	1.3010 09	1.0000 06	3.6690 09	9.9640 07
6.40	9.5110 01	2.8600-06	1.0000 06	3.5730 08	1.3010 09	1.0000 06	3.6590 09	9.9630 07
6.60	9.5120 01	3.0810-06	1.0000 06	3.6700 08	1.3010 09	1.0000 06	3.6480 09	9.9620 07
6.80	9.5120 01	3.1930-06	1.0000 06	3.7680 08	1.3010 09	1.0000 06	3.6380 09	9.9610 07
7.00	9.5130 01	3.3060-06	1.0000 06	3.8650 08	1.3010 09	1.0000 06	3.6280 09	9.9600 07
7.20	9.5140 01	3.4190-06	1.0000 06	3.9610 08	1.3010 09	1.0000 06	3.6180 09	9.9590 07
7.40	9.5150 01	3.5320-06	1.0000 06	4.0570 08	1.3020 09	1.0000 06	3.6080 09	9.9580 07
7.60	9.5150 01	3.6460-06	1.0000 06	4.1530 08	1.3020 09	1.0000 06	3.5980 09	9.9570 07
7.80	9.5150 01	3.7600-06	1.0000 06	4.2480 08	1.3020 09	1.0000 06	3.5880 09	9.9560 07
8.00	9.5160 01	3.8750-06	1.0000 06	4.3430 08	1.3020 09	1.0000 06	3.5780 09	9.9550 07
8.20	9.5170 01	3.9890-06	1.0000 06	4.4380 08	1.3020 09	1.0000 06	3.5680 09	9.9540 07
8.40	9.5180 01	4.1040-06	1.0000 06	4.5320 08	1.3020 09	1.0000 06	3.5590 09	9.9520 07
8.60	9.5190 01	4.2190-06	1.0000 06	4.6250 08	1.3020 09	1.0000 06	3.5490 09	9.9510 07
8.80	9.5200 01	4.3340-06	1.0000 06	4.7180 08	1.3020 09	1.0000 06	3.5390 09	9.9500 07
9.00	9.5210 01	4.4490-06	1.0000 06	4.8110 08	1.3030 09	1.0000 06	3.5290 09	9.9490 07
9.20	9.5220 01	4.5640-06	1.0000 06	4.9030 08	1.3030 09	1.0000 06	3.5200 09	9.9480 07
9.40	9.5230 01	4.6790-06	1.0000 06	5.0860 08	1.3030 09	1.0000 06	3.5100 09	9.9470 07
9.60	9.5250 01	4.7940-06	1.0000 06	5.1770 08	1.3030 09	1.0000 06	3.5000 09	9.9460 07
9.80	9.5260 01	4.9080-06	1.0000 06	5.2670 08	1.3030 09	1.0000 06	3.4910 09	9.9450 07
10.00							3.4810 09	9.9440 07

T=700 K, H= 20 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	1.060 06	1.090-14
2	2*N03 >>> 2*N02 + O2	2.570-14	7.440-49
3	N02 + N03 >>> N02 + N0 + O2	5.510-14	3.970-34
4	N03 + N0 >>> 2*N02	1.900-11	2.230-19
5	N0 + O3 >>> N02 + O2	2.650-13	4.540-28
6	N02 + O3 >>> N03 + O2	3.620-15	6.560-22
7	HN03 + M >>> H0 + N02 + M	8.810-02	1.470-13
8	HN03 + H0 >>> H2O + N03	8.000-14	4.740-10
9	O + O + M >>> O2 + M	2.090-16	3.010-20
10	O + O2 + M >>> O3 + M	1.220-16	7.380 00
11	O + O3 >>> 2*O2	7.110-13	8.630-43
12	O + N0 + M >>> N02 + M	1.980-14	4.700-11
13	O + N02 >>> N0 + O2	1.110-11	8.500-27
14	O + N02 + M >>> N03 + M	5.560-14	4.440-24
15	H0 + H0 >>> H2O + O	4.530-12	2.040-16
16	O2 + 2*N0 >>> 2*N02	7.040-39	9.370-21
17	N02 + H-NU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.200-11	2.110-15
19	O + H02 >>> H0 + O2	3.910-11	2.190-28
20	O2 + H + M >>> H02 + M	6.050-15	1.360-05
21	O3 + H >>> H0 + O2	4.780-11	5.620-37
22	O3 + H0 >>> H02 + O2	3.590-13	8.310-26
23	O3 + H02 >>> H0 + 2*O2	1.180-14	1.120-49
24	H + H0 + M >>> H2O + M	3.060-14	7.360-27
25	H + H02 >>> 2*H0	1.080-10	5.870-24
26	H + H02 >>> H2 + O2	2.550-11	8.030-29
27	H + H2O >>> H2 + H0	6.560-17	8.900-13
28	H + H2O2 >>> H2 + H02	3.060-13	1.770-18
29	H + H2O2 >>> H0 + H2O	3.980-13	1.020-35

30	2*H0 + M >>> H2O2 + M	2.510-14	2.300-04
31	H0 + H02 >>> H2O + O2	4.060-11	7.730-33
32	2*H02 >>> H2O2 + O2	8.320-12	3.020-24
33	H02 + H2O >>> H2O2 + H0	2.720-21	3.070-12
34	NO + H + M >>> HNO + M	6.400-15	4.830-06
35	NO + H0 >>> NO2 + H	2.230-21	2.020-10
36	NO + H0 + M >>> HNO2 + M	1.050-13	1.300-02
37	NO + H02 >>> NO2 + H0	3.600-12	2.680-14
38	H + H + M >>> H2 + M	5.560-16	4.650-25
39	HNO4 + M >>> H02 + NO2 + M	1.310 06	1.230-14
40	CLN03 + M >>> CLO + NO2 + M	8.250 05	5.490-15

HAPP RESIDENCE TIME STUDY

TIME (s)	N205	N02	N03	N0	O3	O2	HNO3	HO	H2O
0.0	4.0000 08	4.0000 09	5.0000 05	1.5000 09	4.5000 12	3.8000 17	4.0000 09	1.0000 06	3.8000 12
0.20	2.2770-02	4.8420 09	4.5740 08	9.1700 09	3.4220 12	3.8000 17	3.9300 09	2.3030 09	3.8000 12
0.40	1.3440-02	2.7990 09	4.6720 08	1.1270 10	3.0120 12	3.8000 17	3.8620 09	2.3720 09	3.8000 12
0.60	1.0820-02	2.2690 09	4.6390 08	1.1870 10	2.4480 12	3.8000 17	3.7940 09	2.4260 09	3.8000 12
0.80	9.9410-03	2.1190 09	4.5660 08	1.2100 10	2.4250 12	3.8000 17	3.7280 09	2.4710 09	3.8000 12
1.00	9.5560-03	2.0770 09	4.4760 08	1.2210 10	2.2070 12	3.8000 17	3.6630 09	2.5070 09	3.8000 12
1.20	9.3210-03	2.0710 09	4.3780 08	1.2290 10	2.0240 12	3.8000 17	3.5990 09	2.5360 09	3.8000 12
1.40	9.1320-03	2.0740 09	4.2760 08	1.2350 10	1.8680 12	3.8000 17	3.5360 09	2.5590 09	3.8000 12
1.60	8.9570-03	2.0900 09	4.1700 08	1.2410 10	1.7330 12	3.8000 17	3.4740 09	2.5760 09	3.8000 12
1.80	8.7840-03	2.1040 09	4.0620 08	1.2470 10	1.6160 12	3.8000 17	3.4140 09	2.5900 09	3.8000 12
2.00	8.6080-03	2.1180 09	3.9540 08	1.2530 10	1.5120 12	3.8000 17	3.3540 09	2.5990 09	3.8000 12
2.20	8.4310-03	2.1330 09	3.8460 08	1.2580 10	1.4210 12	3.8000 17	3.2950 09	2.6050 09	3.8000 12
2.40	8.2510-03	2.1480 09	3.7380 08	1.2630 10	1.3490 12	3.8000 17	3.2380 09	2.6080 09	3.8000 12
2.60	8.0690-03	2.1620 09	3.6310 08	1.2690 10	1.2660 12	3.8000 17	3.1810 09	2.6090 09	3.8000 12
2.80	7.8860-03	2.1770 09	3.5250 08	1.2740 10	1.2000 12	3.8000 17	3.1260 09	2.6070 09	3.8000 12
3.00	7.7020-03	2.1910 09	3.4210 08	1.2790 10	1.1410 12	3.8000 17	3.0710 09	2.6030 09	3.8000 12
3.20	7.5180-03	2.2050 09	3.3180 08	1.2840 10	1.0860 12	3.8000 17	3.0170 09	2.5980 09	3.8000 12
3.40	7.3350-03	2.2180 09	3.2170 08	1.2890 10	1.0360 12	3.8000 17	2.9650 09	2.5910 09	3.8000 12
3.60	7.1530-03	2.2320 09	3.1180 08	1.2930 10	9.9080 11	3.8000 17	2.9130 09	2.5820 09	3.8000 12
3.80	6.9710-03	2.2450 09	3.0220 08	1.2980 10	9.4870 11	3.8000 17	2.8620 09	2.5730 09	3.8000 12
4.00	6.7920-03	2.2580 09	2.9270 08	1.3030 10	9.0980 11	3.8000 17	2.8120 09	2.5620 09	3.8000 12
4.20	6.6150-03	2.2700 09	2.8350 08	1.3070 10	8.7380 11	3.8000 17	2.7630 09	2.5510 09	3.8000 12
4.40	6.4400-03	2.2820 09	2.7450 08	1.3120 10	8.4040 11	3.8000 17	2.7150 09	2.5390 09	3.8000 12
4.60	6.2670-03	2.2940 09	2.6580 08	1.3160 10	8.0920 11	3.8000 17	2.6680 09	2.5260 09	3.8000 12
4.80	6.0970-03	2.3060 09	2.5730 08	1.3200 10	7.8010 11	3.8000 17	2.6210 09	2.5120 09	3.8000 12
5.00	5.9300-03	2.3170 09	2.4910 08	1.3240 10	7.5290 11	3.8000 17	2.5750 09	2.4980 09	3.8000 12
5.20	5.7670-03	2.3280 09	2.4100 08	1.3280 10	7.2740 11	3.8000 17	2.5300 09	2.4840 09	3.8000 12
5.40	5.6060-03	2.3380 09	2.3330 08	1.3330 10	7.0350 11	3.8000 17	2.4860 09	2.4690 09	3.8000 12
5.60	5.4490-03	2.3490 09	2.2570 08	1.3370 10	6.8090 11	3.8000 17	2.4430 09	2.4540 09	3.8000 12
5.80	5.2950-03	2.3590 09	2.1840 08	1.3400 10	6.5970 11	3.8000 17	2.4000 09	2.4390 09	3.8000 12
6.00	5.1450-03	2.3690 09	2.1130 08	1.3440 10	6.3960 11	3.8000 17	2.3580 09	2.4230 09	3.8000 12
6.20	4.9980-03	2.3780 09	2.0450 08	1.3480 10	6.2070 11	3.8000 17	2.3170 09	2.4080 09	3.8010 12
6.40	4.8550-03	2.3880 09	1.9780 08	1.3520 10	6.0270 11	3.8000 17	2.2770 09	2.3920 09	3.8010 12
6.60	4.7150-03	2.3970 09	1.9140 08	1.3550 10	5.8560 11	3.8000 17	2.2370 09	2.3760 09	3.8010 12
6.80	4.5790-03	2.4060 09	1.8520 08	1.3590 10	5.6950 11	3.8000 17	2.1980 09	2.3600 09	3.8010 12
7.00	4.4470-03	2.4140 09	1.7920 08	1.3630 10	5.5410 11	3.8000 17	2.1600 09	2.3440 09	3.8010 12
7.20	4.3180-03	2.4230 09	1.7340 08	1.3660 10	5.3950 11	3.8000 17	2.1220 09	2.3280 09	3.8010 12
7.40	4.1930-03	2.4310 09	1.6780 08	1.3690 10	5.2550 11	3.8000 17	2.0850 09	2.3120 09	3.8010 12
7.60	4.0710-03	2.4390 09	1.6240 08	1.3730 10	5.1220 11	3.8000 17	2.0490 09	2.2960 09	3.8010 12
7.80	3.9530-03	2.4470 09	1.5720 08	1.3760 10	4.9950 11	3.8000 17	2.0130 09	2.2800 09	3.8010 12
8.00	3.8380-03	2.4540 09	1.5210 08	1.3790 10	4.8740 11	3.8000 17	1.9780 09	2.2640 09	3.8010 12
8.20	3.7260-03	2.4620 09	1.4730 08	1.3820 10	4.7580 11	3.8000 17	1.9430 09	2.2480 09	3.8010 12
8.40	3.6180-03	2.4690 09	1.4260 08	1.3850 10	4.6460 11	3.8000 17	1.9100 09	2.2320 09	3.8010 12
8.60	3.5130-03	2.4760 09	1.3810 08	1.3880 10	4.5400 11	3.8000 17	1.8760 09	2.2160 09	3.8010 12
8.80	3.4110-03	2.4820 09	1.3370 08	1.3910 10	4.4370 11	3.8000 17	1.8440 09	2.2000 09	3.8010 12
9.00	3.3130-03	2.4890 09	1.2950 08	1.3940 10	4.3390 11	3.8000 17	1.8110 09	2.1850 09	3.8010 12
9.20	3.2170-03	2.4950 09	1.2540 08	1.3970 10	4.2450 11	3.8000 17	1.7800 09	2.1690 09	3.8010 12
9.40	3.1250-03	2.5020 09	1.2150 08	1.4000 10	4.1540 11	3.8000 17	1.7490 09	2.1540 09	3.8010 12
9.60	3.0350-03	2.5080 09	1.1780 08	1.4030 10	4.0670 11	3.8000 17	1.7180 09	2.1380 09	3.8010 12
9.80	2.9490-03	2.5140 09	1.1410 08	1.4050 10	3.9830 11	3.8000 17	1.6890 09	2.1230 09	3.8010 12
10.00	2.8650-03	2.5190 09	1.1060 08	1.4080 10	3.9020 11	3.8000 17	1.6590 09	2.1080 09	3.8010 12

TIME (S)	N	H	H2	H02	H202	HNO	HNO2	HN04	CLNO3
0.0	1.0000 06	1.0000 06	1.0000 06	2.0000 07	1.4000 09	1.0000 04	1.0000 06	4.0000 09	1.0000 08
0.20	5.2230 11	1.5480 07	1.1300 06	1.8480 09	1.4070 09	1.0000 06	1.2520 06	8.3630-02	4.3450-03
0.40	4.6210 11	1.4190 07	1.2730 06	1.9090 09	1.3110 09	1.0000 06	1.7600 06	4.9930-02	2.5110-03
0.60	4.1390 11	1.3060 07	1.4090 06	1.9600 09	1.3160 09	1.0010 06	2.3390 06	4.1570-02	2.0360-03
0.80	3.7450 11	1.2080 07	1.5360 06	2.0020 09	1.3200 09	1.0010 06	2.9480 06	3.9640-02	1.9010-03
1.00	3.4170 11	1.1220 07	1.6570 06	2.0360 09	1.3250 09	1.0010 06	3.5730 06	3.9530-02	1.8640-03
1.20	3.1400 11	1.0460 07	1.7710 06	2.0630 09	1.3300 09	1.0010 06	4.2100 06	3.9950-02	1.8590-03
1.40	2.9020 11	9.7770 06	1.8780 06	2.0850 09	1.3350 09	1.0010 06	4.8560 06	4.0500-02	1.8650-03
1.60	2.6970 11	9.1670 06	1.9790 06	2.1020 09	1.3400 09	1.0010 06	5.5080 06	4.1060-02	1.8750-03
1.80	2.5180 11	8.6160 06	2.0750 06	2.1150 09	1.3450 09	1.0010 06	6.1660 06	4.1590-02	1.8880-03
2.00	2.3590 11	8.1170 06	2.1660 06	2.1240 09	1.3510 09	1.0020 06	6.8280 06	4.2060-02	1.9010-03
2.20	2.2190 11	7.6620 06	2.2510 06	2.1300 09	1.3560 09	1.0020 06	7.4940 06	4.2470-02	1.9140-03
2.40	2.0940 11	7.2460 06	2.3320 06	2.1340 09	1.3610 09	1.0020 06	8.1620 06	4.2830-02	1.9270-03
2.60	1.9810 11	6.8650 06	2.4080 06	2.1350 09	1.3670 09	1.0020 06	8.8310 06	4.3150-02	1.9400-03
2.80	1.8790 11	6.5140 06	2.4810 06	2.1340 09	1.3720 09	1.0020 06	9.5010 06	4.3420-02	1.9530-03
3.00	1.7870 11	6.1910 06	2.5500 06	2.1310 09	1.3770 09	1.0020 06	1.0170 07	4.3640-02	1.9660-03
3.20	1.7030 11	5.8910 06	2.6150 06	2.1270 09	1.3830 09	1.0020 06	1.0840 07	4.3830-02	1.9790-03
3.40	1.6250 11	5.6130 06	2.6770 06	2.1210 09	1.3880 09	1.0020 06	1.1510 07	4.3980-02	1.9910-03
3.60	1.5550 11	5.3550 06	2.7350 06	2.1140 09	1.3930 09	1.0020 06	1.2180 07	4.4100-02	2.0030-03
3.80	1.4890 11	5.1150 06	2.7910 06	2.1060 09	1.3980 09	1.0030 06	1.2850 07	4.4190-02	2.0140-03
4.00	1.4290 11	4.8900 06	2.8440 06	2.0970 09	1.4030 09	1.0030 06	1.3510 07	4.4250-02	2.0260-03
4.20	1.3730 11	4.6800 06	2.8940 06	2.0870 09	1.4080 09	1.0030 06	1.4170 07	4.4290-02	2.0370-03
4.40	1.3210 11	4.4830 06	2.9420 06	2.0770 09	1.4130 09	1.0030 06	1.4830 07	4.4300-02	2.0480-03
4.60	1.2720 11	4.2990 06	2.9880 06	2.0650 09	1.4180 09	1.0030 06	1.5490 07	4.4290-02	2.0590-03
4.80	1.2270 11	4.1260 06	3.0310 06	2.0540 09	1.4230 09	1.0030 06	1.6150 07	4.4260-02	2.0690-03
5.00	1.1850 11	3.9630 06	3.0720 06	2.0420 09	1.4280 09	1.0030 06	1.6800 07	4.4220-02	2.0790-03
5.20	1.1450 11	3.8090 06	3.1120 06	2.0290 09	1.4330 09	1.0030 06	1.7440 07	4.4160-02	2.0890-03
5.40	1.1070 11	3.6640 06	3.1490 06	2.0170 09	1.4370 09	1.0030 06	1.8090 07	4.4080-02	2.0980-03
5.60	1.0720 11	3.5270 06	3.1850 06	2.0030 09	1.4420 09	1.0030 06	1.8730 07	4.3990-02	2.1080-03
5.80	1.0390 11	3.3980 06	3.2200 06	1.9900 09	1.4460 09	1.0030 06	1.9360 07	4.3890-02	2.1170-03
6.00	1.0080 11	3.2750 06	3.2520 06	1.9770 09	1.4500 09	1.0030 06	2.0000 07	4.3770-02	2.1260-03
6.20	9.7810 10	3.1590 06	3.2840 06	1.9630 09	1.4550 09	1.0030 06	2.0620 07	4.3650-02	2.1340-03
6.40	9.5000 10	3.0440 06	3.3140 06	1.9490 09	1.4590 09	1.0030 06	2.1250 07	4.3510-02	2.1430-03
6.60	8.9800 10	2.9450 06	3.3420 06	1.9350 09	1.4630 09	1.0030 06	2.1870 07	4.3360-02	2.1510-03
6.80	8.9800 10	2.8450 06	3.3700 06	1.9210 09	1.4670 09	1.0040 06	2.2480 07	4.3210-02	2.1590-03
7.00	8.7390 10	2.7510 06	3.3960 06	1.9070 09	1.4710 09	1.0040 06	2.3090 07	4.3050-02	2.1670-03
7.20	8.5100 10	2.6610 06	3.4210 06	1.8930 09	1.4750 09	1.0040 06	2.3700 07	4.2880-02	2.1740-03
7.40	8.2910 10	2.5750 06	3.4450 06	1.8790 09	1.4790 09	1.0040 06	2.4300 07	4.2710-02	2.1820-03
7.60	8.0820 10	2.4930 06	3.4690 06	1.8650 09	1.4820 09	1.0040 06	2.4900 07	4.2530-02	2.1890-03
7.80	7.8830 10	2.4150 06	3.4910 06	1.8510 09	1.4860 09	1.0040 06	2.5490 07	4.2340-02	2.1960-03
8.00	7.6930 10	2.3410 06	3.5120 06	1.8370 09	1.4890 09	1.0040 06	2.6080 07	4.2150-02	2.2020-03
8.20	7.5110 10	2.2700 06	3.5320 06	1.8230 09	1.4930 09	1.0040 06	2.6660 07	4.1960-02	2.2090-03
8.40	7.3360 10	2.2020 06	3.5520 06	1.8100 09	1.4960 09	1.0040 06	2.7240 07	4.1760-02	2.2150-03
8.60	7.1690 10	2.1370 06	3.5710 06	1.7960 09	1.5000 09	1.0040 06	2.7820 07	4.1560-02	2.2220-03
8.80	7.0080 10	2.0740 06	3.5890 06	1.7820 09	1.5030 09	1.0040 06	2.8390 07	4.1350-02	2.2280-03
9.00	6.8540 10	2.0150 06	3.6060 06	1.7680 09	1.5060 09	1.0040 06	2.8950 07	4.1150-02	2.2340-03
9.20	6.7060 10	1.9570 06	3.6230 06	1.7550 09	1.5090 09	1.0040 06	2.9510 07	4.0940-02	2.2390-03
9.40	6.5640 10	1.9020 06	3.6390 06	1.7410 09	1.5120 09	1.0040 06	3.0070 07	4.0720-02	2.2450-03
9.60	6.4260 10	1.8500 06	3.6540 06	1.7280 09	1.5150 09	1.0040 06	3.0620 07	4.0510-02	2.2500-03
9.80	6.2940 10	1.7990 06	3.6690 06	1.7150 09	1.5180 09	1.0040 06	3.1160 07	4.0290-02	2.2560-03
10.00	6.1670 10	1.7500 06	3.6840 06	1.7020 09	1.5210 09	1.0040 06	3.1700 07	4.0080-02	2.2610-03

T= 800 K, H=20 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	5.840 06	8.150-15
2	2*N03 >>> 2*N02 + O2	3.980-14	5.950-48
3	N02 + N03 >>> N02 + N0 + O2	6.590-14	3.130-36
4	N03 + N0 >>> 2*N02	1.900-11	1.740-18
5	N0 + O3 >>> N02 + O2	3.430-13	4.420-26
6	N02 + O3 >>> N03 + O2	5.610-15	9.780-21
7	HN03 + M >>> H0 + N02 + M	4.180 00	9.230-14
8	HN03 + H0 >>> H2O + N03	8.000-14	2.430-18
9	O + O + M >>> O2 + M	1.560-16	9.070-16
10	O + O2 + M >>> O3 + M	9.750-17	4.970 01
11	O + O3 >>> 2*O2	1.070-12	6.110-39
12	O + N0 + M >>> N02 + M	1.560-14	1.490-08
13	O + N02 >>> N0 + O2	1.170-11	5.550-25
14	O + N02 + M >>> N03 + M	4.860-14	3.890-24
15	H0 + H0 >>> H2O + O	5.000-12	1.060-15
16	O2 + 2*N0 >>> 2*N02	6.400-39	8.780-29
17	N02 + H-NU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.200-11	9.550-15
19	O + H02 >>> H0 + O2	4.280-11	3.430-26
20	O2 + H + M >>> H02 + M	4.840-15	7.220-04
21	O3 + H >>> H0 + O2	5.250-11	6.310-34
22	O3 + H0 >>> H02 + O2	4.300-13	3.240-24
23	O3 + H02 >>> H0 + 2*O2	1.480-14	2.350-48
24	H + H0 + M >>> H2O + M	1.890-14	1.870-22
25	H + H02 >>> 2*H0	1.280-10	2.160-22
26	H + H02 >>> H2 + O2	2.710-11	1.450-26
27	H + H2O >>> H2 + H0	4.090-16	1.410-12
28	H + H2O2 >>> H2 + H02	3.930-13	9.470-18
29	H + H2O2 >>> H0 + H2O	5.110-13	6.020-39

30	2*H0 + M >>> H2O2 + M	1.870-14	1.540-02
31	H0 + H02 >>> H2O + O2	4.440-11	5.700-36
32	2*H02 >>> H2O2 + O2	9.100-12	1.530-22
33	H02 + H2O >>> H2O2 + H0	5.190-20	3.460-12
34	NO + H + M >>> HNO + M	5.300-15	3.150-04
35	NO + H0 >>> NO2 + H	3.300-20	2.300-10
36	NO + H0 + M >>> HNO2 + M	7.510-14	7.160-01
37	NO + H02 >>> NO2 + H0	4.460-12	7.660-14
38	H + H + M >>> H2 + M	4.860-16	5.130-21
39	HNO4 + M >>> H02 + NO2 + M	6.970 06	9.130-15
40	CLN03 + M >>> CLO + NO2 + M	5.340 06	3.200-15

HAPP RESIDENCE TIME STUDY

TIME(S)	N205	N02	N03	N0	03	02	HN03	HO	M20
0.0	4.0000 08	8.0000 09	5.0000 05	1.5000 09	4.5000 12	3.8000 17	4.0000 09	1.0000 06	3.8000 12
0.20	5.4690-04	8.6620 08	4.5210 08	1.5350 10	1.2870 12	3.8000 17	1.7350 09	6.9220 09	3.7990 12
0.40	3.9990-04	6.5620 08	4.3640 08	1.6550 10	9.5390 11	3.8000 17	7.5280 08	9.0480 09	3.7980 12
0.60	3.3610-04	5.7830 08	4.1620 08	1.7080 10	7.4970 11	3.8000 17	3.2660 08	1.0170 10	3.7970 12
0.80	2.9390-04	5.3320 08	3.9470 08	1.7320 10	6.4270 11	3.8000 17	1.4170 08	1.0730 10	3.7970 12
1.00	2.6550-04	5.0950 08	3.7320 08	1.7450 10	5.1410 11	3.8000 17	6.1540 07	1.0980 10	3.7970 12
1.20	2.4550-04	4.9920 08	3.5220 08	1.7510 10	4.4040 11	3.8000 17	2.6760 07	1.1040 10	3.7970 12
1.40	2.3030-04	4.9670 08	3.3210 08	1.7550 10	3.8340 11	3.8000 17	1.1680 07	1.0980 10	3.7970 12
1.60	2.1780-04	4.9850 08	3.1290 08	1.7570 10	3.3810 11	3.8000 17	5.1320 06	1.0840 10	3.7970 12
1.80	2.0670-04	5.0250 08	2.9460 08	1.7590 10	3.0140 11	3.8000 17	2.2940 06	1.0660 10	3.7970 12
2.00	1.9660-04	5.0770 08	2.7730 08	1.7600 10	2.7100 11	3.8000 17	1.0620 06	1.0450 10	3.7970 12
2.20	1.8700-04	5.1320 08	2.6100 08	1.7610 10	2.4550 11	3.8000 17	5.2690 05	1.0220 10	3.7970 12
2.40	1.7790-04	5.1880 08	2.4550 08	1.7620 10	2.2390 11	3.8000 17	2.9400 05	9.9830 09	3.7970 12
2.60	1.6910-04	5.2430 08	2.3100 08	1.7630 10	2.0540 11	3.8000 17	1.9210 05	9.7430 09	3.7970 12
2.80	1.6070-04	5.2950 08	2.1730 08	1.7640 10	1.8930 11	3.8000 17	1.4700 05	9.5040 09	3.7980 12
3.00	1.5250-04	5.3450 08	2.0440 08	1.7650 10	1.7530 11	3.8000 17	1.2650 05	9.2700 09	3.7980 12
3.20	1.4470-04	5.3920 08	1.9220 08	1.7650 10	1.6300 11	3.8000 17	1.1660 05	9.0410 09	3.7980 12
3.40	1.3720-04	5.4370 08	1.8080 08	1.7660 10	1.5210 11	3.8000 17	1.1120 05	8.8180 09	3.7980 12
3.60	1.3010-04	5.4790 08	1.7000 08	1.7670 10	1.4240 11	3.8000 17	1.0790 05	8.6030 09	3.7980 12
3.80	1.2320-04	5.5180 08	1.5990 08	1.7670 10	1.3370 11	3.8000 17	1.0540 05	8.3950 09	3.7980 12
4.00	1.1660-04	5.5550 08	1.5040 08	1.7680 10	1.2590 11	3.8000 17	1.0330 05	8.1960 09	3.7980 12
4.20	1.1040-04	5.5900 08	1.4140 08	1.7680 10	1.1880 11	3.8000 17	1.0140 05	8.0040 09	3.7990 12
4.40	1.0440-04	5.6230 08	1.3300 08	1.7690 10	1.1240 11	3.8000 17	9.9620 04	7.8200 09	3.7990 12
4.60	9.8770-05	5.6540 08	1.2510 08	1.7700 10	1.0650 11	3.8000 17	9.7870 04	7.6440 09	3.7990 12
4.80	9.3390-05	5.6830 08	1.1770 08	1.7700 10	1.0120 11	3.8000 17	9.6170 04	7.4750 09	3.7990 12
5.00	8.8290-05	5.7110 08	1.1070 08	1.7700 10	9.6300 10	3.8000 17	9.4510 04	7.3120 09	3.7990 12
5.20	8.3450-05	5.7370 08	1.0420 08	1.7710 10	9.1790 10	3.8000 17	9.2900 04	7.1570 09	3.7990 12
5.40	7.8860-05	5.7610 08	9.8020 07	1.7710 10	8.7630 10	3.8000 17	9.1330 04	7.0080 09	3.7990 12
5.60	7.4520-05	5.7850 08	9.2250 07	1.7720 10	8.3780 10	3.8000 17	8.9800 04	6.8650 09	3.7990 12
5.80	7.0410-05	5.8070 08	8.6820 07	1.7720 10	8.0210 10	3.8000 17	8.8320 04	6.7280 09	3.7990 12
6.00	6.6520-05	5.8290 08	8.1730 07	1.7720 10	7.6890 10	3.8000 17	8.6880 04	6.5970 09	3.8000 12
6.20	6.2850-05	5.8490 08	7.6950 07	1.7730 10	7.3800 10	3.8000 17	8.5480 04	6.4710 09	3.8000 12
6.40	5.9380-05	5.8680 08	7.2470 07	1.7730 10	7.0910 10	3.8000 17	8.4130 04	6.3500 09	3.8000 12
6.60	5.6110-05	5.8870 08	6.8250 07	1.7730 10	6.8210 10	3.8000 17	8.2810 04	6.2330 09	3.8000 12
6.80	5.3020-05	5.9050 08	6.4300 07	1.7730 10	6.5680 10	3.8000 17	8.1540 04	6.1210 09	3.8000 12
7.00	5.0100-05	5.9220 08	6.0580 07	1.7740 10	6.3300 10	3.8000 17	8.0310 04	6.0140 09	3.8000 12
7.20	4.7350-05	5.9390 08	5.7100 07	1.7740 10	6.1070 10	3.8000 17	7.9120 04	5.9100 09	3.8000 12
7.40	4.4760-05	5.9550 08	5.3820 07	1.7740 10	5.8970 10	3.8000 17	7.7970 04	5.8100 09	3.8000 12
7.60	4.2310-05	5.9710 08	5.0750 07	1.7740 10	5.6980 10	3.8000 17	7.6850 04	5.7140 09	3.8000 12
7.80	4.0010-05	5.9860 08	4.7860 07	1.7740 10	5.5110 10	3.8000 17	7.5770 04	5.6220 09	3.8000 12
8.00	3.7830-05	6.0000 08	4.5150 07	1.7740 10	5.3340 10	3.8000 17	7.4720 04	5.5330 09	3.8000 12
8.20	3.5790-05	6.0150 08	4.2410 07	1.7750 10	5.1660 10	3.8000 17	7.3710 04	5.4460 09	3.8000 12
8.40	3.3860-05	6.0290 08	4.0220 07	1.7750 10	5.0070 10	3.8000 17	7.2730 04	5.3630 09	3.8000 12
8.60	3.2040-05	6.0420 08	3.7970 07	1.7750 10	4.8560 10	3.8000 17	7.1770 04	5.2830 09	3.8000 12
8.80	3.0330-05	6.0560 08	3.5870 07	1.7750 10	4.7120 10	3.8000 17	7.0850 04	5.2050 09	3.8010 12
9.00	2.8720-05	6.0690 08	3.3890 07	1.7750 10	4.5760 10	3.8000 17	6.9960 04	5.1300 09	3.8010 12
9.20	2.7200-05	6.0820 08	3.2030 07	1.7750 10	4.4460 10	3.8000 17	6.9100 04	5.0580 09	3.8010 12
9.40	2.5770-05	6.0950 08	3.0280 07	1.7750 10	4.3220 10	3.8000 17	6.8260 04	4.9880 09	3.8010 12
9.60	2.4430-05	6.1080 08	2.8640 07	1.7750 10	4.2040 10	3.8000 17	6.7450 04	4.9200 09	3.8010 12
9.80	2.3160-05	6.1200 08	2.7100 07	1.7750 10	4.0910 10	3.8000 17	6.6660 04	4.8540 09	3.8010 12
10.00	2.1970-05	6.1330 08	2.5650 07	1.7750 10	3.9820 10	3.8000 17	6.5900 04	4.7900 09	3.8010 12

TIME (S)	n	H	H2	H02	H202	HNO	HNO2	HNO4	CLN03
0.0	1.0000 06	1.0000 06	1.0000 06	2.0000 07	1.3000 09	1.0000 04	1.0000 06	4.0000 09	1.0000 08
0.20	1.7340 12	9.1050 07	1.7390 06	2.2310 09	1.2980 09	1.0010 04	1.7770 06	2.5290-03	7.0160-05
0.40	1.2820 12	8.8310 07	3.0670 06	2.9530 09	1.3000 09	1.0020 06	3.3690 06	2.5360-03	5.3150-05
0.60	1.0050 12	7.8050 07	4.5200 06	3.3330 09	1.3060 09	1.0040 06	5.2050 06	2.5230-03	4.6840-05
0.80	8.2140 11	6.7340 07	5.8890 06	3.5250 09	1.3140 09	1.0050 06	7.0400 06	2.4600-03	4.3180-05
1.00	6.8920 11	5.7860 07	7.1050 06	3.6070 09	1.3240 09	1.0060 06	8.7520 06	2.4050-03	4.1270-05
1.20	5.9030 11	4.9850 07	8.1590 06	3.6240 09	1.3340 09	1.0070 04	1.0280 07	2.3680-03	4.0430-05
1.40	5.1390 11	4.3170 07	9.0610 06	3.6010 09	1.3440 09	1.0080 04	1.1620 07	2.3410-03	4.0230-05
1.60	4.5300 11	3.7610 07	9.8310 06	3.5520 09	1.3530 09	1.0090 06	1.2750 07	2.3170-03	4.0380-05
1.80	4.0370 11	3.2960 07	1.0490 07	3.4880 09	1.3620 09	1.0090 06	1.3700 07	2.2940-03	4.0700-05
2.00	3.6300 11	2.9060 07	1.1050 07	3.4140 09	1.3700 09	1.0100 06	1.4470 07	2.2680-03	4.1120-05
2.20	3.2890 11	2.5760 07	1.1530 07	3.3340 09	1.3770 09	1.0100 06	1.5080 07	2.2390-03	4.1570-05
2.40	2.9990 11	2.2950 07	1.1940 07	3.2520 09	1.3830 09	1.0110 04	1.5560 07	2.2080-03	4.2020-05
2.60	2.7510 11	2.0550 07	1.2290 07	3.1640 09	1.3880 09	1.0110 06	1.5920 07	2.1740-03	4.2460-05
2.80	2.5360 11	1.8480 07	1.2600 07	3.0870 09	1.3930 09	1.0110 06	1.6170 07	2.1390-03	4.2890-05
3.00	2.3490 11	1.6700 07	1.2860 07	3.0060 09	1.3970 09	1.0110 06	1.6330 07	2.1030-03	4.3290-05
3.20	2.1840 11	1.5140 07	1.3090 07	2.9270 09	1.4000 09	1.0120 06	1.6410 07	2.0660-03	4.3670-05
3.40	2.0380 11	1.3780 07	1.3290 07	2.8500 09	1.4020 09	1.0120 06	1.6430 07	2.0280-03	4.4040-05
3.60	1.9080 11	1.2590 07	1.3460 07	2.7760 09	1.4040 09	1.0120 06	1.6390 07	1.9910-03	4.4370-05
3.80	1.7910 11	1.1540 07	1.3610 07	2.7050 09	1.4050 09	1.0120 06	1.6310 07	1.9540-03	4.4690-05
4.00	1.6860 11	1.0600 07	1.3750 07	2.6370 09	1.4060 09	1.0120 06	1.6180 07	1.9170-03	4.4990-05
4.20	1.5920 11	9.7760 06	1.3860 07	2.5710 09	1.4060 09	1.0130 04	1.6030 07	1.8810-03	4.5280-05
4.40	1.5060 11	9.0360 06	1.3970 07	2.5080 09	1.4060 09	1.0130 04	1.5850 07	1.8450-03	4.5540-05
4.60	1.4280 11	8.3740 06	1.4060 07	2.4470 09	1.4060 09	1.0130 06	1.5650 07	1.8110-03	4.5790-05
4.80	1.3560 11	7.7780 06	1.4130 07	2.3890 09	1.4050 09	1.0130 06	1.5430 07	1.7770-03	4.6030-05
5.00	1.2900 11	7.2410 06	1.4200 07	2.3340 09	1.4040 09	1.0130 06	1.5210 07	1.7440-03	4.6250-05
5.20	1.2300 11	6.7560 06	1.4270 07	2.2810 09	1.4020 09	1.0130 06	1.4970 07	1.7120-03	4.6460-05
5.40	1.1740 11	6.3160 06	1.4320 07	2.2290 09	1.4010 09	1.0130 06	1.4730 07	1.6810-03	4.6670-05
5.60	1.1230 11	5.9160 06	1.4370 07	2.1810 09	1.3990 09	1.0130 06	1.4480 07	1.6510-03	4.6860-05
5.80	1.0750 11	5.5510 06	1.4410 07	2.1340 09	1.3960 09	1.0130 06	1.4240 07	1.6220-03	4.7040-05
6.00	1.0300 11	5.2180 06	1.4450 07	2.0890 09	1.3940 09	1.0130 06	1.3990 07	1.5930-03	4.7210-05
6.20	9.8890 10	4.9120 06	1.4480 07	2.0450 09	1.3910 09	1.0130 06	1.3740 07	1.5660-03	4.7370-05
6.40	9.5030 10	4.6320 06	1.4510 07	2.0040 09	1.3880 09	1.0130 06	1.3500 07	1.5390-03	4.7530-05
6.60	9.1410 10	4.3740 06	1.4530 07	1.9640 09	1.3850 09	1.0130 04	1.3260 07	1.5130-03	4.7680-05
6.80	8.8020 10	4.1370 06	1.4550 07	1.9260 09	1.3820 09	1.0130 06	1.3020 07	1.4880-03	4.7830-05
7.00	8.4840 10	3.9170 06	1.4570 07	1.8890 09	1.3790 09	1.0130 06	1.2790 07	1.4640-03	4.7970-05
7.20	8.1850 10	3.7140 06	1.4590 07	1.8530 09	1.3750 09	1.0130 06	1.2570 07	1.4410-03	4.8100-05
7.40	7.9030 10	3.5260 06	1.4600 07	1.8190 09	1.3720 09	1.0130 06	1.2340 07	1.4180-03	4.8230-05
7.60	7.6380 10	3.3510 06	1.4610 07	1.7860 09	1.3680 09	1.0130 06	1.2130 07	1.3960-03	4.8360-05
7.80	7.3870 10	3.1890 06	1.4620 07	1.7550 09	1.3640 09	1.0130 06	1.1920 07	1.3740-03	4.8480-05
8.00	7.1500 10	3.0370 06	1.4630 07	1.7240 09	1.3600 09	1.0130 06	1.1710 07	1.3540-03	4.8600-05
8.20	6.9250 10	2.8960 06	1.4630 07	1.6940 09	1.3570 09	1.0130 06	1.1510 07	1.3340-03	4.8720-05
8.40	6.7120 10	2.7640 06	1.4640 07	1.6660 09	1.3530 09	1.0130 06	1.1320 07	1.3140-03	4.8830-05
8.60	6.5090 10	2.6410 06	1.4640 07	1.6380 09	1.3480 09	1.0130 06	1.1130 07	1.2960-03	4.8940-05
8.80	6.3170 10	2.5250 06	1.4640 07	1.6120 09	1.3440 09	1.0130 06	1.0950 07	1.2770-03	4.9050-05
9.00	6.1340 10	2.4170 06	1.4640 07	1.5860 09	1.3400 09	1.0130 06	1.0770 07	1.2600-03	4.9160-05
9.20	5.9600 10	2.3150 06	1.4640 07	1.5610 09	1.3360 09	1.0130 06	1.0600 07	1.2430-03	4.9260-05
9.40	5.7940 10	2.2190 06	1.4640 07	1.5370 09	1.3310 09	1.0130 06	1.0440 07	1.2260-03	4.9370-05
9.60	5.6350 10	2.1290 06	1.4640 07	1.5130 09	1.3270 09	1.0130 06	1.0270 07	1.2100-03	4.9470-05
9.80	5.4840 10	2.0440 06	1.4640 07	1.4910 09	1.3230 09	1.0130 06	1.0120 07	1.1940-03	4.9570-05
10.00	5.3390 10	1.9640 06	1.4640 07	1.4690 09	1.3180 09	1.0130 06	9.9660 06	1.1790-03	4.9670-05

T= 250 K, H= 25 km

19 torr

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N2O5} + \text{M} \gg \text{N2O} + \text{NO3} + \text{M}$	4.050-05	1.270-13
2	$2^*\text{NO3} \gg 2^*\text{NO2} + \text{O2}$	4.710-17	7.460-62
3	$\text{NO2} + \text{NO3} \gg \text{NO2} + \text{NO} + \text{O2}$	4.210-15	1.190-34
4	$\text{NO3} + \text{NO} \gg 2^*\text{NO2}$	1.900-11	3.290-37
5	$\text{NO} + \text{O3} \gg \text{NO2} + \text{O2}$	6.360-15	1.080-56
6	$\text{NO2} + \text{O3} \gg \text{NO3} + \text{O2}$	6.650-14	8.330-39
7	$\text{HNO3} + \text{M} \gg \text{HO} + \text{NO2} + \text{M}$	5.360-28	2.460-12
8	$\text{HNO3} + \text{HO} \gg \text{H2O} + \text{NO3}$	2.000-14	2.800-29
9	$\text{O} + \text{O} + \text{M} \gg \text{O2} + \text{M}$	2.690-15	1.440-57
10	$\text{O} + \text{O2} + \text{M} \gg \text{O3} + \text{M}$	5.760-16	1.620-17
11	$\text{O} + \text{O3} \gg 2^*\text{O2}$	1.920-15	0.0
12	$\text{O} + \text{NO} + \text{M} \gg \text{NO2} + \text{M}$	1.130-13	8.390-48
13	$\text{O} + \text{NO2} \gg \text{NO} + \text{O2}$	5.120-12	6.270-53
14	$\text{O} + \text{NO2} + \text{M} \gg \text{NO3} + \text{M}$	7.070-14	5.660-24
15	$\text{HO} + \text{HO} \gg \text{H2O} + \text{O}$	1.090-12	9.770-27
16	$\text{O2} + 2^*\text{NO} \gg 2^*\text{NO2}$	2.750-38	3.880-35
17	$\text{NO2} + \text{H-NU} \gg \text{HO} + \text{O}$	0.0	0.0
18	$\text{O} + \text{HO} \gg \text{H} + \text{O2}$	4.200-11	7.730-25
19	$\text{O} + \text{HO2} \gg \text{HO} + \text{O2}$	1.080-11	5.450-60
20	$\text{O2} + \text{H} + \text{M} \gg \text{HO2} + \text{M}$	2.780-14	3.570-31
21	$\text{O3} + \text{H} \gg \text{HO} + \text{O2}$	1.270-11	0.0
22	$\text{O3} + \text{HO} \gg \text{HO2} + \text{O2}$	2.750-14	1.040-48
23	$\text{O3} + \text{HO2} \gg \text{HO} + 2^*\text{O2}$	4.450-16	2.410-64
24	$\text{H} + \text{HO} + \text{M} \gg \text{H2O} + \text{M}$	5.670-13	0.0
25	$\text{H} + \text{HO2} \gg 2^*\text{HO}$	9.400-12	1.620-46
26	$\text{H} + \text{HO2} \gg \text{H2} + \text{O2}$	1.040-11	2.550-61
27	$\text{H} + \text{H2O} \gg \text{H2} + \text{HO}$	2.340-24	1.140-15
28	$\text{H} + \text{H2O2} \gg \text{H2} + \text{HO2}$	8.360-15	5.620-29
29	$\text{H} + \text{H2O2} \gg \text{HO} + \text{H2O}$	1.090-14	1.350-75

30	$2^*H_2O + M \ggg H_2O_2 + M$	3.230-13	8.680-32
31	$H_2O + H_2O_2 \ggg H_2O + O_2$	1.120-11	3.890-74
32	$2^*H_2O_2 \ggg H_2O_2 + O_2$	2.300-12	8.580-49
33	$H_2O_2 + H_2O \ggg H_2O_2 + H_2O$	1.020-39	5.490-13
34	$NO + H + M \ggg HNO + M$	1.760-14	2.750-33
35	$NO + H_2O \ggg NO_2 + H$	3.050-34	3.010-11
36	$NO + H_2O + M \ggg HNO_2 + M$	2.310-12	8.770-29
37	$NO + H_2O_2 \ggg NO_2 + H_2O$	1.650-13	7.470-21
38	$H + H + M \ggg H_2 + M$	7.070-16	0.0
39	$HNO_4 + M \ggg H_2O_2 + NO_2 + M$	1.090-05	2.020-13
40	$CLNO_3 + M \ggg ClO + NO_2 + M$	3.260-07	1.590-19

HAPP RESIDENCE TIME STUDY

TIME (S)	H2O5	H2O2	H2O3	H2O	H1	H2	H4O3	H0	H2O
0.0	7.0000 08	6.5000 09	2.0000 06	7.0000 08	4.2000 12	1.7000 17	2.0000 09	1.5000 06	1.9000 12
0.20	7.0000 08	6.5040 09	2.0310 06	6.9630 08	4.2000 12	1.7000 17	2.0000 09	1.4800 06	1.9000 12
0.40	7.0000 08	6.5070 09	2.0630 06	6.9260 08	4.2000 12	1.7000 17	2.0000 09	1.4510 06	1.9000 12
0.60	7.0000 08	6.5110 09	2.0940 06	6.8890 08	4.2000 12	1.7000 17	2.0000 09	1.4220 06	1.9000 12
0.80	7.0000 08	6.5150 09	2.1250 06	6.8520 08	4.2000 12	1.7000 17	2.0000 09	1.3930 06	1.9000 12
1.00	7.0000 08	6.5180 09	2.1560 06	6.8150 08	4.2000 12	1.7000 17	2.0000 09	1.3660 06	1.9000 12
1.20	7.0000 08	6.5220 09	2.1870 06	6.7780 08	4.2000 12	1.7000 17	2.0000 09	1.3390 06	1.9000 12
1.40	7.0000 08	6.5250 09	2.2180 06	6.7430 08	4.2000 12	1.7000 17	2.0000 09	1.3130 06	1.9000 12
1.60	7.0000 08	6.5290 09	2.2490 06	6.7070 08	4.2000 12	1.7000 17	2.0000 09	1.2870 06	1.9000 12
1.80	7.0000 08	6.5320 09	2.2800 06	6.6710 08	4.2000 12	1.7000 17	2.0000 09	1.2630 06	1.9000 12
2.00	7.0000 08	6.5360 09	2.3110 06	6.6350 08	4.2000 12	1.7000 17	2.0000 09	1.2390 06	1.9000 12
2.20	7.0000 08	6.5400 09	2.3420 06	6.6000 08	4.2000 12	1.7000 17	2.0000 09	1.2150 06	1.9000 12
2.40	7.0000 08	6.5430 09	2.3730 06	6.5650 08	4.2000 12	1.7000 17	2.0000 09	1.1930 06	1.9000 12
2.60	7.0000 08	6.5460 09	2.4040 06	6.5300 08	4.2000 12	1.7000 17	2.0000 09	1.1700 06	1.9000 12
2.80	7.0000 08	6.5500 09	2.4350 06	6.4950 08	4.2000 12	1.7000 17	2.0000 09	1.1490 06	1.9000 12
3.00	7.0000 08	6.5530 09	2.4650 06	6.4600 08	4.2000 12	1.7000 17	2.0000 09	1.1280 06	1.9000 12
3.20	7.0000 08	6.5570 09	2.4960 06	6.4260 08	4.2000 12	1.7000 17	2.0000 09	1.1070 06	1.9000 12
3.40	7.0000 08	6.5600 09	2.5270 06	6.3910 08	4.2000 12	1.7000 17	2.0000 09	1.0870 06	1.9000 12
3.60	7.0000 08	6.5630 09	2.5580 06	6.3570 08	4.2000 12	1.7000 17	2.0000 09	1.0680 06	1.9000 12
3.80	7.0000 08	6.5670 09	2.5880 06	6.3230 08	4.2000 12	1.7000 17	2.0000 09	1.0490 06	1.9000 12
4.00	7.0000 08	6.5700 09	2.6190 06	6.2900 08	4.2000 12	1.7000 17	2.0000 09	1.0310 06	1.9000 12
4.20	7.0000 08	6.5730 09	2.6500 06	6.2560 08	4.2000 12	1.7000 17	2.0000 09	1.0130 06	1.9000 12
4.40	7.0000 08	6.5770 09	2.6800 06	6.2230 08	4.2000 12	1.7000 17	2.0000 09	9.9570 05	1.9000 12
4.60	7.0000 08	6.5800 09	2.7110 06	6.1890 08	4.2000 12	1.7000 17	2.0000 09	9.7880 05	1.9000 12
4.80	7.0000 08	6.5830 09	2.7410 06	6.1560 08	4.2000 12	1.7000 17	2.0000 09	9.6280 05	1.9000 12
5.00	7.0000 08	6.5870 09	2.7720 06	6.1240 08	4.2000 12	1.7000 17	2.0000 09	9.4630 05	1.9000 12
5.20	7.0000 08	6.5900 09	2.8020 06	6.0910 08	4.2000 12	1.7000 17	2.0000 09	9.3070 05	1.9000 12
5.40	7.0000 08	6.5930 09	2.8330 06	6.0580 08	4.2000 12	1.7000 17	2.0000 09	9.1560 05	1.9000 12
5.60	7.0000 08	6.5960 09	2.8630 06	6.0260 08	4.2000 12	1.7000 17	2.0000 09	9.0080 05	1.9000 12
5.80	7.0000 08	6.5990 09	2.8940 06	5.9940 08	4.2000 12	1.7000 17	2.0000 09	8.8640 05	1.9000 12
6.00	7.0000 08	6.6020 09	2.9240 06	5.9620 08	4.2000 12	1.7000 17	2.0000 09	8.7240 05	1.9000 12
6.20	7.0000 08	6.6060 09	2.9540 06	5.9300 08	4.2000 12	1.7000 17	2.0000 09	8.5880 05	1.9000 12
6.40	7.0000 08	6.6090 09	2.9850 06	5.8980 08	4.2000 12	1.7000 17	2.0000 09	8.4550 05	1.9000 12
6.60	7.0000 08	6.6120 09	3.0150 06	5.8670 08	4.2000 12	1.7000 17	2.0000 09	8.3260 05	1.9000 12
6.80	7.0000 08	6.6150 09	3.0450 06	5.8360 08	4.2000 12	1.7000 17	2.0000 09	8.2010 05	1.9000 12
7.00	7.0000 08	6.6180 09	3.0760 06	5.8040 08	4.2000 12	1.7000 17	2.0000 09	8.0780 05	1.9000 12
7.20	7.0000 08	6.6210 09	3.1060 06	5.7730 08	4.2000 12	1.7000 17	2.0000 09	7.9590 05	1.9000 12
7.40	7.0000 08	6.6240 09	3.1360 06	5.7430 08	4.2000 12	1.7000 17	2.0000 09	7.8430 05	1.9000 12
7.60	7.0000 08	6.6270 09	3.1660 06	5.7120 08	4.2000 12	1.7000 17	2.0000 09	7.7310 05	1.9000 12
7.80	7.0000 08	6.6300 09	3.1970 06	5.6810 08	4.2000 12	1.7000 17	2.0000 09	7.6210 05	1.9000 12
8.00	7.0000 08	6.6330 09	3.2270 06	5.6510 08	4.2000 12	1.7000 17	2.0000 09	7.5140 05	1.9000 12
8.20	7.0000 08	6.6360 09	3.2570 06	5.6210 08	4.2000 12	1.7000 17	2.0000 09	7.4100 05	1.9000 12
8.40	7.0000 08	6.6390 09	3.2870 06	5.5910 08	4.2000 12	1.7000 17	2.0000 09	7.3090 05	1.9000 12
8.60	7.0000 08	6.6420 09	3.3170 06	5.5610 08	4.2000 12	1.7000 17	2.0000 09	7.2100 05	1.9000 12
8.80	7.0000 08	6.6450 09	3.3470 06	5.5310 08	4.2000 12	1.7000 17	2.0000 09	7.1140 05	1.9000 12
9.00	7.0000 08	6.6480 09	3.3780 06	5.5020 08	4.2000 12	1.7000 17	2.0000 09	7.0210 05	1.9000 12
9.20	7.0000 08	6.6510 09	3.4080 06	5.4720 08	4.2000 12	1.7000 17	2.0000 09	6.9300 05	1.9000 12
9.40	7.0000 08	6.6540 09	3.4380 06	5.4430 08	4.2000 12	1.7000 17	2.0000 09	6.8420 05	1.9000 12
9.60	7.0000 08	6.6560 09	3.4680 06	5.4140 08	4.2000 12	1.7000 17	2.0000 09	6.7550 05	1.9000 12
9.80	7.0000 08	6.6590 09	3.4980 06	5.3850 08	4.2000 12	1.7000 17	2.0000 09	6.6720 05	1.9000 12
10.00	7.0000 08	6.6620 09	3.5280 06	5.3560 08	4.2000 12	1.7000 17	2.0000 09	6.5900 05	1.9000 12

TIME (S)	U	M	U12	U10	U102	U104	CLN03
0.0	1.0000-04	1.0000-06	2.3000-07	1.0000-04	1.0000-05	3.5000-09	4.0000-08
0.20	9.6510-02	1.0000-06	2.4010-07	1.0000-06	8.0050-05	3.5000-09	4.0000-08
0.40	9.2540-02	1.0000-06	2.4060-07	1.0000-04	8.0100-05	3.5000-09	4.0000-08
0.60	9.1660-02	1.0000-06	2.4060-07	1.0000-06	8.0140-05	3.5000-09	4.0000-08
0.80	9.0770-02	1.0000-06	2.4090-07	1.0000-04	8.0190-05	3.5000-09	4.0000-08
1.00	8.9930-02	1.0000-06	2.4110-07	1.0000-06	8.0230-05	3.5000-09	4.0010-08
1.20	8.9120-02	1.0000-06	2.4130-07	1.0000-04	8.0270-05	3.5000-09	4.0010-08
1.40	8.8360-02	1.0000-06	2.4160-07	1.0000-06	8.0310-05	3.5000-09	4.0010-08
1.60	8.7620-02	1.0000-06	2.4180-07	1.0000-04	8.0350-05	3.5000-09	4.0010-08
1.80	8.6920-02	1.0000-06	2.4200-07	1.0000-06	8.0390-05	3.5000-09	4.0010-08
2.00	8.6260-02	1.0000-06	2.4220-07	1.0000-04	8.0430-05	3.5000-09	4.0010-08
2.20	8.5620-02	1.0000-06	2.4240-07	1.0000-06	8.0470-05	3.5000-09	4.0010-08
2.40	8.5010-02	1.0000-06	2.4260-07	1.0000-04	8.0510-05	3.5000-09	4.0010-08
2.60	8.4430-02	1.0000-06	2.4280-07	1.0000-06	8.0540-05	3.5000-09	4.0020-08
2.80	8.3870-02	1.0000-06	2.4290-07	1.0000-04	8.0580-05	3.5000-09	4.0020-08
3.00	8.3340-02	1.0000-06	2.4310-07	1.0000-06	8.0610-05	3.5000-09	4.0020-08
3.20	8.2830-02	1.0000-06	2.4330-07	1.0000-04	8.0640-05	3.5000-09	4.0020-08
3.40	8.2350-02	1.0000-06	2.4350-07	1.0000-06	8.0680-05	3.5000-09	4.0020-08
3.60	8.1880-02	1.0000-06	2.4360-07	1.0000-04	8.0710-05	3.5000-09	4.0020-08
3.80	8.1440-02	1.0000-06	2.4380-07	1.0000-06	8.0740-05	3.5000-09	4.0020-08
4.00	8.1020-02	1.0000-06	2.4390-07	1.0000-04	8.0770-05	3.5000-09	4.0020-08
4.20	8.0610-02	1.0000-06	2.4410-07	1.0000-06	8.0800-05	3.5000-09	4.0020-08
4.40	8.0220-02	1.0000-06	2.4420-07	1.0000-04	8.0830-05	3.5000-09	4.0030-08
4.60	7.9850-02	1.0000-06	2.4440-07	1.0000-06	8.0860-05	3.5000-09	4.0030-08
4.80	7.9500-02	1.0000-06	2.4450-07	1.0000-04	8.0890-05	3.5000-09	4.0030-08
5.00	7.9160-02	1.0000-06	2.4460-07	1.0000-06	8.0910-05	3.5000-09	4.0030-08
5.20	7.8830-02	1.0000-06	2.4470-07	1.0000-04	8.0940-05	3.5000-09	4.0030-08
5.40	7.8520-02	1.0000-06	2.4490-07	1.0000-06	8.0960-05	3.5000-09	4.0030-08
5.60	7.8220-02	1.0000-06	2.4500-07	1.0000-04	8.0990-05	3.5000-09	4.0030-08
5.80	7.7940-02	1.0000-06	2.4510-07	1.0000-06	8.1010-05	3.5000-09	4.0030-08
6.00	7.7670-02	1.0000-06	2.4520-07	1.0000-04	8.1040-05	3.5000-09	4.0040-08
6.20	7.7400-02	1.0000-06	2.4530-07	1.0000-06	8.1060-05	3.5000-09	4.0040-08
6.40	7.7150-02	1.0000-06	2.4540-07	1.0000-04	8.1090-05	3.5000-09	4.0040-08
6.60	7.6910-02	1.0000-06	2.4550-07	1.0000-06	8.1110-05	3.5000-09	4.0040-08
6.80	7.6680-02	1.0000-06	2.4560-07	1.0000-04	8.1130-05	3.5000-09	4.0040-08
7.00	7.6460-02	1.0000-06	2.4570-07	1.0000-06	8.1150-05	3.5000-09	4.0040-08
7.20	7.6250-02	1.0000-06	2.4580-07	1.0000-04	8.1170-05	3.5000-09	4.0040-08
7.40	7.6050-02	1.0000-06	2.4590-07	1.0000-06	8.1200-05	3.5000-09	4.0040-08
7.60	7.5850-02	1.0000-06	2.4600-07	1.0000-04	8.1220-05	3.5000-09	4.0050-08
7.80	7.5650-02	1.0000-06	2.4610-07	1.0000-06	8.1240-05	3.5000-09	4.0050-08
8.00	7.5480-02	1.0000-06	2.4620-07	1.0000-04	8.1260-05	3.5000-09	4.0050-08
8.20	7.5310-02	1.0000-06	2.4630-07	1.0000-06	8.1280-05	3.5000-09	4.0050-08
8.40	7.5150-02	1.0000-06	2.4640-07	1.0000-04	8.1300-05	3.5000-09	4.0050-08
8.60	7.4990-02	1.0000-06	2.4650-07	1.0000-06	8.1310-05	3.5000-09	4.0050-08
8.80	7.4830-02	1.0000-06	2.4660-07	1.0000-04	8.1330-05	3.5000-09	4.0050-08
9.00	7.4690-02	1.0000-06	2.4670-07	1.0000-06	8.1350-05	3.5000-09	4.0050-08
9.20	7.4550-02	1.0000-06	2.4680-07	1.0000-04	8.1370-05	3.5000-09	4.0050-08
9.40	7.4410-02	1.0000-06	2.4690-07	1.0000-06	8.1390-05	3.5000-09	4.0060-08
9.60	7.4280-02	1.0000-06	2.4680-07	1.0000-04	8.1400-05	3.5000-09	4.0060-08
9.80	7.4160-02	1.0000-06	2.4690-07	1.0000-06	8.1420-05	3.5000-09	4.0060-08
10.00	7.4040-02	1.0000-06	2.4690-07	1.0000-04	8.1440-05	3.5000-09	4.0060-08

T=300 K, H= 25 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$N_2O_5 + M \ggg NO_2 + NO_3 + M$	3.270-11	5.940-14
2	$2^*NO_3 \ggg 2^*NO_2 + O_2$	2.410-16	1.750-54
3	$NO_2 + NO_3 \ggg NO_2 + NO + O_2$	4.210-15	4.920-35
4	$NO_3 + NO \ggg 2^*NO_2$	1.900-11	6.980-29
5	$NO + O_3 \ggg NO_2 + O_2$	1.670-14	2.850-49
6	$NO_2 + O_3 \ggg NO_3 + O_2$	3.410-17	2.000-34
7	$HN_3 + M \ggg H_2 + N_2 + M$	4.010-21	1.300-12
8	$HN_3 + H_2 \ggg H_2O + N_2$	4.000-14	1.250-26
9	$O + O + M \ggg O_2 + M$	1.240-15	5.480-58
10	$O + O_2 + M \ggg O_3 + M$	3.420-16	2.740-09
11	$O + O_3 \ggg 2^*O_2$	8.900-15	0.0
12	$O + NO + M \ggg NO_2 + M$	6.400-14	2.510-38
13	$O + NO_2 \ggg NO + O_2$	6.250-12	3.730-46
14	$O + NO_2 + M \ggg NO_3 + M$	5.490-14	4.710-24
15	$H_2 + H_2 \ggg H_2O + H_2$	1.570-12	4.630-24
16	$O_2 + 2^*NO \ggg 2^*NO_2$	1.930-34	2.260-31
17	$NO_2 + H-NU \ggg NO + O$	0.0	0.0
18	$O + H_2 \ggg H + O_2$	4.200-11	2.160-22
19	$O + H_2O \ggg H_2 + O_2$	1.510-11	8.510-52
20	$O_2 + H + M \ggg H_2O_2 + M$	1.660-14	1.360-24
21	$O_3 + H \ggg H_2O + O_2$	1.790-11	1.620-69
22	$O_3 + H_2 \ggg H_2O_2 + O_2$	5.350-14	8.980-43
23	$O_3 + H_2O \ggg H_2O + 2^*O_2$	1.040-15	1.530-63
24	$H + H_2 + M \ggg H_2O + M$	2.940-13	2.380-74
25	$H + H_2O \ggg 2^*H_2O$	1.770-11	1.140-40
26	$H + H_2O_2 \ggg H_2 + O_2$	1.310-11	6.800-53
27	$H + H_2O \ggg H_2 + H_2O$	2.180-25	6.410-14
28	$H + H_2O_2 \ggg H_2 + H_2O_2$	2.130-14	2.960-24
29	$H + H_2O_2 \ggg H_2O + H_2O$	2.760-14	2.950-64

30	$2^*H_2O + M \ggg H_2O_2 + M$	1.480-13	1.080-24
31	$H_2O + H_2O_2 \ggg H_2O + O_2$	1.570-11	1.980-63
32	$2^*H_2O_2 \ggg H_2O_2 + O_2$	4.210-12	1.940-42
33	$H_2O_2 + H_2O \ggg H_2O_2 + H_2O$	6.110-35	8.570-13
34	$NO + H + M \ggg HNO + M$	1.200-14	3.040-26
35	$NO + HO \ggg NO_2 + H$	7.190-34	4.920-11
36	$NO + HO + M \ggg HNO_2 + M$	9.200-13	5.170-22
37	$NO + H_2O_2 \ggg NO_2 + H_2O$	3.660-13	3.740-19
38	$H + H + M \ggg H_2 + M$	5.890-16	5.070-69
39	$HNO_4 + M \ggg H_2O + NO_2 + M$	7.360-03	8.820-14
40	$CLNO_3 + M \ggg ClO + NO_2 + M$	4.760-04	7.770-14

HAPP RESIDENCE TIME STUDY

TIME (S)	1205	1002	904	700	34	02	1003	10	1900	12
0.0	7.0000	6.5000	2.0000	7.0000	4.2000	1.7000	2.0000	1.5000	1.9000	12
0.20	6.9950	6.5150	2.0000	6.9950	4.2000	1.7000	2.0000	1.4800	1.9000	12
0.40	6.9910	6.5300	3.2750	6.8000	4.2000	1.7000	2.0000	1.4400	1.9000	12
0.60	6.9860	6.5450	3.9100	6.7110	4.2000	1.7000	2.0000	1.4070	1.9000	12
0.80	6.9820	6.5600	4.5440	6.6170	4.2000	1.7000	2.0000	1.3800	1.9000	12
1.00	6.9770	6.5750	5.1760	6.5250	4.2000	1.7000	2.0000	1.3590	1.9000	12
1.20	6.9730	6.5900	5.8070	6.4340	4.2000	1.7000	2.0000	1.3430	1.9000	12
1.40	6.9680	6.6040	6.4370	6.3440	4.2000	1.7000	2.0000	1.3330	1.9000	12
1.60	6.9630	6.6190	7.0660	6.2550	4.2000	1.7000	2.0000	1.3280	1.9000	12
1.80	6.9590	6.6320	7.6930	6.1640	4.2000	1.7000	2.0000	1.3270	1.9000	12
2.00	6.9540	6.6460	8.3190	6.0720	4.2000	1.7000	2.0000	1.3310	1.9000	12
2.20	6.9500	6.6600	8.9440	5.9970	4.2000	1.7000	2.0000	1.3390	1.9000	12
2.40	6.9450	6.6740	9.5680	5.9130	4.2000	1.7000	2.0000	1.3510	1.9000	12
2.60	6.9410	6.6870	1.0190	5.8300	4.2000	1.7000	2.0000	1.3580	1.9000	12
2.80	6.9360	6.7010	1.0410	5.7480	4.2000	1.7000	2.0000	1.3870	1.9000	12
3.00	6.9320	6.7140	1.1430	5.6680	4.2000	1.7000	2.0000	1.4110	1.9000	12
3.20	6.9270	6.7270	1.2050	5.5890	4.2000	1.7000	2.0000	1.4370	1.9000	12
3.40	6.9230	6.7410	1.2670	5.5100	4.2000	1.7000	2.0000	1.4670	1.9000	12
3.60	6.9180	6.7540	1.3290	5.4330	4.2000	1.7000	2.0000	1.5000	1.9000	12
3.80	6.9140	6.7670	1.3910	5.3570	4.2000	1.7000	2.0000	1.5350	1.9000	12
4.00	6.9090	6.7790	1.4520	5.2820	4.2000	1.7000	2.0000	1.5730	1.9000	12
4.20	6.9050	6.7920	1.5140	5.2080	4.2000	1.7000	2.0000	1.6140	1.9000	12
4.40	6.9000	6.8050	1.5750	5.1350	4.2000	1.7000	2.0000	1.6570	1.9000	12
4.60	6.8960	6.8170	1.6370	5.0630	4.2000	1.7000	2.0000	1.7020	1.9000	12
4.80	6.8910	6.8300	1.6980	4.9920	4.2000	1.7000	2.0000	1.7500	1.9000	12
5.00	6.8870	6.8420	1.7590	4.9220	4.2000	1.7000	2.0000	1.7990	1.9000	12
5.20	6.8820	6.8540	1.8210	4.8530	4.2000	1.7000	2.0000	1.8510	1.9000	12
5.40	6.8780	6.8660	1.8820	4.7850	4.2000	1.7000	2.0000	1.9040	1.9000	12
5.60	6.8730	6.8780	1.9430	4.7180	4.2000	1.7000	2.0000	1.9590	1.9000	12
5.80	6.8690	6.8900	2.0040	4.6520	4.2000	1.7000	2.0000	2.0150	1.9000	12
6.00	6.8640	6.9020	2.0650	4.5870	4.2000	1.7000	2.0000	2.0740	1.9000	12
6.20	6.8600	6.9130	2.1260	4.5220	4.2000	1.7000	2.0000	2.1330	1.9000	12
6.40	6.8550	6.9250	2.1860	4.4590	4.2000	1.7000	2.0000	2.1940	1.9000	12
6.60	6.8510	6.9360	2.2470	4.3960	4.2000	1.7000	2.0000	2.2560	1.9000	12
6.80	6.8460	6.9480	2.3080	4.3350	4.2000	1.7000	2.0000	2.3200	1.9000	12
7.00	6.8420	6.9590	2.3690	4.2740	4.2000	1.7000	2.0000	2.3840	1.9000	12
7.20	6.8370	6.9700	2.4290	4.2140	4.2000	1.7000	2.0000	2.4500	1.9000	12
7.40	6.8330	6.9810	2.4900	4.1540	4.2000	1.7000	2.0000	2.5160	1.9000	12
7.60	6.8280	6.9920	2.5500	4.0960	4.2000	1.7000	2.0000	2.5840	1.9000	12
7.80	6.8240	7.0030	2.6110	4.0390	4.2000	1.7000	2.0000	2.6520	1.9000	12
8.00	6.8200	7.0140	2.6710	3.9820	4.2000	1.7000	2.0000	2.7210	1.9000	12
8.20	6.8150	7.0250	2.7320	3.9260	4.2000	1.7000	2.0000	2.7910	1.9000	12
8.40	6.8110	7.0350	2.7920	3.8710	4.2000	1.7000	2.0000	2.8620	1.9000	12
8.60	6.8060	7.0460	2.8520	3.8160	4.2000	1.7000	2.0000	2.9330	1.9000	12
8.80	6.8020	7.0560	2.9130	3.7620	4.2000	1.7000	2.0000	3.0050	1.9000	12
9.00	6.7970	7.0670	2.9730	3.7090	4.2000	1.7000	2.0000	3.0780	1.9000	12
9.20	6.7930	7.0770	3.0330	3.6570	4.2000	1.7000	2.0000	3.1510	1.9000	12
9.40	6.7890	7.0870	3.0940	3.6060	4.2000	1.7000	2.0000	3.2250	1.9000	12
9.60	6.7840	7.0970	3.1540	3.5550	4.2000	1.7000	2.0000	3.2990	1.9000	12
9.80	6.7800	7.1080	3.2140	3.5050	4.2000	1.7000	2.0000	3.3730	1.9000	12
10.00	6.7750	7.1180	3.2740	3.4560	4.2000	1.7000	2.0000	3.4480	1.9000	12

T= 700 K, H= 25 km, 19 torr

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N2O5} + \text{M} \gg \text{N02} + \text{N03} + \text{M}$	4.800-05	4.940-15
2	$2*\text{N03} \gg 2*\text{N02} + \text{O2}$	2.570-14	7.440-49
3	$\text{N02} + \text{N03} \gg \text{N02} + \text{NO} + \text{O2}$	5.510-14	3.970-36
4	$\text{N03} + \text{NO} \gg 2*\text{N02}$	1.900-11	2.230-15
5	$\text{NO} + \text{O3} \gg \text{N02} + \text{O2}$	2.650-13	4.540-28
6	$\text{N02} + \text{O3} \gg \text{N03} + \text{O2}$	3.620-15	6.560-22
7	$\text{HN03} + \text{M} \gg \text{HO} + \text{N02} + \text{M}$	4.010-02	6.700-14
8	$\text{HN03} + \text{HO} \gg \text{H2O} + \text{N03}$	4.000-14	4.740-19
9	$\text{O} + \text{O} + \text{M} \gg \text{O2} + \text{M}$	9.500-17	6.210-21
10	$\text{O} + \text{O2} + \text{M} \gg \text{O3} + \text{M}$	5.550-17	3.360-00
11	$\text{O} + \text{O3} \gg 2*\text{O2}$	7.110-13	8.630-43
12	$\text{O} + \text{NO} + \text{M} \gg \text{N02} + \text{M}$	9.010-15	2.140-11
13	$\text{O} + \text{N02} \gg \text{NO} + \text{O2}$	1.110-11	8.500-27
14	$\text{O} + \text{N02} + \text{M} \gg \text{N03} + \text{M}$	2.530-14	2.020-24
15	$\text{HO} + \text{HO} \gg \text{H2O} + \text{O}$	4.530-12	2.040-16
16	$\text{O2} + 2*\text{NO} \gg 2*\text{N02}$	7.040-39	9.370-21
17	$\text{N02} + \text{H-NH} \gg \text{NO} + \text{O}$	0.0	0.0
18	$\text{O} + \text{HO} \gg \text{H} + \text{O2}$	4.200-11	2.110-15
19	$\text{O} + \text{H02} \gg \text{HO} + \text{O2}$	3.910-11	2.190-28
20	$\text{O2} + \text{H} + \text{M} \gg \text{H02} + \text{M}$	2.750-15	6.180-06
21	$\text{O3} + \text{H} \gg \text{HO} + \text{O2}$	4.780-11	5.620-37
22	$\text{O3} + \text{HO} \gg \text{H02} + \text{O2}$	3.590-13	8.310-26
23	$\text{O3} + \text{H02} \gg \text{HO} + 2*\text{O2}$	1.180-14	1.120-49
24	$\text{H} + \text{HO} + \text{M} \gg \text{H2O} + \text{M}$	1.390-14	3.350-27
25	$\text{H} + \text{H02} \gg 2*\text{H0}$	1.080-10	5.870-24
26	$\text{H} + \text{H02} \gg \text{H2} + \text{O2}$	2.550-11	8.030-25
27	$\text{H} + \text{H2O} \gg \text{H2} + \text{HO}$	6.560-17	8.900-13
28	$\text{H} + \text{H2O2} \gg \text{H2} + \text{H02}$	3.060-13	1.770-19
29	$\text{H} + \text{H2O2} \gg \text{HO} + \text{H2O}$	3.980-13	1.020-35

30	2*H0 + M >>> H2O2 + M	1.140-14	1.040-04
31	H0 + H02 >>> H2O + O2	4.060-11	7.730-33
32	2*H02 >>> H2O2 + O2	4.320-12	3.020-24
33	H02 + H2O >>> H2O2 + H0	2.720-21	3.070-12
34	NO + H + M >>> HNO + M	2.910-15	2.200-06
35	NO + H0 >>> H02 + H	2.230-21	2.020-10
36	NO + H0 + M >>> HNO2 + M	4.760-14	5.930-03
37	NO + H02 >>> NO2 + H0	3.600-12	2.680-14
38	H + H + M >>> H2 + M	2.530-16	2.110-25
39	HNO4 + M >>> H02 + NO2 + M	6.570 05	6.140-14
40	CLN03 + M >>> CLO + NO2 + M	3.750 05	2.590-15

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N00	N03	N01	U1	O2	H003	H0	H20
0.0	7.0000-08	6.5000 09	2.0000 06	7.0000 08	9.2000 12	1.7000 17	2.0000 09	1.5000 06	1.9000 12
0.20	2.5000-02	3.3500 09	7.2000 08	8.4330 09	2.1170 12	1.7000 17	1.9840 09	2.2640 09	1.9000 12
0.40	9.2700-03	1.2600 09	7.1170 08	1.0550 10	2.3000 12	1.7000 17	1.9680 09	2.3040 09	1.9000 12
0.60	6.6000-03	9.4610 08	6.2040 08	1.0910 10	1.9610 12	1.7000 17	1.9530 09	2.3340 09	1.9000 12
0.80	6.0510-03	8.8640 08	6.6420 08	1.1010 10	1.7070 12	1.7000 17	1.9370 09	2.3550 09	1.9000 12
1.00	5.7550-03	8.7450 08	6.4030 08	1.1060 10	1.5100 12	1.7000 17	1.9210 09	2.3700 09	1.9000 12
1.20	5.5300-03	8.7350 08	6.1600 08	1.1100 10	1.1520 12	1.7000 17	1.9060 09	2.3790 09	1.9000 12
1.40	5.3470-03	8.7500 08	5.9400 08	1.1140 10	1.2230 12	1.7000 17	1.8910 09	2.3840 09	1.9000 12
1.60	5.1600-03	8.7920 08	5.7190 08	1.1170 10	1.1160 12	1.7000 17	1.8760 09	2.3850 09	1.9000 12
1.80	4.9960-03	8.8320 08	5.5030 08	1.1210 10	1.0250 12	1.7000 17	1.8610 09	2.3840 09	1.9000 12
2.00	4.8300-03	8.8750 08	5.2950 08	1.1240 10	9.4690 11	1.7000 17	1.8460 09	2.3790 09	1.9000 12
2.20	4.6600-03	8.9100 08	5.0930 08	1.1270 10	8.7960 11	1.7000 17	1.8310 09	2.3730 09	1.9000 12
2.40	4.5120-03	8.9620 08	4.8980 08	1.1300 10	8.2070 11	1.7000 17	1.8160 09	2.3650 09	1.9000 12
2.60	4.3590-03	9.0060 08	4.7100 08	1.1330 10	7.6890 11	1.7000 17	1.8020 09	2.3550 09	1.9000 12
2.80	4.2110-03	9.0490 08	4.5200 08	1.1360 10	7.2290 11	1.7000 17	1.7870 09	2.3450 09	1.9000 12
3.00	4.0670-03	9.0920 08	4.3520 08	1.1390 10	6.8190 11	1.7000 17	1.7730 09	2.3330 09	1.9000 12
3.20	3.9270-03	9.1340 08	4.1830 08	1.1410 10	6.4500 11	1.7000 17	1.7590 09	2.3200 09	1.9000 12
3.40	3.7900-03	9.1750 08	4.0200 08	1.1430 10	6.1160 11	1.7000 17	1.7450 09	2.3070 09	1.9000 12
3.60	3.6500-03	9.2140 08	3.8620 08	1.1460 10	5.8140 11	1.7000 17	1.7310 09	2.2930 09	1.9000 12
3.80	3.5090-03	9.2530 08	3.7110 08	1.1490 10	5.5300 11	1.7000 17	1.7170 09	2.2790 09	1.9000 12
4.00	3.4000-03	9.2900 08	3.5650 08	1.1510 10	5.2850 11	1.7000 17	1.7030 09	2.2640 09	1.9000 12
4.20	3.2820-03	9.3270 08	3.4240 08	1.1530 10	5.0540 11	1.7000 17	1.6900 09	2.2490 09	1.9000 12
4.40	3.1640-03	9.3620 08	3.2890 08	1.1560 10	4.8400 11	1.7000 17	1.6760 09	2.2340 09	1.9000 12
4.60	3.0500-03	9.3960 08	3.1590 08	1.1580 10	4.6430 11	1.7000 17	1.6630 09	2.2180 09	1.9000 12
4.80	2.9390-03	9.4290 08	3.0330 08	1.1600 10	4.4590 11	1.7000 17	1.6490 09	2.2030 09	1.9000 12
5.00	2.8320-03	9.4610 08	2.9130 08	1.1620 10	4.2890 11	1.7000 17	1.6360 09	2.1870 09	1.9000 12
5.20	2.7280-03	9.4910 08	2.7970 08	1.1640 10	4.1310 11	1.7000 17	1.6230 09	2.1720 09	1.9000 12
5.40	2.6200-03	9.5210 08	2.6860 08	1.1660 10	3.9830 11	1.7000 17	1.6100 09	2.1560 09	1.9000 12
5.60	2.5110-03	9.5500 08	2.5790 08	1.1690 10	3.8440 11	1.7000 17	1.5970 09	2.1400 09	1.9000 12
5.80	2.4370-03	9.5770 08	2.4760 08	1.1710 10	3.7140 11	1.7000 17	1.5840 09	2.1240 09	1.9000 12
6.00	2.3450-03	9.6040 08	2.3770 08	1.1730 10	3.5920 11	1.7000 17	1.5720 09	2.1090 09	1.9000 12
6.20	2.2580-03	9.6300 08	2.2820 08	1.1740 10	3.4770 11	1.7000 17	1.5590 09	2.0930 09	1.9000 12
6.40	2.1740-03	9.6550 08	2.1910 08	1.1760 10	3.3690 11	1.7000 17	1.5470 09	2.0780 09	1.9000 12
6.60	2.0920-03	9.6790 08	2.1030 08	1.1780 10	3.2660 11	1.7000 17	1.5340 09	2.0620 09	1.9000 12
6.80	2.0130-03	9.7020 08	2.0190 08	1.1800 10	3.1700 11	1.7000 17	1.5220 09	2.0470 09	1.9000 12
7.00	1.9370-03	9.7250 08	1.9300 08	1.1820 10	3.0780 11	1.7000 17	1.5100 09	2.0320 09	1.9000 12
7.20	1.8640-03	9.7470 08	1.8610 08	1.1840 10	2.9910 11	1.7000 17	1.4980 09	2.0170 09	1.9000 12
7.40	1.7930-03	9.7600 08	1.7860 08	1.1850 10	2.9080 11	1.7000 17	1.4860 09	2.0020 09	1.9000 12
7.60	1.7250-03	9.7800 08	1.7150 08	1.1870 10	2.8300 11	1.7000 17	1.4740 09	1.9890 09	1.9000 12
7.80	1.6590-03	9.8000 08	1.6460 08	1.1890 10	2.7550 11	1.7000 17	1.4620 09	1.9730 09	1.9000 12
8.00	1.5960-03	9.8270 08	1.5800 08	1.1900 10	2.6840 11	1.7000 17	1.4510 09	1.9590 09	1.9000 12
8.20	1.5350-03	9.8440 08	1.5170 08	1.1920 10	2.6160 11	1.7000 17	1.4390 09	1.9450 09	1.9000 12
8.40	1.4760-03	9.8640 08	1.4560 08	1.1930 10	2.5510 11	1.7000 17	1.4270 09	1.9310 09	1.9000 12
8.60	1.4200-03	9.8810 08	1.3900 08	1.1950 10	2.4890 11	1.7000 17	1.4160 09	1.9170 09	1.9010 12
8.80	1.3660-03	9.8900 08	1.3420 08	1.1960 10	2.4300 11	1.7000 17	1.4050 09	1.9030 09	1.9010 12
9.00	1.3130-03	9.9150 08	1.2890 08	1.1980 10	2.3730 11	1.7000 17	1.3930 09	1.8900 09	1.9010 12
9.20	1.2630-03	9.9310 08	1.2370 08	1.1990 10	2.3180 11	1.7000 17	1.3820 09	1.8760 09	1.9010 12
9.40	1.2150-03	9.9470 08	1.1860 08	1.2010 10	2.2660 11	1.7000 17	1.3710 09	1.8630 09	1.9010 12
9.60	1.1680-03	9.9620 08	1.1410 08	1.2020 10	2.2160 11	1.7000 17	1.3600 09	1.8500 09	1.9010 12
9.80	1.1240-03	9.9770 08	1.0960 08	1.2030 10	2.1670 11	1.7000 17	1.3490 09	1.8370 09	1.9010 12
10.00	1.0810-03	9.9920 08	1.0520 08	1.2050 10	2.1210 11	1.7000 17	1.3390 09	1.8250 09	1.9010 12

T=800 K, H= 25 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N}_2\text{O}_5 + \text{M} \gg \text{N}_2\text{O}_4 + \text{NO}_3 + \text{M}$	2.650-00	3.700-15
2	$2^*\text{NO}_3 \gg 2^*\text{NO}_2 + \text{O}_2$	3.940-14	5.950-48
3	$\text{NO}_2 + \text{NO}_3 \gg \text{NO}_2 + \text{NO} + \text{O}_2$	6.590-14	3.130-36
4	$\text{NO}_3 + \text{NO} \gg 2^*\text{NO}_2$	1.900-11	1.740-18
5	$\text{NO} + \text{O}_3 \gg \text{NO}_2 + \text{O}_2$	3.430-13	4.420-26
6	$\text{NO}_2 + \text{O}_3 \gg \text{NO}_3 + \text{O}_2$	5.610-15	9.780-21
7	$\text{HNO}_3 + \text{M} \gg \text{HO} + \text{NO}_2 + \text{M}$	1.900-00	4.200-14
8	$\text{HNO}_3 + \text{HO} \gg \text{H}_2\text{O} + \text{NO}_3$	4.000-14	2.430-18
9	$\text{O} + \text{O} + \text{M} \gg \text{O}_2 + \text{M}$	7.040-17	1.870-16
10	$\text{O} + \text{O}_2 + \text{M} \gg \text{O}_3 + \text{M}$	4.430-17	2.260-01
11	$\text{O} + \text{O}_3 \gg 2^*\text{O}_2$	1.070-12	6.110-39
12	$\text{O} + \text{NO} + \text{M} \gg \text{NO}_2 + \text{M}$	7.110-15	6.780-09
13	$\text{O} + \text{NO}_2 \gg \text{NO} + \text{O}_2$	1.170-11	5.550-25
14	$\text{O} + \text{NO}_2 + \text{M} \gg \text{NO}_3 + \text{M}$	2.210-14	1.770-24
15	$\text{HO} + \text{HO} \gg \text{H}_2\text{O} + \text{O}$	5.000-12	1.060-15
16	$\text{O}_2 + 2^*\text{NO} \gg 2^*\text{NO}_2$	6.400-39	8.780-20
17	$\text{NO}_2 + \text{H-NO} \gg \text{NO} + \text{O}$	0.0	0.0
18	$\text{O} + \text{HO} \gg \text{H} + \text{O}_2$	4.200-11	9.550-15
19	$\text{O} + \text{HO}_2 \gg \text{HO} + \text{O}_2$	4.240-11	3.430-26
20	$\text{O}_2 + \text{H} + \text{M} \gg \text{HO}_2 + \text{M}$	2.200-15	3.280-04
21	$\text{O}_3 + \text{H} \gg \text{HO} + \text{O}_2$	5.250-11	6.310-34
22	$\text{O}_3 + \text{HO} \gg \text{HO}_2 + \text{O}_2$	4.300-13	3.240-24
23	$\text{O}_3 + \text{HO}_2 \gg \text{HO} + 2^*\text{O}_2$	1.480-14	2.350-48
24	$\text{H} + \text{HO} + \text{M} \gg \text{H}_2\text{O} + \text{M}$	4.610-15	8.520-23
25	$\text{H} + \text{HO}_2 \gg 2^*\text{HO}$	1.280-10	2.160-22
26	$\text{H} + \text{HO}_2 \gg \text{H}_2 + \text{O}_2$	2.710-11	1.450-26
27	$\text{H} + \text{H}_2\text{O} \gg \text{H}_2 + \text{HO}$	4.090-10	1.410-12
28	$\text{H} + \text{H}_2\text{O}_2 \gg \text{H}_2 + \text{HO}_2$	3.430-13	9.470-18
29	$\text{H} + \text{H}_2\text{O}_2 \gg \text{HO} + \text{H}_2\text{O}$	5.110-13	6.020-33

30	2*H0 + M >>> H2O2 + H	4.500-15	7.010-03
31	H0 + H02 >>> H2O + O2	4.440-11	5.700-30
32	2*H02 >>> H2O2 + O2	4.100-12	1.530-22
33	H02 + H2O >>> H2O2 + H0	5.190-20	3.460-12
34	N0 + H + M >>> HN0 + H	2.410-15	1.430-04
35	N0 + H0 >>> H02 + H	3.300-20	2.300-10
36	N0 + H0 + M >>> HN02 + H	3.420-14	3.250-01
37	N0 + H02 >>> N02 + H0	4.460-12	7.660-14
38	H + H + M >>> H2 + M	2.210-15	2.330-21
39	HN04 + M >>> H02 + N02 + H	3.480 06	4.550-15
40	CLN03 + M >>> CL0 + N02 + M	2.430 06	1.500-15

HAPP RESIDENCE TIME STUDY

TIME (S)	H205	H202	H203	H20	H20	H203	H20	H20
0.0	7.0000 08	6.5000 09	2.0000 05	1.0000 04	4.0000 12	1.7000 17	2.0000 09	1.5000 06
0.20	2.9500 04	2.9430 04	1.0820 04	1.2140 10	7.3780 11	1.7000 17	1.3680 04	1.5000 06
0.40	2.1530 04	2.2750 08	6.7710 08	1.2580 10	3.8780 11	1.7000 17	9.3600 04	6.3440 09
0.60	2.0090 04	2.2230 08	6.4710 08	1.2990 10	3.8780 11	1.7000 17	6.4030 04	7.5520 09
0.80	1.8680 04	2.1690 04	6.1690 04	1.3230 10	4.1510 11	1.7000 17	6.3800 04	8.3910 09
1.00	1.7380 04	2.1140 04	5.8750 04	1.3460 10	3.3460 11	1.7000 17	2.9960 04	8.9470 09
1.20	1.6210 04	2.0740 04	5.5400 04	1.3530 10	3.1760 11	1.7000 17	2.0500 04	9.4000 09
1.40	1.5180 04	2.0440 04	5.3160 04	1.3620 10	2.9040 11	1.7000 17	1.4020 04	9.6740 09
1.60	1.4270 04	2.0230 04	5.0530 04	1.3690 10	2.7150 11	1.7000 17	9.5420 07	9.8430 09
1.80	1.3490 04	2.0100 04	4.8020 04	1.3750 10	2.2740 11	1.7000 17	6.5630 07	9.9330 09
2.00	1.2770 04	2.0040 04	4.5620 04	1.3800 10	2.0700 11	1.7000 17	4.4900 07	9.9620 09
2.20	1.2130 04	2.0040 04	4.3330 04	1.3830 10	1.4460 11	1.7000 17	3.0730 07	9.9450 09
2.40	1.1540 04	2.0000 04	4.1140 04	1.3860 10	1.7450 11	1.7000 17	2.1030 07	9.8930 09
2.60	1.1000 04	2.0170 08	3.9060 04	1.3890 10	1.6140 11	1.7000 17	1.4400 07	9.8140 09
2.80	1.0500 04	2.0280 08	3.7090 04	1.3910 10	1.4990 11	1.7000 17	9.8660 06	9.7150 09
3.00	1.0030 04	2.0400 04	3.5200 04	1.3930 10	1.3980 11	1.7000 17	6.7620 06	9.6020 09
3.20	9.5810 05	2.0530 04	3.3420 04	1.3950 10	1.3070 11	1.7000 17	6.6400 06	9.4780 09
3.40	9.1530 05	2.0670 08	3.1720 04	1.3970 10	1.2270 11	1.7000 17	3.1870 06	9.3470 09
3.60	8.7430 05	2.0800 04	3.0100 04	1.3980 10	1.1540 11	1.7000 17	2.1940 06	9.2110 09
3.80	8.3500 05	2.0930 04	2.8560 04	1.4000 10	1.0880 11	1.7000 17	1.5140 06	9.0710 09
4.00	7.9730 05	2.1050 08	2.7110 04	1.4010 10	1.0290 11	1.7000 17	1.0490 06	8.9300 09
4.20	7.6100 05	2.1190 08	2.5720 04	1.4020 10	9.7500 10	1.7000 17	7.3100 05	8.7880 09
4.40	7.2600 05	2.1300 04	2.4410 04	1.4030 10	9.2560 10	1.7000 17	5.1310 05	8.6460 09
4.60	6.9240 05	2.1410 04	2.3160 04	1.4050 10	8.8040 10	1.7000 17	3.6400 05	8.5060 09
4.80	6.6010 05	2.1520 04	2.1970 04	1.4060 10	8.3880 10	1.7000 17	2.6180 05	8.3680 09
5.00	6.2910 05	2.1610 04	2.0850 04	1.4070 10	8.0050 10	1.7000 17	1.9170 05	8.2310 09
5.20	5.9930 05	2.1700 04	1.9780 04	1.4080 10	7.6510 10	1.7000 17	1.4370 05	8.0970 09
5.40	5.7070 05	2.1780 04	1.8760 04	1.4090 10	7.3230 10	1.7000 17	1.1060 05	7.9660 09
5.60	5.4320 05	2.1850 04	1.7800 04	1.4100 10	7.0180 10	1.7000 17	8.7830 04	7.8370 09
5.80	5.1700 05	2.1920 04	1.6890 04	1.4100 10	6.7350 10	1.7000 17	7.2090 04	7.7120 09
6.00	4.9180 05	2.1940 04	1.6020 04	1.4110 10	6.4700 10	1.7000 17	6.1160 04	7.5890 09
6.20	4.6770 05	2.2040 04	1.5200 04	1.4120 10	6.2230 10	1.7000 17	5.3520 04	7.4690 09
6.40	4.4470 05	2.2040 04	1.4420 04	1.4130 10	5.9920 10	1.7000 17	4.8140 04	7.3530 09
6.60	4.2280 05	2.2130 04	1.3680 04	1.4130 10	5.7750 10	1.7000 17	4.4300 04	7.2390 09
6.80	4.0180 05	2.2180 04	1.2910 04	1.4150 10	5.5790 10	1.7000 17	3.9460 04	7.1290 09
7.00	3.8180 05	2.2210 04	1.2180 04	1.4150 10	5.3790 10	1.7000 17	3.5680 04	7.0210 09
7.20	3.6280 05	2.2240 04	1.1480 04	1.4160 10	5.1990 10	1.7000 17	3.1890 04	6.9160 09
7.40	3.4460 05	2.2270 04	1.0800 04	1.4160 10	5.0280 10	1.7000 17	2.8150 04	6.8150 09
7.60	3.2730 05	2.2300 04	1.0110 04	1.4170 10	4.8670 10	1.7000 17	2.4690 04	6.7160 09
7.80	3.1040 05	2.2320 04	9.9710 07	1.4170 10	4.7140 10	1.7000 17	2.1480 04	6.6190 09
8.00	2.9510 05	2.2340 04	9.4590 07	1.4170 10	4.5700 10	1.7000 17	1.8480 04	6.5250 09
8.20	2.8020 05	2.2360 04	8.9740 07	1.4180 10	4.4330 10	1.7000 17	1.5750 04	6.4340 09
8.40	2.6600 05	2.2370 04	8.5140 07	1.4180 10	4.3030 10	1.7000 17	1.3000 04	6.3460 09
8.60	2.5250 05	2.2390 04	8.0780 07	1.4190 10	4.1790 10	1.7000 17	1.0250 04	6.2590 09
8.80	2.3970 05	2.2400 04	7.6640 07	1.4190 10	4.0610 10	1.7000 17	0.7500 04	6.1750 09
9.00	2.2760 05	2.2410 04	7.2720 07	1.4200 10	3.9490 10	1.7000 17	0.4750 04	6.0930 09
9.20	2.1600 05	2.2420 04	6.9000 07	1.4200 10	3.8420 10	1.7000 17	0.2000 04	6.0140 09
9.40	2.0500 05	2.2420 04	6.5470 07	1.4200 10	3.7420 10	1.7000 17	0.0270 04	5.9360 09
9.60	1.9460 05	2.2430 04	6.2130 07	1.4210 10	3.6420 10	1.7000 17	0.0000 04	5.8610 09
9.80	1.8470 05	2.2440 04	5.8950 07	1.4210 10	3.5490 10	1.7000 17	0.0000 04	5.7880 09
10.00	1.7530 05	2.2440 04	5.5950 07	1.4210 10	3.4590 10	1.7000 17	0.0000 04	5.7160 09

TIME (S)	C	W	H ₁₂	H ₁₀	H ₀₂	H ₀₄	CLN03
0.0	1.0000 06	1.0000 06	2.3000 07	1.0000 09	0.0000 05	3.5000 09	4.0000 08
0.20	2.3150 12	2.2770 04	4.5540 04	1.0010 04	9.4470 05	3.3400-04	8.4530-05
0.40	1.4410 12	2.4330 04	1.1530 06	1.0020 06	1.3860 06	3.4340-04	6.4460-05
0.60	1.5150 12	2.3760 04	1.3720 09	1.0040 06	1.8940 06	3.5920-04	6.2980-05
0.80	1.2800 12	2.2350 04	1.1540 09	1.0050 06	2.4640 06	4.3390-04	6.1450-05
1.00	1.1030 12	2.0640 04	1.0440 06	1.0060 06	3.0800 06	4.5460-04	6.0010-05
1.20	9.6560 11	1.8930 04	1.1170 09	1.0080 06	3.7060 06	4.6640-04	5.8830-05
1.40	8.5520 11	1.7270 04	1.1760 09	1.0090 06	4.3300 06	4.7290-04	5.7920-05
1.60	7.6520 11	1.5740 04	1.1730 09	1.0100 06	4.9400 06	4.7610-04	5.7310-05
1.80	6.9050 11	1.4340 04	1.1450 09	1.0110 06	5.5270 06	4.7720-04	5.6940-05
2.00	6.2740 11	1.3080 04	1.1010 09	1.0120 06	6.0850 06	4.7720-04	5.6780-05
2.20	5.7410 11	1.1950 04	1.1130 09	1.0130 06	6.6110 06	4.7620-04	5.6790-05
2.40	5.2790 11	1.0930 04	1.1050 09	1.0130 06	7.1030 06	4.7460-04	5.6930-05
2.60	4.8780 11	1.0030 04	1.1790 09	1.0140 06	7.5600 06	4.7250-04	5.7160-05
2.80	4.5270 11	9.2130 07	1.1710 09	1.0150 06	7.9810 06	4.6990-04	5.7460-05
3.00	4.2170 11	8.4850 07	1.1740 09	1.0150 06	8.3670 06	4.6690-04	5.7800-05
3.20	3.9420 11	7.8310 07	1.1250 09	1.0160 06	8.7200 06	4.6360-04	5.8170-05
3.40	3.6960 11	7.2430 07	1.1700 09	1.0160 06	9.0390 06	4.5980-04	5.8550-05
3.60	3.4760 11	6.7130 07	1.1740 09	1.0170 06	9.3270 06	4.5570-04	5.8940-05
3.80	3.2770 11	6.2340 07	1.1640 09	1.0170 06	9.5850 06	4.5130-04	5.9310-05
4.00	3.0970 11	5.8010 07	1.1620 09	1.0170 06	9.8140 06	4.4670-04	5.9680-05
4.20	2.9340 11	5.4080 07	1.1590 09	1.0180 06	1.0020 07	4.4180-04	6.0030-05
4.40	2.7840 11	5.0500 07	1.1560 09	1.0180 06	1.0190 07	4.3680-04	6.0360-05
4.60	2.6470 11	4.7250 07	1.1540 09	1.0180 06	1.0350 07	4.3160-04	6.0670-05
4.80	2.5220 11	4.4280 07	1.1510 09	1.0190 06	1.0480 07	4.2620-04	6.0960-05
5.00	2.4060 11	4.1560 07	1.1480 09	1.0190 06	1.0590 07	4.2080-04	6.1230-05
5.20	2.2990 11	3.9040 07	1.1460 09	1.0190 06	1.0680 07	4.1530-04	6.1480-05
5.40	2.2000 11	3.6790 07	1.1430 09	1.0190 06	1.0760 07	4.0980-04	6.1710-05
5.60	2.1080 11	3.4690 07	1.1410 09	1.0200 06	1.0820 07	4.0420-04	6.1920-05
5.80	2.0230 11	3.2750 07	1.1390 09	1.0200 06	1.0860 07	3.9870-04	6.2110-05
6.00	1.9430 11	3.0960 07	1.1360 09	1.0200 06	1.0890 07	3.9310-04	6.2280-05
6.20	1.8690 11	2.9310 07	1.1340 09	1.0200 06	1.0910 07	3.8760-04	6.2440-05
6.40	1.7990 11	2.7780 07	1.1320 09	1.0200 06	1.0910 07	3.8210-04	6.2580-05
6.60	1.7340 11	2.6360 07	1.1300 09	1.0200 06	1.0900 07	3.7670-04	6.2710-05
6.80	1.6730 11	2.5040 07	1.1280 09	1.0210 06	1.0890 07	3.7140-04	6.2830-05
7.00	1.6150 11	2.3810 07	1.1250 09	1.0210 06	1.0860 07	3.6610-04	6.2930-05
7.20	1.5600 11	2.2670 07	1.1240 09	1.0210 06	1.0830 07	3.6090-04	6.3020-05
7.40	1.5090 11	2.1600 07	1.1220 09	1.0210 06	1.0790 07	3.5570-04	6.3110-05
7.60	1.4610 11	2.0600 07	1.1200 09	1.0210 06	1.0750 07	3.5070-04	6.3180-05
7.80	1.4150 11	1.9670 07	1.1180 09	1.0210 06	1.0690 07	3.4570-04	6.3240-05
8.00	1.3710 11	1.8800 07	1.1160 09	1.0210 06	1.0630 07	3.4090-04	6.3300-05
8.20	1.3300 11	1.7980 07	1.1140 09	1.0210 06	1.0570 07	3.3610-04	6.3350-05
8.40	1.2910 11	1.7210 07	1.1120 09	1.0220 06	1.0500 07	3.3150-04	6.3390-05
8.60	1.2540 11	1.6440 07	1.1100 09	1.0220 06	1.0430 07	3.2690-04	6.3430-05
8.80	1.2180 11	1.5810 07	1.1080 09	1.0220 06	1.0360 07	3.2240-04	6.3460-05
9.00	1.1850 11	1.5170 07	1.1050 09	1.0220 06	1.0280 07	3.1810-04	6.3490-05
9.20	1.1530 11	1.4560 07	1.1070 09	1.0220 06	1.0200 07	3.1380-04	6.3510-05
9.40	1.1220 11	1.3940 07	1.1050 09	1.0220 06	1.0120 07	3.0960-04	6.3540-05
9.60	1.0930 11	1.3450 07	1.1040 09	1.0220 06	1.0040 07	3.0550-04	6.3550-05
9.80	1.0650 11	1.2950 07	1.1020 09	1.0220 06	9.9520 06	3.0150-04	6.3570-05
10.00	1.0380 11	1.2460 07	1.1010 09	1.0220 06	9.8650 06	2.9770-04	6.3580-05

T=250 K, H=30 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$N_2O_5 + M \ggg NO_2 + NO_3 + M$	1.940-06	6.060-14
2	$2^*NO_3 \ggg 2^*NO_2 + O_2$	4.710-17	7.460-67
3	$NO_2 + NO_3 \ggg NO_2 + NO + O_2$	4.210-15	1.190-34
4	$NO_3 + NO \ggg 2^*NO_2$	1.900-11	3.290-32
5	$HO + O_3 \ggg HO_2 + O_2$	6.360-15	1.080-56
6	$NO_2 + O_3 \ggg NO_3 + O_2$	6.650-14	8.330-39
7	$HO_2 + M \ggg HO + NO_2 + M$	2.570-24	1.180-12
8	$HO_2 + HO \ggg H_2O + NO_3$	8.000-14	2.800-24
9	$O + O + M \ggg O_2 + M$	1.290-15	3.300-54
10	$O + O_2 + M \ggg O_3 + M$	2.760-16	7.740-17
11	$O + O_3 \ggg 2^*O_2$	1.920-15	0.0
12	$O + NO + M \ggg NO_2 + M$	5.430-14	4.020-44
13	$O + NO_2 \ggg NO + O_2$	5.120-12	6.270-53
14	$O + NO_2 + M \ggg NO_3 + M$	3.390-14	2.710-24
15	$HO + HO \ggg H_2O + O$	1.090-12	9.770-27
16	$O_2 + 2^*NO \ggg 2^*NO_2$	2.750-38	3.880-35
17	$NO_2 + H-NU \ggg NO + O$	0.0	0.0
18	$O + HO \ggg H + O_2$	4.200-11	7.730-25
19	$O + HO_2 \ggg HO + O_2$	1.080-11	5.450-60
20	$O_2 + M + M \ggg HO_2 + M$	1.330-14	1.710-31
21	$O_3 + M \ggg HO_2 + O_2$	1.270-11	0.0
22	$O_3 + HO \ggg HO_2 + O_2$	2.750-14	1.040-44
23	$O_3 + HO_2 \ggg HO + 2^*O_2$	4.450-16	2.410-68
24	$H + HO + M \ggg H_2O + M$	2.710-13	0.0
25	$H + HO_2 \ggg 2^*HO$	9.400-12	1.620-46
26	$H + HO_2 \ggg H_2 + O_2$	1.040-11	2.550-61
27	$H + H_2O \ggg H_2 + HO$	2.340-24	1.140-15
28	$H + H_2O_2 \ggg H_2 + HO_2$	8.360-15	5.620-29
29	$H + H_2O_2 \ggg HO + H_2O$	1.090-14	1.350-75
30	$2^*HO + M \ggg H_2O_2 + M$	1.550-13	4.160-32
31	$HO + HO_2 \ggg H_2O + O_2$	1.120-11	3.890-74
32	$2^*HO_2 \ggg H_2O_2 + O_2$	2.300-12	8.580-44
33	$HO_2 + H_2O \ggg H_2O_2 + HO$	1.020-39	5.490-13
34	$NO + M + M \ggg HNO + M$	8.430-15	1.320-33
35	$NO + HO \ggg NO_2 + H$	3.050-34	3.010-11
36	$NO + HO + M \ggg HNO_2 + M$	1.110-12	4.200-24
37	$NO + HO_2 \ggg NO_2 + HO$	1.650-13	7.470-21
38	$H + H + M \ggg H_2 + M$	3.390-16	0.0
39	$HNO_4 + M \ggg HO_2 + NO_2 + M$	5.800-06	1.070-13
40	$CLNO_3 + M \ggg ClO + NO_2 + M$	1.560-07	0.150-14

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	U0	U3	O2	4403	H0	H20
0.0	4.0000 08	4.2000 09	3.0000 06	4.5000 08	2.5000 12	7.8000 16	4.0000 08	3.0000 06	1.2000 12
0.20	4.0000 08	4.2010 09	3.0000 06	4.4840 08	2.5000 12	7.8000 16	4.0000 08	2.9910 06	1.2000 12
0.40	4.0000 08	4.2030 09	3.0140 06	4.4710 08	2.5000 12	7.8000 16	4.0000 08	2.9820 06	1.2000 12
0.60	4.0000 08	4.2040 09	3.0270 06	4.4570 08	2.5000 12	7.8000 16	4.0000 08	2.9140 06	1.2000 12
0.80	4.0000 08	4.2060 09	3.0350 06	4.4430 08	2.5000 12	7.8000 16	4.0000 08	2.8770 06	1.2000 12
1.00	4.0000 08	4.2070 09	3.0440 06	4.4290 08	2.5000 12	7.8000 16	4.0000 08	2.8400 06	1.2000 12
1.20	4.0000 08	4.2080 09	3.0530 06	4.4150 08	2.5000 12	7.8000 16	4.0000 08	2.8040 06	1.2000 12
1.40	4.0000 08	4.2100 09	3.0620 06	4.4010 08	2.5000 12	7.8000 16	4.0000 08	2.7680 06	1.2000 12
1.60	4.0000 08	4.2110 09	3.0710 06	4.3870 08	2.5000 12	7.8000 16	4.0000 08	2.7330 06	1.2000 12
1.80	4.0000 08	4.2130 09	3.0800 06	4.3730 08	2.5000 12	7.8000 16	4.0000 08	2.6990 06	1.2000 12
2.00	4.0000 08	4.2140 09	3.0890 06	4.3590 08	2.5000 12	7.8000 16	4.0000 08	2.6650 06	1.2000 12
2.20	4.0000 08	4.2150 09	3.0980 06	4.3450 08	2.5000 12	7.8000 16	4.0000 08	2.6310 06	1.2000 12
2.40	4.0000 08	4.2170 09	3.1070 06	4.3310 08	2.5000 12	7.8000 16	4.0000 08	2.5980 06	1.2000 12
2.60	4.0000 08	4.2180 09	3.1160 06	4.3170 08	2.5000 12	7.8000 16	4.0000 08	2.5650 06	1.2000 12
2.80	4.0000 08	4.2190 09	3.1250 06	4.3030 08	2.5000 12	7.8000 16	4.0000 08	2.5330 06	1.2000 12
3.00	4.0000 08	4.2210 09	3.1330 06	4.2900 08	2.5000 12	7.8000 16	4.0000 08	2.5020 06	1.2000 12
3.20	4.0000 08	4.2220 09	3.1420 06	4.2760 08	2.5000 12	7.8000 16	4.0000 08	2.4710 06	1.2000 12
3.40	4.0000 08	4.2240 09	3.1510 06	4.2620 08	2.5000 12	7.8000 16	4.0000 08	2.4400 06	1.2000 12
3.60	4.0000 08	4.2250 09	3.1600 06	4.2490 08	2.5000 12	7.8000 16	4.0000 08	2.4100 06	1.2000 12
3.80	4.0000 08	4.2260 09	3.1690 06	4.2350 08	2.5000 12	7.8000 16	4.0010 08	2.3800 06	1.2000 12
4.00	4.0000 08	4.2280 09	3.1780 06	4.2220 08	2.5000 12	7.8000 16	4.0010 08	2.3510 06	1.2000 12
4.20	4.0000 08	4.2290 09	3.1870 06	4.2080 08	2.5000 12	7.8000 16	4.0010 08	2.3220 06	1.2000 12
4.40	4.0000 08	4.2300 09	3.1960 06	4.1950 08	2.5000 12	7.8000 16	4.0010 08	2.2940 06	1.2000 12
4.60	4.0000 08	4.2310 09	3.2050 06	4.1810 08	2.5000 12	7.8000 16	4.0010 08	2.2660 06	1.2000 12
4.80	4.0000 08	4.2330 09	3.2140 06	4.1680 08	2.5000 12	7.8000 16	4.0010 08	2.2380 06	1.2000 12
5.00	4.0000 08	4.2340 09	3.2230 06	4.1550 08	2.5000 12	7.8000 16	4.0010 08	2.2110 06	1.2000 12
5.20	4.0000 08	4.2350 09	3.2320 06	4.1420 08	2.5000 12	7.8000 16	4.0010 08	2.1840 06	1.2000 12
5.40	4.0000 08	4.2370 09	3.2410 06	4.1280 08	2.5000 12	7.8000 16	4.0010 08	2.1580 06	1.2000 12
5.60	4.0000 08	4.2380 09	3.2500 06	4.1150 08	2.5000 12	7.8000 16	4.0010 08	2.1320 06	1.2000 12
5.80	4.0000 08	4.2390 09	3.2590 06	4.1020 08	2.5000 12	7.8000 16	4.0010 08	2.1070 06	1.2000 12
6.00	4.0000 08	4.2410 09	3.2680 06	4.0890 08	2.5000 12	7.8000 16	4.0010 08	2.0810 06	1.2000 12
6.20	4.0000 08	4.2420 09	3.2770 06	4.0760 08	2.5000 12	7.8000 16	4.0010 08	2.0570 06	1.2000 12
6.40	4.0000 08	4.2430 09	3.2860 06	4.0630 08	2.5000 12	7.8000 16	4.0010 08	2.0320 06	1.2000 12
6.60	4.0000 08	4.2440 09	3.2950 06	4.0500 08	2.5000 12	7.8000 16	4.0010 08	2.0080 06	1.2000 12
6.80	4.0000 08	4.2460 09	3.3040 06	4.0370 08	2.5000 12	7.8000 16	4.0010 08	1.9850 06	1.2000 12
7.00	4.0000 08	4.2470 09	3.3130 06	4.0240 08	2.5000 12	7.8000 16	4.0010 08	1.9610 06	1.2000 12
7.20	4.0000 08	4.2480 09	3.3220 06	4.0110 08	2.5000 12	7.8000 16	4.0010 08	1.9380 06	1.2000 12
7.40	4.0000 08	4.2500 09	3.3310 06	3.9990 08	2.5000 12	7.8000 16	4.0010 08	1.9160 06	1.2000 12
7.60	4.0000 08	4.2510 09	3.3410 06	3.9860 08	2.5000 12	7.8000 16	4.0010 08	1.8940 06	1.2000 12
7.80	4.0000 08	4.2520 09	3.3500 06	3.9730 08	2.5000 12	7.8000 16	4.0010 08	1.8720 06	1.2000 12
8.00	4.0000 08	4.2530 09	3.3590 06	3.9610 08	2.5000 12	7.8000 16	4.0010 08	1.8500 06	1.2000 12
8.20	4.0000 08	4.2550 09	3.3690 06	3.9480 08	2.5000 12	7.8000 16	4.0010 08	1.8290 06	1.2000 12
8.40	4.0000 08	4.2560 09	3.3770 06	3.9350 08	2.5000 12	7.8000 16	4.0010 08	1.8080 06	1.2000 12
8.60	4.0000 08	4.2570 09	3.3860 06	3.9230 08	2.5000 12	7.8000 16	4.0010 08	1.7870 06	1.2000 12
8.80	4.0000 08	4.2590 09	3.3950 06	3.9100 08	2.5000 12	7.8000 16	4.0010 08	1.7670 06	1.2000 12
9.00	4.0000 08	4.2600 09	3.4040 06	3.8980 08	2.5000 12	7.8000 16	4.0010 08	1.7470 06	1.2000 12
9.20	4.0000 08	4.2610 09	3.4130 06	3.8850 08	2.5000 12	7.8000 16	4.0010 08	1.7270 06	1.2000 12
9.40	4.0000 08	4.2620 09	3.4220 06	3.8730 08	2.5000 12	7.8000 16	4.0010 08	1.7080 06	1.2000 12
9.60	4.0000 08	4.2630 09	3.4320 06	3.8610 08	2.5000 12	7.8000 16	4.0010 08	1.6880 06	1.2000 12
9.80	4.0000 08	4.2640 09	3.4410 06	3.8480 08	2.5000 12	7.8000 16	4.0010 08	1.6700 06	1.2000 12
10.00	4.0000 08	4.2660 09	3.4500 06	3.8360 08	2.5000 12	7.8000 16	4.0010 08	1.6510 06	1.2000 12

TIME (s)	n	M	M2	402	M202	MNO	MNO2	MNO4	CLNO3
0.0	1.0000-06	1.0000-06	1.0000-06	2.5000-07	1.2000-09	1.0000-06	8.0000-05	1.5000-09	2.0000-08
0.25	1.1410-04	1.6050-03	1.0000-06	2.5000-07	1.2000-09	1.0000-06	8.0030-05	1.5000-09	2.0000-08
0.50	1.4031-02	2.4420-05	1.0000-06	2.6040-07	1.2000-09	1.0000-06	8.0060-05	1.5000-09	2.0000-08
0.75	2.9300-00	3.4400-06	1.0000-06	2.6100-07	1.2000-09	1.0000-06	8.0090-05	1.5000-09	2.0000-08
1.00	5.4150-01	3.1210-06	1.0000-06	2.6100-07	1.2000-09	1.0000-06	8.0120-05	1.5000-09	2.0000-08
1.25	4.9900-01	1.0770-04	1.0000-06	2.6100-07	1.2000-09	1.0000-06	8.0150-05	1.5000-09	2.0000-08
1.50	4.8020-01	3.0340-06	1.0000-06	2.6100-07	1.2000-09	1.0000-06	8.0170-05	1.5000-09	2.0000-08
1.75	4.7810-01	2.9970-06	1.0000-06	2.6200-07	1.2000-09	1.0000-06	8.0200-05	1.5000-09	2.0000-08
2.00	4.6330-01	2.9580-06	1.0000-06	2.6230-07	1.2000-09	1.0000-06	8.0230-05	1.5000-09	2.0000-08
2.25	4.5480-01	2.9190-06	1.0000-06	2.6260-07	1.2000-09	1.0000-06	8.0260-05	1.5000-09	2.0000-08
2.50	4.4550-01	2.8810-06	1.0000-06	2.6290-07	1.2000-09	1.0000-06	8.0280-05	1.5000-09	2.0000-08
2.75	4.4050-01	2.8440-06	1.0000-06	2.6320-07	1.2000-09	1.0000-06	8.0300-05	1.5000-09	2.0000-08
3.00	4.3170-01	2.8080-06	1.0000-06	2.6350-07	1.2000-09	1.0000-06	8.0330-05	1.5000-09	2.0000-08
3.25	4.2320-01	2.7710-06	1.0000-06	2.6370-07	1.2000-09	1.0000-06	8.0350-05	1.5000-09	2.0000-08
3.50	4.1490-01	2.7360-06	1.0000-06	2.6400-07	1.2000-09	1.0000-06	8.0380-05	1.5000-09	2.0000-08
3.75	4.0690-01	2.7010-06	1.0000-06	2.6430-07	1.2000-09	1.0000-06	8.0400-05	1.5000-09	2.0000-08
4.00	3.9900-01	2.6670-06	1.0000-06	2.6460-07	1.2000-09	1.0000-06	8.0430-05	1.5000-09	2.0000-08
4.25	3.9140-01	2.6330-06	1.0000-06	2.6480-07	1.2000-09	1.0000-06	8.0450-05	1.5000-09	2.0000-08
4.50	3.8400-01	2.6000-06	1.0000-06	2.6510-07	1.2000-09	1.0000-06	8.0470-05	1.5000-09	2.0000-08
4.75	3.7680-01	2.5670-06	1.0000-06	2.6530-07	1.2000-09	1.0000-06	8.0500-05	1.5000-09	2.0000-08
5.00	3.6940-01	2.5350-06	1.0000-06	2.6560-07	1.2000-09	1.0000-06	8.0520-05	1.5000-09	2.0000-08
5.25	3.6290-01	2.5030-06	1.0000-06	2.6580-07	1.2000-09	1.0000-06	8.0540-05	1.5000-09	2.0000-08
5.50	3.5630-01	2.4720-06	1.0000-06	2.6610-07	1.2000-09	1.0000-06	8.0560-05	1.5000-09	2.0000-08
5.75	3.4980-01	2.4410-06	1.0000-06	2.6630-07	1.2000-09	1.0000-06	8.0580-05	1.5000-09	2.0000-08
6.00	3.4350-01	2.4110-06	1.0000-06	2.6660-07	1.2000-09	1.0000-06	8.0600-05	1.5000-09	2.0000-08
6.25	3.3740-01	2.3810-06	1.0000-06	2.6680-07	1.2000-09	1.0000-06	8.0620-05	1.5000-09	2.0000-08
6.50	3.3150-01	2.3520-06	1.0000-06	2.6700-07	1.2000-09	1.0000-06	8.0640-05	1.5000-09	2.0000-08
6.75	3.2570-01	2.3230-06	1.0000-06	2.6720-07	1.2000-09	1.0000-06	8.0660-05	1.5000-09	2.0000-08
7.00	3.2000-01	2.2950-06	1.0000-06	2.6750-07	1.2000-09	1.0000-06	8.0680-05	1.5000-09	2.0000-08
7.25	3.1450-01	2.2670-06	1.0000-06	2.6770-07	1.2000-09	1.0000-06	8.0700-05	1.5000-09	2.0000-08
7.50	3.0920-01	2.2390-06	1.0000-06	2.6790-07	1.2000-09	1.0000-06	8.0720-05	1.5000-09	2.0000-08
7.75	3.0400-01	2.2120-06	1.0000-06	2.6810-07	1.2000-09	1.0000-06	8.0740-05	1.5000-09	2.0000-08
8.00	2.9890-01	2.1860-06	1.0000-06	2.6830-07	1.2000-09	1.0000-06	8.0760-05	1.5000-09	2.0000-08
8.25	2.9400-01	2.1590-06	1.0000-06	2.6850-07	1.2000-09	1.0000-06	8.0780-05	1.5000-09	2.0000-08
8.50	2.8920-01	2.1340-06	1.0000-06	2.6870-07	1.2000-09	1.0000-06	8.0790-05	1.5000-09	2.0000-08
8.75	2.8450-01	2.1080-06	1.0000-06	2.6890-07	1.2000-09	1.0000-06	8.0810-05	1.5000-09	2.0000-08
9.00	2.8000-01	2.0830-06	1.0000-06	2.6910-07	1.2000-09	1.0000-06	8.0830-05	1.5000-09	2.0000-08
9.25	2.7560-01	2.0590-06	1.0000-06	2.6930-07	1.2000-09	1.0000-06	8.0850-05	1.5000-09	2.0000-08
9.50	2.7130-01	2.0340-06	1.0000-06	2.6950-07	1.2000-09	1.0000-06	8.0860-05	1.5000-09	2.0000-08
9.75	2.6710-01	2.0100-06	1.0000-06	2.6960-07	1.2000-09	1.0000-06	8.0880-05	1.5000-09	2.0000-08
10.00	2.6310-01	1.9870-06	1.0000-06	2.6980-07	1.2000-09	1.0000-06	8.0900-05	1.5000-09	2.0000-08
10.25	2.5910-01	1.9640-06	1.0000-06	2.7000-07	1.2000-09	1.0000-06	8.0910-05	1.5000-09	2.0000-08
10.50	2.5520-01	1.9410-06	1.0000-06	2.7020-07	1.2000-09	1.0000-06	8.0930-05	1.5000-09	2.0000-08
10.75	2.5150-01	1.9190-06	1.0000-06	2.7030-07	1.2000-09	1.0000-06	8.0940-05	1.5000-09	2.0000-08
11.00	2.4790-01	1.8960-06	1.0000-06	2.7050-07	1.2000-09	1.0000-06	8.0960-05	1.5000-09	2.0000-08
11.25	2.4420-01	1.8750-06	1.0000-06	2.7070-07	1.2000-09	1.0000-06	8.0970-05	1.5000-09	2.0000-08
11.50	2.4080-01	1.8530-06	1.0000-06	2.7080-07	1.2000-09	1.0000-06	8.0990-05	1.5000-09	2.0000-08
11.75	2.3740-01	1.8320-06	1.0000-06	2.7100-07	1.2000-09	1.0000-06	8.1000-05	1.5000-09	2.0000-08
12.00	2.3410-01	1.8110-06	1.0000-06	2.7120-07	1.2000-09	1.0000-06	8.1020-05	1.5000-09	2.0000-08
12.25	2.3080-01	1.7910-06	1.0000-06	2.7130-07	1.2000-09	1.0000-06	8.1030-05	1.5000-09	2.0000-08
12.50	2.2760-01	1.7710-06	1.0000-06	2.7150-07	1.2000-09	1.0000-06	8.1050-05	1.5000-09	2.0000-08

T=300 K, H=30 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	1.570-03	2.840-14
2	2*N03 >>> 2*N02 + O2	2.410-16	1.750-58
3	N02 + N03 >>> N02 + N0 + O2	8.210-15	4.920-34
4	N03 + N0 >>> 2*N02	1.900-11	6.980-29
5	N0 + O3 >>> N02 + O2	1.670-14	2.850-49
6	N02 + O3 >>> N03 + O2	3.410-17	2.000-34
7	HN03 + M >>> H0 + N02 + M	1.920-21	6.220-13
8	HN03 + H0 >>> H20 + N03	8.000-14	1.250-26
9	O + O + M >>> O2 + M	5.900-16	1.260-54
10	O + O2 + M >>> O3 + M	1.640-16	1.310-09
11	O + O3 >>> 2*O2	8.900-15	0.0
12	O + N0 + M >>> N02 + M	3.060-14	1.200-38
13	O + N02 >>> N0 + O2	6.250-12	3.730-44
14	O + N02 + M >>> N03 + M	2.820-14	2.260-24
15	H0 + H0 >>> H20 + O	1.570-12	4.630-24
16	O2 + 2*N0 >>> 2*N02	1.930-38	2.260-31
17	N02 + H-NU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.200-11	2.160-22
19	O + H02 >>> H0 + O2	1.510-11	8.510-52
20	O2 + H + M >>> H02 + M	7.960-15	6.500-25
21	O3 + H >>> H0 + O2	1.790-11	1.620-69
22	O3 + H0 >>> H02 + O2	5.350-14	8.980-43
23	O3 + H02 >>> H0 + 2*O2	1.040-15	1.530-63
24	H + H0 + M >>> H20 + M	1.410-13	1.140-75
25	H + H02 >>> 2*H0	1.770-11	1.140-40
26	H + H02 >>> H2 + O2	1.310-11	6.800-53
27	H + H20 >>> H2 + H0	2.180-25	6.410-15
28	H + H202 >>> H2 + H02	2.130-14	2.960-26
29	H + H202 >>> H0 + H20	2.760-14	2.950-65
30	2*H0 + M >>> H202 + M	7.080-14	5.170-24
31	H0 + H02 >>> H20 + O2	1.570-11	1.980-63
32	2*H02 >>> H202 + O2	3.210-12	1.980-42
33	H02 + H20 >>> H202 + H0	6.110-35	8.570-17
34	N0 + H + M >>> HN0 + M	5.750-15	1.470-26
35	N0 + H0 >>> N02 + H	7.190-34	4.920-11
36	N0 + H0 + M >>> HN02 + M	4.410-13	2.470-22
37	N0 + H02 >>> N02 + H0	3.660-13	3.740-19
38	H + H + M >>> H2 + M	2.820-16	2.430-69
39	HN04 + M >>> H02 + N02 + M	3.880-03	4.650-14
40	CLN03 + M >>> CLO + N02 + M	2.240-04	3.940-14

HAPP RESIDENCE TIME STUDY

TIME (S)	H2O5	H2O7	H2O3	N2	U3	O2	HNO3	H2O	H2O
0.0	4.0000 08	4.2000 09	3.0000 06	4.5000 04	2.5000 12	7.8000 16	4.0000 08	3.0000 06	1.2000 12
0.20	3.9940 04	4.2050 09	3.1420 06	4.4730 04	2.5000 12	7.8000 16	4.0000 08	2.9990 06	1.2000 12
0.40	3.9970 08	4.2100 04	3.3430 06	4.4750 04	2.5000 12	7.8000 16	4.0000 08	2.9420 06	1.2000 12
0.60	3.9990 08	4.2150 09	3.5740 06	4.3880 08	2.5000 12	7.8000 16	4.0000 08	2.8660 06	1.2000 12
0.80	3.9950 08	4.2200 09	3.7650 06	4.3720 08	2.5000 12	7.8000 16	4.0000 08	2.8040 06	1.2000 12
1.00	3.9940 08	4.2250 09	3.9560 06	4.3160 04	2.5000 12	7.8000 16	4.0000 08	2.7470 06	1.2000 12
1.20	3.9920 08	4.2290 09	4.1460 06	4.2400 08	2.5000 12	7.8000 16	4.0000 08	2.6900 06	1.2000 12
1.40	3.9910 08	4.2340 09	4.3370 06	4.2440 08	2.5000 12	7.8000 16	4.0000 08	2.6350 06	1.2000 12
1.60	3.9900 08	4.2390 09	4.5270 06	4.2080 08	2.5000 12	7.8000 16	4.0000 08	2.5820 06	1.2000 12
1.80	3.9890 08	4.2440 09	4.7170 06	4.1730 08	2.5000 12	7.8000 16	4.0000 08	2.5310 06	1.2000 12
2.00	3.9870 08	4.2490 09	4.9060 06	4.1380 08	2.5000 12	7.8000 16	4.0000 08	2.4830 06	1.2000 12
2.20	3.9860 08	4.2530 09	5.0950 06	4.1040 08	2.5000 12	7.8000 16	4.0000 08	2.4360 06	1.2000 12
2.40	3.9850 08	4.2580 09	5.2840 06	4.0700 08	2.5000 12	7.8000 16	4.0000 08	2.3910 06	1.2000 12
2.60	3.9840 08	4.2620 09	5.4740 06	4.0360 08	2.5000 12	7.8000 16	4.0000 08	2.3480 06	1.2000 12
2.80	3.9820 08	4.2670 09	5.6630 06	4.0020 08	2.5000 12	7.8000 16	4.0000 08	2.3070 06	1.2000 12
3.00	3.9810 08	4.2710 09	5.8520 06	3.9690 08	2.5000 12	7.8000 16	4.0000 08	2.2680 06	1.2000 12
3.20	3.9800 08	4.2760 09	6.0400 06	3.9350 08	2.5000 12	7.8000 16	4.0000 08	2.2300 06	1.2000 12
3.40	3.9790 08	4.2800 09	6.2290 06	3.9030 08	2.5000 12	7.8000 16	4.0000 08	2.1940 06	1.2000 12
3.60	3.9780 08	4.2850 09	6.4170 06	3.8700 08	2.5000 12	7.8000 16	4.0000 08	2.1590 06	1.2000 12
3.80	3.9750 08	4.2890 09	6.6050 06	3.8380 08	2.5000 12	7.8000 16	4.0000 08	2.1270 06	1.2000 12
4.00	3.9750 08	4.2940 09	6.7930 06	3.8060 08	2.5000 12	7.8000 16	4.0000 08	2.0950 06	1.2000 12
4.20	3.9740 08	4.2990 09	6.9800 06	3.7740 08	2.5000 12	7.8000 16	4.0000 08	2.0650 06	1.2000 12
4.40	3.9730 08	4.3030 09	7.1680 06	3.7420 08	2.5000 12	7.8000 16	4.0000 08	2.0370 06	1.2000 12
4.60	3.9710 08	4.3070 09	7.3550 06	3.7110 08	2.5000 12	7.8000 16	4.0000 08	2.0100 06	1.2000 12
4.80	3.9700 08	4.3110 09	7.5430 06	3.6800 08	2.5000 12	7.8000 16	4.0000 08	1.9840 06	1.2000 12
5.00	3.9690 08	4.3150 09	7.7300 06	3.6490 08	2.5000 12	7.8000 16	4.0000 08	1.9590 06	1.2000 12
5.20	3.9680 08	4.3200 09	7.9170 06	3.6190 08	2.5000 12	7.8000 16	4.0000 08	1.9360 06	1.2000 12
5.40	3.9660 08	4.3240 09	8.1030 06	3.5890 08	2.5000 12	7.8000 16	4.0000 08	1.9140 06	1.2000 12
5.60	3.9650 08	4.3290 09	8.2900 06	3.5590 08	2.5000 12	7.8000 16	4.0000 08	1.8940 06	1.2000 12
5.80	3.9640 08	4.3340 09	8.4770 06	3.5290 08	2.5000 12	7.8000 16	4.0000 08	1.8740 06	1.2000 12
6.00	3.9630 08	4.3370 09	8.6630 06	3.4990 08	2.5000 12	7.8000 16	4.0000 08	1.8560 06	1.2000 12
6.20	3.9610 08	4.3410 09	8.8490 06	3.4700 08	2.5000 12	7.8000 16	4.0000 08	1.8380 06	1.2000 12
6.40	3.9600 08	4.3450 09	9.0360 06	3.4410 08	2.5000 12	7.8000 16	4.0000 08	1.8220 06	1.2000 12
6.60	3.9590 08	4.3490 09	9.2220 06	3.4120 08	2.5000 12	7.8000 16	4.0000 08	1.8070 06	1.2000 12
6.80	3.9580 08	4.3530 09	9.4080 06	3.3840 08	2.5000 12	7.8000 16	4.0000 08	1.7920 06	1.2000 12
7.00	3.9560 08	4.3570 09	9.5930 06	3.3550 08	2.5000 12	7.8000 16	4.0000 08	1.7790 06	1.2000 12
7.20	3.9550 08	4.3610 09	9.7790 06	3.3270 08	2.5000 12	7.8000 16	4.0000 08	1.7670 06	1.2000 12
7.40	3.9540 08	4.3650 09	9.9650 06	3.3000 08	2.5000 12	7.8000 16	4.0000 08	1.7560 06	1.2000 12
7.60	3.9530 08	4.3690 09	1.0150 07	3.2720 08	2.5000 12	7.8000 16	4.0000 08	1.7450 06	1.2000 12
7.80	3.9510 08	4.3730 09	1.0340 07	3.2450 08	2.5000 12	7.8000 16	4.0000 08	1.7360 06	1.2000 12
8.00	3.9500 08	4.3770 09	1.0520 07	3.2170 08	2.5000 12	7.8000 16	4.0000 08	1.7270 06	1.2000 12
8.20	3.9490 08	4.3810 09	1.0710 07	3.1910 08	2.5000 12	7.8000 16	4.0000 08	1.7190 06	1.2000 12
8.40	3.9480 08	4.3840 09	1.0890 07	3.1640 08	2.5000 12	7.8000 16	4.0000 08	1.7120 06	1.2000 12
8.60	3.9470 08	4.3880 09	1.1080 07	3.1370 08	2.5000 12	7.8000 16	4.0000 08	1.7050 06	1.2000 12
8.80	3.9450 08	4.3920 09	1.1260 07	3.1110 08	2.5000 12	7.8000 16	4.0000 08	1.7000 06	1.2000 12
9.00	3.9440 08	4.3960 09	1.1450 07	3.0850 08	2.5000 12	7.8000 16	4.0010 08	1.6950 06	1.2000 12
9.20	3.9430 08	4.4000 09	1.1630 07	3.0590 08	2.5000 12	7.8000 16	4.0010 08	1.6910 06	1.2000 12
9.40	3.9420 08	4.4030 09	1.1820 07	3.0340 08	2.5000 12	7.8000 16	4.0010 08	1.6870 06	1.2000 12
9.60	3.9400 08	4.4070 09	1.2000 07	3.0080 08	2.5000 12	7.8000 16	4.0010 08	1.6850 06	1.2000 12
9.80	3.9390 08	4.4110 09	1.2180 07	2.9830 08	2.5000 12	7.8000 16	4.0010 08	1.6820 06	1.2000 12
10.00	3.9380 08	4.4150 09	1.2370 07	2.9580 08	2.5000 12	7.8000 16	4.0010 08	1.6810 06	1.2000 12

TIME (S)	N	M	M2	H02	M202	MNO	MNO2	MNO4	CLM03
0.0	1.0000 06	1.0000 06	1.0000 06	2.5000 07	1.2000 09	1.0000 06	8.0000 05	1.5000 09	2.0000 08
0.20	7.7210 04	1.4420-02	1.0000 06	2.7100 07	1.2000 09	1.0000 04	8.0010 05	1.4990 09	2.0000 04
0.40	5.1810 03	1.1930-03	1.0000 06	2.8500 07	1.2000 09	1.0000 06	8.0020 05	1.4940 09	2.0000 04
0.60	7.1330 02	1.5430-04	1.0000 06	2.9610 07	1.2000 09	1.0000 06	8.0030 05	1.4970 09	2.0000 04
0.80	2.9240 02	7.8900-05	1.0000 06	3.0820 07	1.2000 09	1.0000 04	8.0050 05	1.4950 09	2.0000 08
1.00	2.5900 02	7.1500-05	1.0000 04	3.2040 07	1.2000 09	1.0000 04	8.0060 05	1.4940 09	2.0000 06
1.20	2.5740 02	6.9570-05	1.0000 06	3.3250 07	1.2000 09	1.0000 06	8.0070 05	1.4930 09	2.0000 06
1.40	2.5720 02	6.8120-05	1.0000 06	3.4460 07	1.2000 09	1.0000 06	8.0080 05	1.4920 09	1.9990 08
1.60	2.5710 02	6.6740-05	1.0000 06	3.5660 07	1.2000 09	1.0000 06	8.0090 05	1.4910 09	1.9990 08
1.80	2.5710 02	6.5430-05	1.0000 06	3.6860 07	1.2000 09	1.0000 06	8.0100 05	1.4900 09	1.9990 08
2.00	2.5710 02	6.4160-05	1.0000 06	3.8060 07	1.2000 09	1.0000 04	8.0110 05	1.4880 09	1.9990 04
2.20	2.5710 02	6.2950-05	1.0000 06	3.9260 07	1.2000 09	1.0000 04	8.0120 05	1.4870 09	1.9990 08
2.40	2.5700 02	6.1740-05	1.0000 06	4.0450 07	1.2000 09	1.0000 06	8.0130 05	1.4850 09	1.9990 08
2.60	2.5700 02	6.0670-05	1.0000 06	4.1640 07	1.2000 09	1.0000 06	8.0140 05	1.4840 09	1.9990 08
2.80	2.5700 02	5.9610-05	1.0000 06	4.2820 07	1.2000 09	1.0000 06	8.0150 05	1.4830 09	1.9990 08
3.00	2.5690 02	5.8590-05	1.0000 06	4.4010 07	1.2000 09	1.0000 04	8.0160 05	1.4820 09	1.9990 08
3.20	2.5690 02	5.7610-05	1.0000 06	4.5190 07	1.2000 09	1.0000 06	8.0170 05	1.4800 09	1.9990 04
3.40	2.5690 02	5.6670-05	1.0000 06	4.6370 07	1.2000 09	1.0000 06	8.0180 05	1.4790 09	1.9990 08
3.60	2.5690 02	5.5780-05	1.0000 06	4.7540 07	1.2000 09	1.0000 06	8.0190 05	1.4780 09	1.9990 08
3.80	2.5690 02	5.4930-05	1.0000 06	4.8720 07	1.2000 09	1.0000 06	8.0200 05	1.4770 09	1.9980 08
4.00	2.5680 02	5.4110-05	1.0000 06	4.9890 07	1.2000 09	1.0000 06	8.0210 05	1.4760 09	1.9980 08
4.20	2.5680 02	5.3340-05	1.0000 06	5.1060 07	1.2000 09	1.0000 06	8.0220 05	1.4750 09	1.9980 08
4.40	2.5680 02	5.2600-05	1.0000 06	5.2220 07	1.2000 09	1.0000 06	8.0230 05	1.4740 09	1.9980 08
4.60	2.5680 02	5.1900-05	1.0000 06	5.3380 07	1.2000 09	1.0000 06	8.0240 05	1.4730 09	1.9980 08
4.80	2.5680 02	5.1230-05	1.0000 06	5.4540 07	1.2000 09	1.0000 04	8.0250 05	1.4720 09	1.9980 08
5.00	2.5680 02	5.0600-05	1.0000 06	5.5700 07	1.2000 09	1.0000 06	8.0260 05	1.4710 09	1.9980 08
5.20	2.5680 02	5.0000-05	1.0000 06	5.6860 07	1.2000 09	1.0000 06	8.0270 05	1.4700 09	1.9980 08
5.40	2.5670 02	4.9430-05	1.0000 06	5.8010 07	1.2000 09	1.0000 06	8.0280 05	1.4690 09	1.9980 08
5.60	2.5670 02	4.8890-05	1.0000 06	5.9160 07	1.2000 09	1.0000 06	8.0290 05	1.4680 09	1.9980 08
5.80	2.5670 02	4.8390-05	1.0000 06	6.0310 07	1.2000 09	1.0000 06	8.0300 05	1.4670 09	1.9980 08
6.00	2.5670 02	4.7910-05	1.0000 06	6.1460 07	1.2000 09	1.0000 04	8.0310 05	1.4660 09	1.9980 08
6.20	2.5670 02	4.7460-05	1.0000 06	6.2600 07	1.2000 09	1.0000 06	8.0320 05	1.4650 09	1.9980 08
6.40	2.5670 02	4.7040-05	1.0000 06	6.3740 07	1.2000 09	1.0000 04	8.0330 05	1.4640 09	1.9980 08
6.60	2.5670 02	4.6640-05	1.0000 06	6.4890 07	1.2000 09	1.0000 06	8.0340 05	1.4630 09	1.9980 08
6.80	2.5670 02	4.6270-05	1.0000 06	6.6020 07	1.2000 09	1.0000 06	8.0350 05	1.4620 09	1.9970 08
7.00	2.5670 02	4.5930-05	1.0000 06	6.7160 07	1.2000 09	1.0000 06	8.0360 05	1.4610 09	1.9970 08
7.20	2.5670 02	4.5610-05	1.0000 06	6.8290 07	1.2000 09	1.0000 06	8.0370 05	1.4600 09	1.9970 08
7.40	2.5670 02	4.5320-05	1.0000 06	6.9420 07	1.2000 09	1.0000 06	8.0380 05	1.4590 09	1.9970 08
7.60	2.5670 02	4.5050-05	1.0000 06	7.0550 07	1.2000 09	1.0000 04	8.0390 05	1.4580 09	1.9970 08
7.80	2.5670 02	4.4800-05	1.0000 06	7.1680 07	1.2000 09	1.0000 06	8.0400 05	1.4570 09	1.9970 08
8.00	2.5660 02	4.4570-05	1.0000 06	7.2810 07	1.2000 09	1.0000 06	8.0410 05	1.4560 09	1.9970 08
8.20	2.5660 02	4.4370-05	1.0000 06	7.3930 07	1.2000 09	1.0000 06	8.0420 05	1.4550 09	1.9970 08
8.40	2.5660 02	4.4180-05	1.0000 06	7.5050 07	1.2000 09	1.0000 04	8.0430 05	1.4540 09	1.9970 08
8.60	2.5660 02	4.4020-05	1.0000 06	7.6170 07	1.2000 09	1.0000 06	8.0440 05	1.4530 09	1.9970 08
8.80	2.5660 02	4.3870-05	1.0000 06	7.7290 07	1.2000 09	1.0000 04	8.0450 05	1.4520 09	1.9970 08
9.00	2.5660 02	4.3750-05	1.0000 06	7.8410 07	1.2000 09	1.0000 06	8.0460 05	1.4510 09	1.9970 08
9.20	2.5660 02	4.3640-05	1.0000 06	7.9520 07	1.2000 09	1.0000 06	8.0470 05	1.4500 09	1.9970 08
9.40	2.5660 02	4.3550-05	1.0000 06	8.0630 07	1.2000 09	1.0000 06	8.0480 05	1.4490 09	1.9970 08
9.60	2.5660 02	4.3480-05	1.0000 06	8.1750 07	1.2000 09	1.0000 06	8.0490 05	1.4480 09	1.9960 08
9.80	2.5660 02	4.3420-05	1.0000 06	8.2850 07	1.2000 09	1.0000 06	8.0500 05	1.4470 09	1.9960 08
10.00	2.5660 02	4.3380-05	1.0000 06	8.3960 07	1.2000 09	1.0000 06	8.0510 05	1.4460 09	1.9960 08

T=700 K, H=30 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$N_2O_5 + M \ggg N_2O + NO_3 + M$	2.300-05	2.360-14
2	$2^*NO_3 \ggg 2^*NO_2 + O_2$	2.570-14	7.440-49
3	$NO_2 + NO_3 \ggg NO_2 + NO + O_2$	5.510-14	3.970-34
4	$NO_3 + NO \ggg 2^*NO_2$	1.900-11	2.230-19
5	$NO + O_3 \ggg NO_2 + O_2$	2.650-13	4.540-28
6	$NO_2 + O_3 \ggg NO_3 + O_2$	3.620-15	6.560-22
7	$HNO_3 + M \ggg HO + NO_2 + M$	1.920-02	3.210-14
8	$HNO_3 + HO \ggg H_2O + NO_3$	8.000-14	4.740-19
9	$O + O + M \ggg O_2 + M$	4.550-17	1.420-21
10	$O + O_2 + M \ggg O_3 + M$	2.660-17	1.610-00
11	$O + O_3 \ggg 2^*O_2$	7.110-13	8.630-43
12	$O + NO + M \ggg NO_2 + M$	4.320-15	1.020-11
13	$O + NO_2 \ggg NO + O_2$	1.110-11	8.500-27
14	$O + NO_2 + M \ggg NO_3 + M$	1.210-14	9.680-25
15	$HO + HO \ggg H_2O + O$	4.530-12	2.040-16
16	$O_2 + 2^*NO \ggg 2^*NO_2$	7.040-39	9.370-21
17	$NO_2 + H-NU \ggg NO + O$	0.0	0.0
18	$O + HO \ggg H + O_2$	4.200-11	2.110-15
19	$O + NO_2 \ggg HO + O_2$	3.910-11	2.190-28
20	$O_2 + H + M \ggg HO_2 + M$	1.320-15	2.960-06
21	$O_3 + H \ggg HO + O_2$	4.780-11	5.620-37
22	$O_3 + HO \ggg HO_2 + O_2$	3.590-13	8.310-26
23	$O_3 + NO_2 \ggg HO + 2^*O_2$	1.180-14	1.120-49
24	$H + HO + M \ggg H_2O + M$	6.670-15	1.600-27
25	$H + NO_2 \ggg 2^*HO$	1.080-10	5.870-24
26	$H + NO_2 \ggg H_2 + O_2$	2.550-11	8.030-20
27	$H + H_2O \ggg H_2 + HO$	6.560-17	8.900-13
28	$H + H_2O_2 \ggg H_2 + HO_2$	3.060-13	1.770-18
29	$H + H_2O_2 \ggg HO + H_2O$	3.980-13	1.020-15
30	$2^*HO + M \ggg H_2O_2 + M$	5.470-15	5.000-05
31	$HO + NO_2 \ggg H_2O + O_2$	4.060-11	7.730-33
32	$2^*HO_2 \ggg H_2O_2 + O_2$	11.320-12	3.020-24
33	$HO_2 + H_2O \ggg H_2O_2 + HO$	2.720-21	3.070-12
34	$NO + H + M \ggg HNO + M$	1.390-15	1.050-06
35	$NO + HO \ggg NO_2 + H$	2.230-21	2.020-10
36	$NO + HO + M \ggg HNO_2 + M$	2.280-14	2.840-03
37	$NO + NO_2 \ggg NO_2 + NO$	3.600-12	2.680-14
38	$H + H + M \ggg H_2 + M$	1.210-16	1.010-25
39	$HNO_4 + M \ggg HO_2 + NO_2 + M$	3.350-05	3.130-14
40	$CLNO_3 + M \ggg CLO + NO_2 + M$	1.800-05	1.270-15

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	N0	U3	O2	MMO3	MO	M20
0.0	4.0000-04	4.2000-09	3.0000-06	4.5000-04	2.5000-12	7.8000-16	4.0000-08	3.0000-06	1.2000-12
0.20	4.4400-02	3.4400-09	4.1100-08	3.7700-09	1.4400-12	7.8000-16	3.9850-08	1.0560-09	1.2000-12
0.40	5.0770-03	1.2040-09	4.1100-08	3.7700-09	1.4400-12	7.8000-16	3.9750-08	1.0560-09	1.2000-12
0.60	2.1210-03	5.1210-08	4.0290-08	6.2430-09	1.1620-12	7.8000-16	3.9340-08	1.1570-09	1.2000-12
0.80	1.3250-03	3.2700-08	3.9420-08	6.4380-09	9.8530-11	7.8000-16	3.9390-08	1.1890-09	1.2000-12
1.00	1.0740-03	2.7130-08	3.8530-08	6.5040-09	8.6030-11	7.8000-16	3.9240-08	1.2210-09	1.2000-12
1.20	9.7140-04	2.5120-08	3.7430-08	6.5340-09	7.6900-11	7.8000-16	3.9090-08	1.2520-09	1.2000-12
1.40	9.1100-04	2.4200-08	3.6750-08	6.5530-09	6.9370-11	7.8000-16	3.8940-08	1.2800-09	1.2000-12
1.60	8.4170-04	2.3900-08	3.5890-08	6.5670-09	6.4420-11	7.8000-16	3.8790-08	1.3050-09	1.2000-12
1.80	8.5540-04	2.3750-08	3.5040-08	6.5780-09	5.4460-11	7.8000-16	3.8640-08	1.3280-09	1.2000-12
2.00	8.3370-04	2.3710-08	3.4210-08	6.5890-09	5.4240-11	7.8000-16	3.8490-08	1.3480-09	1.2000-12
2.20	8.1460-04	2.3740-08	3.3390-08	6.5980-09	5.0580-11	7.8000-16	3.8340-08	1.3660-09	1.2000-12
2.40	7.9700-04	2.3790-08	3.2590-08	6.6070-09	4.7380-11	7.8000-16	3.8200-08	1.3820-09	1.2000-12
2.60	7.8030-04	2.3870-08	3.1910-08	6.6150-09	4.4550-11	7.8000-16	3.8050-08	1.3960-09	1.2000-12
2.80	7.6430-04	2.3950-08	3.1050-08	6.6230-09	4.2030-11	7.8000-16	3.7900-08	1.4080-09	1.2000-12
3.00	7.4870-04	2.4040-08	3.0300-08	6.6310-09	3.9770-11	7.8000-16	3.7760-08	1.4190-09	1.2000-12
3.20	7.3340-04	2.4130-08	2.9570-08	6.6390-09	3.7730-11	7.8000-16	3.7610-08	1.4280-09	1.2000-12
3.40	7.1850-04	2.4230-08	2.8860-08	6.6470-09	3.5840-11	7.8000-16	3.7470-08	1.4360-09	1.2000-12
3.60	7.0380-04	2.4320-08	2.8160-08	6.6540-09	3.4200-11	7.8000-16	3.7320-08	1.4430-09	1.2000-12
3.80	6.8930-04	2.4410-08	2.7470-08	6.6620-09	3.2660-11	7.8000-16	3.7180-08	1.4500-09	1.2000-12
4.00	6.7510-04	2.4500-08	2.6810-08	6.6690-09	3.1240-11	7.8000-16	3.7040-08	1.4550-09	1.2000-12
4.20	6.6100-04	2.4590-08	2.6160-08	6.6760-09	2.9940-11	7.8000-16	3.6890-08	1.4590-09	1.2000-12
4.40	6.4720-04	2.4680-08	2.5520-08	6.6830-09	2.8740-11	7.8000-16	3.6750-08	1.4630-09	1.2000-12
4.60	6.3350-04	2.4760-08	2.4900-08	6.6890-09	2.7620-11	7.8000-16	3.6610-08	1.4660-09	1.2000-12
4.80	6.2020-04	2.4840-08	2.4290-08	6.6960-09	2.6580-11	7.8000-16	3.6470-08	1.4680-09	1.2000-12
5.00	6.0700-04	2.4920-08	2.3700-08	6.7030-09	2.5610-11	7.8000-16	3.6330-08	1.4700-09	1.2000-12
5.20	5.9400-04	2.5000-08	2.3120-08	6.7090-09	2.4710-11	7.8000-16	3.6190-08	1.4710-09	1.2000-12
5.40	5.8120-04	2.5080-08	2.2550-08	6.7150-09	2.3860-11	7.8000-16	3.6050-08	1.4710-09	1.2000-12
5.60	5.6850-04	2.5150-08	2.2000-08	6.7210-09	2.3070-11	7.8000-16	3.5910-08	1.4710-09	1.2000-12
5.80	5.5620-04	2.5220-08	2.1460-08	6.7270-09	2.2320-11	7.8000-16	3.5770-08	1.4710-09	1.2000-12
6.00	5.4400-04	2.5290-08	2.0930-08	6.7330-09	2.1620-11	7.8000-16	3.5640-08	1.4710-09	1.2000-12
6.20	5.3210-04	2.5360-08	2.0410-08	6.7390-09	2.0960-11	7.8000-16	3.5500-08	1.4690-09	1.2000-12
6.40	5.2030-04	2.5420-08	1.9910-08	6.7450-09	2.0330-11	7.8000-16	3.5360-08	1.4680-09	1.2000-12
6.60	5.0870-04	2.5490-08	1.9420-08	6.7500-09	1.9740-11	7.8000-16	3.5230-08	1.4660-09	1.2000-12
6.80	4.9740-04	2.5550-08	1.8940-08	6.7560-09	1.9180-11	7.8000-16	3.5090-08	1.4640-09	1.2000-12
7.00	4.8620-04	2.5610-08	1.8470-08	6.7610-09	1.8650-11	7.8000-16	3.4960-08	1.4620-09	1.2000-12
7.20	4.7520-04	2.5660-08	1.8020-08	6.7670-09	1.8140-11	7.8000-16	3.4820-08	1.4600-09	1.2000-12
7.40	4.6450-04	2.5720-08	1.7570-08	6.7720-09	1.7660-11	7.8000-16	3.4690-08	1.4570-09	1.2000-12
7.60	4.5390-04	2.5770-08	1.7140-08	6.7770-09	1.7200-11	7.8000-16	3.4550-08	1.4540-09	1.2000-12
7.80	4.4350-04	2.5820-08	1.6710-08	6.7820-09	1.6760-11	7.8000-16	3.4420-08	1.4510-09	1.2000-12
8.00	4.3340-04	2.5870-08	1.6300-08	6.7870-09	1.6350-11	7.8000-16	3.4290-08	1.4470-09	1.2000-12
8.20	4.2340-04	2.5920-08	1.5900-08	6.7920-09	1.5950-11	7.8000-16	3.4160-08	1.4440-09	1.2000-12
8.40	4.1360-04	2.5960-08	1.5500-08	6.7960-09	1.5570-11	7.8000-16	3.4030-08	1.4400-09	1.2000-12
8.60	4.0410-04	2.6010-08	1.5120-08	6.8010-09	1.5200-11	7.8000-16	3.3890-08	1.4370-09	1.2000-12
8.80	3.9470-04	2.6050-08	1.4740-08	6.8060-09	1.4850-11	7.8000-16	3.3760-08	1.4330-09	1.2000-12
9.00	3.8550-04	2.6090-08	1.4380-08	6.8100-09	1.4520-11	7.8000-16	3.3630-08	1.4290-09	1.2000-12
9.20	3.7650-04	2.6130-08	1.4020-08	6.8150-09	1.4200-11	7.8000-16	3.3500-08	1.4240-09	1.2000-12
9.40	3.6760-04	2.6170-08	1.3670-08	6.8190-09	1.3890-11	7.8000-16	3.3380-08	1.4200-09	1.2000-12
9.60	3.5900-04	2.6200-08	1.3330-08	6.8230-09	1.3590-11	7.8000-16	3.3250-08	1.4160-09	1.2000-12
9.80	3.5050-04	2.6240-08	1.3000-08	6.8270-09	1.3300-11	7.8000-16	3.3120-08	1.4110-09	1.2000-12
10.00	3.4220-04	2.6270-08	1.2670-08	6.8320-09	1.3030-11	7.8000-16	3.2990-08	1.4070-09	1.2000-12

TIME (S)	U	M	M2	MU2	M2U2	MNO	MNO2	MNO4	CLW03
0.0	1.0000 06	1.0000 06	1.0000 06	2.5000 07	1.2000 09	1.0000 06	0.0000 05	1.5000 09	2.0000 08
0.20	4.0270 11	5.9430 07	1.0800 04	4.3640 08	1.2010 09	1.0000 06	0.0520 05	1.4150 02	6.0110 03
0.40	4.2430 11	6.0470 07	1.2400 04	3.3440 04	1.2010 09	1.0000 06	0.0810 05	1.3730 03	2.0870 03
0.60	6.5000 11	9.7670 07	1.4300 06	4.0800 04	1.2000 09	1.0000 06	0.0860 05	1.9530 03	8.6740 04
0.80	4.2520 11	9.9240 07	1.6500 06	4.2830 08	1.1900 09	1.0000 06	0.0910 05	1.3090 03	5.6670 04
1.00	5.8510 11	9.7290 07	1.8600 06	4.4730 08	1.1900 09	1.0010 06	0.0910 05	1.1350 03	4.7020 04
1.20	5.4220 11	9.3900 07	2.1100 06	4.6460 08	1.1900 09	1.0010 06	0.0910 05	1.0910 03	4.3530 04
1.40	5.0150 11	8.9740 07	2.3450 06	4.8010 08	1.1900 09	1.0010 06	0.0910 05	1.0890 03	4.2070 04
1.60	4.6450 11	8.5590 07	2.5690 06	4.9390 04	1.1970 09	1.0010 06	0.0910 05	1.1040 03	4.1430 04
1.80	4.3160 11	8.1550 07	2.7890 06	5.0630 04	1.1960 09	1.0010 06	0.0910 05	1.1240 03	4.1160 04
2.00	4.0250 11	7.7730 07	3.0030 04	5.1730 04	1.1960 09	1.0010 06	0.0910 05	1.1470 03	4.1100 04
2.20	3.7470 11	7.4150 07	3.2110 06	5.2710 04	1.1950 09	1.0020 06	0.0910 05	1.1700 03	4.1130 04
2.40	3.5380 11	7.0810 07	3.4130 06	5.3590 04	1.1950 09	1.0020 06	0.0910 05	1.1920 03	4.1230 04
2.60	3.3330 11	6.7700 07	3.6090 06	5.4360 04	1.1940 09	1.0020 06	0.0910 05	1.2130 03	4.1360 04
2.80	3.1590 11	6.4810 07	3.7980 06	5.5060 04	1.1940 09	1.0020 06	0.0910 05	1.2330 03	4.1510 04
3.00	2.9840 11	6.2110 07	3.9820 06	5.5680 04	1.1930 09	1.0020 06	0.0910 05	1.2510 03	4.1660 04
3.20	2.8350 11	5.9590 07	4.1600 06	5.6220 04	1.1930 09	1.0020 06	0.0910 05	1.2680 03	4.1820 04
3.40	2.6980 11	5.7230 07	4.3320 06	5.6710 04	1.1920 09	1.0020 06	0.0910 05	1.2840 03	4.1980 04
3.60	2.5740 11	5.5020 07	4.4980 06	5.7140 04	1.1920 09	1.0020 06	0.0910 05	1.2990 03	4.2140 04
3.80	2.4600 11	5.2940 07	4.6600 06	5.7520 04	1.1910 09	1.0020 06	0.0910 05	1.3120 03	4.2300 04
4.00	2.3550 11	5.0990 07	4.8160 06	5.7850 04	1.1900 09	1.0030 06	0.0910 05	1.3250 03	4.2460 04
4.20	2.2590 11	4.9150 07	4.9670 06	5.8140 04	1.1900 09	1.0030 06	0.0910 05	1.3360 03	4.2610 04
4.40	2.1690 11	4.7410 07	5.1130 06	5.8390 04	1.1890 09	1.0030 06	0.0910 05	1.3470 03	4.2760 04
4.60	2.0840 11	4.5770 07	5.2550 06	5.8600 04	1.1890 09	1.0030 06	0.0910 05	1.3560 03	4.2910 04
4.80	2.0090 11	4.4220 07	5.3920 06	5.8780 04	1.1880 09	1.0030 06	0.0910 05	1.3650 03	4.3050 04
5.00	1.9370 11	4.2750 07	5.5250 06	5.8930 04	1.1880 09	1.0030 06	0.0910 05	1.3730 03	4.3190 04
5.20	1.8690 11	4.1360 07	5.6530 06	5.9050 04	1.1870 09	1.0030 06	0.0910 05	1.3800 03	4.3330 04
5.40	1.8060 11	4.0030 07	5.7780 06	5.9150 04	1.1870 09	1.0030 06	0.0910 05	1.3870 03	4.3460 04
5.60	1.7460 11	3.8770 07	5.8990 06	5.9220 04	1.1860 09	1.0030 06	0.0910 05	1.3920 03	4.3590 04
5.80	1.6910 11	3.7570 07	6.0160 06	5.9280 04	1.1860 09	1.0030 06	0.0910 05	1.3960 03	4.3710 04
6.00	1.6380 11	3.6430 07	6.1290 06	5.9310 04	1.1850 09	1.0030 06	0.0910 05	1.4020 03	4.3830 04
6.20	1.5880 11	3.5340 07	6.2390 06	5.9320 04	1.1850 09	1.0030 06	0.0910 05	1.4060 03	4.3950 04
6.40	1.5410 11	3.4300 07	6.3460 06	5.9310 04	1.1840 09	1.0040 06	0.0910 05	1.4100 03	4.4060 04
6.60	1.4970 11	3.3310 07	6.4490 06	5.9290 04	1.1840 09	1.0040 06	0.0910 05	1.4130 03	4.4170 04
6.80	1.4550 11	3.2360 07	6.5500 06	5.9260 04	1.1830 09	1.0040 06	0.0910 05	1.4150 03	4.4270 04
7.00	1.4150 11	3.1450 07	6.6470 06	5.9200 04	1.1830 09	1.0040 06	0.0910 05	1.4170 03	4.4370 04
7.20	1.3770 11	3.0580 07	6.7410 06	5.9140 04	1.1820 09	1.0040 06	0.0910 05	1.4190 03	4.4470 04
7.40	1.3410 11	2.9740 07	6.8330 06	5.9070 04	1.1820 09	1.0040 06	0.0910 05	1.4200 03	4.4570 04
7.60	1.3060 11	2.8940 07	6.9220 06	5.8980 04	1.1810 09	1.0040 06	0.0910 05	1.4210 03	4.4660 04
7.80	1.2730 11	2.8170 07	7.0090 06	5.8890 04	1.1810 09	1.0040 06	0.0910 05	1.4210 03	4.4750 04
8.00	1.2420 11	2.7440 07	7.0930 06	5.8780 04	1.1800 09	1.0040 06	0.0910 05	1.4210 03	4.4830 04
8.20	1.2120 11	2.6730 07	7.1740 06	5.8670 04	1.1800 09	1.0040 06	0.0910 05	1.4210 03	4.4910 04
8.40	1.1830 11	2.6040 07	7.2530 06	5.8540 04	1.1790 09	1.0040 06	0.0910 05	1.4210 03	4.4990 04
8.60	1.1560 11	2.5390 07	7.3300 06	5.8410 04	1.1790 09	1.0040 06	0.0910 05	1.4200 03	4.5070 04
8.80	1.1290 11	2.4760 07	7.4050 06	5.8280 04	1.1780 09	1.0040 06	0.0910 05	1.4190 03	4.5140 04
9.00	1.1040 11	2.4150 07	7.4780 06	5.8130 04	1.1780 09	1.0040 06	0.0910 05	1.4180 03	4.5210 04
9.20	1.0800 11	2.3560 07	7.5490 06	5.7980 04	1.1770 09	1.0040 06	0.0910 05	1.4160 03	4.5280 04
9.40	1.0570 11	2.3000 07	7.6180 06	5.7830 04	1.1770 09	1.0040 06	0.0910 05	1.4150 03	4.5350 04
9.60	1.0340 11	2.2450 07	7.6840 06	5.7670 04	1.1770 09	1.0040 06	0.0910 05	1.4130 03	4.5410 04
9.80	1.0130 11	2.1930 07	7.7500 06	5.7500 04	1.1760 09	1.0040 06	0.0910 05	1.4100 03	4.5470 04
10.00	9.9180 10	2.1420 07	7.8130 06	5.7330 04	1.1760 09	1.0040 06	0.0910 05	1.4080 03	4.5530 04

T= 800 K, H=30 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> NO2 + NO3 + M	1.27D-06	1.77D-15
2	2*N03 >>> 2*N02 + O2	3.44D-14	5.95D-44
3	N02 + N03 >>> N02 + N0 + O2	6.59D-14	3.13D-34
4	N03 + N0 >>> 2*NO2	1.98D-11	1.74D-18
5	N0 + O3 >>> N02 + O2	3.43D-13	4.42D-26
6	N02 + O3 >>> N03 + O2	5.61D-15	9.78D-21
7	HN03 + M >>> H0 + N02 + M	9.09D-01	2.01D-14
8	HN03 + H0 >>> H2O + N03	9.00D-14	2.43D-18
9	O + O + M >>> O2 + M	3.39D-17	4.30D-17
10	O + O2 + M >>> O3 + M	2.12D-17	1.08D-01
11	O + O3 >>> 2*O2	1.07D-12	6.11D-39
12	O + N0 + M >>> N02 + M	3.40D-15	3.25D-09
13	O + N02 >>> N0 + O2	1.17D-11	5.55D-25
14	O + N02 + M >>> N03 + M	1.06D-14	8.47D-25
15	H0 + H0 >>> H2O + O	5.00D-12	1.06D-15
16	O2 + 2*N0 >>> 2*N02	6.40D-39	8.78D-20
17	N02 + H-HU >>> N0 + O	0.0	0.0
18	O + H0 >>> H + O2	4.20D-11	9.55D-15
19	O + H02 >>> H0 + O2	4.28D-11	3.43D-24
20	O2 + H + M >>> H02 + M	1.05D-15	1.57D-04
21	O3 + H >>> H0 + O2	5.25D-11	6.31D-34
22	O3 + H0 >>> H02 + O2	4.30D-13	3.24D-24
23	O3 + H02 >>> H0 + 2*O2	1.48D-14	2.35D-44
24	H + H0 + M >>> H2O + M	4.12D-15	4.08D-23
25	H + H02 >>> 2*H0	1.28D-10	2.16D-22
26	H + H02 >>> H2 + O2	2.71D-11	1.45D-26
27	H + H2O >>> H2 + H0	4.09D-16	1.41D-12
28	H + H2O2 >>> H2 + H02	3.93D-13	9.47D-18
29	H + H2O2 >>> H0 + H2O	5.11D-13	6.02D-33
30	2*H0 + M >>> H2O2 + M	4.07D-15	3.36D-03
31	H0 + H02 >>> H2O + O2	4.44D-11	5.70D-30
32	2*H02 >>> H2O2 + O2	4.10D-12	1.53D-22
33	H02 + H2O >>> H2O2 + H0	5.19D-20	3.46D-17
34	N0 + H + M >>> HN0 + M	1.15D-15	6.86D-05
35	N0 + H0 >>> N02 + H	3.30D-20	2.30D-10
36	N0 + H0 + M >>> HN02 + M	1.64D-14	1.54D-01
37	N0 + H02 >>> N02 + H0	4.46D-12	7.66D-14
38	H + H + M >>> H2 + M	1.06D-16	1.12D-21
39	HN04 + M >>> H02 + N02 + M	1.76D-06	2.31D-15
40	CLN03 + M >>> CLO + N02 + M	1.16D-06	7.36D-16

HAPP RESIDENCE TIME STUDY

TIME (s)	N205	H02	H03	H0	U3	O2	HNO3	MO	H2U
0.0	4.0000-04	4.2000 09	1.0000 04	4.5000 04	2.5000 12	7.8000 14	4.0000 04	1.0000 04	1.2000 12
0.20	2.5400-04	4.4800 08	4.0710 08	6.3600 04	3.4700 11	7.8000 16	3.5300 04	1.8700 09	1.2000 12
0.40	3.3100-05	5.0700 07	1.0700 08	6.8100 04	2.1000 11	7.8000 14	2.7000 08	2.6200 09	1.1900 12
0.60	2.6300-05	4.8600 07	1.8700 08	6.8400 04	2.0400 11	7.8000 14	2.7000 08	3.1200 09	1.1900 12
0.80	2.5200-05	4.7700 07	3.7700 08	6.9300 04	1.4900 11	7.8000 14	1.9300 04	3.9500 09	1.1900 12
1.00	2.4530-05	4.7700 07	3.6700 08	6.9700 04	1.7600 11	7.8000 16	1.6100 08	4.5100 09	1.1900 12
1.20	2.3620-05	4.7600 07	3.5810 08	7.0100 04	1.6500 11	7.8000 16	1.3400 08	5.0210 09	1.1900 12
1.40	2.3140-05	4.7500 07	3.4870 08	7.0400 04	1.5500 11	7.8000 16	1.1200 08	5.4600 09	1.1900 12
1.60	2.2690-05	4.7400 07	3.3950 08	7.0700 04	1.4410 11	7.8000 16	9.3300 07	5.8670 09	1.1900 12
1.80	2.1870-05	4.7380 07	3.3060 08	7.0970 04	1.3770 11	7.8000 16	7.7800 07	6.2180 09	1.1900 12
2.00	2.1290-05	4.7370 07	3.2180 08	7.1180 04	1.3010 11	7.8000 16	6.4800 07	6.5260 09	1.1900 12
2.20	2.0730-05	4.7400 07	3.1330 08	7.1370 04	1.2320 11	7.8000 16	5.4100 07	6.7960 09	1.1900 12
2.40	2.0210-05	4.7460 07	3.0490 08	7.1550 04	1.1680 11	7.8000 16	4.5100 07	7.0310 09	1.1900 12
2.60	1.9710-05	4.7550 07	2.9680 08	7.1700 04	1.1100 11	7.8000 16	3.7600 07	7.2340 09	1.1900 12
2.80	1.9230-05	4.7670 07	2.8880 08	7.1840 04	1.0500 11	7.8000 16	3.1350 07	7.4080 09	1.1900 12
3.00	1.8770-05	4.7820 07	2.8110 08	7.1970 04	1.0060 11	7.8000 14	2.6140 07	7.5550 09	1.1900 12
3.20	1.8330-05	4.7990 07	2.7350 08	7.2080 04	9.6050 10	7.8000 14	2.1790 07	7.6800 09	1.1900 12
3.40	1.7910-05	4.8180 07	2.6620 08	7.2190 04	9.1400 10	7.8000 16	1.8170 07	7.7830 09	1.1900 12
3.60	1.7500-05	4.8380 07	2.5900 08	7.2290 04	8.7480 10	7.8000 16	1.5150 07	7.8660 09	1.1900 12
3.80	1.7100-05	4.8600 07	2.5200 08	7.2380 04	8.4160 10	7.8000 16	1.2630 07	7.9330 09	1.1900 12
4.00	1.6720-05	4.8830 07	2.4520 08	7.2470 04	8.0720 10	7.8000 16	1.0530 07	7.9850 09	1.1900 12
4.20	1.6350-05	4.9070 07	2.3850 08	7.2550 04	7.7510 10	7.8000 16	8.7800 06	8.0220 09	1.1900 12
4.40	1.5980-05	4.9320 07	2.3210 08	7.2620 04	7.4510 10	7.8000 16	7.3210 06	8.0480 09	1.1900 12
4.60	1.5630-05	4.9580 07	2.2580 08	7.2690 04	7.1490 10	7.8000 16	6.1050 06	8.0620 09	1.1900 12
4.80	1.5280-05	4.9830 07	2.1960 08	7.2760 04	6.9050 10	7.8000 16	5.0910 06	8.0670 09	1.1900 12
5.00	1.4940-05	5.0080 07	2.1370 08	7.2830 04	6.6560 10	7.8000 16	4.2450 06	8.0630 09	1.1900 12
5.20	1.4610-05	5.0330 07	2.0780 08	7.2890 04	6.4220 10	7.8000 16	3.5410 06	8.0520 09	1.1900 12
5.40	1.4290-05	5.0590 07	2.0220 08	7.2950 04	6.2010 10	7.8000 16	2.9530 06	8.0330 09	1.1900 12
5.60	1.3960-05	5.0830 07	1.9670 08	7.3000 04	5.9920 10	7.8000 16	2.4640 06	8.0090 09	1.1900 12
5.80	1.3640-05	5.1080 07	1.9130 04	7.3060 04	5.7950 10	7.8000 16	2.0560 06	7.9800 09	1.1900 12
6.00	1.3330-05	5.1320 07	1.8610 08	7.3110 04	5.6090 10	7.8000 16	1.7150 06	7.9460 09	1.1900 12
6.20	1.3030-05	5.1560 07	1.8100 08	7.3160 04	5.4320 10	7.8000 16	1.4310 06	7.9080 09	1.1900 12
6.40	1.2730-05	5.1780 07	1.7600 08	7.3210 04	5.2650 10	7.8000 16	1.1950 06	7.8660 09	1.1900 12
6.60	1.2430-05	5.2010 07	1.7120 08	7.3260 04	5.1060 10	7.8000 16	9.9780 05	7.8220 09	1.1900 12
6.80	1.2140-05	5.2220 07	1.6650 08	7.3300 04	4.9550 10	7.8000 16	8.3340 05	7.7740 09	1.1900 12
7.00	1.1860-05	5.2430 07	1.6200 08	7.3350 04	4.8110 10	7.8000 16	6.9630 05	7.7250 09	1.1900 12
7.20	1.1580-05	5.2630 07	1.5750 08	7.3390 04	4.6740 10	7.8000 16	5.8210 05	7.6730 09	1.1900 12
7.40	1.1300-05	5.2820 07	1.5320 08	7.3430 04	4.5440 10	7.8000 16	4.8680 05	7.6200 09	1.1900 12
7.60	1.1030-05	5.3010 07	1.4900 08	7.3470 04	4.4200 10	7.8000 16	4.0740 05	7.5660 09	1.1900 12
7.80	1.0750-05	5.3190 07	1.4490 08	7.3510 04	4.3010 10	7.8000 16	3.4120 05	7.5110 09	1.1900 12
8.00	1.0500-05	5.3360 07	1.4090 08	7.3550 04	4.1800 10	7.8000 16	2.8600 05	7.4540 09	1.1900 12
8.20	1.0240-05	5.3520 07	1.3710 08	7.3590 04	4.0790 10	7.8000 16	2.3990 05	7.3970 09	1.1900 12
8.40	9.9910-06	5.3680 07	1.3330 08	7.3620 04	3.9750 10	7.8000 16	2.0150 05	7.3400 09	1.1900 12
8.60	9.7430-06	5.3830 07	1.2960 08	7.3660 04	3.8760 10	7.8000 16	1.6950 05	7.2820 09	1.1900 12
8.80	9.4990-06	5.3970 07	1.2600 08	7.3690 04	3.7810 10	7.8000 14	1.4280 05	7.2230 09	1.1900 12
9.00	9.2600-06	5.4100 07	1.2260 08	7.3730 04	3.6890 10	7.8000 16	1.2060 05	7.1650 09	1.1900 12
9.20	9.0260-06	5.4230 07	1.1920 08	7.3760 04	3.6020 10	7.8000 16	1.0200 05	7.1070 09	1.1900 12
9.40	8.7970-06	5.4350 07	1.1590 08	7.3790 04	3.5180 10	7.8000 16	8.6510 04	7.0490 09	1.1900 12
9.60	8.5720-06	5.4460 07	1.1270 08	7.3820 04	3.4370 10	7.8000 16	7.3590 04	6.9920 09	1.1900 12
9.80	8.3520-06	5.4560 07	1.0960 08	7.3850 04	3.3590 10	7.8000 16	6.2800 04	6.9330 09	1.1900 12
10.00	8.1370-06	5.4660 07	1.0660 08	7.3880 04	3.2840 10	7.8000 16	5.3600 04	6.8750 09	1.1900 12

T=250 K, H=35 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$N_2O_5 + M \ggg N_2O + NO_3 + M$	9.800-07	3.060-14
2	$2^*NO_3 \ggg 2^*NO_2 + O_2$	4.710-17	7.460-62
3	$NO_2 + NO_3 \ggg NO_2 + NO + O_2$	4.210-15	1.190-34
4	$NO_3 + NO \ggg 2^*NO_2$	1.900-11	3.290-32
5	$NO + O_3 \ggg NO_2 + O_2$	6.360-15	1.080-56
6	$HNO_2 + O_3 \ggg NO_3 + O_2$	6.650-18	8.330-39
7	$HNO_3 + M \ggg HO + NO_2 + M$	1.300-28	5.960-13
8	$HNO_3 + HO \ggg H_2O + NO_3$	8.000-14	2.800-29
9	$O + O + M \ggg O_2 + M$	6.520-16	8.420-59
10	$O + O_2 + M \ggg O_3 + M$	1.400-16	3.910-13
11	$O + O_3 \ggg 2^*O_2$	1.920-15	0.0
12	$O + NO + M \ggg NO_2 + M$	2.740-14	2.030-48
13	$O + NO_2 \ggg NO + O_2$	5.120-12	6.270-53
14	$O + NO_2 + M \ggg NO_3 + M$	1.710-14	1.370-24
15	$HO + HO \ggg H_2O + O$	1.090-12	9.770-27
16	$O_2 + 2^*NO \ggg 2^*HNO_2$	2.750-38	3.880-35
17	$NO_2 + H-NH \ggg NO + O$	0.0	0.0
18	$O + HO \ggg H + O_2$	4.200-11	7.730-25
19	$O + HNO_2 \ggg HO + O_2$	1.080-11	5.450-60
20	$O_2 + H + M \ggg HO_2 + M$	6.740-15	8.640-32
21	$O_3 + H \ggg HO + O_2$	1.270-11	0.0
22	$O_3 + HO \ggg HO_2 + O_2$	2.750-14	1.040-48
23	$O_3 + HNO_2 \ggg HO + 2^*O_2$	4.450-16	2.410-68
24	$H + HO + M \ggg H_2O + M$	1.370-13	0.0
25	$H + HNO_2 \ggg 2^*HO$	9.400-12	1.620-46
26	$H + HNO_2 \ggg H_2 + O_2$	1.040-11	2.550-61
27	$H + H_2O \ggg H_2 + HO$	2.340-28	1.140-15
28	$H + H_2O_2 \ggg H_2 + HO_2$	8.360-15	5.620-29
29	$H + H_2O_2 \ggg HO + H_2O$	1.090-14	1.340-75
30	$2^*HO + M \ggg H_2O_2 + M$	7.470-14	2.110-11
31	$HO + HNO_2 \ggg H_2O + O_2$	1.120-11	3.890-74
32	$2^*HNO_2 \ggg H_2O_2 + O_2$	2.300-12	8.580-49
33	$HNO_2 + H_2O \ggg H_2O_2 + HO$	1.020-34	5.490-17
34	$NO + H + M \ggg HNO + M$	4.260-15	6.670-34
35	$NO + HO \ggg NO_2 + H$	3.050-38	3.010-11
36	$NO + HO + M \ggg HNO_2 + M$	5.600-13	2.120-20
37	$NO + HNO_2 \ggg NO_2 + HO$	1.650-13	7.470-21
38	$H + H + M \ggg H_2 + M$	1.710-16	0.0
39	$HNO_4 + M \ggg HNO_2 + NO_2 + M$	3.140-06	5.800-14
40	$CLNO_3 + M \ggg ClO + NO_2 + M$	7.900-08	4.320-14

TIME (s)	1205	1202	NU3	NU	U3	O2	HNO3	H2O	H2O
0.0	5.5000	2.2000	2.0000	5.4500	1.3000	3.4000	5.0000	5.0000	5.3000
0.20	5.5000	2.2010	2.0000	5.4510	1.3000	3.4000	5.0000	5.0250	5.3000
0.40	5.5000	2.2020	1.9999	5.4420	1.3000	3.4000	5.0000	4.9900	5.3000
0.60	5.5000	2.2030	1.9999	5.4420	1.3000	3.4000	5.0000	4.9550	5.3000
0.80	5.5000	2.2040	1.9999	5.4440	1.3000	3.4000	5.0010	4.9200	5.3000
1.00	5.5000	2.2050	1.9999	5.4450	1.3000	3.4000	5.0010	4.8850	5.3000
1.20	5.5000	2.2059	1.9980	5.4450	1.3000	3.4000	5.0010	4.8510	5.3000
1.40	5.5000	2.2064	1.9980	5.4460	1.3000	3.4000	5.0010	4.8170	5.3000
1.60	5.5000	2.2070	1.9979	5.4470	1.3000	3.4000	5.0010	4.7830	5.3000
1.80	5.5000	2.2080	1.9979	5.4480	1.3000	3.4000	5.0010	4.7490	5.3000
2.00	5.5000	2.2090	1.9979	5.4490	1.3000	3.4000	5.0010	4.7160	5.3000
2.20	5.5000	2.2100	1.9966	5.4500	1.3000	3.4000	5.0010	4.6830	5.3000
2.40	5.5000	2.2110	1.9966	5.3910	1.3000	3.4000	5.0020	4.6510	5.3000
2.60	5.5000	2.2120	1.9966	5.3920	1.3000	3.4000	5.0020	4.6180	5.3000
2.80	5.5000	2.2130	1.9959	5.3740	1.3000	3.4000	5.0020	4.5860	5.3000
3.00	5.5000	2.2140	1.9959	5.3650	1.3000	3.4000	5.0020	4.5540	5.3000
3.20	5.5000	2.2150	1.9959	5.3560	1.3000	3.4000	5.0020	4.5220	5.3000
3.40	5.5000	2.2159	1.9940	5.3470	1.3000	3.4000	5.0020	4.4910	5.3000
3.60	5.5000	2.2160	1.9940	5.3380	1.3000	3.4000	5.0020	4.4590	5.3000
3.80	5.5000	2.2170	1.9940	5.3290	1.3000	3.4000	5.0020	4.4280	5.3000
4.00	5.5000	2.2180	1.9940	5.3200	1.3000	3.4000	5.0020	4.3970	5.3000
4.20	5.5000	2.2190	1.9940	5.3110	1.3000	3.4000	5.0030	4.3670	5.3000
4.40	5.5000	2.2200	1.9940	5.3030	1.3000	3.4000	5.0030	4.3370	5.3000
4.60	5.5000	2.2210	1.9930	5.2940	1.3000	3.4000	5.0030	4.3060	5.3000
4.80	5.5000	2.2219	1.9930	5.2850	1.3000	3.4000	5.0030	4.2770	5.3000
5.00	5.5000	2.2220	1.9930	5.2760	1.3000	3.4000	5.0030	4.2470	5.3000
5.20	5.5000	2.2230	1.9930	5.2670	1.3000	3.4000	5.0030	4.2170	5.3000
5.40	5.5000	2.2240	1.9930	5.2570	1.3000	3.4000	5.0030	4.1880	5.3000
5.60	5.5000	2.2250	1.9920	5.2500	1.3000	3.4000	5.0030	4.1590	5.3000
5.80	5.5000	2.2260	1.9920	5.2410	1.3000	3.4000	5.0030	4.1310	5.3000
6.00	5.5000	2.2270	1.9920	5.2330	1.3000	3.4000	5.0040	4.1020	5.3000
6.20	5.5000	2.2280	1.9920	5.2240	1.3000	3.4000	5.0040	4.0740	5.3000
6.40	5.5000	2.2290	1.9920	5.2150	1.3000	3.4000	5.0040	4.0460	5.3000
6.60	5.5000	2.2299	1.9920	5.2070	1.3000	3.4000	5.0040	4.0180	5.3000
6.80	5.5000	2.2309	1.9920	5.1980	1.3000	3.4000	5.0040	3.9900	5.3000
7.00	5.5000	2.2319	1.9920	5.1890	1.3000	3.4000	5.0040	3.9620	

TIME (S)	C	H	H2	H2O	H2O2	HNO2	HNO3	CLNO3
0.0	1.0000 04	1.0000 04	1.0000 04	1.7000 07	2.0000 08	7.0000 05	2.3000 08	4.0000 07
0.20	3.4530 05	2.7240-01	1.0000 06	1.7470 07	2.0000 08	7.0040 05	2.3000 08	4.0000 07
0.40	1.1920 05	9.3470-02	1.0000 06	1.8010 07	2.0000 08	7.0060 05	2.3000 08	4.0000 07
0.60	4.1170 04	3.2040-02	1.0000 06	1.8040 07	2.0000 08	7.0090 05	2.3000 08	4.0000 07
0.80	1.4220 04	3.1010-02	1.0000 06	1.8070 07	2.0000 08	7.0120 05	2.3000 08	4.0000 07
1.00	4.9120 03	3.7900-03	1.0000 06	1.8100 07	2.0000 08	7.0150 05	2.3000 08	4.0000 07
1.20	1.4990 03	1.3150-03	1.0000 06	1.8130 07	2.0000 08	7.0180 05	2.3000 08	4.0000 07
1.40	5.9980 02	4.6440-04	1.0000 06	1.8160 07	2.0000 08	7.0210 05	2.3000 08	4.0000 07
1.60	2.0680 02	1.7530-04	1.0000 06	1.8190 07	2.0000 08	7.0240 05	2.3000 08	4.0000 07
1.80	7.4500 01	7.5380-05	1.0000 06	1.8220 07	2.0000 08	7.0270 05	2.3000 08	4.0000 07
2.00	2.8780 01	4.0980-05	1.0000 06	1.8250 07	2.0000 08	7.0300 05	2.3000 08	4.0000 07
2.20	1.2450 01	2.9050-05	1.0000 06	1.8280 07	2.0000 08	7.0330 05	2.3000 08	4.0000 07
2.40	7.4430 00	2.4820-05	1.0000 06	1.8310 07	2.0000 08	7.0360 05	2.3000 08	4.0000 07
2.60	5.5010 00	2.3240-05	1.0000 06	1.8340 07	2.0000 08	7.0390 05	2.3000 08	4.0000 07
2.80	4.7910 00	2.2570-05	1.0000 06	1.8370 07	2.0000 08	7.0420 05	2.3000 08	4.0000 07
3.00	1.5060 00	2.2210-05	1.0000 06	1.8400 07	2.0000 08	7.0450 05	2.3000 08	4.0000 07
3.20	4.3690 00	2.1960-05	1.0000 06	1.8430 07	2.0000 08	7.0480 05	2.3000 08	4.0000 07
3.40	4.2840 00	2.1740-05	1.0000 06	1.8460 07	2.0000 08	7.0510 05	2.3000 08	4.0000 07
3.60	4.2170 00	2.1550-05	1.0000 06	1.8490 07	2.0000 08	7.0540 05	2.3000 08	4.0000 07
3.80	4.1560 00	2.1360-05	1.0000 06	1.8520 07	2.0000 08	7.0570 05	2.3000 08	4.0000 07
4.00	4.0990 00	2.1170-05	1.0000 06	1.8550 07	2.0000 08	7.0600 05	2.3000 08	4.0000 07
4.20	4.0430 00	2.0980-05	1.0000 06	1.8580 07	2.0000 08	7.0630 05	2.3000 08	4.0000 07
4.40	3.9880 00	2.0800-05	1.0000 06	1.8610 07	2.0000 08	7.0660 05	2.3000 08	4.0000 07
4.60	3.9340 00	2.0620-05	1.0000 06	1.8640 07	2.0000 08	7.0690 05	2.3000 08	4.0000 07
4.80	3.8810 00	2.0440-05	1.0000 06	1.8670 07	2.0000 08	7.0720 05	2.3000 08	4.0000 07
5.00	3.8290 00	2.0270-05	1.0000 06	1.8700 07	2.0000 08	7.0750 05	2.3000 08	4.0000 07
5.20	3.7770 00	2.0090-05	1.0000 06	1.8730 07	2.0000 08	7.0780 05	2.3000 08	4.0000 07
5.40	3.7260 00	1.9920-05	1.0000 06	1.8760 07	2.0000 08	7.0810 05	2.3000 08	4.0000 07
5.60	3.6760 00	1.9750-05	1.0000 06	1.8790 07	2.0000 08	7.0840 05	2.3000 08	4.0000 07
5.80	3.6270 00	1.9580-05	1.0000 06	1.8820 07	2.0000 08	7.0870 05	2.3000 08	4.0000 07
6.00	3.5780 00	1.9420-05	1.0000 06	1.8850 07	2.0000 08	7.0900 05	2.3000 08	4.0000 07
6.20	3.5300 00	1.9250-05	1.0000 06	1.8880 07	2.0000 08	7.0930 05	2.3000 08	4.0000 07
6.40	3.4830 00	1.9090-05	1.0000 06	1.8910 07	2.0000 08	7.0960 05	2.3000 08	4.0000 07
6.60	3.4360 00	1.8930-05	1.0000 06	1.8940 07	2.0000 08	7.0990 05	2.3000 08	4.0000 07
6.80	3.3900 00	1.8770-05	1.0000 06	1.8970 07	2.0000 08	7.1020 05	2.3000 08	4.0000 07
7.00	3.3450 00	1.8610-05	1.0000 06	1.9000 07	2.0000 08	7.1050 05	2.3000 08	4.0000 07
7.20	3.3000 00	1.8460-05	1.0000 06	1.9030 07	2.0000 08	7.1080 05	2.3000 08	4.0000 07
7.40	3.2560 00	1.8300-05	1.0000 06	1.9060 07	2.0000 08	7.1110 05	2.3000 08	4.0000 07
7.60	3.2130 00	1.8150-05	1.0000 06	1.9090 07	2.0000 08	7.1140 05	2.3000 08	4.0000 07
7.80	3.1700 00	1.8000-05	1.0000 06	1.9120 07	2.0000 08	7.1170 05	2.3000 08	4.0000 07
8.00	3.1280 00	1.7850-05	1.0000 06	1.9150 07	2.0000 08	7.1200 05	2.3000 08	4.0000 07
8.20	3.0870 00	1.7710-05	1.0000 06	1.9180 07	2.0000 08	7.1230 05	2.3000 08	4.0000 07
8.40	3.0460 00	1.7560-05	1.0000 06	1.9210 07	2.0000 08	7.1260 05	2.3000 08	4.0000 07
8.60	3.0060 00	1.7420-05	1.0000 06	1.9240 07	2.0000 08	7.1290 05	2.3000 08	4.0000 07
8.80	2.9660 00	1.7280-05	1.0000 06	1.9270 07	2.0000 08	7.1320 05	2.3000 08	4.0000 07
9.00	2.9270 00	1.7140-05	1.0000 06	1.9300 07	2.0000 08	7.1350 05	2.3000 08	4.0000 07
9.20	2.8880 00	1.7000-05	1.0000 06	1.9330 07	2.0000 08	7.1380 05	2.3000 08	4.0000 07
9.40	2.8500 00	1.6860-05	1.0000 06	1.9360 07	2.0000 08	7.1410 05	2.3000 08	4.0000 07
9.60	2.8130 00	1.6730-05	1.0000 06	1.9390 07	2.0000 08	7.1440 05	2.3000 08	4.0000 07
9.80	2.7760 00	1.6590-05	1.0000 06	1.9420 07	2.0000 08	7.1470 05	2.3000 08	4.0000 07
10.00	2.7390 00	1.6460-05	1.0000 06	1.9450 07	2.0000 08	7.1500 05	2.3000 08	4.0000 07

T=300 K, H=35 km

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N2O5} + \text{M} \ggg \text{NO2} + \text{NO3} + \text{M}$	7.930-04	1.440-14
2	$2^*\text{NO3} \ggg 2^*\text{NO2} + \text{O2}$	2.410-16	1.750-58
3	$\text{NO2} + \text{NO3} \ggg \text{NO2} + \text{NO} + \text{O2}$	8.210-15	4.920-35
4	$\text{NO3} + \text{NO} \ggg 2^*\text{NO2}$	1.900-11	6.980-29
5	$\text{NO} + \text{O3} \ggg \text{NO2} + \text{O2}$	1.670-14	2.850-49
6	$\text{NO2} + \text{O3} \ggg \text{NO3} + \text{O2}$	3.410-17	2.000-34
7	$\text{HNO3} + \text{M} \ggg \text{HO} + \text{NO2} + \text{M}$	9.700-22	3.150-13
8	$\text{HNO3} + \text{HO} \ggg \text{H2O} + \text{NO3}$	8.000-14	1.250-26
9	$\text{O} + \text{O} + \text{M} \ggg \text{O2} + \text{M}$	2.980-16	3.210-50
10	$\text{O} + \text{O2} + \text{M} \ggg \text{O3} + \text{M}$	8.280-17	6.650-10
11	$\text{O} + \text{O3} \ggg 2^*\text{O2}$	8.900-15	0.0
12	$\text{O} + \text{NO} + \text{M} \ggg \text{NO2} + \text{M}$	1.550-14	6.070-39
13	$\text{O} + \text{NO2} \ggg \text{NO} + \text{O2}$	6.250-12	3.730-46
14	$\text{O} + \text{NO2} + \text{M} \ggg \text{NO3} + \text{M}$	1.430-14	1.140-24
15	$\text{HO} + \text{HO} \ggg \text{H2O} + \text{O}$	1.570-12	4.630-24
16	$\text{O2} + 2^*\text{NO} \ggg 2^*\text{NO2}$	1.930-38	2.260-31
17	$\text{NO2} + \text{H-NU} \ggg \text{NO} + \text{O}$	0.0	0.0
18	$\text{O} + \text{HO} \ggg \text{H} + \text{O2}$	4.200-11	2.160-22
19	$\text{O} + \text{HO2} \ggg \text{HO} + \text{O2}$	1.510-11	8.510-52
20	$\text{O2} + \text{H} + \text{M} \ggg \text{HO2} + \text{M}$	4.030-15	3.280-25
21	$\text{O3} + \text{H} \ggg \text{HO} + \text{O2}$	1.790-11	1.620-69
22	$\text{O3} + \text{HO} \ggg \text{HO2} + \text{O2}$	5.350-14	8.980-43
23	$\text{O3} + \text{HO2} \ggg \text{HO} + 2^*\text{O2}$	1.040-15	1.530-63
24	$\text{H} + \text{HO} + \text{M} \ggg \text{H2O} + \text{M}$	7.120-14	5.770-76
25	$\text{H} + \text{HO2} \ggg 2^*\text{HO}$	1.770-11	1.140-40
26	$\text{H} + \text{HO2} \ggg \text{H2} + \text{O2}$	1.310-11	6.800-53
27	$\text{H} + \text{H2O} \ggg \text{H2} + \text{HO}$	2.180-25	6.410-15
28	$\text{H} + \text{H2O2} \ggg \text{H2} + \text{HO2}$	2.130-14	2.960-26
29	$\text{H} + \text{H2O2} \ggg \text{HO} + \text{H2O}$	2.760-14	2.950-65
30	$2^*\text{HO} + \text{M} \ggg \text{H2O2} + \text{M}$	3.540-14	2.610-45
31	$\text{HO} + \text{HO2} \ggg \text{H2O} + \text{O2}$	1.570-11	1.980-63
32	$2^*\text{HO2} \ggg \text{H2O2} + \text{O2}$	3.210-12	1.980-42
33	$\text{HO2} + \text{H2O} \ggg \text{H2O2} + \text{HO}$	6.110-35	8.570-13
34	$\text{NO} + \text{H} + \text{M} \ggg \text{HNO} + \text{M}$	2.910-15	7.450-27
35	$\text{NO} + \text{HO} \ggg \text{NO2} + \text{H}$	7.190-34	4.920-11
36	$\text{NO} + \text{HO} + \text{M} \ggg \text{HNO2} + \text{M}$	2.230-13	1.250-22
37	$\text{NO} + \text{HO2} \ggg \text{NO2} + \text{HO}$	3.660-13	3.740-19
38	$\text{H} + \text{H} + \text{M} \ggg \text{H2} + \text{M}$	1.430-16	1.230-69
39	$\text{HNO4} + \text{M} \ggg \text{HO2} + \text{NO2} + \text{M}$	2.090-03	2.500-14
40	$\text{CLNO3} + \text{M} \ggg \text{ClO} + \text{NO2} + \text{M}$	1.150-04	2.870-14

AD-A085 198

PERKIN-ELMER CORP. NORWALK CONN. ELECTRO-OPTICAL DIV. F/G 13/2
HIGH ALTITUDE POLLUTION PROGRAM STRATOSPHERIC MEASUREMENT SYSTEM--ETC(U)
FEB 80 N H MACOY, R WEINGARTEN, A PIRES DOT-FA77WA-4080
PE-14262 FAA/EE-80-11 NL

UNCLASSIFIED

3 OF 3
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AUG 81



END
DATE
FILMED
6-80
DTIC

HAPP RESIDENCY TIME STUDY

TIME (s)	N205	N02	N03	N0	U3	O2	MNO3	M0	M20
0.0	5.5000 07	2.2000 09	2.0000 06	5.5000 08	1.3000 12	3.8000 16	5.0000 07	5.0000 06	5.3000 11
0.20	5.4990 07	2.2020 09	2.0240 06	5.4750 08	1.3000 12	3.8000 16	5.0000 07	5.0060 06	5.3000 11
0.40	5.4980 07	2.2050 09	2.0480 06	5.4570 08	1.3000 12	3.8000 16	5.0000 07	5.0000 06	5.3000 11
0.60	5.4970 07	2.2070 09	2.0720 06	5.4290 08	1.3000 12	3.8000 16	5.0000 07	4.9350 06	5.3000 11
0.80	5.4970 07	2.2100 09	2.0960 06	5.4050 08	1.3000 12	3.8000 16	5.0000 07	4.8720 06	5.3000 11
1.00	5.4960 07	2.2120 09	2.1200 06	5.3820 08	1.3000 12	3.8000 16	5.0000 07	4.8090 06	5.3000 11
1.20	5.4950 07	2.2150 09	2.1440 06	5.3580 08	1.3000 12	3.8000 16	5.0000 07	4.7470 06	5.3000 11
1.40	5.4940 07	2.2170 09	2.1680 06	5.3350 08	1.3000 12	3.8000 16	5.0000 07	4.6860 06	5.3000 11
1.60	5.4930 07	2.2200 09	2.1920 06	5.3120 08	1.3000 12	3.8000 16	5.0010 07	4.6260 06	5.3000 11
1.80	5.4920 07	2.2220 09	2.2160 06	5.2890 08	1.3000 12	3.8000 16	5.0010 07	4.5660 06	5.3000 11
2.00	5.4910 07	2.2240 09	2.2400 06	5.2660 08	1.3000 12	3.8000 16	5.0010 07	4.5060 06	5.3000 11
2.20	5.4900 07	2.2270 09	2.2640 06	5.2430 08	1.3000 12	3.8000 16	5.0010 07	4.4500 06	5.3000 11
2.40	5.4900 07	2.2290 09	2.2870 06	5.2200 08	1.3000 12	3.8000 16	5.0010 07	4.3940 06	5.3000 11
2.60	5.4890 07	2.2310 09	2.3110 06	5.1970 08	1.3000 12	3.8000 16	5.0010 07	4.3380 06	5.3000 11
2.80	5.4880 07	2.2340 09	2.3350 06	5.1750 08	1.3000 12	3.8000 16	5.0010 07	4.2830 06	5.3000 11
3.00	5.4870 07	2.2340 09	2.3590 06	5.1520 08	1.3000 12	3.8000 16	5.0010 07	4.2290 06	5.3000 11
3.20	5.4860 07	2.2380 09	2.3830 06	5.1300 08	1.3000 12	3.8000 16	5.0010 07	4.1750 06	5.3000 11
3.40	5.4850 07	2.2410 09	2.4070 06	5.1070 08	1.3000 12	3.8000 16	5.0010 07	4.1230 06	5.3000 11
3.60	5.4840 07	2.2430 09	2.4310 06	5.0850 08	1.3000 12	3.8000 16	5.0010 07	4.0710 06	5.3000 11
3.80	5.4830 07	2.2450 09	2.4540 06	5.0630 08	1.3000 12	3.8000 16	5.0010 07	4.0200 06	5.3000 11
4.00	5.4830 07	2.2480 09	2.4780 06	5.0410 08	1.3000 12	3.8000 16	5.0010 07	3.9700 06	5.3000 11
4.20	5.4820 07	2.2500 09	2.5020 06	5.0190 08	1.3000 12	3.8000 16	5.0010 07	3.9200 06	5.3000 11
4.40	5.4810 07	2.2520 09	2.5260 06	4.9970 08	1.3000 12	3.8000 16	5.0010 07	3.8710 06	5.3000 11
4.60	5.4800 07	2.2540 09	2.5500 06	4.9760 08	1.3000 12	3.8000 16	5.0010 07	3.8230 06	5.3000 11
4.80	5.4790 07	2.2570 09	2.5740 06	4.9540 08	1.3000 12	3.8000 16	5.0010 07	3.7760 06	5.3000 11
5.00	5.4790 07	2.2590 09	2.5970 06	4.9330 08	1.3000 12	3.8000 16	5.0020 07	3.7290 06	5.3000 11
5.20	5.4770 07	2.2610 09	2.6210 06	4.9110 08	1.3000 12	3.8000 16	5.0020 07	3.6830 06	5.3000 11
5.40	5.4770 07	2.2630 09	2.6450 06	4.8900 08	1.3000 12	3.8000 16	5.0020 07	3.6380 06	5.3000 11
5.60	5.4760 07	2.2660 09	2.6690 06	4.8680 08	1.3000 12	3.8000 16	5.0020 07	3.5930 06	5.3000 11
5.80	5.4750 07	2.2680 09	2.6930 06	4.8470 08	1.3000 12	3.8000 16	5.0020 07	3.5490 06	5.3000 11
6.00	5.4740 07	2.2700 09	2.7160 06	4.8260 08	1.3000 12	3.8000 16	5.0020 07	3.5060 06	5.3000 11
6.20	5.4730 07	2.2720 09	2.7400 06	4.8050 08	1.3000 12	3.8000 16	5.0020 07	3.4630 06	5.3000 11
6.40	5.4720 07	2.2740 09	2.7640 06	4.7840 08	1.3000 12	3.8000 16	5.0020 07	3.4210 06	5.3000 11
6.60	5.4710 07	2.2770 09	2.7880 06	4.7640 08	1.3000 12	3.8000 16	5.0020 07	3.3800 06	5.3000 11
6.80	5.4700 07	2.2790 09	2.8110 06	4.7430 08	1.3000 12	3.8000 16	5.0020 07	3.3390 06	5.3000 11
7.00	5.4700 07	2.2810 09	2.8350 06	4.7220 08	1.3000 12	3.8000 16	5.0020 07	3.2990 06	5.3000 11
7.20	5.4690 07	2.2830 09	2.8590 06	4.7020 08	1.3000 12	3.8000 16	5.0020 07	3.2600 06	5.3000 11
7.40	5.4680 07	2.2850 09	2.8830 06	4.6810 08	1.3000 12	3.8000 16	5.0020 07	3.2210 06	5.3000 11
7.60	5.4670 07	2.2870 09	2.9060 06	4.6610 08	1.3000 12	3.8000 16	5.0020 07	3.1820 06	5.3000 11
7.80	5.4660 07	2.2890 09	2.9300 06	4.6410 08	1.3000 12	3.8000 16	5.0020 07	3.1450 06	5.3000 11
8.00	5.4650 07	2.2910 09	2.9540 06	4.6200 08	1.3000 12	3.8000 16	5.0020 07	3.1070 06	5.3000 11
8.20	5.4640 07	2.2940 09	2.9780 06	4.6000 08	1.3000 12	3.8000 16	5.0020 07	3.0710 06	5.3000 11
8.40	5.4640 07	2.2960 09	3.0010 06	4.5800 08	1.3000 12	3.8000 16	5.0020 07	3.0350 06	5.3000 11
8.60	5.4630 07	2.2990 09	3.0250 06	4.5600 08	1.3000 12	3.8000 16	5.0020 07	2.9990 06	5.3000 11
8.80	5.4620 07	2.3000 09	3.0490 06	4.5410 08	1.3000 12	3.8000 16	5.0020 07	2.9640 06	5.3000 11
9.00	5.4610 07	2.3020 09	3.0730 06	4.5210 08	1.3000 12	3.8000 16	5.0020 07	2.9300 06	5.3000 11
9.20	5.4600 07	2.3040 09	3.0960 06	4.5010 08	1.3000 12	3.8000 16	5.0030 07	2.8960 06	5.3000 11
9.40	5.4590 07	2.3060 09	3.1200 06	4.4820 08	1.3000 12	3.8000 16	5.0030 07	2.8620 06	5.3000 11
9.60	5.4580 07	2.3080 09	3.1440 06	4.4620 08	1.3000 12	3.8000 16	5.0030 07	2.8290 06	5.3000 11
9.80	5.4580 07	2.3100 09	3.1680 06	4.4420 08	1.3000 12	3.8000 16	5.0030 07	2.7970 06	5.3000 11
10.00	5.4570 07	2.3120 09	3.1910 06	4.4230 08	1.3000 12	3.8000 16	5.0030 07	2.7650 06	5.3000 11

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T=700 K, H=35 km

R- NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> NO2 + NO3 + M	1.160-05	1.200-14
2	2*NO3 >>> 2*NO2 + O2	2.570-14	7.440-49
3	NO2 + NO3 >>> NO2 + NO + O2	5.510-14	3.970-36
4	NO3 + NO >>> 2*NO2	1.900-11	2.230-19
5	NO + O3 >>> NO2 + O2	2.650-13	4.540-24
6	NO2 + O3 >>> NO3 + O2	3.620-15	6.560-22
7	HNO3 + M >>> HO + NO2 + M	9.700-03	1.620-14
8	HNO3 + HO >>> H2O + NO3	8.000-14	4.740-19
9	O + O + M >>> O2 + M	2.300-17	3.640-22
10	O + O2 + M >>> O3 + M	1.340-17	8.120-01
11	O + O3 >>> 2*O2	7.110-13	8.630-43
12	O + NO + M >>> NO2 + M	2.180-15	5.170-12
13	O + NO2 >>> NO + O2	1.110-11	8.500-27
14	O + NO2 + M >>> NO3 + M	6.110-15	4.890-25
15	HO + HO >>> H2O + O	4.530-12	2.040-16
16	O2 + 2*NO >>> 2*NO2	7.040-39	9.370-21
17	NO2 + H-NU >>> NO + O	0.0	0.0
18	O + HO >>> H + O2	4.200-11	2.110-15
19	O + HNO2 >>> HO + O2	3.910-11	2.190-24
20	O2 + H + M >>> HNO2 + M	6.660-16	1.500-04
21	O3 + H >>> HO + O2	4.780-11	5.620-37
22	O3 + HO >>> HNO2 + O2	3.590-13	8.310-26
23	O3 + HNO2 >>> HO + 2*O2	1.140-14	1.120-49
24	H + HO + M >>> H2O + M	3.370-15	8.100-28
25	H + HNO2 >>> 2*HO	1.080-10	5.870-24
26	H + HNO2 >>> H2 + O2	2.550-11	8.030-29
27	H + H2O >>> H2 + HO	6.560-17	8.900-13
28	H + H2O2 >>> H2 + HNO2	3.060-13	1.770-18
29	H + H2O2 >>> HO + H2O	3.980-13	1.020-35
30	2*HO + M >>> H2O2 + M	2.760-15	2.530-45
31	HO + HNO2 >>> H2O + O2	4.060-11	7.730-33
32	2*HNO2 >>> H2O2 + O2	11.320-12	3.020-24
33	HNO2 + H2O >>> H2O2 + HO	2.720-21	3.070-12
34	NO + H + M >>> HNO + M	7.040-16	5.320-07
35	NO + HO >>> NO2 + H	2.230-21	2.020-10
36	NO + HO + M >>> HNO2 + M	1.150-14	1.430-03
37	HO + HNO2 >>> NO2 + HO	3.600-12	2.680-14
38	H + H + M >>> H2 + M	6.110-17	5.120-26
39	HNO4 + M >>> HNO2 + NO2 + M	1.760-05	1.650-15
40	CLNO3 + M >>> CLO + NO2 + M	9.080-04	6.530-16

HAPP RESIDENCE TIME STUDY

TIME (S)	1205	1202	1201	1200	1159	1158	1157	1156	1155	1154	1153	1152	1151	1150	1149	1148	1147	1146	1145	1144	1143	1142	1141	1140	1139	1138	1137	1136	1135	1134	1133	1132	1131	1130	1129	1128	1127	1126	1125	1124	1123	1122	1121	1120	1119	1118	1117	1116	1115	1114	1113	1112	1111	1110	1109	1108	1107	1106	1105	1104	1103	1102	1101	1100	1099	1098	1097	1096	1095	1094	1093	1092	1091	1090	1089	1088	1087	1086	1085	1084	1083	1082	1081	1080	1079	1078	1077	1076	1075	1074	1073	1072	1071	1070	1069	1068	1067	1066	1065	1064	1063	1062	1061	1060	1059	1058	1057	1056	1055	1054	1053	1052	1051	1050	1049	1048	1047	1046	1045	1044	1043	1042	1041	1040	1039	1038	1037	1036	1035	1034	1033	1032	1031	1030	1029	1028	1027	1026	1025	1024	1023	1022	1021	1020	1019	1018	1017	1016	1015	1014	1013	1012	1011	1010	1009	1008	1007	1006	1005	1004	1003	1002	1001	1000	999	998	997	996	995	994	993	992	991	990	989	988	987	986	985	984	983	982	981	980	979	978	977	976	975	974	973	972	971	970	969	968	967	966	965	964	963	962	961	960	959	958	957	956	955	954	953	952	951	950	949	948	947	946	945	944	943	942	941	940	939	938	937	936	935	934	933	932	931	930	929	928	927	926	925	924	923	922	921	920	919	918	917	916	915	914	913	912	911	910	909	908	907	906	905	904	903	902	901	900	899	898	897	896	895	894	893	892	891	890	889	888	887	886	885	884	883	882	881	880	879	878	877	876	875	874	873	872	871	870	869	868	867	866	865	864	863	862	861	860	859	858	857	856	855	854	853	852	851	850	849	848	847	846	845	844	843	842	841	840	839	838	837	836	835	834	833	832	831	830	829	828	827	826	825	824	823	822	821	820	819	818	817	816	815	814	813	812	811	810	809	808	807	806	805	804	803	802	801	800	799	798	797	796	795	794	793	792	791	790	789	788	787	786	785	784	783	782	781	780	779	778	777	776	775	774	773	772	771	770	769	768	767	766	765	764	763	762	761	760	759	758	757	756	755	754	753	752	751	750	749	748	747	746	745	744	743	742	741	740	739	738	737	736	735	734	733	732	731	730	729	728	727	726	725	724	723	722	721	720	719	718	717	716	715	714	713	712	711	710	709	708	707	706	705	704	703	702	701	700	699	698	697	696	695	694	693	692	691	690	689	688	687	686	685	684	683	682	681	680	679	678	677	676	675	674	673	672	671	670	669	668	667	666	665	664	663	662	661	660	659	658	657	656	655	654	653	652	651	650	649	648	647	646	645	644	643	642	641	640	639	638	637	636	635	634	633	632	631	630	629	628	627	626	625	624	623	622	621	620	619	618	617	616	615	614	613	612	611	610	609	608	607	606	605	604	603	602	601	600	599	598	597	596	595	594	593	592	591	590	589	588	587	586	585	584	583	582	581	580	579	578	577	576	575	574	573	572	571	570	569	568	567	566	565	564	563	562	561	560	559	558	557	556	555	554	553	552	551	550	549	548	547	546	545	544	543	542	541	540	539	538	537	536	535	534	533	532	531	530	529	528	527	526	525	524	523	522	521	520	519	518	517	516	515	514	513	512	511	510	509	508	507	506	505	504	503	502	501	500	499	498	497	496	495	494	493	492	491	490	489	488	487	486	485	484	483	482	481	480	479	478	477	476	475	474	473	472	471	470	469	468	467	466	465	464	463	462	461	460	459	458	457	456	455	454	453	452	451	450	449	448	447	446	445	444	443	442	441	440	439	438	437	436	435	434	433	432	431	430	429	428	427	426	425	424	423	422	421	420	419	418	417	416	415	414	413	412	411	410	409	408	407	406	405	404	403	402	401	400	399	398	397	396	395	394	393	392	391	390	389	388	387	386	385	384	383	382	381	380	379	378	377	376	375	374	373	372	371	370	369	368	367	366	365	364	363	362	361	360	359	358	357	356	355	354	353	352	351	350	349	348	347	346	345	344	343	342	341	340	339	338	337	336	335	334	333	332	331	330	329	328	327	326	325	324	323	322	321	320	319	318	317	316	315	314	313	312	311	310	309	308	307	306	305	304	303	302	301	300	299	298	297	296	295	294	293	292	291	290	289	288	287	286	285	284	283	282	281	280	279	278	277	276	275	274	273	272	271	270	269	268	267	266	265	264	263	262	261	260	259	258	257	256	255	254	253	252	251	250	249	248	247	246	245	244	243	242	241	240	239	238	237	236	235	234	233	232	231	230	229	228	227	226	225	224	223	222	221	220	219	218	217	216	215	214	213	212	211	210	209	208	207	206	205	204	203	202	201	200	199	198	197	196	195	194	193	192	191	190	189	188	187	186	185	184	183	182	181	180	179	178	177	176	175	174	173	172	171	170	169	168	167	166	165	164	163	162	161	160	159	158	157	156	155	154	153	152	151	150	149	148	147	146	145	144	143	142	141	140	139	138	137	136	135	134	133	132	131	130	129	128	127	126	125	124	123	122	121	120	119	118	117	116	115	114	113	112	111	110	109	108	107	106	105	104	103	102	101	100	99	98	97	96	95	94	93	92	91	90	89	88	87	86	85	84	83	82	81	80	79	78	77	76	75	74	73	72	71	70	69	68	67	66	65	64	63	62	61	60	59	58	57	56	55	54	53	52	51	50	49	48	47	46	45	44	43	42	41	40	39	38	37	36	35	34	33	32	31	30	29	28	27	26	25	24	23	22	21	20	19	18	17	16	15	14	13	12	11	10	9	8	7	6	5	4	3	2	1	0
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TIME (s)	N	M	M2	HQ2	HQ2Z	HNO	HNO2	HNO4	CLNO3
0.0	1.0000	0.0	1.0000	1.7000	2.6000	1.0000	7.0000	2.3000	4.0000
0.20	1.6950	1.1	1.0010	1.3000	2.6010	1.0000	6.9990	2.5530	1.3590
0.40	2.7630	1.1	1.0080	5.3540	2.6010	1.0000	7.0020	5.6360	8.5730
0.60	3.4440	1.1	1.0100	4.6690	2.6000	1.0000	7.0100	3.0530	4.7130
0.80	3.8560	1.1	1.0170	4.6690	2.6000	1.0000	7.0210	1.6980	2.5180
1.00	4.0930	1.1	1.0240	4.9710	2.7990	1.0000	7.0340	1.0240	1.4310
1.20	4.2100	1.1	1.0330	5.3070	2.7990	1.0000	7.0480	0.6750	9.1660
1.40	4.2450	1.1	1.0440	5.6590	2.7980	1.0000	7.0630	5.4550	6.6750
1.60	4.2750	1.1	1.0560	6.0140	2.7980	1.0000	7.0790	4.6730	5.3180
1.80	4.3030	1.1	1.0690	6.3710	2.7980	1.0000	7.0960	4.2650	4.6340
2.00	4.3290	1.1	1.0840	6.7240	2.7970	1.0000	7.1140	4.0450	4.1640
2.20	4.3530	1.1	1.1000	7.0710	2.7970	1.0000	7.1330	3.9280	3.8450
2.40	4.3740	1.1	1.1170	7.4110	2.7960	1.0000	7.1530	3.8730	3.6180
2.60	4.3930	1.1	1.1350	7.7420	2.7960	1.0000	7.1730	3.8600	3.4510
2.80	4.4100	1.1	1.1540	8.0630	2.7950	1.0000	7.1950	3.8750	3.3270
3.00	4.4250	1.1	1.1740	8.3750	2.7950	1.0000	7.2170	3.9120	3.2340
3.20	4.4390	1.1	1.1950	8.6780	2.7940	1.0000	7.2400	3.9660	3.1640
3.40	4.4520	1.1	1.2160	8.9710	2.7940	1.0000	7.2640	4.0320	3.1110
3.60	4.4640	1.1	1.2380	9.2540	2.7930	1.0000	7.2890	4.1080	3.0730
3.80	4.4750	1.1	1.2610	9.5280	2.7930	1.0000	7.3140	4.1910	3.0440
4.00	4.4850	1.1	1.2840	9.7940	2.7920	1.0000	7.3400	4.2790	3.0250
4.20	4.4940	1.1	1.3080	1.0050	2.7910	1.0000	7.3670	4.3720	3.0110
4.40	4.5030	1.1	1.3320	1.0300	2.7910	1.0000	7.3950	4.4670	3.0020
4.60	4.5110	1.1	1.3560	1.0540	2.7900	1.0000	7.4230	4.5640	2.9960
4.80	4.5190	1.1	1.3810	1.0770	2.7900	1.0000	7.4510	4.6620	2.9920
5.00	4.5260	1.1	1.4060	1.0990	2.7890	1.0000	7.4810	4.7600	2.9900
5.20	4.5330	1.1	1.4320	1.1210	2.7880	1.0000	7.5110	4.8580	3.0030
5.40	4.5390	1.1	1.4570	1.1420	2.7880	1.0000	7.5410	4.9550	3.0030
5.60	4.5450	1.1	1.4830	1.1630	2.7870	1.0000	7.5720	5.0510	3.0080
5.80	4.5500	1.1	1.5090	1.1840	2.7870	1.0000	7.6030	5.1460	3.0130
6.00	4.5550	1.1	1.5350	1.2010	2.7860	1.0000	7.6350	5.2400	3.0200
6.20	4.5600	1.1	1.5610	1.2280	2.7850	1.0000	7.6680	5.3320	3.0260
6.40	4.5640	1.1	1.5880	1.2380	2.7850	1.0010	7.7010	5.4220	3.0330
6.60	4.5680	1.1	1.6140	1.2550	2.7840	1.0010	7.7340	5.5110	3.0400
6.80	4.5720	1.1	1.6400	1.2720	2.7830	1.0010	7.7680	5.5970	3.0470
7.00	4.5750	1.1	1.6670	1.2880	2.7830	1.0010	7.8020	5.6820	3.0540
7.20	4.5780	1.1	1.6930	1.3030	2.7820	1.0010	7.8370	5.7640	3.0610
7.40	4.5810	1.1	1.7200	1.3190	2.7810	1.0010	7.8720	5.8450	3.0680
7.60	4.5840	1.1	1.7460	1.3330	2.7810	1.0010	7.9070	5.9240	3.0750
7.80	4.5870	1.1	1.7730	1.3480	2.7800	1.0010	7.9430	6.0010	3.0820
8.00	4.5900	1.1	1.7990	1.3620	2.7790	1.0010	7.9790	6.0760	3.0890
8.20	4.5930	1.1	1.8250	1.3750	2.7790	1.0010	8.0150	6.1490	3.0960
8.40	4.5960	1.1	1.8520	1.3880	2.7780	1.0010	8.0520	6.2210	3.1020
8.60	4.5990	1.1	1.8780	1.4010	2.7780	1.0010	8.0890	6.2900	3.1080
8.80	4.6020	1.1	1.9040	1.4130	2.7770	1.0010	8.1270	6.3580	3.1150
9.00	4.6050	1.1	1.9300	1.4250	2.7760	1.0010	8.1640	6.4240	3.1210
9.20	4.6080	1.1	1.9560	1.4360	2.7750	1.0010	8.2020	6.4890	3.1270
9.40	4.6110	1.1	1.9820	1.4480	2.7750	1.0010	8.2410	6.5520	3.1330
9.60	4.6140	1.1	2.0080	1.4590	2.7740	1.0010	8.2790	6.6130	3.1390
9.80	4.6170	1.1	2.0340	1.4690	2.7740	1.0010	8.3180	6.6730	3.1450
10.00	4.6200	1.1	2.0590	1.4790	2.7730	1.0010	8.3570	6.7310	3.1500

T=800 K, H=35 km

R-NAME	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N2O5} + \text{M} \gg \text{N02} + \text{N03} + \text{M}$	6.42E-05	8.97E-16
2	$2\text{N03} \gg 2\text{N02} + \text{O2}$	3.98E-14	5.95E-48
3	$\text{N02} + \text{N03} \gg \text{N02} + \text{N0} + \text{O2}$	6.59E-14	3.13E-36
4	$\text{N03} + \text{N0} \gg 2\text{N02}$	1.90E-11	1.74E-18
5	$\text{N0} + \text{O3} \gg \text{N02} + \text{O2}$	3.43E-13	4.42E-26
6	$\text{N02} + \text{O3} \gg \text{N03} + \text{O2}$	5.61E-15	9.78E-21
7	$\text{HNO3} + \text{M} \gg \text{H0} + \text{N02} + \text{M}$	4.60E-01	1.02E-14
8	$\text{HNO3} + \text{H0} \gg \text{H2O} + \text{N03}$	8.00E-14	2.43E-18
9	$\text{O} + \text{O} + \text{M} \gg \text{O2} + \text{M}$	1.71E-17	1.10E-17
10	$\text{O} + \text{O2} + \text{M} \gg \text{O3} + \text{M}$	1.07E-17	5.47E-00
11	$\text{O} + \text{O3} \gg 2\text{O2}$	1.07E-12	6.11E-39
12	$\text{O} + \text{N0} + \text{M} \gg \text{N02} + \text{M}$	1.72E-15	1.64E-09
13	$\text{O} + \text{N02} \gg \text{N0} + \text{O2}$	1.17E-11	5.55E-25
14	$\text{O} + \text{N02} + \text{M} \gg \text{N03} + \text{M}$	5.35E-15	4.28E-25
15	$\text{H0} + \text{H0} \gg \text{H2O} + \text{O}$	5.00E-12	1.06E-15
16	$\text{O2} + 2\text{N0} \gg 2\text{N02}$	6.40E-39	8.78E-28
17	$\text{N02} + \text{H-NU} \gg \text{N0} + \text{O}$	0.0	0.0
18	$\text{O} + \text{H0} \gg \text{H} + \text{O2}$	4.20E-11	9.55E-15
19	$\text{O} + \text{H02} \gg \text{H0} + \text{O2}$	4.28E-11	3.43E-26
20	$\text{O2} + \text{H} + \text{M} \gg \text{H02} + \text{M}$	5.33E-18	7.95E-05
21	$\text{O3} + \text{H} \gg \text{H0} + \text{O2}$	5.25E-11	6.31E-34
22	$\text{O3} + \text{H0} \gg \text{H02} + \text{O2}$	4.30E-13	3.24E-24
23	$\text{O3} + \text{H02} \gg \text{H0} + 2\text{O2}$	1.48E-14	2.35E-48
24	$\text{H} + \text{H0} + \text{M} \gg \text{H2O} + \text{M}$	2.08E-15	2.06E-29
25	$\text{H} + \text{H02} \gg 2\text{H0}$	1.28E-10	2.16E-22
26	$\text{H} + \text{H02} \gg \text{H2} + \text{O2}$	2.71E-11	1.45E-26
27	$\text{H} + \text{H2O} \gg \text{H2} + \text{H0}$	4.09E-16	1.41E-12
28	$\text{H} + \text{H2O2} \gg \text{H2} + \text{H02}$	3.93E-13	9.47E-18
29	$\text{H} + \text{H2O2} \gg \text{H0} + \text{H2O}$	5.11E-13	6.02E-39
30	$2\text{H0} + \text{M} \gg \text{H2O2} + \text{M}$	2.06E-15	1.70E-03
31	$\text{H0} + \text{H02} \gg \text{H2O} + \text{O2}$	4.44E-11	5.70E-30
32	$2\text{H02} \gg \text{H2O2} + \text{O2}$	9.10E-12	1.53E-22
33	$\text{H02} + \text{H2O} \gg \text{H2O2} + \text{H0}$	5.19E-20	3.46E-12
34	$\text{N0} + \text{H} + \text{M} \gg \text{HNO} + \text{M}$	5.84E-16	3.47E-05
35	$\text{N0} + \text{H0} \gg \text{N02} + \text{H}$	3.30E-20	2.30E-10
36	$\text{N0} + \text{H0} + \text{M} \gg \text{HNO2} + \text{M}$	8.27E-15	7.88E-02
37	$\text{N0} + \text{H02} \gg \text{H02} + \text{H0}$	4.46E-12	7.66E-14
38	$\text{H} + \text{H} + \text{M} \gg \text{H2} + \text{M}$	5.35E-17	5.64E-22
39	$\text{HNO4} + \text{M} \gg \text{H02} + \text{N02} + \text{M}$	9.25E-05	1.21E-15
40	$\text{CLN03} + \text{M} \gg \text{CL0} + \text{N02} + \text{M}$	5.87E-05	3.77E-16

TIME (s)	12/25	12/22	11/23	11/10	11/4	10/2	10/0	9/0	8/0	7/0
0.0	5.5900	2.2000	2.0000	5.5000	1.3000	3.4000	5.0000	5.0000	5.0000	5.3000
0.20	7.3770-05	4.9220	5.9210	2.1450	4.1900	3.8000	3.1730	3.1730	3.1730	5.2990
0.40	1.1670-05	1.4300	2.7400	3.7400	5.5000	3.0000	4.4500	4.4500	4.4500	5.2990
0.60	2.2050-05	2.6900	3.0500	3.0500	4.5000	3.0000	6.4860	6.4860	6.4860	5.2970
0.80	9.2190-07	1.1410	5.7840	3.0780	6.6110	3.4000	8.5300	8.5300	8.5300	5.2960
1.00	7.1380-07	0.9400	3.0940	3.0940	6.0110	3.8000	3.1580	3.1580	3.1580	5.2950
1.20	6.5540-07	8.4310	3.6520	3.0880	5.7450	3.8000	2.8810	2.8810	2.8810	5.2940
1.40	6.6640-07	8.2880	5.5460	3.0920	5.6130	3.8000	2.6280	2.6280	2.6280	5.2930
1.60	5.3420-07	6.2250	6.5210	3.0950	5.4990	3.8000	1.5510	1.5510	1.5510	5.2930
1.80	6.2130-07	4.1880	5.4570	3.0970	5.3750	3.8000	2.1860	2.1860	2.1860	5.2920
2.00	6.1450-07	4.1590	6.3940	3.1000	5.2660	3.8000	1.9940	1.9940	1.9940	5.2910
2.20	6.0560-07	4.1360	5.3310	3.1020	5.1600	3.8000	1.8590	1.8590	1.8590	5.2910
2.40	6.0710-07	4.1160	6.2690	3.1050	5.0540	3.8000	1.6190	1.6190	1.6190	5.2900
2.60	5.8900-07	4.1000	5.2670	3.1070	4.9680	3.8000	1.5140	1.5140	1.5140	5.2890
2.80	5.8120-07	4.0870	5.1460	3.1090	4.9420	3.8000	1.3810	1.3810	1.3810	5.2880
3.00	5.7730-07	4.0780	5.0880	3.1110	4.7480	3.8000	1.2590	1.2590	1.2590	5.2870
3.20	5.6640-07	4.0680	5.0270	3.1120	4.6770	3.8000	1.1490	1.1490	1.1490	5.2860
3.40	5.5940-07	4.0630	4.9680	3.1140	4.5890	3.8000	1.0480	1.0480	1.0480	5.2860
3.60	5.5280-07	4.0600	4.9100	3.1150	4.5030	3.8000	9.5570	9.5570	9.5570	5.2850
3.80	5.4610-07	4.0590	4.8520	3.1170	4.4200	3.8000	8.7180	8.7180	8.7180	5.2840
4.00	5.3980-07	4.0600	4.7960	3.1180	4.3380	3.8000	7.9520	7.9520	7.9520	5.2840
4.20	5.3370-07	4.0630	4.7390	3.1200	4.2590	3.8000	7.2530	7.2530	7.2530	5.2830
4.40	5.2770-07	4.0640	4.6840	3.1210	4.1820	3.8000	6.6160	6.6160	6.6160	5.2830
4.60	5.2190-07	4.0740	4.6290	3.1220	4.1080	3.8000	6.0350	6.0350	6.0350	5.2820
4.80	5.1630-07	4.0820	4.5740	3.1230	4.0350	3.8000	5.5050	5.5050	5.5050	5.2820
5.00	5.1080-07	4.0910	4.5210	3.1240	3.9640	3.8000	5.0210	5.0210	5.0210	5.2810
5.20	5.0540-07	4.0910	4.4640	3.1250	3.8950	3.8000	4.5800	4.5800	4.5800	5.2810
5.40	5.0020-07	4.1130	4.4150	3.1260	3.8280	3.8000	4.1780	4.1780	4.1780	5.2800
5.60	4.9510-07	4.1260	4.3630	3.1270	3.7620	3.8000	3.8110	3.8110	3.8110	5.2800
5.80	4.9010-07	4.1400	4.3120	3.1270	3.6990	3.8000	3.4760	3.4760	3.4760	5.2790
6.00	4.8520-07	4.1540	4.2610	3.1290	3.6370	3.8000	3.1710	3.1710	3.1710	5.2790
6.20	4.8040-07	4.1690	4.21							

25 km HNO₃, N₂O₅, O

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	4.05D-06	1.27D-13
2	2*N03 >>> 2*N02 + O2	4.71D-17	7.46D-62
3	N02 + N03 >>> N02 + N0 + O2	4.21D-15	1.19D-34
4	N03 + N0 >>> 2*N02	1.90D-11	3.29D-32
5	N0 + O3 >>> N02 + O2	6.36D-15	1.08D-56
6	N02 + O3 >>> N03 + O2	6.65D-18	8.33D-39
7	HNO3 + M >>> H0 + N02 + M	5.36D-28	2.46D-12
8	HNO3 + H0 >>> H2O + N03	8.00D-14	3.76D-30
9	O + O + M >>> O2 + M	2.69D-15	1.44D-57
10	O + O2 + M >>> O3 + M	5.76D-16	1.62D-12
11	O + O3 >>> 2*O2	1.92D-15	0.0
12	O + N0 + M >>> N02 + M	1.13D-13	8.39D-48
13	O + N02 >>> N0 + O2	5.12D-12	6.27D-53
14	O + N02 + M >>> N03 + M	7.07D-14	5.66D-24
15	H0 + H0 >>> H2O + O	1.09D-12	9.77D-27
16	O2 + 2*N0 >>> 2*N02	2.75D-38	3.88D-35

HAPP RESIDENCE TIME STUDY

TIME (S)	N2O5	NO2	NO3	NO	O3	O2	HNO3	HO	H2O
0.0	6.5000 08	6.3000 09	2.0000 06	7.0000 09	4.3000 12	1.4100 17	3.0000 09	1.0000 06	1.0000 11
0.50	6.5000 08	6.3950 09	1.9610 06	6.9050 09	4.3000 12	1.4100 17	3.0000 09	9.9230 05	1.0000 11
1.00	6.5000 08	6.4890 09	1.9280 06	6.8110 09	4.3000 12	1.4100 17	3.0000 09	9.8450 05	1.0000 11
1.50	6.5000 08	6.5820 09	1.8990 06	6.7180 09	4.3000 12	1.4100 17	3.0000 09	9.7660 05	1.0000 11
2.00	6.5000 08	6.6730 09	1.8750 06	6.6270 09	4.3000 12	1.4100 17	3.0000 09	9.6870 05	1.0000 11
2.50	6.5000 08	6.7630 09	1.8560 06	6.5370 09	4.3000 12	1.4100 17	3.0000 09	9.6080 05	1.0000 11
3.00	6.5000 08	6.8520 09	1.8400 06	6.4480 09	4.2990 12	1.4100 17	3.0000 09	9.5280 05	1.0000 11
3.50	6.5000 08	6.9400 09	1.8280 06	6.3600 09	4.2990 12	1.4100 17	3.0000 09	9.4480 05	1.0000 11
4.00	6.5000 08	7.0260 09	1.8200 06	6.2740 09	4.2990 12	1.4100 17	3.0000 09	9.3680 05	1.0000 11
4.50	6.5000 08	7.1110 09	1.8140 06	6.1890 09	4.2990 12	1.4100 17	3.0000 09	9.2870 05	1.0000 11
5.00	6.5000 08	7.1950 09	1.8120 06	6.1050 09	4.2990 12	1.4100 17	3.0000 09	9.2060 05	1.0000 11
5.50	6.5000 08	7.2780 09	1.8120 06	6.0220 09	4.2990 12	1.4100 17	3.0000 09	9.1250 05	1.0000 11
6.00	6.5000 08	7.3600 09	1.8140 06	5.9400 09	4.2990 12	1.4100 17	3.0000 09	9.0430 05	1.0000 11
6.50	6.5000 08	7.4410 09	1.8190 06	5.8590 09	4.2990 12	1.4100 17	3.0000 09	8.9620 05	1.0000 11
7.00	6.5000 08	7.5210 09	1.8260 06	5.7800 09	4.2990 12	1.4100 17	3.0000 09	8.8800 05	1.0000 11
7.50	6.5000 08	7.5990 09	1.8360 06	5.7010 09	4.2990 12	1.4100 17	3.0000 09	8.7980 05	1.0000 11
8.00	6.5000 08	7.6770 09	1.8470 06	5.6230 09	4.2990 12	1.4100 17	3.0000 09	8.7170 05	1.0000 11
8.50	6.5000 08	7.7530 09	1.8590 06	5.5470 09	4.2990 12	1.4100 17	3.0000 09	8.6350 05	1.0000 11
9.00	6.5000 08	7.8280 09	1.8740 06	5.4720 09	4.2980 12	1.4100 17	3.0000 09	8.5530 05	1.0000 11
9.50	6.5000 08	7.9030 09	1.8900 06	5.3970 09	4.2980 12	1.4100 17	3.0000 09	8.4710 05	1.0000 11
10.00	6.5000 08	7.9760 09	1.9080 06	5.3240 09	4.2980 12	1.4100 17	3.0000 09	8.3890 05	1.0000 11

25 km HNO₃, N₂O₅, O, 300°K

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	3.270-03	5.94D-14
2	2*N03 >>> 2*N02 + O2	2.41D-16	1.75D-58
3	N02 + N03 >>> N02 + N0 + O2	8.21D-15	4.92D-35
4	N03 + N0 >>> 2*N02	1.90D-11	6.98D-29
5	N0 + O3 >>> N02 + O2	1.67D-14	2.85D-49
6	N02 + O3 >>> N03 + O2	3.41D-17	2.00D-34
7	HN03 + M >>> H0 + N02 + M	4.01D-21	1.30D-12
8	HN03 + H0 >>> H2O + N03	8.00D-14	1.69D-27
9	O + O + M >>> O2 + M	1.23D-15	5.48D-58
10	O + O2 + M >>> O3 + M	3.42D-16	2.74D-09
11	O + O3 >>> 2*O2	8.90D-15	0.0
12	O + N0 + M >>> N02 + M	6.40D-14	2.51D-38
13	O + N02 >>> N0 + O2	6.25D-12	3.73D-46
14	O + N02 + M >>> N03 + M	5.89D-14	4.71D-24
15	H0 + H0 >>> H2O + O	1.57D-12	4.63D-24
16	O2 + 2*N0 >>> 2*N02	1.93D-38	2.26D-31

HAPP RESIDENCE TIME STUDY

TIME(S)	N2O5	N2O	NO3	NO	O3	O2	HNO3	HO	H2O
0.0	6.500D 08	6.300D 09	2.000D 06	7.000D 09	4.300D 12	1.180D 17	3.000D 09	1.000D 06	1.000D 11
0.50	6.489D 08	6.548D 09	3.358D 06	6.753D 09	4.300D 12	1.180D 17	3.000D 09	9.959D 05	1.000D 11
1.00	6.479D 08	6.787D 09	4.655D 06	6.514D 09	4.300D 12	1.180D 17	3.000D 09	9.916D 05	1.000D 11
1.50	6.468D 08	7.019D 09	5.898D 06	6.284D 09	4.299D 12	1.180D 17	3.000D 09	9.872D 05	1.000D 11
2.00	6.458D 08	7.242D 09	7.096D 06	6.062D 09	4.299D 12	1.180D 17	3.000D 09	9.827D 05	1.000D 11
2.50	6.447D 08	7.457D 09	8.255D 06	5.857D 09	4.299D 12	1.180D 17	3.000D 09	9.780D 05	1.000D 11
3.00	6.437D 08	7.665D 09	9.380D 06	5.641D 09	4.299D 12	1.180D 17	3.000D 09	9.732D 05	1.000D 11
3.50	6.426D 08	7.865D 09	1.048D 07	5.441D 09	4.298D 12	1.180D 17	3.000D 09	9.684D 05	1.000D 11
4.00	6.416D 08	8.059D 09	1.155D 07	5.248D 09	4.298D 12	1.180D 17	3.000D 09	9.634D 05	1.000D 11
4.50	6.405D 08	8.246D 09	1.260D 07	5.053D 09	4.298D 12	1.180D 17	3.000D 09	9.584D 05	1.000D 11
5.00	6.395D 08	8.426D 09	1.364D 07	4.853D 09	4.298D 12	1.180D 17	3.000D 09	9.532D 05	1.000D 11
5.50	6.384D 08	8.600D 09	1.466D 07	4.710D 09	4.298D 12	1.180D 17	3.000D 09	9.480D 05	1.000D 11
6.00	6.374D 08	8.764D 09	1.567D 07	4.544D 09	4.298D 12	1.180D 17	3.000D 09	9.427D 05	1.000D 11
6.50	6.363D 08	8.930D 09	1.667D 07	4.383D 09	4.297D 12	1.180D 17	3.000D 09	9.373D 05	1.000D 11
7.00	6.353D 08	9.086D 09	1.766D 07	4.227D 09	4.297D 12	1.180D 17	3.000D 09	9.319D 05	1.000D 11
7.50	6.343D 08	9.237D 09	1.865D 07	4.078D 09	4.297D 12	1.180D 17	3.000D 09	9.264D 05	1.000D 11
8.00	6.332D 08	9.383D 09	1.963D 07	3.933D 09	4.297D 12	1.180D 17	3.000D 09	9.209D 05	1.000D 11
8.50	6.322D 08	9.523D 09	2.062D 07	3.793D 09	4.297D 12	1.180D 17	3.000D 09	9.153D 05	1.000D 11
9.00	6.312D 08	9.659D 09	2.160D 07	3.659D 09	4.297D 12	1.180D 17	3.000D 09	9.096D 05	1.000D 11
9.50	6.302D 08	9.790D 09	2.258D 07	3.529D 09	4.297D 12	1.180D 17	3.000D 09	9.039D 05	1.000D 11
10.00	6.291D 08	9.916D 09	2.357D 07	3.404D 09	4.296D 12	1.180D 17	3.000D 09	8.982D 05	1.000D 11

25 km HNO₃, N₂O₅, O, 400°K

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> NO2 + NO3 + M	1.33D 01	2.17D-14
2	2*N03 >>> 2*N02 + O2	1.86D-15	2.85D-54
3	NO2 + NO3 >>> NO2 + NO + O2	1.89D-14	1.63D-35
4	NO3 + NO >>> 2*N02	1.90D-11	1.00D-24
5	NO + O3 >>> NO2 + O2	5.60D-14	5.39D-40
6	NO2 + O3 >>> NO3 + O2	2.62D-16	5.96D-29
7	HNO3 + M >>> HO + NO2 + M	1.20D-12	4.75D-13
8	HNO3 + HO >>> H2O + NO3	8.00D-14	3.49D-24
9	O + O + M >>> O2 + M	4.36D-16	5.06D-48
10	O + O2 + M >>> O3 + M	1.68D-16	2.82D-05
11	O + O3 >>> 2*O2	6.05D-14	6.81D-66
12	O + NO + M >>> NO2 + M	2.95D-14	1.65D-26
13	O + NO2 >>> NO + O	8.03D-12	1.10D-37
14	O + NO2 >>> NO + O	4.42D-14	3.54D-24
		2.50D-12	1.02D-20
		1.24D-38	1.09D-26

HAPP RESIDENCE TIME STUDY

TIME(S)	N205	N02	N03	N0	03	02	HNO3	H0	H2O
0.0	6.5000 08	6.3000 09	2.0000 06	7.0000 09	4.3000 12	8.8400 16	3.0000 09	1.0000 06	1.0000 11
0.50	8.4800 05	7.8040 09	6.2160 08	6.1750 09	4.2990 12	8.8400 16	3.0000 09	9.9830 05	1.0000 11
1.00	9.3410 03	8.5650 09	5.9360 08	5.4440 09	4.2980 12	8.8400 16	3.0000 09	9.9640 05	1.0000 11
1.50	8.5610 03	9.2310 09	5.7030 08	4.8010 09	4.2980 12	8.8400 16	3.0000 09	9.9430 05	1.0000 11
2.00	8.8110 03	9.8160 09	5.5160 08	4.2340 09	4.2970 12	8.8400 16	3.0000 09	9.9210 05	1.0000 11
2.50	9.0260 03	1.0330 10	5.3660 08	3.7360 09	4.2970 12	8.8400 16	3.0000 09	9.8980 05	1.0000 11
3.00	9.2170 03	1.0780 10	5.2480 08	3.2960 09	4.2960 12	8.8400 16	3.0000 09	9.8730 05	1.0000 11
3.50	9.3910 03	1.1180 10	5.1570 08	2.9090 09	4.2960 12	8.8400 16	3.0000 09	9.8480 05	1.0000 11
4.00	9.5550 03	1.1530 10	5.0870 08	2.5670 09	4.2950 12	8.8400 16	3.0000 09	9.8220 05	1.0000 11
4.50	9.7130 03	1.1830 10	5.0360 08	2.2660 09	4.2950 12	8.8400 16	3.0000 09	9.7950 05	1.0000 11
5.00	9.8660 03	1.2100 10	5.0020 08	2.0000 09	4.2950 12	8.8400 16	3.0000 09	9.7670 05	1.0000 11
5.50	1.0020 04	1.2340 10	4.9810 08	1.7660 09	4.2950 12	8.8400 16	3.0000 09	9.7390 05	1.0000 11
6.00	1.0170 04	1.2550 10	4.9720 08	1.5590 09	4.2950 12	8.8400 16	3.0000 09	9.7110 05	1.0000 11
6.50	1.0320 04	1.2730 10	4.9730 08	1.3760 09	4.2940 12	8.8400 16	3.0000 09	9.6820 05	1.0000 11
7.00	1.0470 04	1.2890 10	4.9840 08	1.2150 09	4.2940 12	8.8400 16	3.0000 09	9.6530 05	1.0000 11
7.50	1.0620 04	1.3030 10	5.0020 08	1.0730 09	4.2940 12	8.8400 16	3.0000 09	9.6240 05	1.0000 11
8.00	1.0780 04	1.3150 10	5.0270 08	9.4740 08	4.2940 12	8.8400 16	3.0000 09	9.5940 05	1.0000 11
8.50	1.0930 04	1.3260 10	5.0580 08	8.3660 08	4.2940 12	8.8400 16	3.0000 09	9.5640 05	1.0000 11
9.00	1.1090 04	1.3350 10	5.0950 08	7.3880 08	4.2940 12	8.8400 16	3.0000 09	9.5350 05	1.0000 11
9.50	1.1250 04	1.3440 10	5.1360 08	6.5240 08	4.2940 12	8.8400 16	3.0000 09	9.5050 05	1.0000 11
10.00	1.1410 04	1.3510 10	5.1810 08	5.7620 08	4.2940 12	8.8400 16	3.0000 09	9.4750 05	1.0000 11

25 km HNO_3 , N_2O_5 , 0, 550°K

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N}_2\text{O}_5 + \text{M} \gg \text{N}_2\text{O}_2 + \text{N}_2\text{O}_3 + \text{M}$	1.10D-04	8.79D-15
2	$2^*\text{N}_2\text{O}_3 \gg \text{N}_2\text{O}_2 + \text{O}_2$	9.88D-15	7.98D-51
3	$\text{N}_2\text{O}_2 + \text{N}_2\text{O}_3 \gg \text{N}_2\text{O}_2 + \text{N}_2\text{O}_3 + \text{O}_2$	3.73D-14	6.64D-36
4	$\text{N}_2\text{O}_3 + \text{N}_2\text{O}_3 \gg \text{N}_2\text{O}_2$	1.90D-11	2.54D-21
5	$\text{N}_2\text{O}_2 + \text{O}_3 \gg \text{N}_2\text{O}_2 + \text{O}_2$	1.50D-13	2.09D-32
6	$\text{N}_2\text{O}_2 + \text{O}_3 \gg \text{N}_2\text{O}_3 + \text{O}_2$	1.40D-15	1.81D-24
7	$\text{HNO}_3 + \text{M} \gg \text{H}_2\text{O} + \text{N}_2\text{O}_2 + \text{M}$	7.06D-06	1.56D-13
8	$\text{HNO}_3 + \text{H}_2\text{O} \gg \text{H}_2\text{O} + \text{N}_2\text{O}_3$	8.00D-14	1.80D-21
9	$\text{O} + \text{O} + \text{M} \gg \text{O}_2 + \text{M}$	1.72D-16	9.45D-31
10	$\text{O} + \text{O}_2 + \text{M} \gg \text{O}_3 + \text{M}$	8.61D-17	4.97D-02
11	$\text{O} + \text{O}_3 \gg \text{O}_2 + \text{O}_2$	2.90D-13	3.43D-51
12	$\text{O} + \text{N}_2\text{O}_2 + \text{M} \gg \text{N}_2\text{O}_2 + \text{M}$	1.44D-14	7.09D-17
13	$\text{O} + \text{N}_2\text{O}_2 \gg \text{N}_2\text{O}_2 + \text{O}_2$	9.85D-12	9.33D-31
14	$\text{O} + \text{N}_2\text{O}_2 + \text{M} \gg \text{N}_2\text{O}_3 + \text{M}$	3.21D-14	2.57D-24
15	$\text{H}_2\text{O} + \text{H}_2\text{O} \gg \text{H}_2\text{O} + \text{O}$	3.65D-12	5.56D-18
16	$\text{O}_2 + 2^*\text{N}_2\text{O} \gg \text{N}_2\text{O}_2$	8.65D-39	6.75D-23

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	N0	O3	O2	HNO3	H0	H2O
0.0	6.5000 08	6.3000 09	2.0000 06	7.0000 09	4.3000 12	6.4300 16	3.0000 09	1.0000 06	1.0000 11
0.50	4.1750 00	8.1630 09	6.3910 08	5.8000 09	4.2400 12	6.4300 16	3.0000 09	1.0100 06	1.0000 11
1.00	4.4110 00	8.6850 09	6.3460 08	5.2820 09	4.1990 12	6.4300 16	3.0000 09	1.0200 06	1.0000 11
1.50	4.5640 00	8.9970 09	6.3380 08	4.9710 09	4.1590 12	6.4300 16	3.0000 09	1.0300 06	1.0000 11
2.00	4.6700 00	9.1870 09	6.3510 08	4.7800 09	4.1200 12	6.4300 16	3.0000 09	1.0400 06	1.0000 11
2.50	4.7460 00	9.3030 09	6.3740 08	4.6620 09	4.0820 12	6.4300 16	3.0000 09	1.0500 06	1.0000 11
3.00	4.8020 00	9.3730 09	6.4020 08	4.5890 09	4.0440 12	6.4300 16	3.0000 09	1.0590 06	1.0000 11
3.50	4.8460 00	9.4150 09	6.4310 08	4.5440 09	4.0070 12	6.4300 16	3.0000 09	1.0690 06	1.0000 11
4.00	4.8800 00	9.4400 09	6.4590 08	4.5170 09	3.9710 12	6.4300 16	3.0000 09	1.0790 06	1.0000 11
4.50	4.9070 00	9.4530 09	6.4850 08	4.5010 09	3.9350 12	6.4300 16	3.0000 09	1.0890 06	1.0000 11
5.00	4.9280 00	9.4600 09	6.5090 08	4.4910 09	3.9000 12	6.4300 16	3.0000 09	1.0990 06	1.0000 11
5.50	4.9450 00	9.4630 09	6.5290 08	4.4860 09	3.8650 12	6.4300 16	3.0000 09	1.1090 06	1.0000 11
6.00	4.9580 00	9.4630 09	6.5470 08	4.4840 09	3.8310 12	6.4300 16	3.0000 09	1.1180 06	1.0000 11
6.50	4.9680 00	9.4620 09	6.5610 08	4.4840 09	3.7970 12	6.4300 16	3.0000 09	1.1280 06	1.0000 11
7.00	4.9760 00	9.4600 09	6.5720 08	4.4850 09	3.7640 12	6.4300 16	3.0000 09	1.1380 06	1.0000 11
7.50	4.9800 00	9.4570 09	6.5800 08	4.4870 09	3.7310 12	6.4300 16	3.0000 09	1.1480 06	1.0000 11
8.00	4.9830 00	9.4540 09	6.5850 08	4.4900 09	3.6990 12	6.4300 16	3.0000 09	1.1570 06	1.0000 11
8.50	4.9830 00	9.4510 09	6.5870 08	4.4920 09	3.6680 12	6.4300 16	3.0000 09	1.1670 06	1.0000 11
9.00	4.9810 00	9.4480 09	6.5870 08	4.4950 09	3.6370 12	6.4300 16	3.0000 09	1.1770 06	1.0000 11
9.50	4.9770 00	9.4450 09	6.5840 08	4.4980 09	3.6060 12	6.4300 16	3.0000 09	1.1870 06	1.0000 11
10.00	4.9720 00	9.4430 09	6.5790 08	4.5020 09	3.5760 12	6.4300 16	3.0000 09	1.1960 06	1.0000 11

25 km HNO₃, N₂O₅, O, 600°K

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	N2O5 + M >>> N02 + N03 + M	4.80D 04	7.07D-15
2	2*N03 >>> 2*N02 + O2	1.43D-14	4.66D-50
3	N02 + N03 >>> N02 + N0 + O2	4.34D-14	5.43D-36
4	N03 + N0 >>> 2*N02	1.90D-11	1.45D-20
5	N0 + O3 >>> N02 + O2	1.87D-13	1.02D-30
6	N02 + O3 >>> N03 + O2	2.02D-15	1.79D-23
7	HNO3 + M >>> H0 + N02 + M	2.10D-04	1.15D-13
8	HNO3 + H0 >>> H2O + N03	8.00D-14	7.21D-21
9	O + O + M >>> O2 + M	1.37D-16	6.30D-27
10	O + O2 + M >>> O3 + M	7.31D-17	2.58D-01
11	O + O3 >>> 2*O2	4.11D-13	6.34D-48
12	O + N0 + M >>> N02 + M	1.21D-14	9.65D-15
13	O + N02 >>> N0 + O2	1.03D-11	3.23D-29
14	O + N02 + M >>> N03 + M	2.95D-14	2.36D-24
15	H0 + H0 >>> H2O + O	3.97D-12	2.26D-17
16	O2 + 2*N0 >>> 2*N02	7.98D-39	4.63D-22

HAPP RESIDENCE TIME STUDY

TIME (S)	N2O5	NO2	NO3	NO	O3	O2	HN03	HO	H2O
0.0	6.5000 08	6.3000 09	2.0000 06	7.0000 09	4.3000 12	5.8900 16	3.0000 09	1.0000 06	1.0000 11
0.50	5.4730-01	5.7580 09	6.4600 08	8.1990 09	3.9290 12	5.8900 16	3.0000 09	1.3140 06	1.0000 11
1.00	4.2660-01	4.6430 09	6.2430 08	9.3350 09	3.6640 12	5.8900 16	2.9990 09	1.6290 06	1.0000 11
1.50	3.7400-01	4.2700 09	5.9510 08	9.7370 09	3.4310 12	5.8900 16	2.9990 09	1.9430 06	1.0000 11
2.00	3.4330-01	4.1340 09	5.6440 08	9.9050 09	3.2240 12	5.8900 16	2.9990 09	2.2570 06	1.0000 11
2.50	3.1990-01	4.0700 09	5.3420 08	9.9990 09	3.0370 12	5.8900 16	2.9980 09	2.5710 06	1.0000 11
3.00	2.9950-01	4.0310 09	5.0500 08	1.0070 10	2.8690 12	5.8900 16	2.9980 09	2.8850 06	1.0000 11
3.50	2.8090-01	4.0020 09	4.7710 08	1.0130 10	2.7170 12	5.8900 16	2.9980 09	3.1990 06	1.0000 11
4.00	2.6360-01	3.9780 09	4.5050 08	1.0180 10	2.5780 12	5.8900 16	2.9970 09	3.5120 06	1.0000 11
4.50	2.4750-01	3.9560 09	4.2520 08	1.0220 10	2.4520 12	5.8900 16	2.9970 09	3.8260 06	1.0000 11
5.00	2.3250-01	3.9360 09	4.0140 08	1.0270 10	2.3370 12	5.8900 16	2.9970 09	4.1390 06	1.0000 11
5.50	2.1840-01	3.9180 09	3.7880 08	1.0310 10	2.2300 12	5.8900 16	2.9970 09	4.4520 06	1.0000 11
6.00	2.0530-01	3.9020 09	3.5750 08	1.0350 10	2.1330 12	5.8900 16	2.9960 09	4.7650 06	1.0000 11
6.50	1.9300-01	3.8860 09	3.3750 08	1.0380 10	2.0420 12	5.8900 16	2.9960 09	5.0780 06	1.0000 11
7.00	1.8150-01	3.8710 09	3.1870 08	1.0420 10	1.9590 12	5.8900 16	2.9960 09	5.3910 06	1.0000 11
7.50	1.7090-01	3.8580 09	3.0100 08	1.0450 10	1.8810 12	5.8900 16	2.9950 09	5.7030 06	1.0000 11
8.00	1.6090-01	3.8450 09	2.8450 08	1.0480 10	1.8090 12	5.8900 16	2.9950 09	6.0160 06	1.0000 11
8.50	1.5160-01	3.8320 09	2.6890 08	1.0510 10	1.7420 12	5.8900 16	2.9950 09	6.3280 06	1.0000 11
9.00	1.4300-01	3.8210 09	2.5440 08	1.0530 10	1.6790 12	5.8900 16	2.9940 09	6.6400 06	1.0000 11
9.50	1.3490-01	3.8100 09	2.4070 08	1.0560 10	1.6210 12	5.8900 16	2.9940 09	6.9520 06	1.0000 11
10.00	1.2740-01	3.7990 09	2.2800 08	1.0580 10	1.5650 12	5.8900 16	2.9940 09	7.2640 06	1.0000 11

25 km, HNO_3 , N_2O_5 , O, 700°K

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N}_2\text{O}_5 + \text{M} \ggg \text{N}_2\text{O}_2 + \text{NO}_3 + \text{M}$	4.800 05	4.940-15
2	$2^*\text{NO}_3 \ggg 2^*\text{NO}_2 + \text{O}_2$	2.570-14	7.440-49
3	$\text{NO}_2 + \text{NO}_3 \ggg \text{NO}_2 + \text{NO} + \text{O}_2$	5.510-14	3.970-36
4	$\text{NO}_3 + \text{NO} \ggg 2^*\text{NO}_2$	1.900-11	2.230-19
5	$\text{NO} + \text{O}_3 \ggg \text{NO}_2 + \text{O}_2$	2.650-13	4.540-28
6	$\text{NO} + \text{O}_3 \ggg \text{NO}_3 + \text{O}_2$	3.620-15	6.560-22
7	$\text{HNO}_3 + \text{M} \ggg \text{HO} + \text{NO}_2 + \text{M}$	4.010-02	6.700-14
8	$\text{HNO}_3 + \text{HO} \ggg \text{H}_2\text{O} + \text{NO}_3$	8.000-14	6.380-20
9	$\text{O} + \text{O} + \text{M} \ggg \text{O}_2 + \text{M}$	9.500-17	6.210-21
10	$\text{O} + \text{O}_2 + \text{M} \ggg \text{O}_3 + \text{M}$	5.550-17	3.360 00
11	$\text{O} + \text{O}_3 \ggg 2^*\text{O}_2$	7.110-13	8.630-43
12	$\text{O} + \text{NO} + \text{M} \ggg \text{NO}_2 + \text{M}$	9.010-15	2.140-11
13	$\text{O} + \text{NO}_2 \ggg \text{NO} + \text{O}_2$	1.110-11	8.500-27
14	$\text{O} + \text{NO}_2 + \text{M} \ggg \text{NO}_3 + \text{M}$	2.530-14	2.020-24
15	$\text{HO} + \text{HO} \ggg \text{H}_2\text{O} + \text{O}$	4.530-12	2.040-16
16	$\text{O}_2 + 2^*\text{NO} \ggg 2^*\text{NO}_2$	7.040-39	9.370-21

HAPP RESIDENCE TIME STUDY

TIME (S)	N2O5	N2O	N2O3	NO	O3	O2	HNO3	HO	H2O
0.0	6.5000 08	6.3000 09	2.0000 06	7.0000 09	4.3000 12	5.0000 16	3.0000 09	1.0000 06	1.0000 11
0.50	2.4320-03	3.8470 08	6.1520 08	1.3660 10	1.2660 12	5.0000 16	2.9400 09	6.0490 07	9.9990 10
1.00	1.7790-03	3.1780 08	5.4490 08	1.3860 10	8.0020 11	5.0000 16	2.8820 09	1.1880 08	9.9980 10
1.50	1.6020-03	3.2410 08	4.8100 08	1.3970 10	6.0710 11	5.0010 16	2.8250 09	1.7580 08	9.9970 10
2.00	1.4510-03	3.3300 08	4.2390 08	1.4080 10	4.9040 11	5.0010 16	2.7690 09	2.3160 08	9.9960 10
2.50	1.3060-03	3.4060 08	3.7310 08	1.4170 10	4.1120 11	5.0010 16	2.7140 09	2.8620 08	9.9960 10
3.00	1.1700-03	3.4690 08	3.2810 08	1.4270 10	3.5400 11	5.0010 16	2.6600 09	3.3960 08	9.9950 10
3.50	1.0430-03	3.5210 08	2.8830 08	1.4350 10	3.1060 11	5.0010 16	2.6070 09	3.9170 08	9.9950 10
4.00	9.2730-04	3.5640 08	2.5310 08	1.4440 10	2.7660 11	5.0010 16	2.5560 09	4.4250 08	9.9940 10
4.50	8.2210-04	3.6010 08	2.2210 08	1.4510 10	2.4430 11	5.0010 16	2.5050 09	4.9220 08	9.9940 10
5.00	7.2730-04	3.6310 08	1.9490 08	1.4590 10	2.2680 11	5.0010 16	2.4550 09	5.4060 08	9.9940 10
5.50	6.4240-04	3.6570 08	1.7090 08	1.4660 10	2.0800 11	5.0010 16	2.4060 09	5.8780 08	9.9940 10
6.00	5.6670-04	3.6790 08	1.4990 08	1.4730 10	1.9210 11	5.0010 16	2.3590 09	6.3380 08	9.9940 10
6.50	4.9940-04	3.6980 08	1.3140 08	1.4790 10	1.7840 11	5.0010 16	2.3120 09	6.7850 08	9.9930 10
7.00	4.3990-04	3.7140 08	1.1520 08	1.4850 10	1.6650 11	5.0010 16	2.2660 09	7.2210 08	9.9930 10
7.50	3.8730-04	3.7290 08	1.0110 08	1.4910 10	1.5600 11	5.0010 16	2.2210 09	7.6450 08	9.9930 10
8.00	3.4090-04	3.7410 08	8.8660 07	1.4960 10	1.4680 11	5.0010 16	2.1770 09	8.0560 08	9.9930 10
8.50	3.0010-04	3.7530 08	7.7810 07	1.5020 10	1.3860 11	5.0010 16	2.1340 09	8.4560 08	9.9930 10
9.00	2.6430-04	3.7640 08	6.8320 07	1.5070 10	1.3120 11	5.0010 16	2.0910 09	8.8450 08	9.9930 10
9.50	2.3290-04	3.7740 08	6.0030 07	1.5110 10	1.2460 11	5.0010 16	2.0500 09	9.2220 08	9.9930 10
10.00	2.0530-04	3.7840 08	5.2790 07	1.5160 10	1.1860 11	5.0010 16	2.0090 09	9.5870 08	9.9930 10

25 km, HNO_3 , N_2O_5 , O, 800°K

R-NUM	REACTION	FORWARD RATE	BACKWARD RATE
1	$\text{N}_2\text{O}_5 + \text{M} \gg \text{N}_2\text{O}_2 + \text{N}_2\text{O}_3 + \text{M}$	2.65D 06	3.70D-15
2	$2^*\text{N}_2\text{O}_3 \gg \text{N}_2\text{O}_2 + \text{O}_2$	3.98D-14	5.95D-48
3	$\text{N}_2\text{O}_2 + \text{N}_2\text{O}_3 \gg \text{N}_2\text{O}_2 + \text{N}_2\text{O} + \text{O}_2$	6.59D-14	3.13D-36
4	$\text{N}_2\text{O}_3 + \text{N}_2\text{O} \gg \text{N}_2\text{O}_2$	1.90D-11	1.74D-18
5	$\text{N}_2\text{O} + \text{O}_3 \gg \text{N}_2\text{O}_2 + \text{O}_2$	3.43D-13	4.42D-26
6	$\text{N}_2\text{O}_2 + \text{O}_3 \gg \text{N}_2\text{O}_3 + \text{O}_2$	5.61D-15	9.78D-21
7	$\text{HNO}_3 + \text{M} \gg \text{HO} + \text{N}_2\text{O}_2 + \text{M}$	1.90D 00	4.20D-14
8	$\text{HNO}_3 + \text{HO} \gg \text{H}_2\text{O} + \text{N}_2\text{O}_3$	8.00D-14	3.28D-19
9	$\text{O} + \text{O} + \text{M} \gg \text{O}_2 + \text{M}$	7.08D-17	1.87D-16
10	$\text{O} + \text{O}_2 + \text{M} \gg \text{O}_3 + \text{M}$	4.43D-17	2.26D 01
11	$\text{O} + \text{O}_3 \gg \text{N}_2\text{O}_2$	1.07D-12	6.11D-39
12	$\text{O} + \text{N}_2\text{O} + \text{M} \gg \text{N}_2\text{O}_2 + \text{M}$	7.11D-15	6.78D-09
13	$\text{O} + \text{N}_2\text{O}_2 \gg \text{N}_2\text{O} + \text{O}_2$	1.17D-11	5.55D-25
14	$\text{O} + \text{N}_2\text{O}_2 + \text{M} \gg \text{N}_2\text{O}_3 + \text{M}$	2.21D-14	1.77D-24
15	$\text{HO} + \text{HO} \gg \text{H}_2\text{O} + \text{O}$	5.00D-12	1.06D-15
16	$\text{O}_2 + 2^*\text{N}_2\text{O} \gg \text{N}_2\text{O}_2$	6.40D-39	8.78D-20

HAPP RESIDENCE TIME STUDY

TIME (S)	N205	N02	N03	N0	O3	O2	HN03	HO	H2O
0.0	6.5000 08	6.3000 09	2.0000 06	7.0000 09	4.3000 12	4.4000 16	3.0000 09	1.0000 06	1.0000 11
0.50	1.0480-04	1.2700 08	5.9090 08	1.5720 10	2.1410 11	4.4000 16	1.1610 09	1.8330 09	9.9850 10
1.00	6.6970-05	9.4140 07	5.0940 08	1.6550 10	1.7920 11	4.4010 16	4.4950 08	2.5190 09	9.9730 10
1.50	4.7510-05	7.7980 07	4.3620 08	1.5910 10	1.5370 11	4.4010 16	1.7400 08	2.7590 09	9.9640 10
2.00	3.6570-05	7.0310 07	3.7240 08	1.7090 10	1.3440 11	4.4010 16	6.7360 07	2.8270 09	9.9570 10
2.50	2.9560-05	6.6680 07	3.1750 08	1.7190 10	1.1930 11	4.4010 16	2.6080 07	2.8280 09	9.9500 10
3.00	2.4500-05	6.4860 07	2.7040 08	1.7260 10	1.0720 11	4.4010 16	1.0100 07	2.8040 09	9.9450 10
3.50	2.0520-05	6.3820 07	2.3030 08	1.7300 10	9.7280 10	4.4010 16	3.9100 06	2.7710 09	9.9400 10
4.00	1.7270-05	6.3080 07	1.9600 08	1.7340 10	8.8990 10	4.4010 16	1.5160 06	2.7360 09	9.9360 10
4.50	1.4550-05	6.2460 07	1.6690 08	1.7370 10	8.1980 10	4.4010 16	5.8910 05	2.7000 09	9.9330 10
5.00	1.2270-05	6.1860 07	1.4200 08	1.7400 10	7.5980 10	4.4010 16	2.3030 05	2.6640 09	9.9300 10
5.50	1.0350-05	6.1310 07	1.2090 08	1.7420 10	7.0780 10	4.4010 16	9.1350 04	2.6300 09	9.9270 10
6.00	8.7360-06	6.0750 07	1.0300 08	1.7440 10	6.6230 10	4.4010 16	3.7520 04	2.5950 09	9.9240 10
6.50	7.3750-06	6.0210 07	8.7710 07	1.7450 10	6.2230 10	4.4010 16	1.6630 04	2.5620 09	9.9220 10
7.00	6.2300-06	5.9700 07	7.4730 07	1.7470 10	5.8670 10	4.4010 16	8.5040 03	2.5300 09	9.9190 10
7.50	5.2680-06	5.9210 07	6.3710 07	1.7480 10	5.5490 10	4.4010 16	5.3140 03	2.4980 09	9.9170 10
8.00	4.4590-06	5.8760 07	5.4340 07	1.7490 10	5.2640 10	4.4010 16	4.0390 03	2.4670 09	9.9160 10
8.50	3.7780-06	5.8330 07	4.6380 07	1.7500 10	5.0060 10	4.4010 16	3.5060 03	2.4370 09	9.9140 10
9.00	3.2060-06	5.7950 07	3.9620 07	1.7500 10	4.7710 10	4.4010 16	3.2630 03	2.4080 09	9.9120 10
9.50	2.7240-06	5.7590 07	3.3870 07	1.7510 10	4.5580 10	4.4010 16	3.1340 03	2.3790 09	9.9110 10
10.00	2.3180-06	5.7270 07	2.8990 07	1.7520 10	4.3620 10	4.4010 16	3.0520 03	2.3520 09	9.9100 10